The Effect of IGFC Warm Gas Cleanup System Conditions on the Gas-Solid Partitioning and Form of Trace Species in Coal Syngas and Their Interactions with SOFC Anodes



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**Fuel Cell Research Group** 



## • Today's Power Generation Market

- -Coal fired power generation under increased scrutiny
  - More efficient processes with less environmental impact
- -Coal is the U.S.' most abundant and pollutant laden fuel
- -Economics dictates it will be used for years to come
- -New technologies show great promise
  - Gasification Convert coal into coal syngas (CSG)
  - Cleanup Remove contaminants from CSG
  - Fuel Cells Directly convert chemical energy to electricity
- -U.S. DOE investing in development of all technologies



### Coal Gasification

- –Coal is transformed into a syngas by the addition of  $O_2$  and  $H_2O$ .
- -Reactors operate at temperatures up to 1700°C and 70atm.
- -The syngas contains a mixture of  $H_2$ , CO, CO<sub>2</sub>,  $H_2O$ , CH<sub>4</sub>, N<sub>2</sub>, and many trace species.
- $-O_2$  blown entrained flow gasification is used in today's IGCC power plants and anticipated to be used in the future [1].



Solid Oxide Fuel Cell



Figure 1. Solid Oxide Fuel Cell Operation [2].



### SOFCs and Coal Syngas (CSG)

- Recent studies have shown feasibility of operating solid oxide fuel cell (SOFC) systems with coal syngas [4-5].
- Current CSG cleanup systems work very well for the removal of S and CI species, however operating temperature is too low (ambient).
- Future warm/hot gas cleanup systems will operate at much higher temperatures (250-500°C).
- The effect of trace species that may pass through warm gas cleanup conditions is not known.





#### Figure 2. U.S. DOE FutureGen [3].



### • Trace Species in CSG

- -Trace elements contained in coal are classified into three groups base upon their volatility [1].
  - Class I: Least volatile, will remain in the ash.
  - *Class II*: More volatile, partition between condensed and gas phases.
  - Class III: Volatile, show little to no tendency to condense.
- Previous reports have shown the presence of As, P, Sb, Cd, Be, Cr, Hg, K, Se, Na, V, Pb, Zn.



- Thermodynamics are used to determine the condensation behavior of trace species contained in CSG.
  - -Gaseous species are assumed to travel to SOFC module.
  - Solid species are assumed to have a 100% removal efficiency.
  - System temperatures and pressures were varied from 200-500°C and 1-15atm.
- Thermodynamic analyses of the anode was also completed based upon warm gas cleanup results.
  - -Study evaluated anode composition (Ni,  $ZrO_2$ , and  $Y_2O_3$ ).
  - Study completed over SOFC operational temperatures 700-900°C and anticipated pressures 1-15atm.



Table 1. Coal Syngas Composition [6].

Component	Composition (vol%)		
H <sub>2</sub>	29.3		
СО	28.7		
CO <sub>2</sub>	11.8		
N <sub>2</sub>	3.0		
H <sub>2</sub> O	27.2		



#### Table 2. Trace Species Contained in Coal Syngas [1,7-9].

Component	Concentration (ppmv)	Volatility Class	
AsH <sub>3</sub>	0.6	II	
HC1	1	III	
PH <sub>3</sub>	1.91	II	
Sb	0.07	II	
Cd	0.011	II	
Be	0.025	II	
Cr	6	II	
Hg	0.025	II	
K	512	Ι	
Se	0.15	II	
Na	320	Ι	
V	0.025	II	
Pb	0.26	II	
Zn	9	II	



### **Results**

#### Table 3. Trace Species Behavior.

Component	Behavior
As	Gas/Solid
Р	Gas/Solid
Sb	Gas
Cd	Gas/Solid
Be	Solid
Cr	Solid
Hg	Gas
K	Solid
Se	Gas/Solid
Na	Solid
V	Solid
Pb	Gas/Solid
Zn	Solid



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#### Table 4. Trace Species in Anode Fuel.

Component	Concentration (ppmv)	
Sb	0.07	
As	0.6	
Cd	0.011	
Pb	0.26	
Hg	0.025	
Р	1.91	
Se	0.15	



#### Table 5. Outlet Edge Fuel Composition.

Component	Composition (vol%)		
H <sub>2</sub>	4.6		
СО	4.0		
CO <sub>2</sub>	36.5		
N <sub>2</sub>	3.0		
H <sub>2</sub> O	51.9		



## **Anode Evaluation Results**

### Study Results

- Sb, As, and P trace species were found to form secondary Ni phases.
- Cd, Pb, Hg, and Se were not found to form secondary phases in the anode.



## **Sb/Anode Interactions**

$$As_3Sb(g) + Ni(s) \rightarrow NiSb(s) + 0.75As_4(g) \qquad Eq.1$$



Figure 3. Equilibrium Pressures of As<sub>3</sub>Sb Associated with

Equation 1 Over SOFC Operation Conditions.



### **As/Anode Interactions**

$$0.25As_4(g) + Ni(s) \rightarrow NiAs(s) \qquad Eq.2$$



Figure 4. Equilibrium Pressures of As<sub>4</sub> Associated with

Equation 1 Over SOFC Operation Conditions.



### **As/Anode Interactions**

 $AsH_3(g) + Ni(s) \rightarrow NiAs(s) + 1.5H_2(g) \qquad Eq.3$ 



**Figure 5.** Equilibrium Pressures of AsH<sub>3</sub> Associated with Equation 3 Over SOFC Operation Conditions at the Inlet (a) and Outlet (b).



## **P/Anode Interactions**

 $2PH_3(g) + 5Ni(s) \rightarrow Ni_5P_2(s) + 3H_2(g) \qquad Eq.4$ 



**Figure 6.** Equilibrium Pressures of  $PH_3$  Associated with Equation 3 Over SOFC Operation Conditions at the Inlet (a) and Outlet (b).



## **Trace Metal Oxidation**

 $Pb(g)+0.5O_{2}(g) \rightarrow PbO(s) \qquad Eq.5$  $Cd(g)+0.5O_{2}(g) \rightarrow CdO(s) \qquad Eq.6$  $Hg(g)+0.5O_{2}(g) \rightarrow HgO(g) \qquad Eq.7$  $Ni(s)+0.5O_{2}(g) \rightarrow NiO(g) \qquad Eq.8$ 

Table 6. O2 Equilibrium Partial PressuresAssociated with Equations 5-8.

Т(ºС)	pO <sub>2</sub> (syngas)	pO <sub>2</sub> (Eq.5)	pO <sub>2</sub> (Eq.6)	pO <sub>2</sub> (Eq.7)	pO <sub>2</sub> (Eq.8)
700	1.60E-17	9.80E-15	8.50E-07	7.50E+15	5.42E-17
800	6.40E-15	1.00E-10	3.50E-03	1.90E+17	1.20E-14
900	9.80E-13	2.20E-07	3.40E+00	2.70E+18	1.04E-12



### **Thermodynamic Evaluation Conclusions**

- 1. Many trace species in coal syngas will form solid phases over warm gas cleanup conditions. In particular Be, Cr, K, Na, V, and Zn all formed condensed species.
- 2. Thermodynamic evaluations showed that Sb, As, Cd, Hg, P, and Se vapor species form in warm gas cleanup conditions.
- 3. No secondary phase formations between the vapor specie forms and oxide components were found.
- 4. No phase formations between Se vapor species and Ni were found.
- 5. Sb, As, and P vapor species were shown to have the potential to form secondary phases with Ni.
- 6. Oxidation of the fuel species was shown to have a large effect on the amount of secondary Ni phases formed in the anode.
- 7. The oxidation of the trace metal vapor species was shown not to be feasible.



## **Future Testing**

- Experimentally determine the effects of trace syngas species
  - AsH<sub>3</sub> (Fall/Winter '06/'07)
  - PH<sub>3</sub>, Hg, and Sb (Spring '07)
  - Cd, Pb, and Se (Summer '07)



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### Questions

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