Vacuum Assumptions

for

RHIC

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Vacuum Assumptions

Two Sections :

(a) Warm - 25% of Circumference

Equivalent Nitrogen Pressure : 10^{-9} torr

Gas Composition :

CO : 50% and H_2 : 50%

Room Temperature : $300^\circ K$

(b) Cold - 75% of Circumference

Equivalent Nitrogen Pressure : 10^{-11} torr

Gas Composition :

He : 50% and H_2 : 50%

Liquid Helium : $4.2^\circ K$

(2)

The vacuum pressure is measured with a gauge with the following efficiency factors

1.0 for CO

0.5 for H_2

0.5 for He

Therefore : in the warm section

$$(0.5) n_{H_2} + (1.0) n_{CO} = n_{N_2}$$

$$n_{H_2} = n_{CO}$$

and in the cold section

$$(0.5) n_{H_2} + (0.5) n_{He} = n_{N_2}$$

$$n_{H_2} = n_{He}$$

By definition

$$n_{N_2} = 2.687 \times 10^{19} \frac{P_{\text{torr}}}{760} \times \frac{273.15}{T_{\text{ok}}}$$

(3)

$$n_{N_2} = 3.22 \times 10^{16} P_{\text{torr}} / \text{cc}$$

$$= 3.22 \times 10^7 / \text{cc} \quad \underline{\text{warm}} \text{ section}$$

$$n_{N_2} = 2.30 \times 10^{18} P_{\text{torr}} / \text{cc}$$

$$= 2.30 \times 10^7 / \text{cc} \quad \underline{\text{cold}} \text{ section}$$

and

	<u>warm</u>	<u>cold</u>
densities, n	25%	75%
H ₂	$2.1 \times 10^7 / \text{cc}$	$2.3 \times 10^7 / \text{cc}$
He	—	$2.3 \times 10^7 / \text{cc}$
CO	$2.1 \times 10^7 / \text{cc}$	—

with

	<u>Z</u>	<u>A</u>
H ₂	2	2
He	2	4
CO	14	28

or

warmcold

25%

75%

 n_H $4.2 \times 10^7 / \text{cc}$ $4.6 \times 10^7 / \text{cc}$ n_{He}

-

2.3

 n_e

2.1

-

 n_o

2.1

-

with

ZA

H

1

1

He

2

4

C

6

12

O

8

16