

UCRL 39

g 10/A

c. 1

UNIVERSITY OF  
CALIFORNIA

Radiation  
Laboratory

FOR REFERENCE

NOT TO BE TAKEN FROM THIS ROOM

BERKELEY, CALIFORNIA

**Special Review of  
Declassified Reports**

Authorized by USDOE JK Bratton

Unclassified TWX P182206Z May 79

UCR-39 1948

**REPORT PROPERLY DECLASSIFIED**



Official

Date

3-25-80



INDEX NO. U.C.R.L - 39  
This document contains        pgs,  
and        plates of figures.  
This is Copy 10 of 11. Series A

DO NOT REMOVE THIS PAGE

Issued to Information Division  
**RESTRICTED**  
CLASSIFICATION CHANGES TO BE STAMPED HERE  
BY THE DISTRICT ENGINEER  
(CLASSIFIED BY AUTHORITY OF THE U.S. GOVERNMENT)

THIS IS A CLASSIFIED DOCUMENT

1. This document contains restricted data within the meaning of the Atomic Energy Act of 1946 and/or information affecting the national defense of the United States within the meaning of the Espionage Act U. S. C. 31 & 32, as amended. Its transmission or the revelation of its contents in any manner to an unauthorized person is prohibited and may result in severe criminal penalty.
2. Before this document can be given to a person to read, his name must be on the Reading List of those authorized to read material on this subject, or permission must be obtained from the Information Division or the Executive Office.
3. A SECRET or CONFIDENTIAL document is to be kept only in a guarded area. When stored, it must be kept in a locked safe or in a locked filing case with a tumbler lock.
4. A SECRET or CONFIDENTIAL document is not to be copied or otherwise duplicated without permission of the originating office. Extensive notes on the contents of a secret report may be taken only in a bound notebook with numbered pages, having the designation SECRET. This must be safeguarded in the same way as a Secret Report (See 1, 2 and 3). Other notes must be avoided, but if made must be done in a form meaningless to anyone but the writer and later destroyed.
5. The person or office to whom a SECRET or CONFIDENTIAL document is issued is accountable for the document at all times. A system of accountability by signed receipts or a signed record book must always be used. Permission for transmission to another office than that of the assignee must be obtained through the Information Division and recorded there.
6. After a document passes the stage of usefulness to its possessor, it should be returned to the Information Division, whereupon the individual will be relieved of the responsibility for its safe-keeping.
7. Initial and date the pages of this document which describe any work that you have witnessed, adding any pertinent information, such as references to original work records. This document, properly signed, dated and annotated, is important in patent preparation.
8. Each person who reads this document must sign this cover sheet below. Please sign and date the cover sheet after reading the report. Also state by Yes or No whether or not notes have been taken. See paragraph 4 above.

Route to	Read by	Notes?	Date	Route to	Read by	Notes?	Date

CLASSIFICATION CANCELLED BY AUTHORITY OF THE DISTRICT ENGINEER BY THE DECLASSIFICATION COMMITTEE

**RESTRICTED**

UCRL 39

This document consists of pages, Series A.

NEUTRON DEFICIENT ISOTOPES OF TELLURIUM AND ANTIMONY

M. Lindner and I. Perlman

Radiation Laboratory and Department of Chemistry  
University of California, Berkeley, California

While investigating the relative yields for the many reactions resulting from the irradiation of antimony with 200-Mev deuterons in the Berkeley 184-inch cyclotron several previously unreported isotopes of tellurium and antimony were encountered. The tellurium fraction when followed on a thin mica window counter could be resolved into half-life periods of 2.5 hrs, 6.0 days and a small amount of a long-lived component. The 2.5 hour period has not been further characterized with respect to mass number or mode of decay other than to note that the radiation is predominantly electrons. The 6.0-day period is accompanied by positrons which were shown to be due to a 3.5 minute antimony daughter which is undoubtedly the same activity assigned to  $Sb^{118}$  by Risser, Lark-Horowitz and Smith<sup>1</sup>. The positron energy was found to be  $3.1 \pm 0.2$  mev by absorption in beryllium and from the end point of the energy distribution curve taken with a low-resolution beta-ray spectrometer. Gamma activity is also present with this period. The 6.0-day tellurium showed a high abundance of x-rays, little or no conversion electrons and some gamma-ray activity which could be due to the 3.5 minute antimony daughter.

The tellurium fraction contained another component of 4.5-day half-life which could not be observed in the decay curve because of its low abundance but which was detected by means of its 39-hour antimony daughter. The 39-hour antimony showed x-rays of tin (critical absorption with cadmium, silver and palladium), no detectable hard radiation or electrons and is apparently identical with an activity recently assigned to  $Sb^{119}$  by Coleman and Pool<sup>2</sup>.

This document contains restricted data within the meaning of the Espionage Act of 1917, as amended, and the Espionage Act of 1950, as amended, or the revelation of its contents in any manner to an unauthorized person is prohibited and may result in severe civil or criminal penalties. DISTRICT ENGINEER DECLASSIFICATION COMMITTEE

~~RESTRICTED~~

UCRL-39  
Page 2.

In order to produce a more favorable ratio of the 4.5-day tellurium to 6.0-day tellurium, antimony was bombarded with 40-Mev deuterons which from other measurements was predicted to produce the  $d,4n$  reaction in good yield and the  $d,5n$  reaction in poorer yield. If the 4.5-day tellurium is  $\text{Te}^{119}$  and the 6.0-day period is  $\text{Te}^{118}$  the desired result should be achieved. The counting rates for the x-rays of the two activities were found to be in the ratio of 40:1 in the direction anticipated. Since the activity of several days half-life is largely that of the 4.5-day tellurium (except for the positrons of the 3.5-minute  $\text{Sb}^{118}$ ), the negative particles and electromagnetic radiation may be assumed to belong to this isotope. Conversion electrons of 0.2 Mev and 0.5 Mev were measured with the low resolution beta-ray spectrometer and a hard gamma-ray of about 1.5 Mev was detected through a lead absorption curve. Softer gamma radiation could have been present. All tellurium decay curves tailed out into a longer period, probably  $\text{Te}^{121}$ .

The antimony fraction from bombardments of antimony with 200-Mev deuterons showed a number of periods which could be resolved by selectively counting the x-rays through beryllium and the electrons without absorber. There was x-ray activity of a few hours half-life which could be a mixture of the 2.8 hour and 5.1 hour periods reported by Coleman and Pool. The 2.8-day  $\text{Sb}^{122}$  showed up well in the curve taken without absorbers and the 39-hour antimony was prominent in the decay curve taken through beryllium. The yield of the 39-hour antimony was much greater than that which could have grown from its 4.5-day tellurium parent, indicating direct formation. Another antimony isotope of 6.0-day half-life was also observed. This activity is characterized by a preponderance of x-ray and gamma-ray activity with electrons in low abundance. Since the 6.0-day antimony did not appear in antimony removed from tellurium, the tellurium isobar is either stable or of half-life less than 10 minutes. Although the half-life is the same as that of the tellurium mentioned above, it could be separated chemically with antimony from

tellurium and there was no positron activity accompanying it. It is not possible to make an isotopic assignment of this activity. After the decay of the 6.0-day antimony, the curve turned into the 60-day period of  $\text{Sb}^{124}$  which was identified also by its radiation characteristics. The formation of  $\text{Sb}^{124}$  presents the unanswered question as to whether this isotope is formed by the high energy deuterons or by secondary low energy neutrons.

Table 7 lists the isotopes of antimony and tellurium found in the bombardment of antimony with 200-Mev deuterons. The cross sections for their formation will be given in a later publication in which the yields for the large number of products of 200-Mev deuterons on antimony will be discussed.

The assignment of the 4.5-day tellurium - 39-hour antimony isobars to mass number 119 is most reasonable in view of their production with 40-Mev deuterons on antimony, their decay characteristics and the possibilities open. This assignment agrees with that of the 39-hour antimony by Coleman and Pool from other evidence. The most reasonable assignment of the 6.0-day tellurium - 3.5 minute antimony isobars is mass number 118. They were produced with 40-Mev deuterons on antimony which would be impossible for mass number 116 and probably for mass number 117. An odd mass number (117) is also unlikely in view of observed half-life relationships. This isotopic assignment is in apparent conflict with the assignment of the 5.1-hour antimony to mass number 118<sup>2</sup>. However, it is possible that the 3.5-minute antimony and the 5.1-hour antimony are isomeric and that in the decay of the 6.0-day tellurium parent only the upper state (3.5-minute antimony) is formed.

The 6.0-day antimony has not yet been sufficiently characterized to permit a guess of the isotopic assignment but in view of the assignments already made it is not unlikely that this isotope is isomeric with another known antimony activity.

The cooperation of Dr. D. C. Sewell, Mr. J. T. Vale, and all of those whose operation of the 184-inch cyclotron made these irradiations possible is gratefully acknowledged.

This paper is based on work performed under Contract Number W-7405-Eng-48 with the Atomic Energy Commission in connection with the Radiation Laboratory of the University of California, Berkeley, California.

Table I

Isotopes of Sb and Te formed in the  
Irradiation of Antimony with 200-Mev Deuterons

Isotope	Half-life	Decay mech.	Energy (Mev)		Other References
			Particles	$\gamma$ -Rays	
Sb	Several hours	$K, e^-$			(2)
Sb <sup>118</sup>	3.5 minutes	$\beta^+, \gamma$	3.1 <sup>±</sup> 0.2		(1)
Sb <sup>119</sup>	39 hours	K	none		(2)
Sb	6.0 days	$K, e^-, \gamma$ or $IT, e^-, \gamma$		1.1	
*Sb <sup>122</sup>	2.8 days	$\beta^-, \gamma, e^-$			(3)
*Sb <sup>124</sup>	60 days	$\beta^-, \gamma$			(3)
Te <sup>118</sup>	6.0 days	K, no $\gamma$ (?)			
Te <sup>119</sup>	4.5 days	$K, e^-, \gamma$	0.2, 0.5	1.4	
*Te <sup>121</sup>	140 days	$IT, e^-, \gamma$			(4)

\* Isotope identified from properties listed in the literature but not further characterized.

- 
- (1) Risser, Lark-Horowitz and Smith, Phys. Rev. 57, 355(1940).
  - (2) Coleman and Pool, Phys. Rev. 72, 1070(1947).
  - (3) Livingood and Seaborg, Phys. Rev. 55, 414(1939).
  - (4) Pool and Edwards, Phys. Rev. 69, 140(1946).