

“The Durability of Polymeric Encapsulation Materials for Concentrating Photovoltaic Systems”

***David C. Miller¹*, Matt Muller¹, Michael D. Kempe¹, Kenji Araki²,
Cheryl E. Kennedy¹, and Sarah R. Kurtz¹***

1. National Renewable Energy Laboratory (NREL), 15013 Denver West Parkway, Golden, CO, USA 80401

2. Daido Steel Co., Ltd. 2-30 Daido-cho, Minami, Nagoya 457-8545, Japan

* David.Miller@nrel.gov



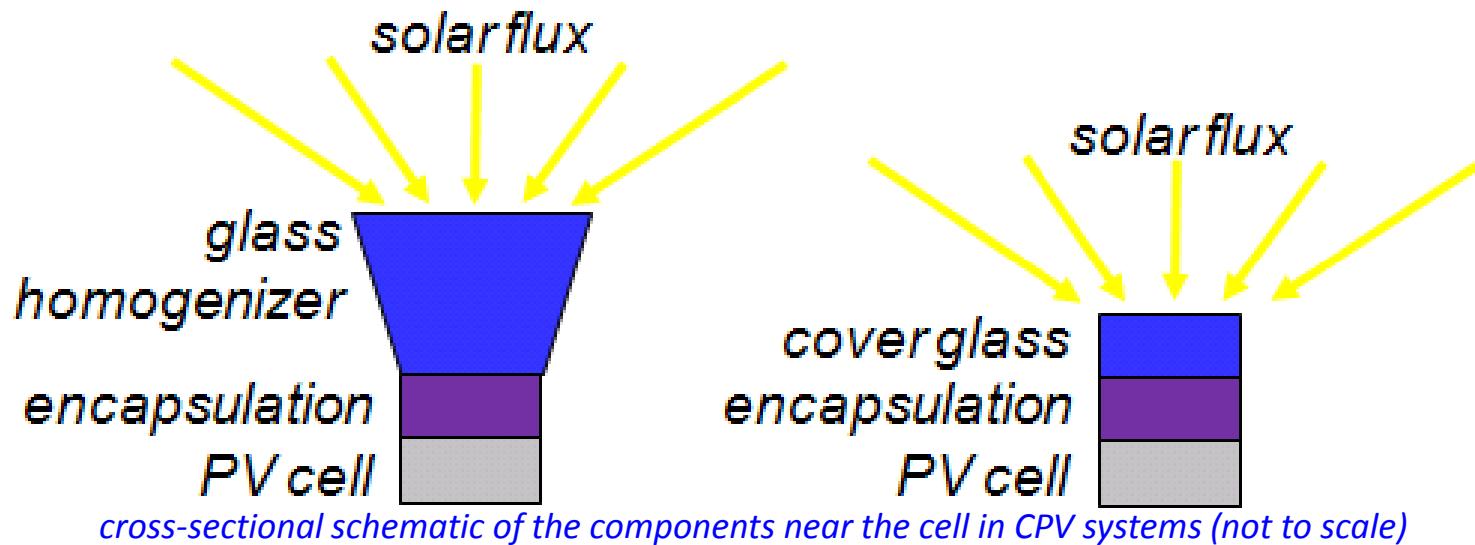
**2012 PV Module Reliability Workshop
(Denver West Marriot, Golden, CO)
8:30-8:50 am, 2012/3/01 (Thursday)
Golden Ballroom**

***-this presentation contains no
proprietary information-***

NREL/PR 5200 54524

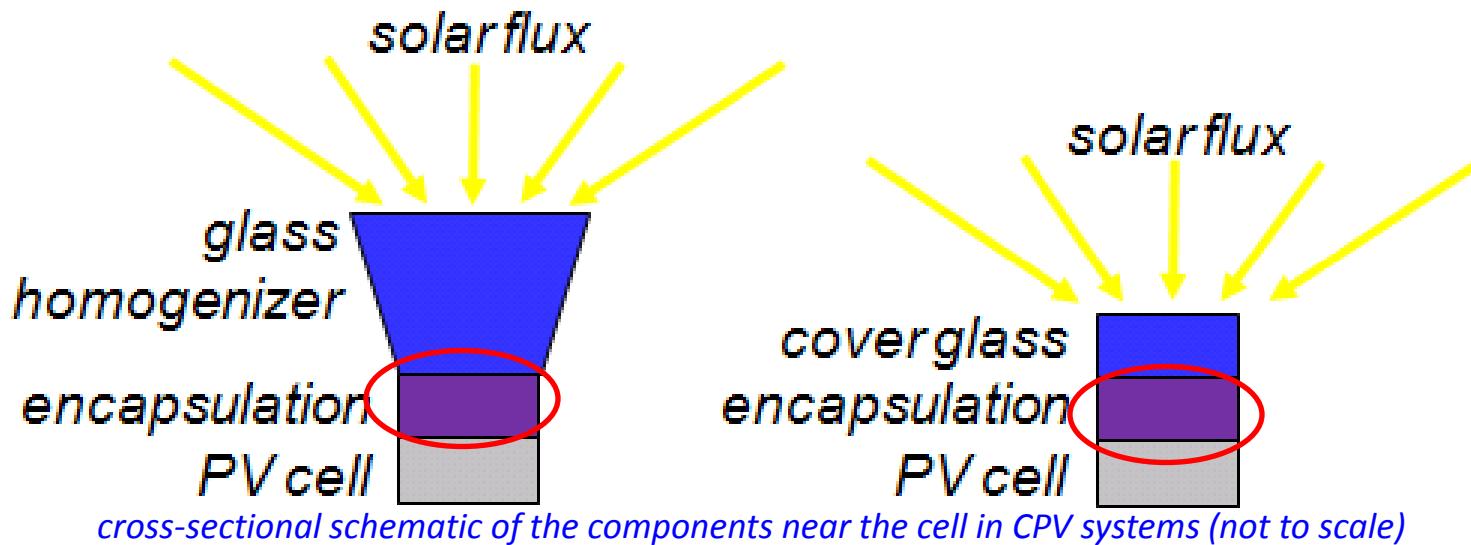
Motivation for the NREL Field Study

- Concentrating Photovoltaic (CPV) modules use cost effective optics (\$) to focus light onto high efficiency ($\eta=44\%$) multijunction cells (\$\$\$\$\$)



Motivation for the NREL Field Study

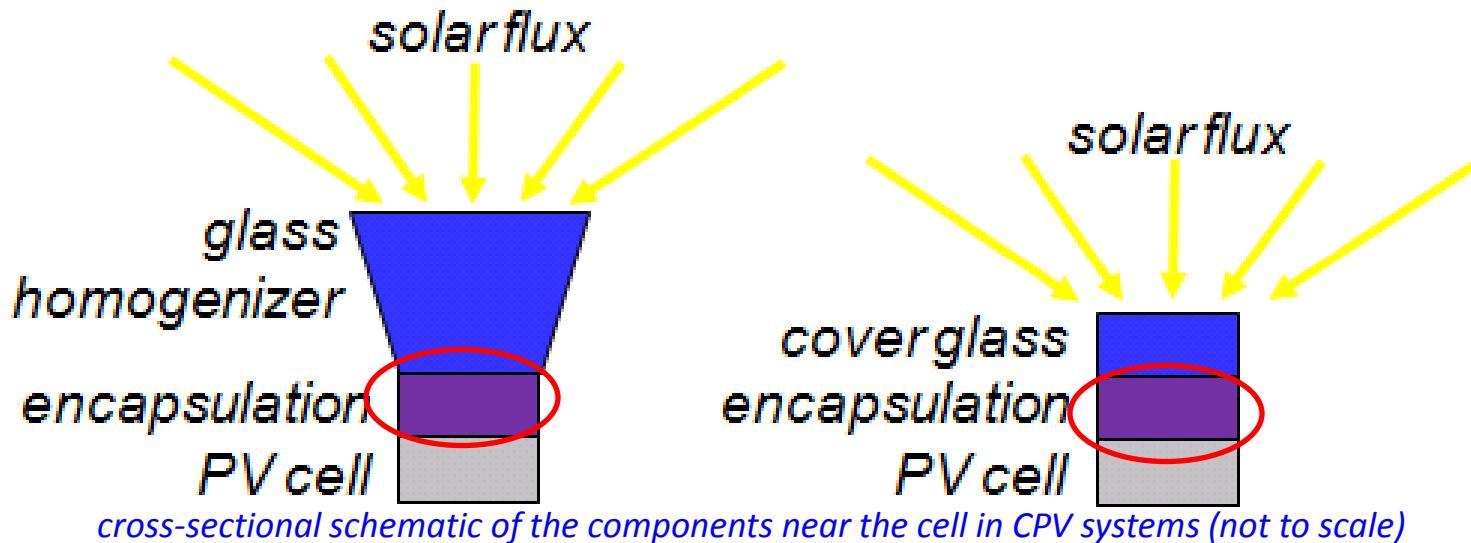
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corrosion prevention, optical coupling : CPV systems typically use encapsulation to adhere optical component(s) or cover glass to the cell

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corrosion prevention, optical coupling : CPV systems typically use encapsulation to adhere optical component(s) or cover glass to the cell

encapsulation durability (30 year field deployment) is unknown:

- identify field failure modes
- gain insight related to failure mechanisms
- distinguish between material types
- identify materials for future study (HALT & qualification tests)

Details of the Experiment (Specimens & Apparatus)

Miller et. al., PIP, DOI: 10.1002/pip.1241.

hydrocarbons
(representative types)

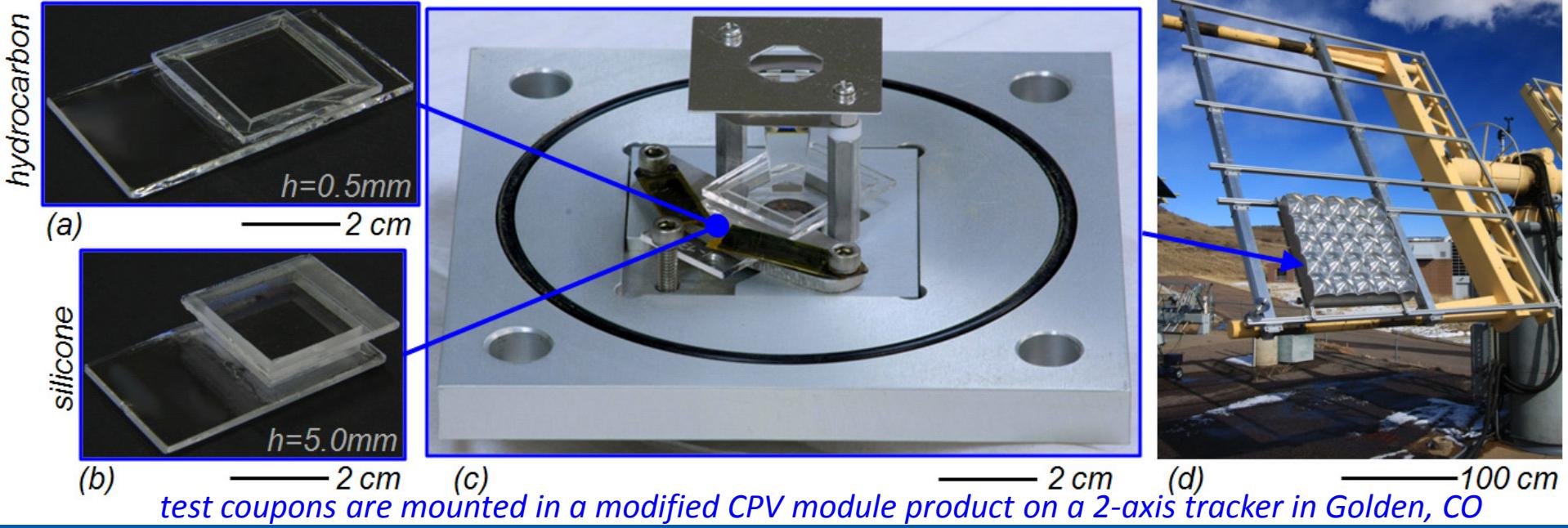
silicones
(representative grades)

MATERIAL	ON-TEST	IN QUEUE
EVA	6	2
ionomer	2	0
polyolefin	1	1
PVB	2	0
TPU	1	2
PDMS	11	5
PPMS	2	1
TOTAL	25	11

test coupons are mounted in a modified CPV module product on a 2-axis tracker in Golden, CO

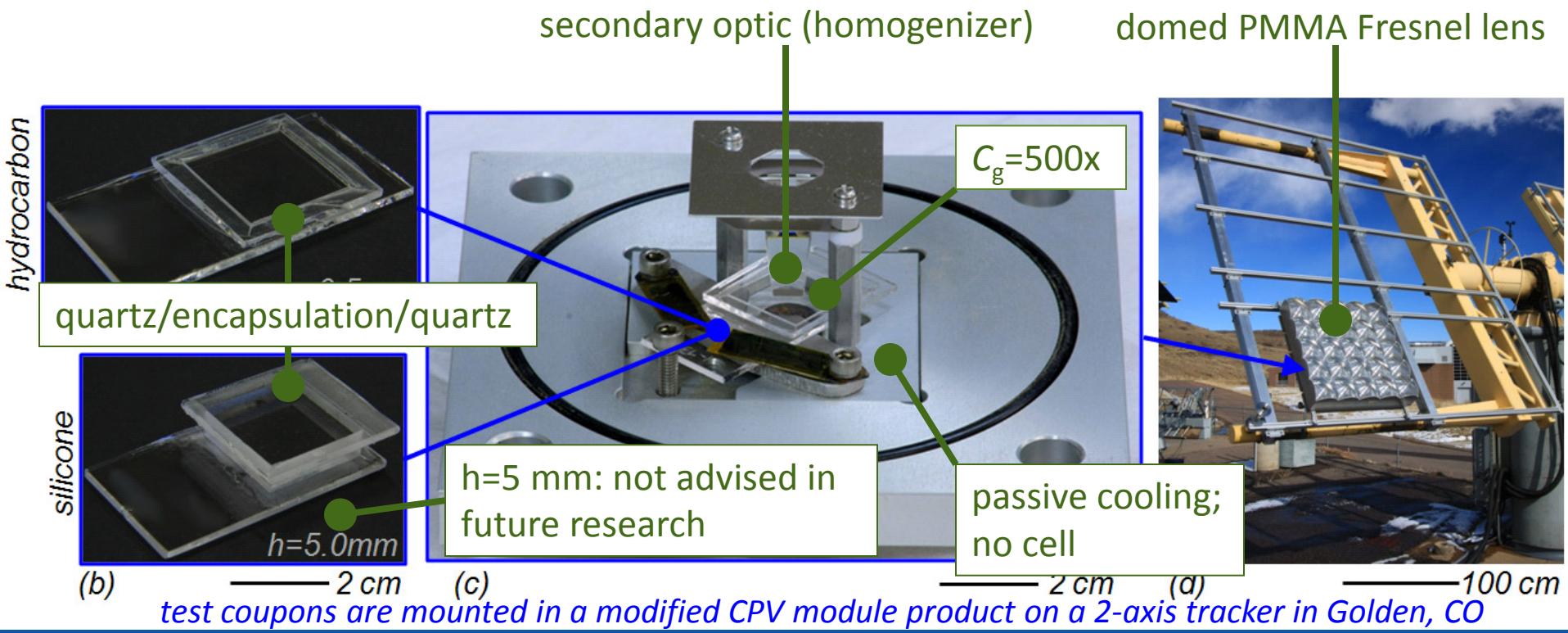
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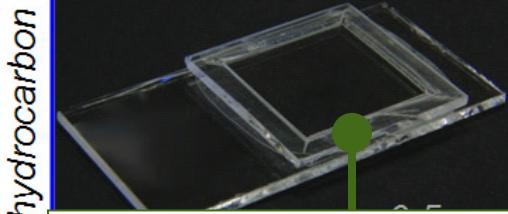
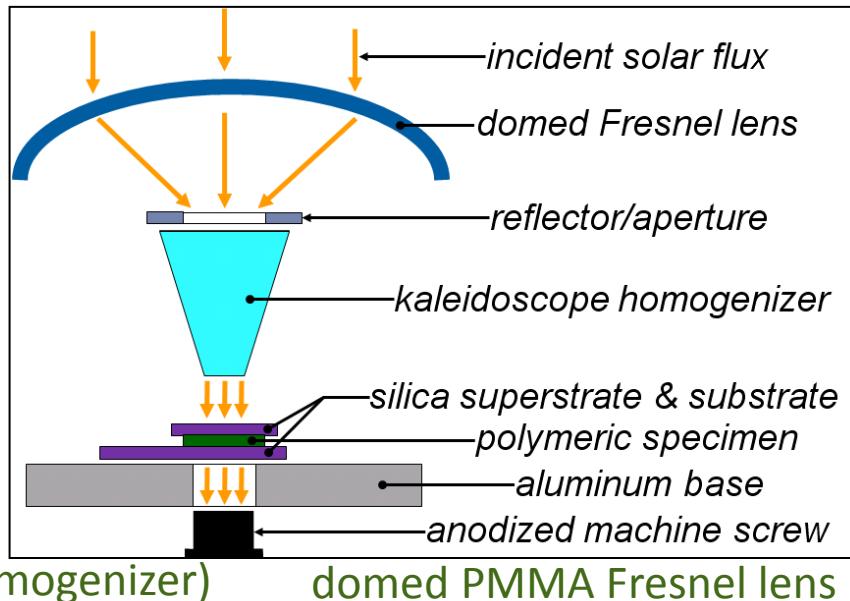
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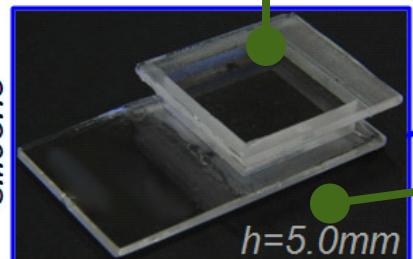
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hydrocarbon

quartz/encapsulation/quartz

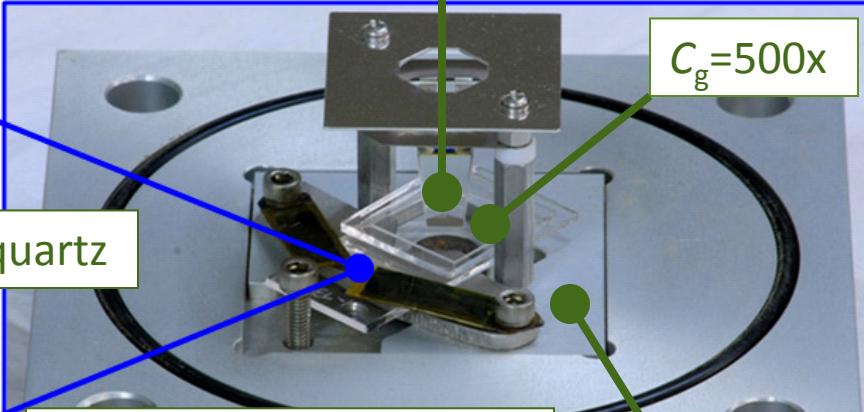


silicone

$h=5.0\text{mm}$

(b)

2 cm



(c)

test coupons are mounted in a modified CPV module product on a 2-axis tracker in Golden, CO



(d)

100 cm

Details of the Experiment (Measurands & Schedule)

“Continuous” measurements:

- ambient conditions (irradiance, temperature, wind...)
- fixture temperature (via thermocouple)

Periodic measurements:

- transmittance ($T[\lambda]$, hemispherical & direct)
 - mass
 - appearance (photograph)
- from $T[\lambda]$, calculate: yellowness index (D65 source, 1964 10° observer), haze, $\lambda_{\text{cut-on}}$...
- fluorescence spectroscopy

Final measurements:

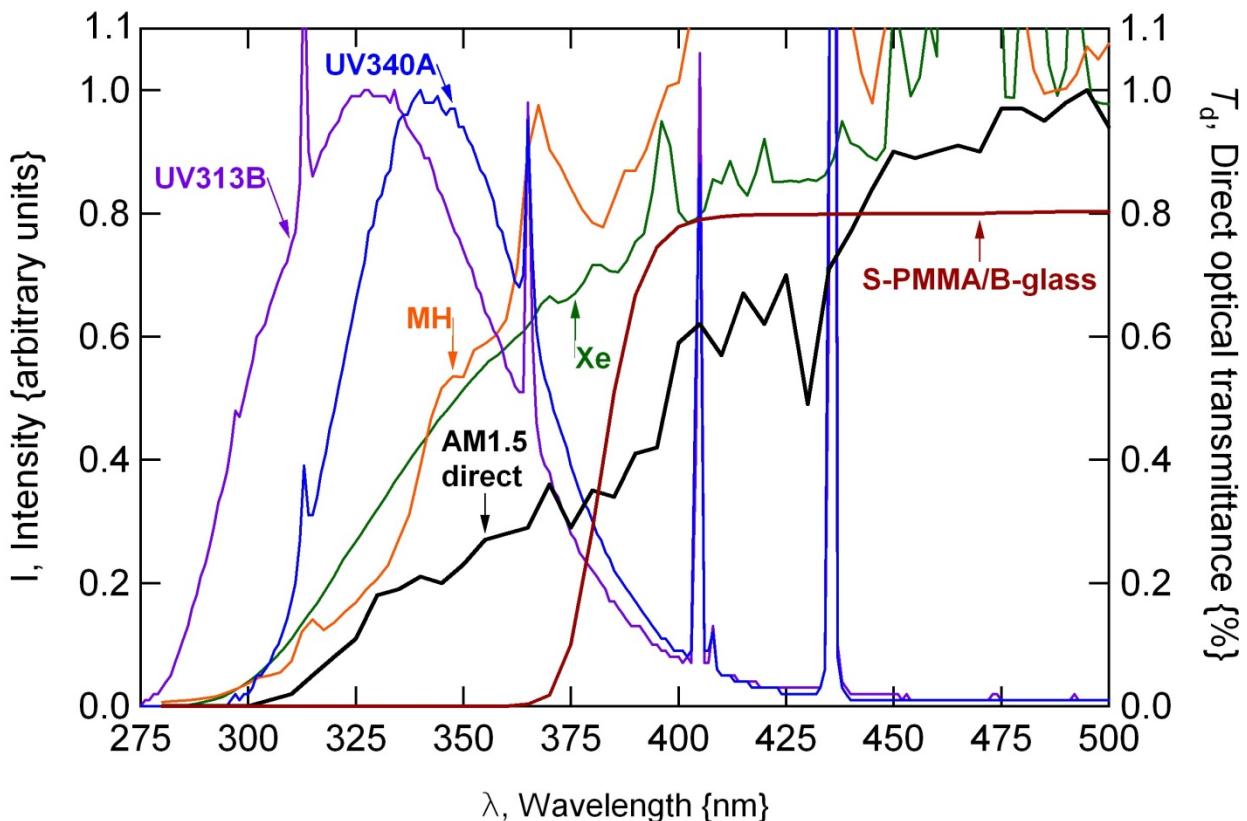
- FTIR, RAMAN, NMR
- TGA, DSC (polymer physics)

Test schedule:

0, 1, 2, 4, 6, 12, 18, 24, 30, 36 ... months

Optical Irradiance May Vary from CPV Transmittance

- PMMA transmits little ($T=1\%$) UV flux, $\lambda > 390$ nm
- Thermal content therefore has increased significance (coupled UV & thermal degradation)



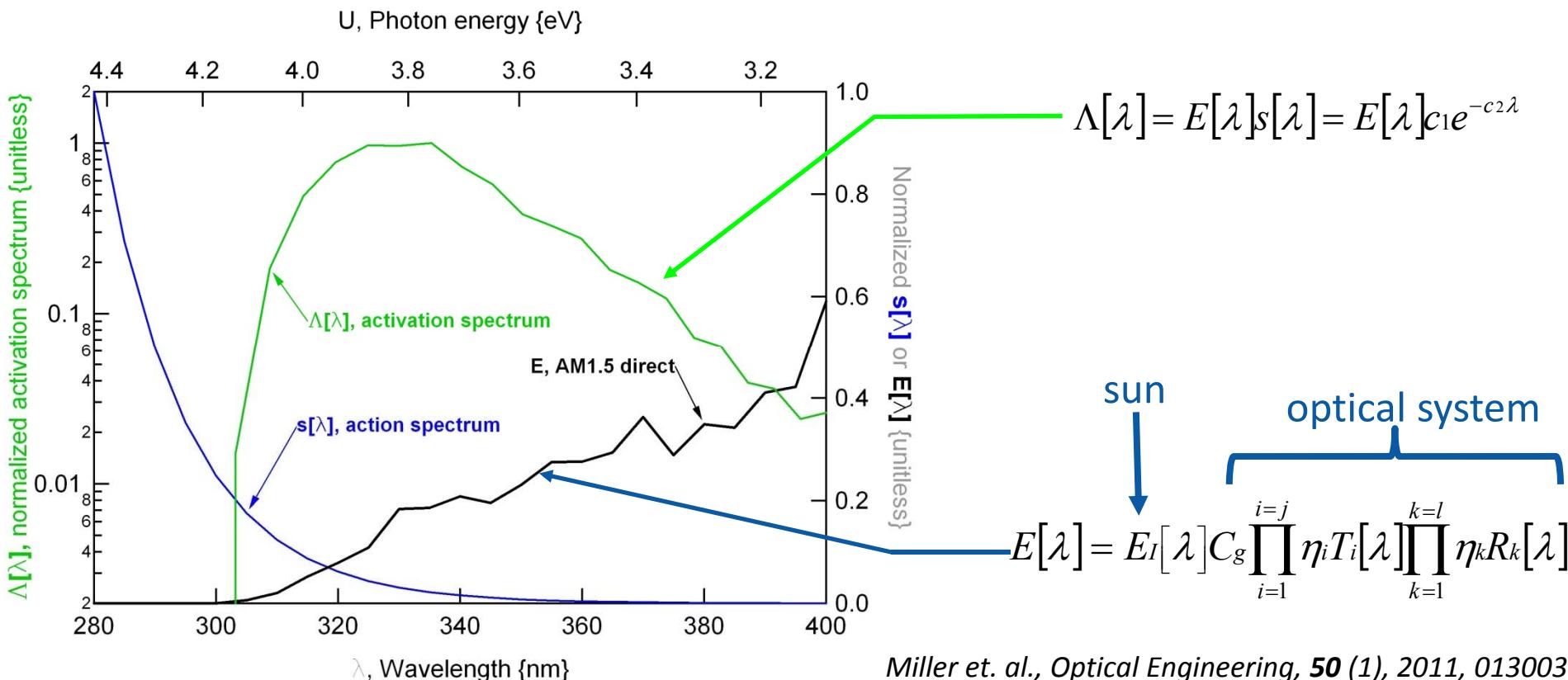
- Some popular indoor sources (UV 313V, UV340A) are completely inappropriate for a PMMA-enabled CPV system
- SoG Fresnel lens is substantially more transmitting ($T=89\%$) of UV

Miller et. al., PIP, DOI: 10.1002/pip.1241

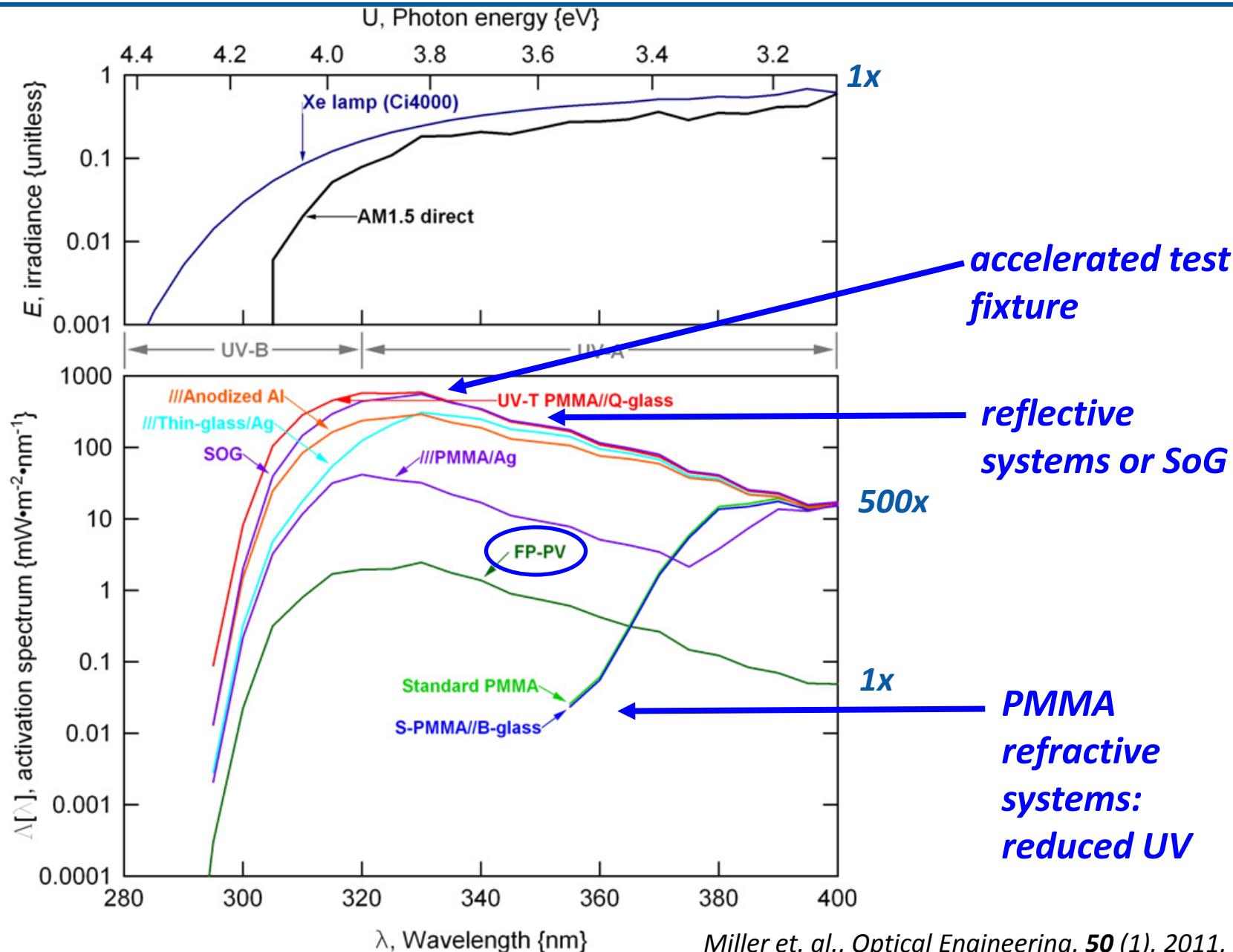
Irradiance for popular optical sources (including the sun) relative to the CPV optical system

UV Radiation: Damaging Dose

- Early weathering studies \Rightarrow total UV dose (damage vs. Joules or hours)
- Activation spectrum instead considers:
 1. characteristics of source & optical system
 2. effectiveness of damage at each λ ("action spectrum")
 3. may be unique to each characteristic (+ and -)



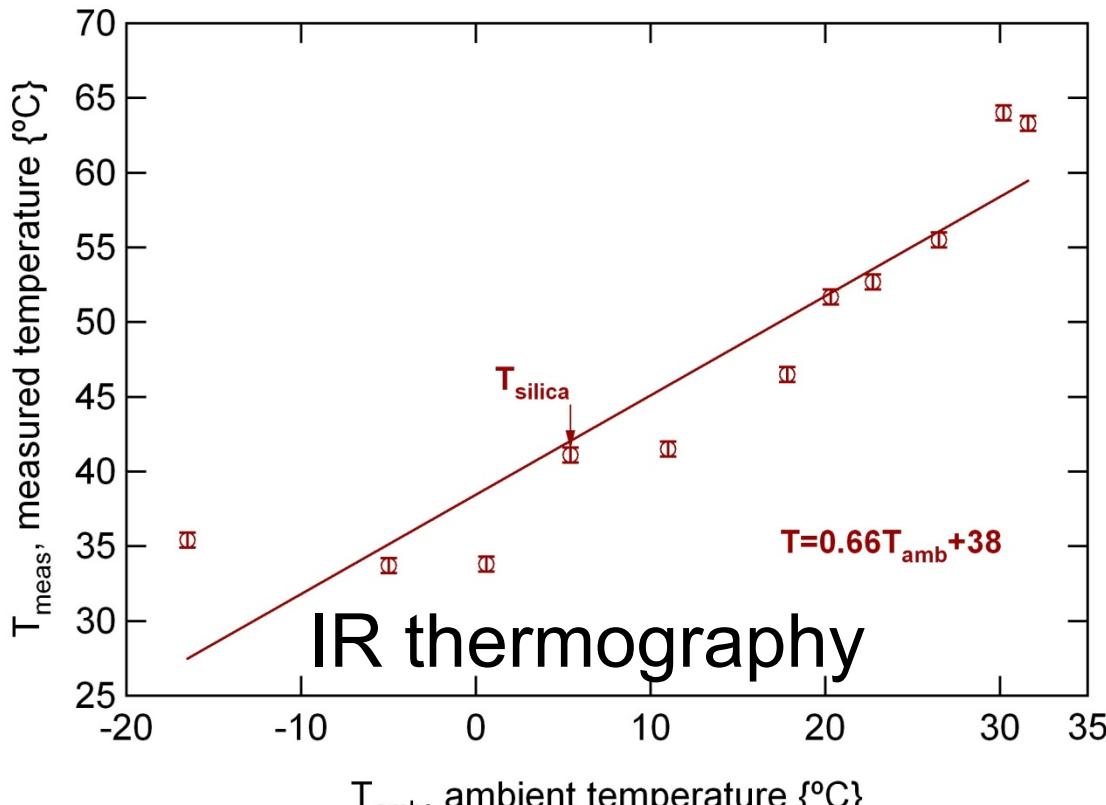
The Optical System Readily Affects UV & IR Dose



Miller et. al., Optical Engineering, 50 (1), 2011, 013003

The Field Conditions (Specimen Temperature)

- Specimen temperature proportional to optical (IR) absorptance (thermal management “system”: conduction to the frame.)
- Measured at solar noon. Factors: T_{amb} , irradiance, wind speed
- $\sim 40^{\circ}\text{C}$ temperature rise observed. $T_{\text{max}} 70\text{-}80^{\circ}\text{C}$ in summer.



PDMS specimen temperature, determined using optical thermography

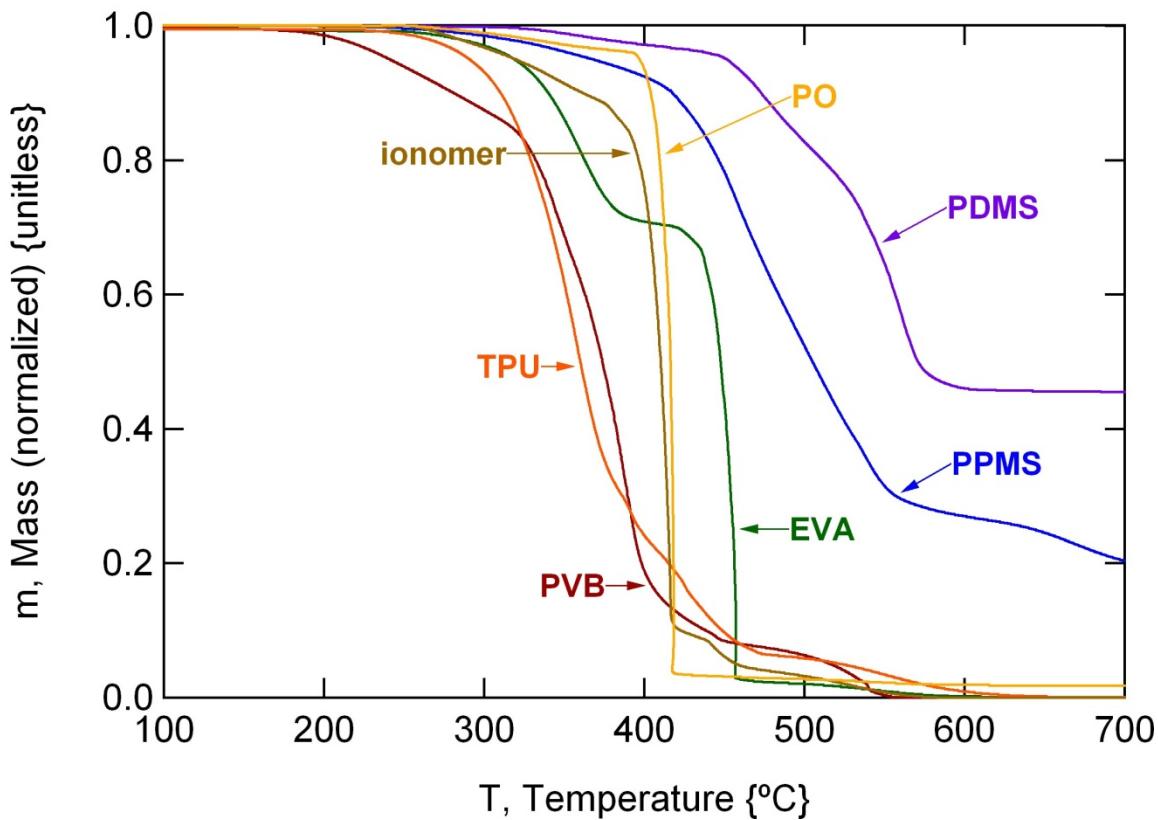
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Thermal Decomposition of the Encapsulation May Occur at High Temperature

- Thermal stability compared using thermogravimetric analysis (TGA) @ $20^{\circ}\text{C}\cdot\text{min}^{-1}$

- Onset of decomposition for hydrocarbons: $200\text{-}300^{\circ}\text{C}$

- Silicones more thermally stable: $T_{\text{onset}} 300\text{-}400^{\circ}\text{C}$



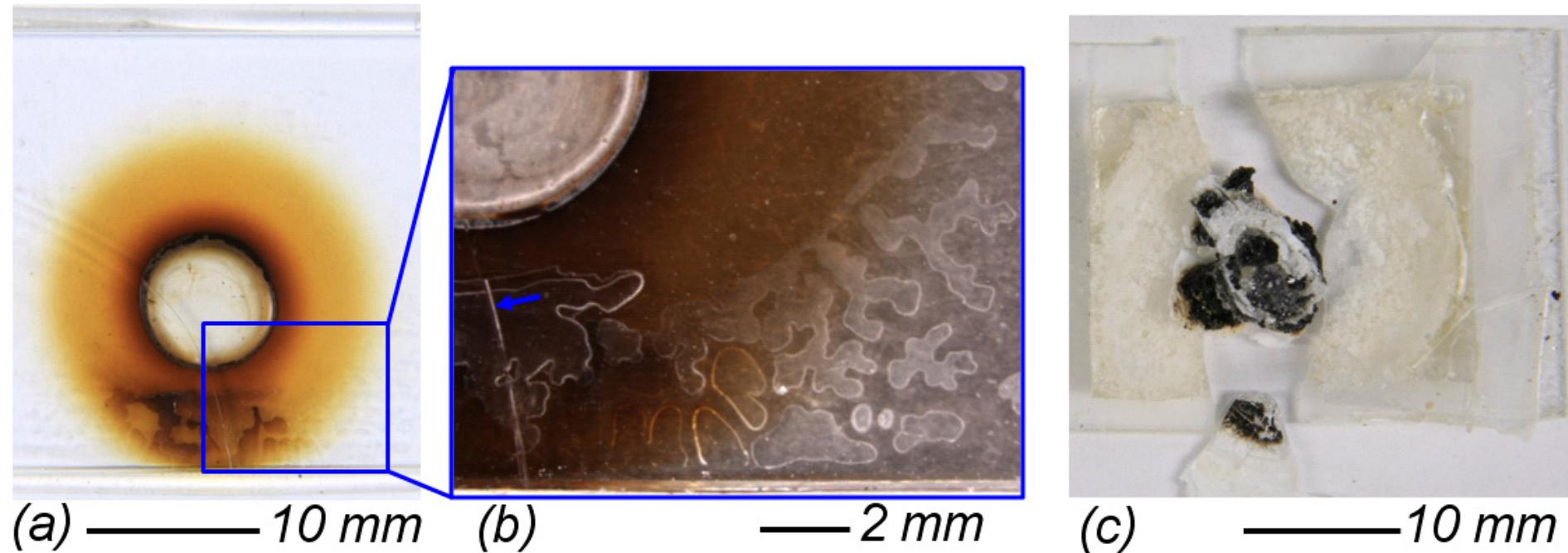
*Thermography data for representative materials from the study
Miller et. al., PIP, DOI: 10.1002/pip.1241*

*Remember T 's for later!

Results of Discovery Experiments

(The Homogenizer)

EVA: without homogenizer, rapid discoloration \Rightarrow combustion



optical images of EVA in (a) & (b), and PDMS in (c).
inset shows: voided center, char, cracked cover-glass, discoloration, delamination

silicone: without homogenizer \Rightarrow combustion

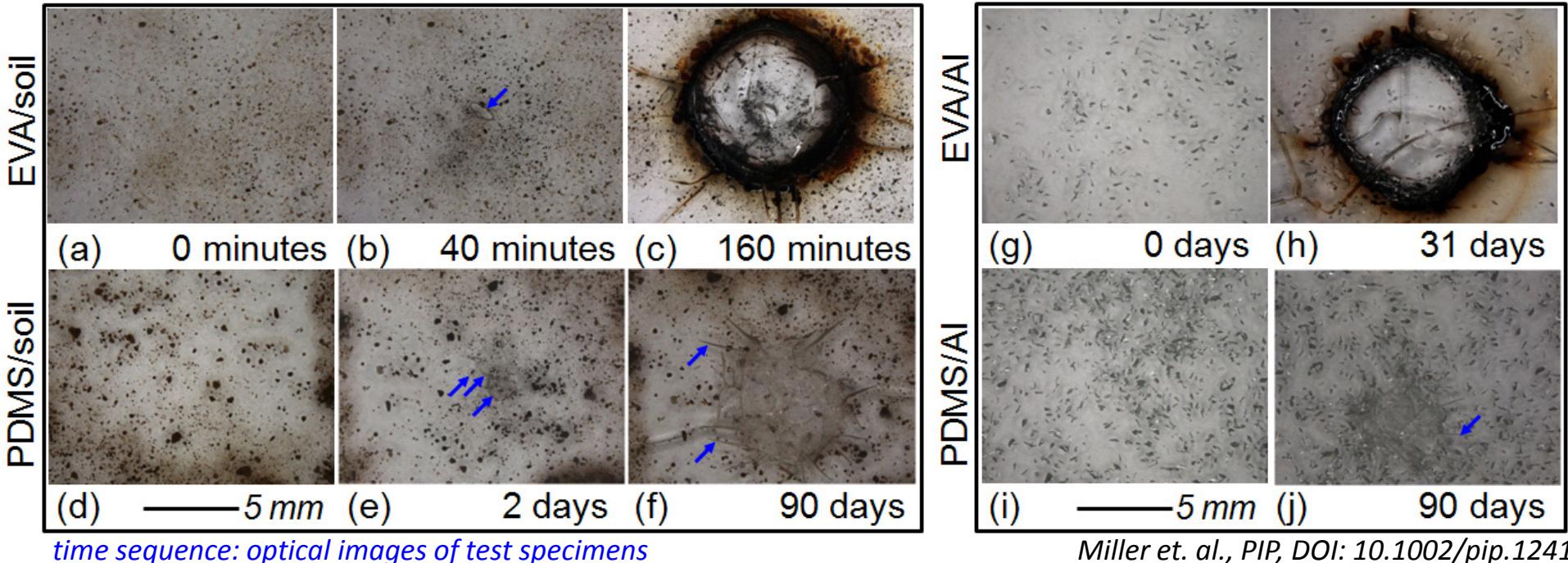
- Likely motivated by local hot spots (10^1 to $10^3 \cdot C_g$)

D.C. Miller, S.R. Kurtz, Solar Energy Materials and Solar Cells, 2011.

Results of Discovery Experiments

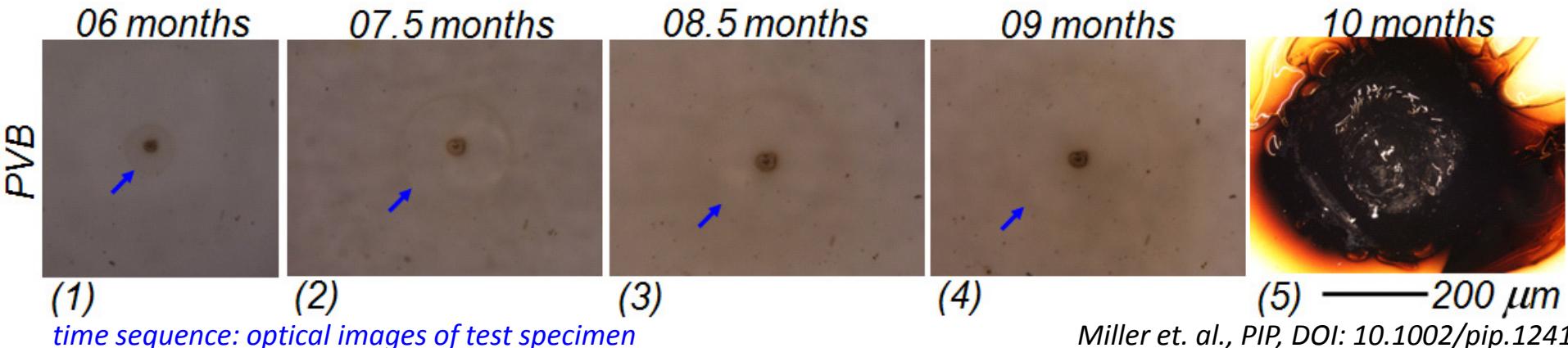
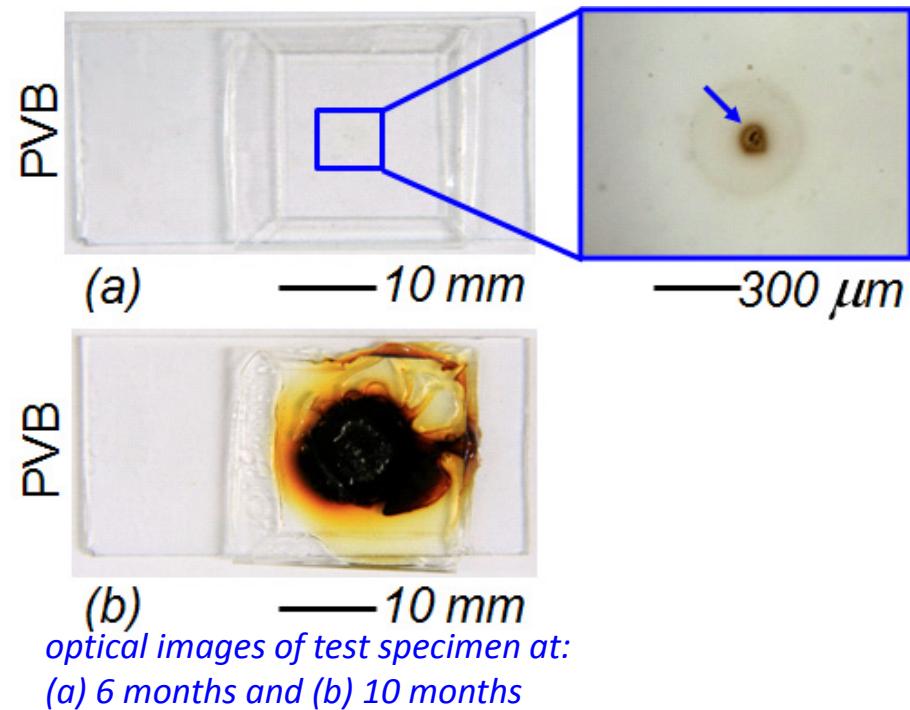
The Effect of Contamination)

- Intentionally introduce soil, Al, PE, or bubbles into EVA or silicone
- EVA: soil, Al, PE motivated localized discoloration \Rightarrow combustion
- silicone: soil, Al \Rightarrow localized cracking. (no primer present)
- elapsed time: minutes – days/weeks
- bubbles: no failure @ $C_g=500$, despite 4% measured $T[\lambda]$ reduction



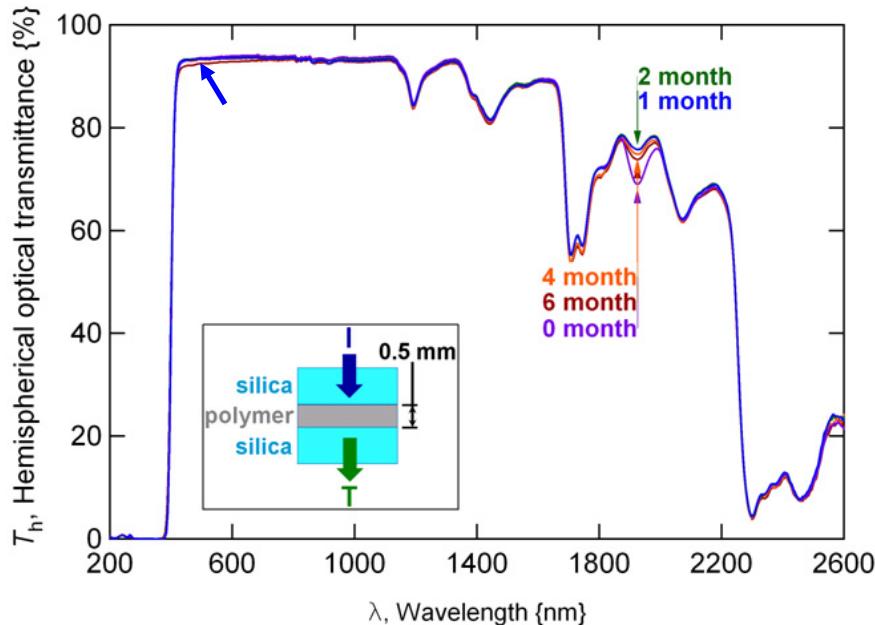
Results of the Formal Experiment (Hydrocarbon Specimens)

- PVB was the first material to demonstrate thermal runaway mediated failure
- The radius of the affected region was seen to slowly grow during the cold winter months



Miller et. al., PIP, DOI: 10.1002/pip.1241

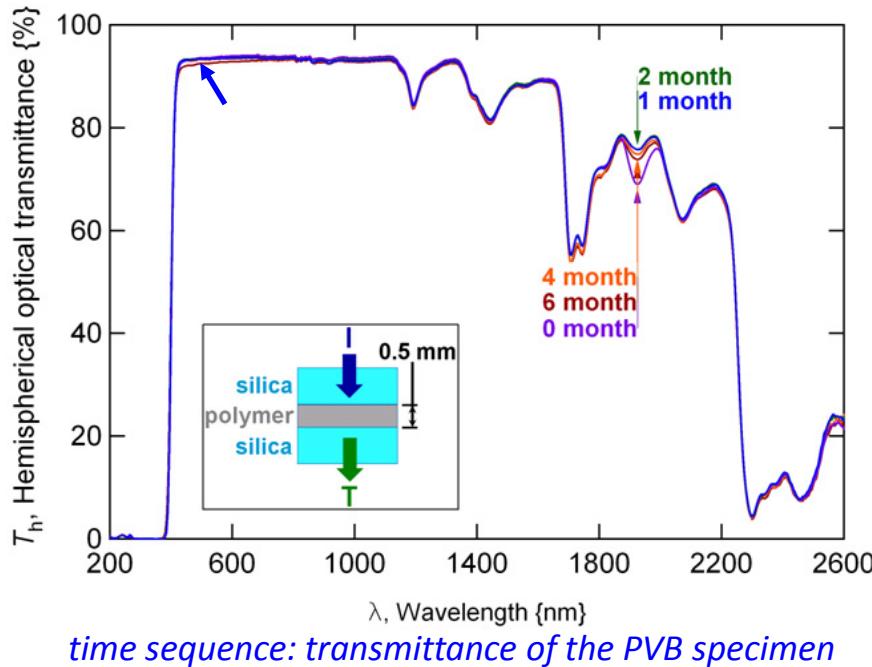
Results of the Formal Experiment (Hydrocarbon Specimens)



- Transmittance & YI not significantly affected, despite impending failure
- A diagnostic characteristic with predictive capability is preferred!!!

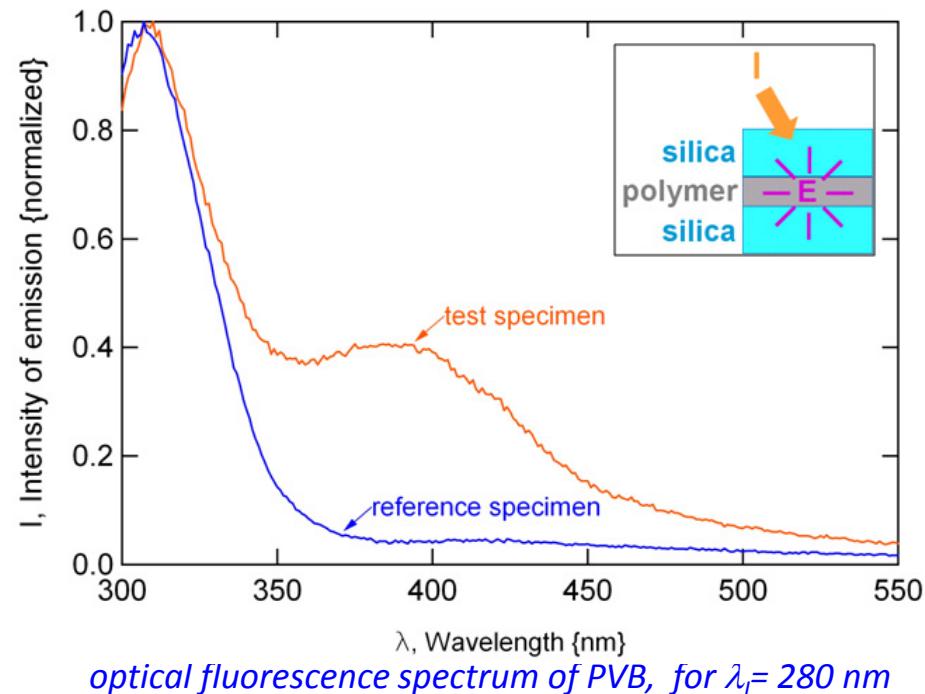
optical fluorescence spectrum of PVB, for $\lambda_f = 280$ nm

Results of the Formal Experiment (Hydrocarbon Specimens)



time sequence: transmittance of the PVB specimen

- Transmittance & YI not significantly affected, despite impending failure
- A diagnostic characteristic with predictive capability is preferred!!!



optical fluorescence spectrum of PVB, for $\lambda_f = 280$ nm

- Optical & Raman spectroscopy clearly indicate fluorescence
- These techniques may help understand the degradation mechanism (e.g., chromophores)

Results of the Formal Experiment

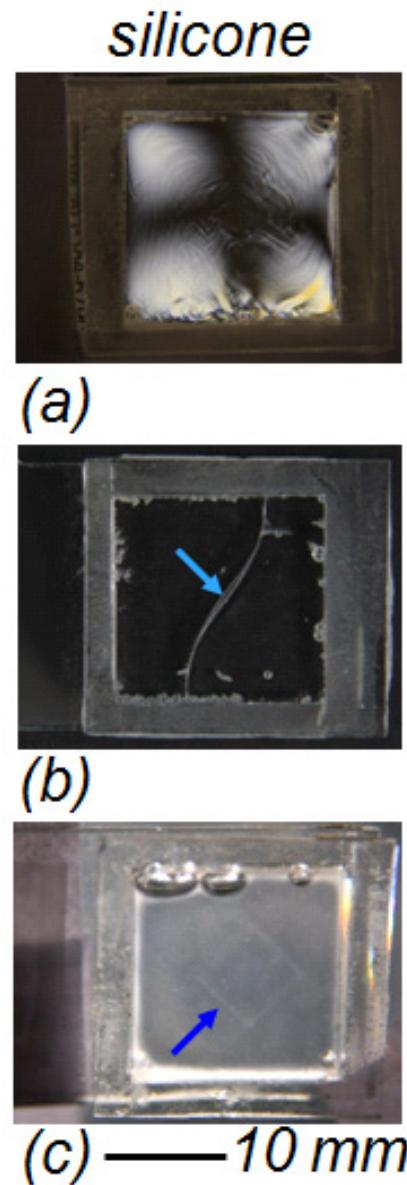
(Silicone Specimens)

- Observations of silicone specimens include: (a) densification, (b) cracking, and (c) haze formation

No mass change with time for the (5) **densified** specimens \Rightarrow likely occurred during molding

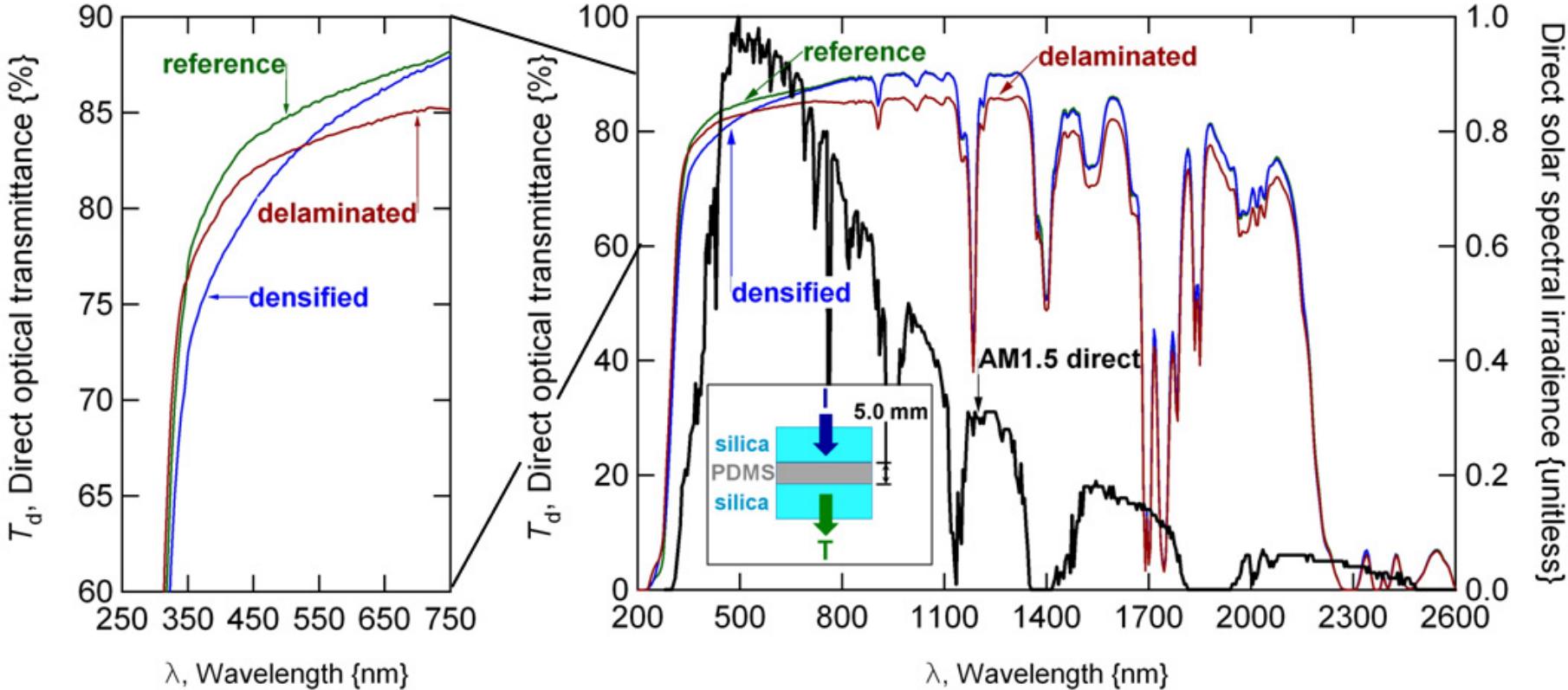
- **Crack** advancement occurred during cold weather periods only \Rightarrow likely motivated by CTE misfit
- Additional fractured specimens may be emerging

Haze formation is attributed to one material's unique formulation



optical images of silicone specimens, including those obtained using (a) cross-polarization or (c) back-lighting

Results of the Formal Experiment (Densified Silicone Specimens)



- Densification is not delamination
- Densification does scatter direct light

Problematic for CPV?

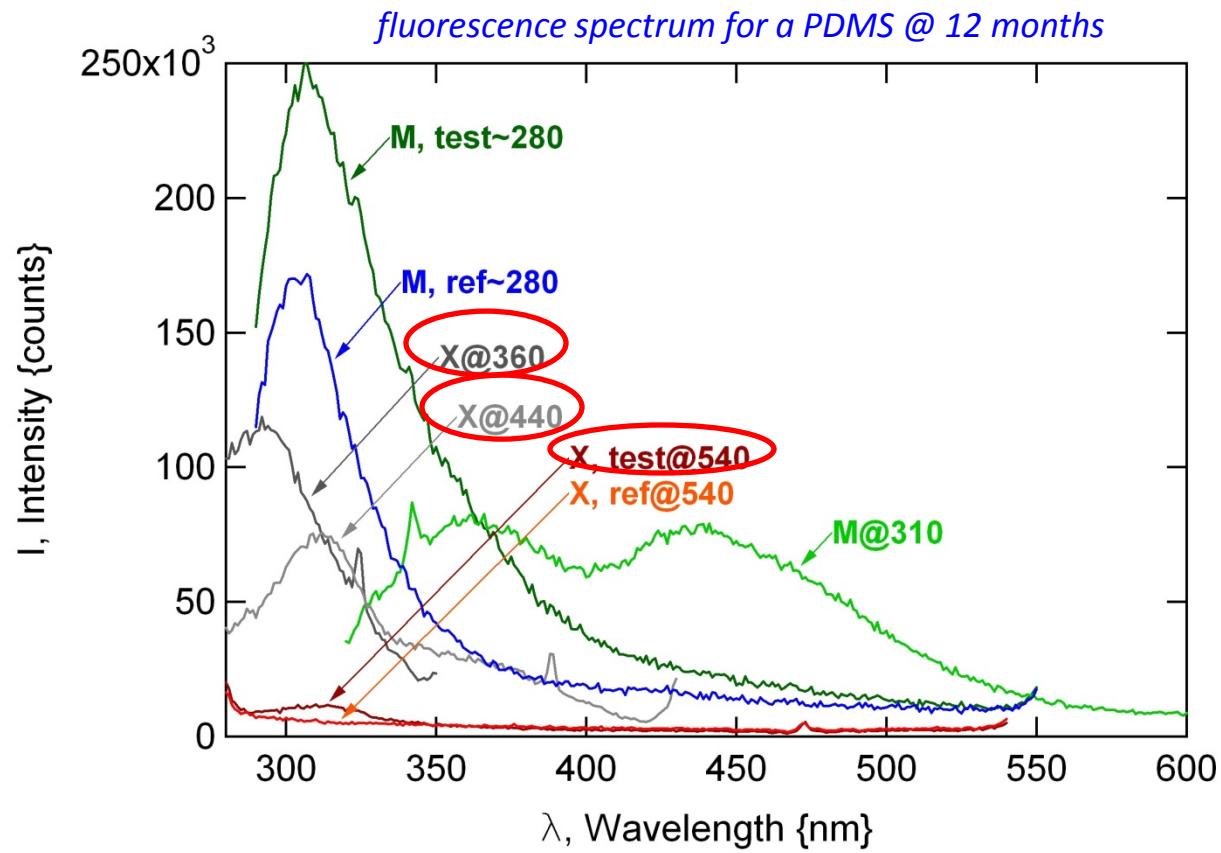
- Current limited condition (blue light)
 - Optical attenuation (less power)
- ⇒ May not be significant in thin bond layers

Fluorescence Identifies the Silicones Are Affected!

- Unexpected new peaks identified for all silicone specimens!

- The particular details location and relative intensity of the new M_t peaks varied with formulation

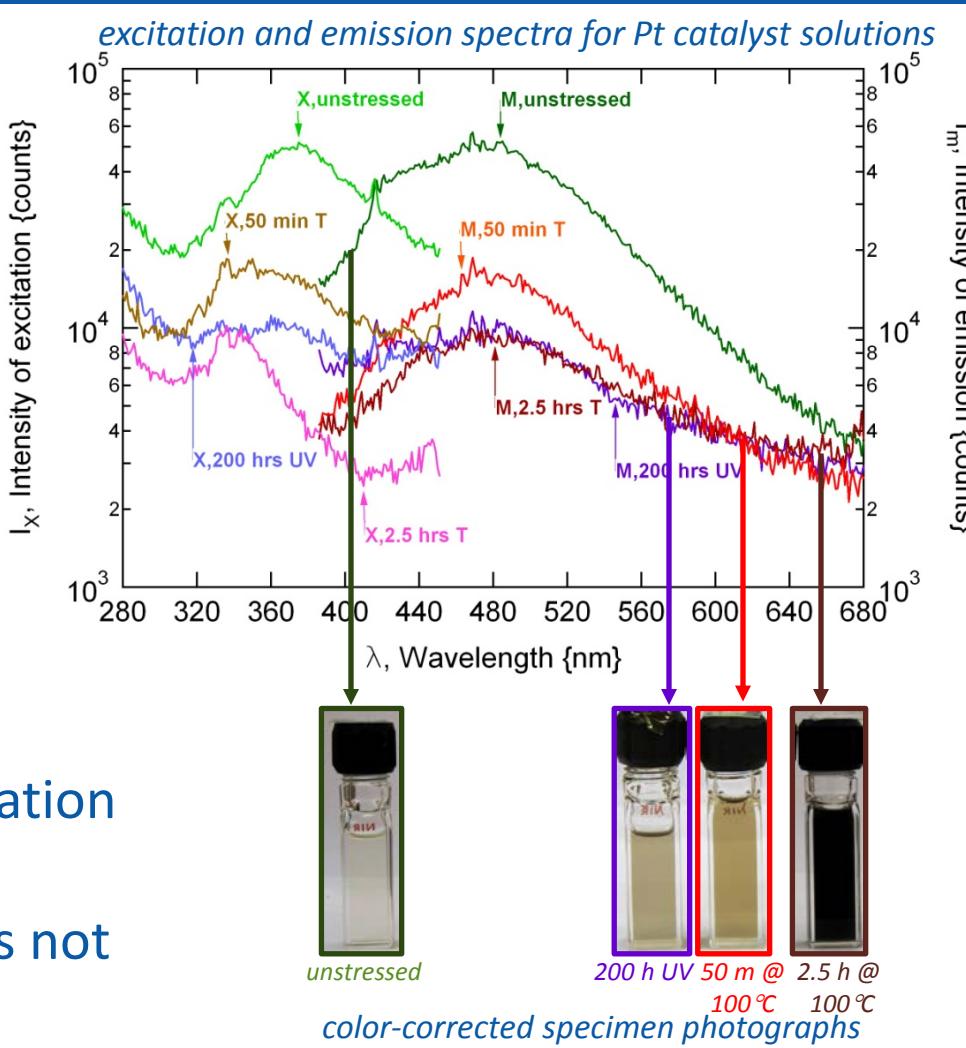
- Attributed to Pt catalyst (working to verify)



- The implications are unclear. PDMS is historically robust in extreme environments. $\lambda_x \dots < 390$ nm for PMMA, ~320 nm for SoG

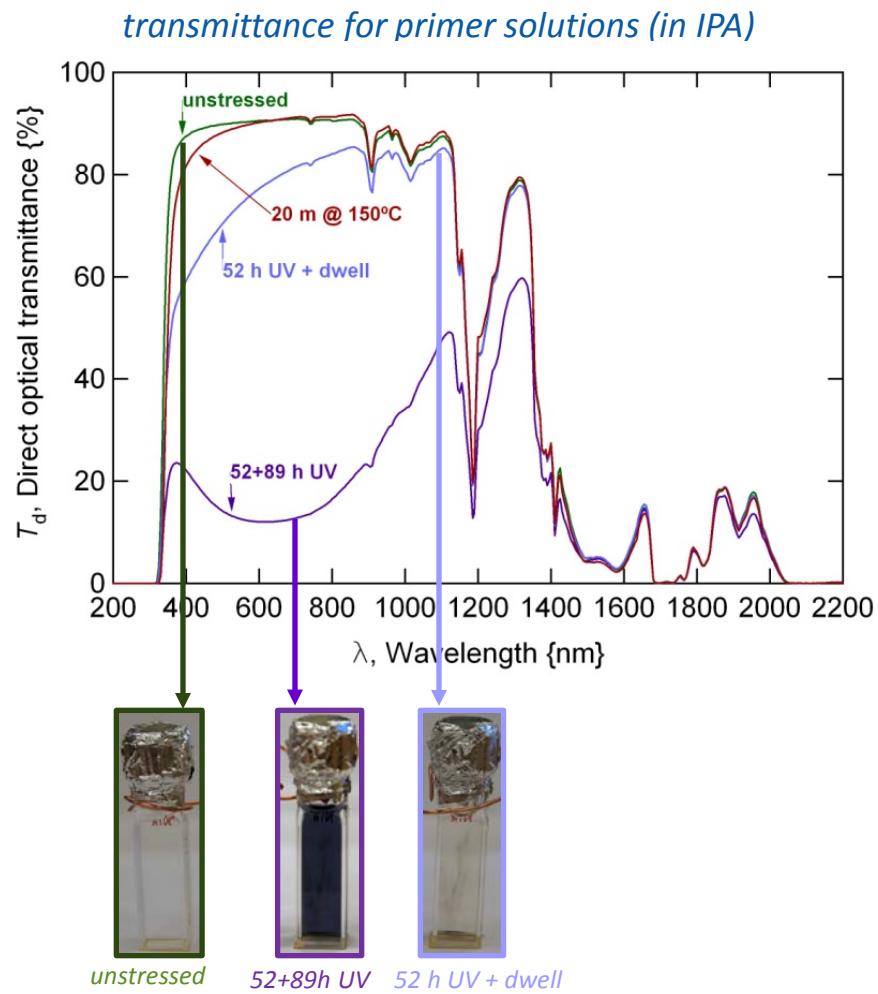
UV and/or Temperature Can Degrade Pt Catalyst

- Karstedt's catalyst, Pt(0), examined in tetramethyldivinyldisiloxane
- Catalyst loses fluorescence with UV or T
- Organometallic literature: mononuclear Pt with ligands → colloidal Pt, 3-5 nm Ø
- Discoloration (optical absorptance) could motivate thermal runaway
- No evidence to date of optical degradation in NREL specimens
- Fluorescence of catalyst solution does not correspond to that in x-linked PDMS
- Alternate pathways: different catalyst type (ligands), peroxide cured silicone, PMMA on glass (PoG) lenses, AR coatings



UV Can Degrade Silicone Primers

- Dow-Corning 92-023 used in all NREL PDMS specimens
- The Ti based primer (on glass) reduces UV transmittance for $\lambda < 300$ nm ($n \text{ TiO}_2 = 2.5$)
- Experiments identify primer is quite photoactive:
 - discoloration with minor fluorescence
- Transparency recovered with time (O_2 facilitated?)
- TiO_2 used in self cleaning coatings.
(UV driven consumption of organic contamination). Affect on PDMS is unclear.
- Alternate pathway: Sn catalyzed primers ($n \text{ SnO} = 2.1$)



Summary & Conclusions

Field study of the durability of polymeric encapsulation materials for CPV

Discovery experiments:

- Quickly confirmed the importance of an optical homogenizer
- Al, soil, polymeric contamination \Rightarrow T runaway & combustion of EVA
- Al, soil contamination \Rightarrow cracking of silicone

Formal experiment:

- 17 of 25 specimens not discussed today!
- 3 of 25 specimens “failed”.

PVB: localized discoloration \Rightarrow thermal runaway \Rightarrow combustion

Fluorescence & Raman spectroscopy may diagnose & provide prediction

Silicone: densification, cracking, haze-formation

Densification affects the direct transmittance

PDMS Fluorescence:

- Working to understand observed peaks; alternative “solutions” identified

*Transmittance of optical system and corresponding activation spectrum of the encapsulation are critical to encapsulation durability

Acknowledgements

- NREL: Dr. Keith Emery, Dr. Daryl Myers, Dr. John Pern, Matt Beach, Christa Loux, Tom Moriconi, Marc Oddo, Bryan Price, Kent Terwilliger, Robert Tirawat



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Paper: “The Durability of Polymeric Encapsulation Materials for Concentrating Photovoltaic Systems”, Prog. Photovoltaics,
DOI: 10.1002/pip.1241.