**RESEARCH ON OIL RECOVERY MECHANISMS
IN HEAVY OIL RESERVOIRS**

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Executive Summary

The United States continues to rely heavily on petroleum fossil fuels as a primary energy source, while domestic reserves dwindle. However, so-called heavy oil (10 to 20 °API) remains an underutilized resource of tremendous potential. Heavy oils are much more viscous than conventional oils. As a result, they are difficult to produce with conventional recovery methods such as pressure depletion and water injection. Thermal recovery is especially important for this class of reservoirs because adding heat, usually via steam injection, generally reduces oil viscosity dramatically. This improves displacement efficiency. The research described here was directed toward improved understanding of thermal and heavy-oil production mechanisms and is categorized into:

- (i) flow and rock properties,
- (ii) in-situ combustion
- (iii) additives to improve mobility control
- (iv) reservoir definition, and
- (v) support services

The scope of activities extended over a three-year period.

Significant work was accomplished in the area of flow properties of steam, water, and oil in consolidated and unconsolidated porous media, transport in fractured porous media, foam generation and flow in homogeneous and heterogeneous porous media, the effects of displacement pattern geometry and mobility ratio on oil recovery, and analytical representation of water influx. Significant results are now described briefly.

In the area of flow properties, Akin *et al* conducted experiments using a computed tomography (CT) scanner and also a simulation study to examine the temperature dependence of water/oil relative permeability curves (Akin *et al.* 1999). In the dynamic mode of relative permeability measurement and at temperatures from about 20 to 80 °C, viscous instabilities dominate the displacement pattern. Interpretations based on assumption of one-dimensional flow are, thus, invalid. Simulations confirm that input relative permeability curves differ greatly from the apparent curves obtained using standard interpretation techniques based on a one-dimensional flow assumption.

An experimental apparatus for investigating relative permeability, capillary pressure, and fluid flow characteristics of low permeability rocks was designed specifically for use in CT scanners (Schembre *et al.* 1998). Experiments and interpretation of results were completed for spontaneous water imbibition experiments in diatomite, chalk, and sandstone. Air/water and oil/water systems were studied with differing initial water saturations.

To better understand multiphase flow and matrix to fracture fluid transfer, experiments on a fractured Berea sandstone system were conducted (Rangel-German *et al.* 1998). The progress of fluid movement was measured with CT scanning, as in the previous experimental studies. Multiphase flow in the fracture system was observable with the CT scanner. Fractures with the narrowest apertures showed the most stable displacement fronts, largest effects of capillary pressure within the fracture, and fastest breakthrough times. Using fine-grid simulations and a commercial simulator, fracture relative permeability and capillary pressure curves were obtained.

Several projects studied aspects of gas mobility in porous media and the effect of steam on porous media. In regard to foam for the control of gas mobility in porous media, Bertin *et al* studied foam generation and transient displacement in heterogeneous porous media (Bertin *et al.* 1998). A composite heterogeneous porous medium was created by centering a nearly homogeneous sandstone within a cylindrical tube and filling the annular space with a more permeable homogeneous sand. Capillary communication between the two zones of the composite core was prevented by the placement of a teflon jacket around the sandstone. Removal of the jacket allowed the heterogeneities to communicate. Using the CT scanner, it is concluded that displacement by foam can be efficient in both cases. Foam can control gas mobility in a manner that helps to equalize gas flow resistance in media with different permeabilities.

Research into foam properties was also conducted by Sagar *et al* (1998). The aim of this work was to visualize and understand the deleterious effects of oil on some foaming agents. A silicon micromodel etched with an accurate representation of the pore network of a Berea sandstone allowed direct observation of foam generation, flow, and destruction. It was found that fluorosurfactants produce strong, static, gas-blocking foams while alpha-olefin sulfonates do not in the presence of crude oil. All experiments were conducted close to atmospheric pressure.

Another area of recent research is the dissolution of a siliceous rock matrix when it is subjected to hot steam condensate (Bhat and Kovscek 1998). For low permeability porous media where oil recovery occurs by steam injection, dissolution of small amounts of rock mineral, transport of the dissolved rock, and reprecipitation in cooler regions of the reservoir could alter the distribution of porosity and permeability. Network models have been employed to understand how permeability shifts with evolving pore topology. This modeling effort has explained previous experimental results where a very strong power-law relationship between permeability and porosity was noted for hot fluid injection into fine-grained diatomite.

In the area of reservoir definition, study has focused on the shape of displacement patterns as a function of mobility ratio and water encroachment into oil and gas reservoirs. Wang *et al* dispelled the common notion that areal sweep efficiency from a staggered-line-drive pattern is always better than that from a five-spot pattern (Wang *et al.* 1999). Using a streamline based simulator, it was found that the five-spot pattern results in superior recovery efficiency compared to other patterns when the mobility ratio is very favorable. Also in this area, Brigham developed an analytical solution to diagnose the effects of water influx from the perimeter of an oil reservoir (Brigham 1998). Essentially, the exact infinite series solutions for various geometries were transformed into very accurate closed form approximations for ease of calculation. Linear, radial, and spherical systems were examined.

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1. Introduction and Objectives

The Stanford University Petroleum Research Institute (SUPRI) studied heavy-oil production mechanisms with the ultimate goal of increasing oil reservoir recovery efficiency. Additional aspects of the research effort covered a broad range of techniques that are useful for all improved oil recovery targets.

The research was organized into five topical areas:

- **Flow and Rock Properties.** The objectives of flow and rock properties were to assess the effect of reservoir parameters, such as temperature and pressure, on relative permeability and capillary pressure. Included in this effort were improvements to existing laboratory techniques to better study heat and fluid transport in porous media at reservoir temperature and pressure. Ultimately, this results in better data for use in predictive numerical models.
- **In-Situ Combustion.** In the area of in-situ combustion, goals were to study parameters affecting combustion field projects such as catalytic metallic salt additives, to study the effect of oxygen concentration on combustion parameters, to model accurately reaction kinetics and flow behavior of combustion processes.
- **Additives to Improve Mobility Control.** Objectives were to study and model the foam process for improving mobility control of injected gases including gravity override and channeling, with emphasis on steam injection. Objectives also included elucidation and modeling of the mechanisms of foam flow in porous media with oil present.
- **Reservoir Definition.** The main goal of reservoir definition was to develop techniques for evaluating the response of oil reservoirs to fluid injection. Reservoir geology and flow processes were studied in concert and this leads to interpretation procedures that allow us to better understand oil reservoirs.
- **Support Services.** In this category, objectives were to provide technical support, economic evaluation, and design and monitoring of field tests. Included here were technology transfer efforts aimed at sharing technology developed in all of the research areas with industry and other researchers.

The following sections summarize our accomplishments, present our technical results in detail, and chronicle our technology transfer efforts.

2. Accomplishments

Accomplishments during the contract period take the form of technical reports, papers presented at technical meetings, and manuscripts submitted to journals. A universally accessible world wide web site was also created to aid in the rapid distribution of technical information.

During this period of research SUPRI also held three Industrial Advisory Committee reviews and participated at DOE's request in 3 memorandum of understanding (MOU) meetings with Venezuela. Next accomplishments are summarized by category.

2.1 Industrial Outreach

- SUPRI Industrial Advisory Committee Review Meeting, Hartley Conference Center, Stanford University, April 15-16, 1999.
- SUPRI Industrial Advisory Committee Review Meeting, Hartley Conference Center, Stanford University, April 16-17, 1998.
- University Research Forum, held jointly with Petroleum Technology Transfer Council, University of Southern California, and the University of California, Berkeley, Los Angeles, CA, May 15, 1997.
- SUPRI Industrial Advisory Committee Review Meeting, room 450 Durand Building, Stanford University, April 4-5, 1997.
- Horizontal Wells and Enhanced Oil Recovery, meeting following the International Thermal Operations and Heavy Oil Symposium, Bakersfield, CA February 12, 1997.
- Establishment of a world wide web site describing our work and providing online distribution of research reports and papers at <http://ekofisk.stanford.edu/supria.html>.

2.2 Technical Reports

- SUPRI TR-119: "Computer Simulation of Single-Well Steam Assisted Gravity Drainage (SW-SAGD)" by K. E. Elliot and A. R. Kavscek (July 1999).
- SUPRI TR-118: "An Analytical Model for Simulating Heavy-Oil Recovery by Cyclic Steam Injection Using Horizontal Wells" by U. K. Diwan and A. R. Kavscek (July 1999).
- SUPRI TR-116: "An Experimental and Theoretical Investigation of Multi- Phase Flow in Fractured Porous Media" by E. R. Rangel-German, L. M. Castanier, and S. Akin (June 1999).
- SUPRI TR-115: "A Study of the Effect of Mobility Ratios on Pattern Displacement Behavior" by Y. Wang, A. R. Kavscek, and W. E. Brigham (December 1998).
- SUPRI TR-114: "Spontaneous Imbibition in Low Permeability Porous Media" by J. M. Schembre, S. Akin, and A. R. Kavscek (December 1998).
- SUPRI TR-113: "Modeling Permeability Alteration in Diatomite Reservoirs During Steam Drive" by S. K. Bhat and A. R. Kavscek (July 1998).
- SUPRI TR-112: "An Experimental Investigation of Foam Flow in Homogeneous and Heterogeneous Porous Media" by O. G. Apaydin, H. J. Bertin, L. M. Castanier, and A.R. Kavscek (July 1998).
- SUPRI TR-110: "Oil-Foam Interactions in a Micromodel" by N. S. Sagar and L. M. Castanier (November 1997).
- SUPRI TR-108: "A Comparison of Mass Rate and Steam Quality Reductions to Optimize Steamflood Performance" by G. L. Messner (April 1997).
- SUPRI TR-107: "CT Imaging Techniques for Two-Phase and Three-Phase In-Situ Saturation Measurements", by B. C. Sharma, W. E. Brigham, and L. M. Castanier (September 1996).

- SUPRI TR-104: "CT Measurements of Two-Phase Flow in Fractured Porous Media", by R.G. Hughes, W. E. Brigham, and L. M. Castanier. (June 1997).

2.3 Annual Reports

- SUPRI TR-117: "SUPRI Heavy Oil Research Program Twenty Second Annual Report" by W. E. Brigham, A. R. Kavscek, and L. M. Castanier (May 1999).
- SUPRI TR-111: "SUPRI Heavy Oil Research Program Twenty First Annual Report" by W. E. Brigham, A. R. Kavscek, and L. M. Castanier (May 1998).
- SUPRI TR-109: "SUPRI Heavy Oil Research Program Twentieth Annual Report" by W. E. Brigham, A. R. Kavscek, and L. M. Castanier (May 1997).

2.4 Conference Presentations

- "Simulation of Early-Time Response of Single-Well Steam-Assisted Gravity Drainage", SPE 54618, by K. T. Elliot and A. R. Kavscek, presented at the SPE Western Regional Meeting, 26-27 May 1999, Anchorage, AK.
- "Multiphase Flow Properties of Fractured Porous Media", SPE 54591, by E. Rangel-German, S. Akin and L. M. Castanier, presented at the SPE Western Regional Meeting, 26-27 May 1999, Anchorage, AK.
- "Imbibition Studies of Low-Permeability Porous Media", SPE 54590, by S. Akin and A. R. Kavscek, presented at the SPE Western Regional Meeting, 26-27 May 1999, Anchorage, AK.
- "Effect of Temperature on Heavy Oil/Water Relative Permeabilities", SPE54120, by S. Akin, L. M. Castanier, and W. E. Brigham presented at the 1999 SPE International Thermal Operations Symposium, 17-19 March 1999, Bakersfield, CA.
- "Reservoir Simulation of Foam Displacement Processes", UNITAR 1998.186, by A. R. Kavscek presented at the 7th Unitar International Conference on Heavy Crude and Tar Sands, 27-31 October 1998, Beijing, China.
- Messner, G.I., L.M. Castanier, and W.E. Brigham: "A Comparison of Mass Rate and Steam Quality Reductions to Optimize Steamflood Performance," presented at the Unitar Symposium, October 27-30, 1998, Beijing, China.
- "Effect of Mobility Ratio on Areal Sweep Efficiency and Pattern Flood Behavior", by Y. Wang, A. R. Kavscek, and W. E. Brigham presented at the 19th Annual International Energy Agency Workshop and Symposium, 4-7 October 1998, Carmel, CA.
- "Experimental and Theoretical Investigation of Multiphase Flow in Fractured Media", by E. R. German, S. Akin, and L. M. Castanier presented at the Geothermal Resources Council Meeting 1998 Annual Meeting, 20-23 September 1998, San Diego, California, USA.
- "Foam Flow in Heterogeneous Porous Media: Effect of Crossflow", SPE 39678, by H. J. Bertin, O. G. Apaydin, L. M. Castanier, and A. R. Kavscek presented at the SPE/DOE Improved Oil Recovery Symposium, 19-22 April 1998, Tulsa, OK, USA.
- Joshi, Sameer, L.M. Castanier, and W.E. Brigham: "Techno-Economic and Risk Evaluation of an EOR Project," SPE 39578, paper presented at the 1998 SPE India Oil and Gas Conference and Exhibition, February 17-19, 1998, New Delhi.

2.5 Journal Publications

- Wang, Y., A. R. Kovscek, and W. E. Brigham, " Effect of Mobility Ratio on Pattern Behavior of a Homogeneous Porous Medium," *In Situ*, **23(1)**, p. 1-20 (1999).
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- Kovscek, A. R., T. W. Patzek, and C. J. Radke. "Mechanistic Foam Flow Simulation in Heterogeneous and Multidimensional Porous Media," *Society of Petroleum Engineers Journal*, **2(4)**, p. 511-526 (1997).
- Castanier, L.M. and W.E. Brigham: "Modifying In-situ Combustion with Metallic Additives," *In Situ* **21 (1)**, 27-45 (1997).
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PROJECT 1: FLOW PROPERTY STUDIES

To assess the influence of different reservoir conditions (temperature and pressure) on the absolute and relative permeability to oil and water and on capillary pressure.

1.1 EFFECT OF TEMPERATURE ON HEAVY OIL/WATER RELATIVE PERMEABILITIES

(S. Akin, L.M. Castanier, and W.E. Brigham)

This paper, SPE 54120, was presented at the 1999 SPE International Thermal Operations Symposium, March 17-19, 1999, Bakersfield, California.

SPE 54120

Effect of Temperature on Heavy Oil/Water Relative Permeabilities

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Abstract

In the first part of this study, the accuracy of the JBN technique for the determination of heavy oil/water relative permeabilities, and the effect of temperature on relative permeabilities is examined by giving numerical as well as experimental examples. Using the JBN technique leads to a false temperature dependence of relative permeability curves. In the second part, we present unsteady state relative permeability experiments with initial brine saturation at differing temperatures conducted using South Belridge sand and heavy oil. A new three step experimental technique and an analysis procedure were developed to test the effect of temperature on relative permeabilities. In this technique, an ambient temperature unsteady-state relative permeability run is conducted in the first stage, and following that the temperature is increased twice (i.e. 122°F and 150°F). Two phase saturation profiles along the sand pack are measured using a CT scanner. A commercial black oil simulator, coupled with a global optimization code is then used to estimate two phase relative permeabilities. Experimental saturation profiles, differential pressure and recovery data collected from both the ambient and higher temperature data are used in the numerical model. It has been observed that a single set of relative permeability curves can represent both the ambient and high temperature parts of the experiment. This suggests that relative permeability is not a function of temperature at least for the system tested.

Introduction

For modeling thermal recovery processes for heavy oil recovery, one needs to know not only the relative permeabilities at the original reservoir temperature but also the effect of increasing temperature on the relative permeability

curves. Over the past three decades, a number of experimental studies have reported contradictory temperature effects on two-phase relative permeabilities as well as residual oil saturations in porous media. Although a number of studies have reported significant effects no consensus has emerged on the generality of such effects or on possible mechanisms causing such effects. Therefore this study aims to determine the effect of temperature on relative permeability curves for a reservoir sand/ heavy-oil/ NaBr water system and to examine the use of unsteady-state relative permeability estimation techniques in such systems.

Edmondson¹ found a reduction in residual oil saturation in Berea cores as the temperature increased. Using refined oils and an unsteady state experimental technique, he noted that the relative permeability ratio of oil to water shifted toward a higher water saturation as the temperature increased.

Poston et al.² investigated the effects of temperature on unconsolidated sand using refined oils. They reported that irreducible water saturation increased, and the residual oil saturation decreased with increasing temperature. Moreover, they concluded that relative permeabilities to both oil and water increased with increasing temperature.

Contradicting previous researchers, two studies^{3,4} have reported temperature independent relative permeability data obtained from unsteady state experiments in unconsolidated and Berea sandstone cores using refined oils. Miller and Ramey⁴ noted various artifacts, such as; material balance errors, viscous instability, clay migration, wettability alteration by miscible cleaning methods, and capillary pressure end effects, that may have affected former conclusions on the effect of temperature on relative permeability data.

Closmann et al.⁵ used the steady state technique to measure altered, unaltered and de-asphalted tar and brine relative permeabilities at elevated temperature using previously frozen Peace River cores. They concluded that, at higher temperatures, tar and water relative permeability curves shifted toward lower water saturations.

Maini and Batcicky⁶ conducted unsteady state experiments at temperatures ranging from room temperature to 522°F using a previously frozen core from a heavy oil reservoir, and stock tank oil and formation brine from the same reservoir. They measured the absolute permeability to formation brine using a transient pressure decay test, and used history-matching calculations to obtain relative permeabilities. They concluded

that, irreducible water saturation increased and residual oil saturation decreased with increasing temperature up to a temperature limit. Moreover, they reported that relative permeability to oil decreased, and relative permeability to brine remained unchanged with increasing temperature.

Polikar et al.⁷ used the steady state technique to measure the relative permeability of Athabasca bitumen using unconsolidated Ottawa sand as the porous medium. Their results showed no significant effect of temperature on relative permeability and saturation data up to 392°F.

Maini and Okazawa⁸ analyzed unsteady state experiments conducted on an unconsolidated silica sand using Bodo stock tank oil and de-ionized water. They history matched production and differential pressure data to obtain relative permeability curves. They concluded that the most reliable data they obtained were at residual oil saturation after three pore volumes injected. They observed increasing relative permeability to water with increasing temperature.

Unlike the previous researchers, Watson and Ertekin⁹ studied the effect of temperature gradient on relative permeability measurements using the steady state technique. They found that different temperature gradients caused differences in both irreducible water saturations and residual

oil saturations. These data suggested that the fired Berea cores become increasingly water-wet, and unfired cores become increasingly oil wet, during imbibition and drainage. Moreover, they concluded that relative permeabilities to both oil and water decreased with increasing temperature. They also observed that the absolute water permeabilities of fired Berea sandstone cores were unaffected by temperature.

Polikar et al.¹⁰ conducted both steady state and unsteady-state relative permeability experiments to find out the effect of temperature on relative permeability. Using Athabasca bitumen and deionized water, they used the JBN¹¹ technique to analyze their results. They observed no significant temperature effects on relative permeability of either phase, nor in the residual oil and irreducible water saturations. They attributed the small changes seen to core heterogeneity. Moreover, they reported that steady state and unsteady-state techniques resulted in similar relative permeability curves.

More recently, Muqem et al.¹² conducted steady state two phase and three phase relative permeability experiments at 167°F and 257°F (75°C and 125°C) using an unconsolidated silica sand with refined oil, brine and nitrogen gas. The measured relative permeabilities showed no significant temperature effect.

Table 1 - A Summary Of The Recent Experimental Investigations Of Temperature Effects On Relative Permeability.

Ref.	Technique	Fluids	Core	Temperature Range	Saturation Measurement	End-Point Saturation	Relative Permeability
1	Unsteady State (Welge)	Refined Oils (No 5 & 15)	Berea (1382°F)	room-500°F	graduated cylinders	$S_{or} \downarrow T \uparrow$	k_{rw} shifted to lower S_w
3	Unsteady State (JBN)	Refined Oil	Unconsolidated Ottawa (170-200)	room-250°F	photo cell + frequency counter	independent	independent
4	Unsteady State (JBN)	Refined Oil (Blandol)	Unconsolidated Ottawa (100-200) + Berea (932°F)	room-300°F	separator + graduated cylinders	independent	independent
5	Steady State	Altered-unaltered-deasphalted tar	Frozen Peace River Cores	385°F	electrical resistance	not presented	k_{rw} shifted to lower S_w
6	Unsteady State (History Matching)	Stock Tank Oil	Frozen Core from a Hvy Oil Res	room-522°F	separator + fractional collector	$S_{wirr} \uparrow T \uparrow$ $S_{or} \downarrow T \uparrow$	$k_{ro} \downarrow T \uparrow$ $k_{rw} \rightarrow T \uparrow$
7	Steady State	Athabasca Bitumen	Unconsolidated Ottawa (~200)	room-482°F	material balance	independent	independent
8	Unsteady State (History Matching)	Bodo Stock Tank Oil+3.5% ether	Unconsolidated Silica Sand (140-170)	room-392°F	separator + fractional collector	not presented	$k_{rw} \uparrow T \uparrow$ $k_{ro} \downarrow T \uparrow$
10	Steady State	Refined Oil (Soltrol 170)	Berea (1832°F) + unfired	room-300°F	material balance	$S_{wirr} \uparrow T \uparrow$	$k_{ro}, k_{rw} \downarrow T \uparrow$
11	Steady & Unsteady State (JBN)	Athabasca Bitumen	Unconsolidated Silica Sand (170-230)	212°F-482°F	material balance	independent	independent
12	Steady State	Refined Oil+N ₂	Unconsolidated Silica Sand (140-200)	167°F&257°F	separator	not presented	independent
13	Unsteady State (JBN)	SJV Oil-GulfLight + SJV	Unconsolidated preserved cores + Berea (925°F)	room-300°F	graduated cylinders	independent	k_{rw} shifted to lower S_w

Kumar and Inouye¹³ used the JBN technique to analyze unsteady state relative permeability experiments conducted at differing temperatures. They tried to obtain low temperature analogues to high temperature relative permeability data. They concluded that relative permeability data obtained at ambient conditions can be used for high temperatures as long as the viscosity ratios and the wettabilities were similar. They also reported that residual oil saturation and irreducible water saturation are independent of temperature and are a function of viscosity ratio.

The results obtained from the aforementioned studies are summarized in Table 1. It can be observed that there is no conclusive agreement on the effect of temperature on oil/water relative permeabilities. Although some studies show an effect of temperature, others disagree with that by attributing changes observed to viscous instabilities and to other experimental errors. Basically, the reasons for divergence of experimental data may be summarized as follows:

- 1.Errors in saturation measurements.
- 2.Errors caused by neglect of capillary pressure end effects.
- 3.Wettability variations with differing oils, and brines.
- 4.Assumptions made to relate experimental procedures and/or calculations.
- 5.Inadequacy of mathematical models used to represent multiphase flow conditions.

It is clear that, unless the problems stated above are overcome, a better understanding of the effects of temperature on relative permeabilities will not be achieved. In the first part of this study, the accuracy of the JBN technique for the determination of heavy oil/water relative permeabilities and its implication on the effect of temperature on relative permeabilities is discussed by giving numerical as well as experimental examples. In the second part of the study, a new experimental technique is presented to tackle the effect of the temperature problem in a qualitative manner. In this technique, an ambient temperature unsteady-state relative permeability run is conducted in the first stage, and following that, the temperature is increased twice (i.e. 122°F and 150°F). Two phase saturation profiles along the sand pack are measured using a CT scanner. A commercial black oil simulator, coupled with a global optimization code is then used to estimate two phase relative permeabilities. It has been observed that a single set of relative permeability curves can represent both the ambient and high temperature parts of the experiment. This suggests that relative permeability is not a function of temperature at least over the temperature range studied on the system tested.

Application of the Unsteady-State Relative Permeability Estimation Technique to Heavy Oil Systems

Most of the parameters used in thermal simulations are obtained from laboratory measurements. For instance, water/oil relative permeability curves are determined from corefloods involving formation brine and oil. There is,

however, uncertainty associated with this type of data since the high viscosity of heavy oils makes the measurement difficult, and the unspecified heterogeneity of the core material complicates flow paths and saturation distributions.. The principal objective of this section is to examine whether or not a Buckley-Leverett type displacement model (i.e. JBN technique) could be used to describe the behavior of such unstable displacements. Another purpose is to find out what happens if such a technique is used to estimate oil/water relative permeabilities. To achieve this goal, several hypothetical, unstable and immiscible displacements corresponding to experiments conducted at different temperatures were simulated using a commercial simulator. The level of instability in these experiments was quantified by a dimensionless stability number, N_{is} proposed by Peters and Flock¹⁴.

$$N_{is} = \frac{(M - 1)(V - V_c)\mu_w d^2}{N_w k_{wor} \sigma}$$

According to Peters and Flock, a core-flood will be stable (piston-like displacement, and no viscous fingering) if the dimensionless stability number is less than 13.56. The properties of the porous medium and the fluids used in the hypothetical experiments as well as the stability numbers for no initial water saturation are presented in Tables 2 and 3. It can be observed that even at high temperatures, these core-floods are highly unstable showing significant viscous fingering. However, capillary end effects were kept at a minimum as the scaling coefficients proposed by Rapoport and Leas¹⁵ were higher than the critical value (i.e. $L u \mu_w$, ft-ft/s-cp is greater than 3.5) for all experiments.

A commercial black oil simulator, ECLIPSE¹⁶, was used to simulate the experimental conditions given in Table 2. Three dimensional, radial geometry was used to simulate the experiments. To account for viscous fingers, center grids had smaller diameter increments compared to the outer ones. The first and last grid planes in the z direction had higher transmissibility compared to other grids to reproduce the experimental inlet and outlet lines together with spider-web shaped end plates. Production and pressure differential history across the core were collected. Unsteady-state relative permeabilities were then computed by applying the JBN technique¹¹ using the equations given below:

$$f_o = \frac{1}{1 + \frac{k_{rw}\mu_o}{k_{ro}\mu_w}} = \frac{dN_p}{dW_i}$$

$$S_{w2} = S_{wi} + N_p - W_i f_o$$

Table 2 - Properties Of Porous Medium And Fluids Used For Hypothetical Experiments.

Porous Medium	Fluid Properties
	Oil
Permeability = 5.3 Darcies	Viscosity @70 F=220 cp
Porosity = 0.394	Density @70 F=0.878 gm/cc
Length = 17.8 cm	Water
Diameter = 2.54 cm	Viscosity at 70 F = 0.97 cp
	Density at 70 F = 1.00 gm/cc

Table 3 - Stability Numbers For Different Experiments With No Initial Water Saturation.

Experiment	1	2	3
Temperature, °F	70	122	150
Stability Number	40167	6650	3174
Rapoport & Leas Number	14.21	7.82	5.97

Figure 1 gives the comparison of the saturation distributions at 0.193PV of water injected obtained from simulations along the hypothetical core (L=17.8 cm) at three different temperatures (70, 122 and 150°F) and at four different locations with a zero initial water saturation. It can be observed that the front is not stable, and moreover, a fingering pattern is visible in the center of the core for all cases. Note that the same set of relative permeability curves was used to generate these data.

Fig.1- Comparison of saturation distributions at different temperatures: top 70°F, middle 122°F, bottom 150°F, at 0.193 PV injected.

The relative permeability curves obtained using JBN analysis for no initial water saturation are compared to the input curves in Fig 2. Notice that although the same relative permeability curves were used to simulate core-floods at different temperatures, different apparent relative permeability curves were obtained using the JBN technique. Interestingly, the resulting relative permeability curves show a false temperature dependence. That is to say, there is a considerable increase in the calculated relative permeability to oil and water at a constant saturation as temperature increases. The end point oil saturation also decreases and the corresponding water relative permeability increases as temperature increases as observed in Table 4. This false temperature dependence agrees with the general trend that oil and water relative permeability curves shift to the right as reported in studies by Sinnokrot¹⁷, Weinbrandt et al.¹⁸, and others^{5,6}. Moreover, these false findings agree with the concept of decreased residual oil saturation with a temperature increase.

Fig. 2- Comparison of JBN derived relative permeabilities to input data for differing temperatures with no initial water saturation.

Table 4 - End Point Oil Saturation And Relative Permeability To Water At Different Temperatures Without Initial Water Saturation.

Temperature (°F)	S _{or}	K _{rw}
70	0.292	0.0538
122	0.255	0.0764
150	0.212	0.161
input	0.200	0.172

Three similar hypothetical experiments were simulated to see the effect of initial water saturation. The stability numbers were much smaller compared to the no initial water saturation case, but still indicated unstable displacements as observed in Table 5. In these cases, the calculated relative permeability curves showed similar false temperature effects, and also failed to find the input relative permeability curves as shown in Fig 3; the differences were somewhat smaller for oil relative permeabilities but larger for water relative permeabilities. Similar to the no initial water saturation cases, the calculated residual oil saturation decreased as temperature increased as observed in Table 6.

Table 5 - Stability Numbers For Different Experiments With Initial Water Saturation.

Experiment	1	2	3
Temperature, °F	70	122	150
Stability Number	582.3	96.4	46
Rapoport & Leas Number	14.21	7.82	5.97

Table 6 - End Point Oil Saturation And Relative Permeability To Water At Different Temperatures With Initial Water Saturation.

Temperature (°F)	S _{or}	K _{rw}
70	0.281	0.0684
122	0.244	0.0995
150	0.204	0.2119
input	0.200	0.172

Another common technique used to estimate relative permeability curves is one dimensional history matching. To estimate relative permeability curves with the history

matching or parameter estimation approach, an objective function as shown below, is constructed as a weighted sum of squared differences between the measured data and calculations from a one-dimensional mathematical model of the experiment.

$$J = \left\{ \sum_{i=1}^N W_{pi} (\Delta P_i^{obs} - \Delta P_i^{cal})^2 + \sum_{i=1}^N W_{qi} (Q_{ti}^{obs} - Q_{ti}^{cal})^2 + \sum_{k=1}^M \sum_{i=1}^N W_{si} (S_{i,k}^{obs} - S_{i,k}^{cal})^2 \right\}$$

Fig. 3- Comparison of JBN derived relative permeabilities to input data for different temperatures for 40% initial water saturation.

For a typical displacement experiment, the measured data might consist of a set of pressure drops across the core, the recovery histories of the displaced phase, and if available the internal saturation profiles¹⁹.

By using a one dimensional radial grid definition, the same hypothetical, aforementioned data for no initial water saturation was used to generate relative permeability curves by matching the pressure and production response for differing temperatures. Figure 4 compares the estimated relative permeability curves with the input data. As the temperature

increased, oil relative permeability increased and water relative permeability decreased at the same saturation. Therefore, it can be concluded that one dimensional history matching is also showing a false temperature effect.

Fig. 4- Comparison of history matched relative permeabilities to input data for different temperatures for no initial water saturation.

The above findings clearly show that JBN, and like techniques can lead to erroneous results. It seems that the viscous instability effects get smaller if the core is initially at residual water saturation. Another important finding is that because of viscous instabilities, the JBN technique also leads to a false temperature effect regardless of initial water saturation.

A possible, alternative solution to the problem of finding the effect of temperature on heavy-oil/water relative permeabilities is to introduce a new experimental technique that consists of three different parts. In the first part, a conventional oil/water relative permeability experiment is conducted until no more oil is produced. Following that the temperature is increased twice, and at each of these higher temperatures injection is continued until residual oil saturation. The pressure, saturation and production data collected during the ambient part of the experiment is used to calculate heavy oil/water relative permeability curves using the aforementioned history matching procedure. Then, using these relative permeability curves, the rest of the experiment (i.e. higher temperature parts) is simulated and compared to

experimental data. There are two possible outcomes of this procedure:

1. Both parts of the data (i.e. cold and hot) are matched by using only one set of relative permeability curves, which would indicate that the flow functions do not depend on temperature.
2. The algorithm fails to represent both parts of the experimental data (i.e. cold and hot) using a single set of relative permeability curves, which would indicate that the flow functions depend on temperature.

This technique provides a qualitative result to the effect of temperature on oil/water relative permeabilities. Yet, at the same time the higher temperature parts of the experiment provide some quantitative information on the residual oil saturation and end point relative permeabilities.

Results and Discussion

Two different sets of experiments were conducted using the proposed experimental procedure and analysis. In the first set of experiments, clean Ottawa sand and a high viscosity mineral oil were used. Table 7 gives the properties of the porous medium and the fluids used in these experiments. Since high temperature relative permeability experiments require core holders and sleeves that can withstand high temperature and pressure, a core holder was designed that consists of an aluminum outer shell with internal adjustable centralizers which hold and centralize an aluminum core sleeve (1 3/4 in ID, and 16 in long) as shown in Fig 5. The core sleeve can operate safely up to 482°F and 1000 psi. With its current design the core holder is suitable for CT monitored high temperature experiments using a heating circulator for the annulus.

Table 7 - Properties Of The Sand And Fluids Used In The First Set Of Experiments.

Porous Medium	Fluid Properties
Ottawa Sand (mesh 100-200)	Mineral Oil (Kaydol)
Permeability = 5.3 Darcies	Viscosity @ 70°F=220 cp
Porosity = 0.31%	Density @ 70°F=0.878 gm/cc
Length = 40.64 cm	Water (8% weight KBr)
Diameter = 4.445 cm	Viscosity at 70°F=1.075 cp
	Density at 70°F=1.05 gm/cc

Fig. 5- Schematic diagram of the core holder used in the experiments.

Following the aforementioned procedure a displacement experiment was conducted with initial water saturation. The first part of the experiment was performed at room temperature and then the temperature was increased to 122°F and, then to 150°F. CT scans were taken to monitor saturation changes at differing times. Figure 6 displays 3D reconstructions of water saturation distributions at different stages of the experiment at differing temperatures. The left column of this figure gives images taken at 0.606, 1.01, 6.06, and 8.33 pore volumes of water injected from the ambient temperature part of the experiment. It can be observed that, because of the high viscosity ratio, water fingers were formed and the flood front was not stable. The effect of temperature is presented at the right column of the same figure. As can be seen, average water saturation does not change much. Figure 7 shows average oil saturation profiles obtained by averaging CT derived saturations plotted at differing times. It can be noticed that temperature increase has a minimal affect on the end point saturations. We can say that residual oil saturation is independent of temperature. Following that, a nonlinear optimization code¹⁹ coupled with a commercial simulator¹⁶ in 2-D radial mode was used to history match the ambient part of the experimental data in a least squares manner and the relative permeability curves presented in Fig 8 were obtained. Figure 9 gives pressure differential and production data fits to all parts of the experiment using these curves. It can be observed that the experiment can be modeled using the relative permeability curves obtained from the cold part. This indicates that heavy oil/water relative permeabilities do not depend on temperature for the system in consideration at this temperature range.

Variations in capillary pressure and relative permeability with temperature may be expected for a rock/fluid system whose wettability is temperature dependent. Sinnokrot et al.¹⁷ discussed that relative permeabilities and capillary pressures may exhibit temperature dependence for systems whose fluids contain large concentrations of wettability altering compounds like asphaltenes. To check such a possibility another

experiment with initial water saturation was conducted using South Belridge sand and oil. This field was selected since it is a major target for thermal recovery operations like steam flooding and in-situ combustion²⁰. Table 8 gives the properties of the porous medium and the fluids used in the experiment. The sand was packed without cleaning to preserve the original wettability. The initial temperature of the experiment was kept at the reservoir temperature of 95°F (35°C)²⁰ to have a reasonable starting viscosity ratio as observed in Fig 10. The rest of the experimental procedure was similar to the previous one such that the temperature was increased to 122°F and, then to 150°F after the initial part. However, because of experimental difficulties water injection at higher temperatures was not continued down to residual oil saturation.

Fig. 6- 3-D reconstruction of water saturation distribution at different stages of the experiment 1 (Left 70°F, right top 70°F, middle 122°F and bottom 150°F).

Fig. 7- Oil saturation distribution at different stages of Experiment 1.

Fig. 8- Relative permeabilities obtained by automated history matching technique for Experiment 1.

Fig. 9- Pressure and production data fit to Experiment 1.

Fig. 10- South Belridge oil/water viscosity ratio temperature chart for Experiment 2.

Figure 11 gives re-constructions of the water saturation profiles obtained using the CT scan images. A higher level of instability is apparent compared to the previous experiment because of the high viscosity ratio. Compared to the previous experiment the effect of temperature is much greater as can also be observed from the average saturation profiles plotted in Fig 12. This was due to the short duration of each hot part of the experiment. However, a nice match was obtained to experimental pressure data shown in Fig 13 using the relative permeability curves presented in Fig 14 which is based on the cold temperature match. Therefore, it was concluded that relative permeability is not a function of temperature at least for the systems presented and for the temperature range studied. It also appears that wettability alteration is not an important factor at this temperature level. Further study at higher temperatures is required.

Fig. 11- Planar reconstruction of water saturation distribution at different stages of the Experiment 2 (Left 70°F, right top 70°F, middle two 122°F and bottom 150°F).

Fig. 12- Oil saturation distribution at different stages of the Experiment 2.

Figure 11 gives re-constructions of the water saturation profiles obtained using the CT scan images. A higher level of

instability is apparent compared to the previous experiment because of the high viscosity ratio. Compared to the previous experiment the effect of temperature is much greater as can also be observed from the average saturation profiles plotted in Fig 12. This was due to the short duration of each hot part of the experiment. However, a nice match was obtained to experimental pressure data shown in Fig 13 using the relative permeability curves presented in Fig 14 which is based on the cold temperature match. Therefore, it was concluded that relative permeability is not a function of temperature at least for the systems presented and for the temperature range studied. It also appears that wettability alteration is not an important factor at this temperature level. Further study at higher temperatures is required.

Fig. 13- Pressure differential data fit to Experiment 2.

Conclusions

Based on hypothetical heavy-oil/water numerical simulations it was observed that the use of the JBN technique for the estimation of heavy-oil/water relative permeabilities leads to erroneous results and a false temperature effect. A new experimental technique is presented to tackle the problem of the effect of temperature on oil/water relative permeability. It was found that temperature does not have any effect on the relative permeabilities for the systems and temperatures used in this study. These findings show that previously observed apparent temperature dependence of endpoint saturations and relative permeabilities is basically caused by a decrease in the oil/water viscosity ratio with temperature.

Fig. 14- Relative permeabilities obtained by automated history matching technique for Experiment 2.

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Nomenclature

d = Core diameter, ft.
 f = Fractional flow, fraction
 g = Gravitational acceleration
 k = Relative permeability, fraction
 u = Darcy velocity, ft/sec
 I_r = Relative injectivity, fraction

J = Objective function
 M = Mobility ratio or number of grids
 N = Number of data points
 N_p = Production, PV
 N_w = Wettability number, dimensionless
 Q = Volume of displaced phase recovered, PV
 Δp = Pressure differential across the core, psi
 S = Saturation, fraction
 V_c = Characteristic velocity, ft/sec
 W = Injection, PV or weighing factor
 μ = Viscosity, cp
 ρ = Density, lb/ft³
 σ = Oil-water interfacial tension, dyne/cm

Subscripts and superscripts

cal = Calculated
 i, j, k = Indices
 o = Oil
 obs = Observed
 or = Residual oil
 p = Produced
 w = Water
 $w2$ = End point

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1.2 MULTIPHASE-FLOW PROPERTIES OF FRACTURED POROUS MEDIA

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Abstract

The fluid transfer parameters between matrix and fracture are not well known. Consequently, simulation of fractured reservoirs uses, in general, very crude and unproved hypothesis such as zero capillary pressure in the fracture and/or relative permeability functions that are linear with saturation. In order to improve the understanding of flow in fractured media, an experimental study was conducted and numerical simulation used to interpret experimental results.

A laboratory flow apparatus was built to obtain data on water-air imbibition and oil-water drainage displacements in fractured sandstone systems. During the experiments, porosity and saturation were measured along the core utilizing a Computerized Tomography (CT) scanner. Saturation images were reconstructed in 3-D to observe how matrix-fracture interaction occurred. Differences in fluid saturations and relative permeabilities caused by changes of fracture width have also been analyzed.

In the case of water-air imbibition, fracture systems with narrower fracture apertures showed more stable fronts and slower water breakthrough than the wide fracture systems. However, the final water saturation was higher in wide fracture systems, thus showing that capillary pressure in the narrow fracture has more effect on fluid distribution in the matrix. During oil-water drainage, oil saturations were higher in the blocks near the thin fracture, again showing the effect of fracture capillary pressure. Oil fingering was observed in the wide fracture.

Fine-grid simulations of the experiments using a commercial reservoir simulator were performed. Relative permeability and capillary pressure curves were obtained by history matching the experiments. The results showed that the assumption of fracture relative permeability equal to phase

saturation is incorrect. We found that both capillary and viscous forces affect the process. The matrix capillary pressure obtained by matching an experiment showed lower values than reported in the literature.

Introduction

Flow equations describing fractured porous media are usually written assuming that reservoir behavior is dominated by the transfer of fluid from the matrix to high conductivity fractures, that are often entirely responsible for flow between blocks and flow to wells.

Conventional representation of these equations assumes the knowledge of both rock and fluid properties, capillary pressure, and relative permeabilities. Any of these parameters can be obtained for the matrix by laboratory work, but not for the fractures. To obtain more data on properties such as fracture capillary pressure, fracture relative permeabilities and/or saturation distributions, an experimental and numerical study was conducted

In general, several authors (Kazemi and Merrill (1979), Beckner (1990), Gilman et al. (1994)) have assumed that fracture capillary pressure is negligible. Others have shown experimentally that capillary continuity becomes important when gravity provides a driving force (Horie et al. (1988), Firoozabadi and Hauge (1990), Labastie (1990), Firoozabadi and Markeset (1992). Kazemi (1990) states his belief that capillary continuity is prevalent in the vertical direction and has suggested that, to reduce the number of equations to solve, fractured reservoir simulators should use the dual-permeability formulation for the z direction and the dual-porosity formulation for the x and y directions.

For this work, detailed measurements of pressure, rate, and saturation distribution were performed, and attempts to measure phase distribution inside the fracture were also made using a (CT) scanner. This research resulted in a much better understanding of the physical processes that occur when two or three phases flow in a fractured system, compared to previous reported studies (Guzman and Aziz (1993), Hughes (1995)). Our first step was constructing an experimental apparatus capable of reproducing the results obtained by that by Hughes(1995). Thus, we conducted similar experiments using two of the core holders developed by him in order to verify that this new experiment gives consistent results. Once the apparatus was tested, we conducted multiphase flow

experiments using an 8% NaBr brine solution as the wetting phase, and decane as the nonwetting phase. Three stages in each experiment were completed: study of water imbibing into a dry core, decane displacing water in a water-saturated core, and water displacing oil. Differences in fluid saturations and relative permeabilities caused by changes in fracture width were also analyzed.

Simulations of the experiments using a commercial reservoir simulator were performed. Fracture relative permeability and capillary pressure curves were obtained by history matching the experiments. A sensitivity analysis of parameters such as fracture relative permeability, capillary pressure in the fracture, and fracture width was also completed.

Experiment Design

The experiments presented in this work were made using an apparatus similar to the one designed, constructed and used by Hughes (1995). The following subsections give a description of the material and methods used to build the experimental set-up.

Cores. For this work, we used two of the three rectangular blocks of Boise sandstone cores constructed and used by Hughes(1995). Both core holders consist of two 2 15/16 x 1 1/2 x 11 inch blocks. The first core holder is system of the two blocks together, i.e. with no spacer in between the blocks. The second one is a similar configuration, but has a 1-mm thick spacer fastened in place with Epoxy 907 to provide separation between the blocks to simulate a fracture.

Fig. 1- The Core Holder.

Core holders. Due to the rectangular shape and the desire to measure in-situ saturations through the use of the CT scanner, conventional core holders could not be used. A core holder similar to the one designed by Guzman and Aziz (1993), and later developed by Hughes (1995) was used. An epoxy resin surrounds the core, as well as plexiglas end plates and a piece of 3/8 inch Viton that acts as a gasket between the core and the plexiglas end plates as shown in Fig. 1.

Positioning System. The positioning system consists of a moving table with 0.01 mm accuracy that is positioned electronically by means of a control panel. Basically, any flat surface can be attached to this table using screws. This system was one of the major improvements to previous designs (Guzman, 1993; Hughes, 1995). In this way, we can obtain cross-sectional images of the core at the same location at any desired time of the experiment. This could not be performed in other studies because the standard patient table of a CT Scanner of this type lacks comparable accuracy.

Flow System. The flow system (Fig. 2) allows fluids to be directed to any port or combination of ports in the experiment such that: 1) It can be directed to calibrate the pressure transducers, 2) It can be used to inject from one end and to produce from the opposite end, 3) It can be used to inject into one or more of the ports on the top and bottom of the core holder, or 4) To bypass the core holder completely. It also consists of two pumps that inject 0.01 to 9.99 cm³/min in 0.01 increments. Plumbing downstream of the pumps allows mixing of the fluids being discharged by each pump. This setup allows injection pressure to be monitored with a test gauge and recirculation to measure pump output rates.

Fig. 2- Flow System (After Hughes, 1995.)

Production Measurement System. The production measurement system is an adaptation of a design given by Ameri and Wong (1985). Fig. 3 shows this system. The electronic balances are connected to the serial communication ports of a personal computer (PC).

Pressure Measurement System. Pressure drop along the core can be measured between any of the six different positions on the core labeled as T-x. In Fig. 2, one can see that two of the transducers are connected to the inlet and outlet of the core holder. Each transducer is connected to a carrier demodulator which produces a DC signal in the range from -10 to +10 V that is proportional to the pressure drop. The output signals were recorded on a chart recorder and also captured in digital form.

Fig. 3- Production system (after Hughes, 1995.)

CT Scanner. A CT Scanner can be used to calculate porosity, saturation and in some cases, concentration distribution and to track advancing fronts. It can also be used to measure fracture aperture. We use a Picker 1200SX Dual Energy CT scanner.

Fluids. CT numbers are proportional to their densities. A selection of fluids and dopants were employed in order to obtain the best contrast in CT numbers and, hence, most accurate in terms of phase saturation. Decane was used to represent the oil phase, and 8% NaBr, by weight, water solution was used as the water phase.

Experimental Procedure

The first core used had a fracture with no spacer in between the blocks. The total travel distance of the positioning system (accuracy ± 0.01 mm) was then 25.5 cm, resulting in 13 slices including two located at the pressure ports (fourth and eleventh, for the first experiment; and fourth and tenth, for the second experiment.) The second experiment was conducted using a core holder that contained a similar system with two Boise sandstone blocks with a 1 mm thick fracture in between.

Fig. 4 shows the CT scan locations.

Fig. 4- CT Scan locations.

The first step in the experimental procedure was to scan the dry core. This step was very important to verify that the apparatus and the set-up were working properly. For instance, there were some problems with the core holder in a first trial

for the dry scan. We could not keep its position horizontal, because it was quite heavy and held only by one of the sides. Two C clamps were used to strengthen the attachment to the positioning system and, thus, to keep the core holder in the same horizontal position during the experiment. After these improvements, the dry scans were conducted successfully.

The second step was to evaluate how water imbibed into an unsaturated core. This part of the experiment gave us results that can be compared with those obtained by Hughes (1995), and eventually used to evaluate how well the new experiment worked. Starting with a flow rate of $2 \text{ cm}^3/\text{min}$, CT images of the core were taken every 5 minutes for the first 20 minutes of water injection (0.15 PV); after this, images were taken at 30 min (0.22 PV), 45 min (0.33 PV), 1 hr. (0.45 PV), 1hr 30 min (0.67 PV), 2 hr (0.89 PV), 3 hr (1.34 PV), and 4 hours (1.78 PV) after water injection started. Next, the top and bottom ports were opened since we wanted to fill up the core to the maximum water saturation ($S_w=1$.) Common to all time steps, we tried to take images up to one location ahead of the position of the water front.

After 5 hours of water injection (2.25 PV), the flow rate was changed to $0.5 \text{ cm}^3/\text{min}$ in order to inject water for 14 more hours to reach steady state. After 19 hours (8.55 PV), reference 100% water saturated images were taken. Then, at 19 hours and 45 min (8.9 PV) the water rate was changed back to $2 \text{ cm}^3/\text{min}$, and the top and bottom ports were closed. At 21 hours (9.45 PV) a final set of slices was taken to find any possible change in CT numbers. Negligible changes in CT numbers were observed. At 23 hours of water injection, the third step of the experiment started. Table 1 summarizes the flow rates used.

The third step of the experiment involved injecting decane at $2 \text{ cm}^3/\text{min}$. The scanning frequency was intended to be the same as the water injection process. Thus, images of the core were taken every five minutes until the first 20 minutes of oil injection (0.15 PV). After that, more images were taken at 30 min (0.22 PV), 45 min (0.33 PV), 1 hr. (0.45 PV), 1hr 30 min (0.67 PV), 2 hr 30 min (1.13 PV), and 3 hr 45 min. (1.69 PV) after oil injection started. Then, the top and bottom ports were opened at 4 hours (1.79 PV) since we wanted to fill the core up to the maximum possible point. At 4 hr 30 min of oil injection, one more set of images was taken. After 5 hours (2.25 PV) of oil injection, the flow rate was changed to $1 \text{ cm}^3/\text{min}$ for 16 hours (7.2 PV) to reach higher possible oil saturations ($1-S_{wc}$). After 21 hours of oil injection (9.45 PV), a new set of slices was taken, and the ports were closed. Then, at 21 hours and 39 min the oil injection was stopped, and water injection at the maximum pump rate ($9.99 \text{ cm}^3/\text{min}$) was started.

The fourth step of the experiment was the water displacing oil stage. Injecting $9.99 \text{ cm}^3/\text{min}$ of water, images were taken at 5 min (0.04 PV), 15 min (0.13 PV), and 30 min (0.23 PV). At 1hr 10 min (0.53 PV), water injection was stopped, the inlet and outlet lateral ports were also closed, leaving all of the core holder's ports closed. One more set of images was taken after two days of closing the core holder to observe if capillary

equilibrium had been reached. Table 1 summarizes the flow rates used and the timing of each.

The second experiment was performed using the core holder that contained a system with two Boise sandstone blocks with a 1 mm thick fracture in between. The same experimental procedure was followed. Essentially, we wanted to obtain CT images, and then saturation values for equivalent times, so we could compare the differences between the two systems. The rates and timing for the case of two blocks with a wider fracture are shown in Table 2.

Data Analysis

Using the CT numbers obtained from the experiments, porosity and saturation distributions along the core were determined. The most common way to calculate porosity from CT Scanner images is by using the following expression (Withjack, 1988):

$$\phi = \frac{CT_{cw} - CT_{cd}}{CT_w - CT_a} \dots\dots\dots(1)$$

where CT_{cw} is the CT number for a 100% water saturated core at a matrix location, CT_{cd} is the CT number for a dry core at a matrix location, CT_w is the CT number for water, and CT_a is the CT number for air. The CT number for water (8%NaBr solution, for this work) is around 360, while the CT number for air is -1000.

The water and oil saturations were also calculated from the CT images. The following equations show how to evaluate water saturation for the water displacing air case, and for the oil displacing water or water displacing oil case.

$$S_w = \frac{CT_{aw} - CT_{cd}}{CT_{cw} - CT_{cd}} \dots\dots\dots(2)$$

where CT_{aw} is the CT number for water and air saturated core at a matrix location. Similarly, for the case of oil-water systems:

$$S_w = \frac{CT_{ow} - CT_{cw}}{\phi(CT_o - CT_w)} \dots\dots\dots(3)$$

where CT_{ow} is the CT number for a water and oil saturated core at a matrix location, and CT_o is the CT number for oil. The CT number for oil (decane, for this work) is around -272.

Experimental Results

First, we computed the values of porosity using Eq. 1 at each location indicated in Fig. 4. The porosity images for both systems are shown in Fig. 5, where the values below each square correspond to the mean and the standard deviation, respectively, of the porosity values obtained from CT numbers. This step was essential for both experiments because from these images and their corresponding values we

differentiated the regions with higher or lower porosity. The range of values is shown in the color bar in the right hand side in Fig. 4. There, we can also see that the system with the thin fracture has a little higher porosity than the system with the wide fracture, especially in the top block, where values up to 0.13 were observed. We found that the average value for porosity calculated from the scans was 14.5% which matches with the average value 14.35% reported by Hughes (1995); however, this value differs from the average porosity measurements of 25.4% obtained by Guzman and Aziz (1993) and 29.3% obtained by Sumnu (1995). All of these previous studies have also shown areas in the rocks, which have lower permeability.

Water-Air Imbibition. Following the aforementioned procedure, and the flow rates, times and water injection conditions listed in Table 1, several sets of images like those shown in Fig. 6 corresponding to 1.5 hours of water injection (0.67 PV) were obtained. Similar to the porosity images, each square corresponds to water saturation values obtained at one location. The pair of values presented below each square corresponds to the mean and standard deviation of the top, bottom, and both blocks, respectively. The gray scale on the right hand side shows the range of water saturation values. Different profiles, corresponding to different pore volumes of water injected were plotted. The profiles are more or less stable and they appear to be quasi one dimensional. The same procedure was followed for the wide fracture system and CT saturations were obtained. Fig. 7 gives water the saturation distribution at 0.67 PV of water injected. Different profiles, corresponding to different pore volumes of water injected were also plotted. The profiles are less stable than those for the thin fracture system as observed from the 8th through 12th slices. We observed similar profiles for all times.

Oil-Water Drainage. Similar images were obtained for the second stage of the experiments (oil injection). Fig. 8 shows a set of images corresponding to 2.5 hours (1.13 PV) of oil injection. Here, the pairs of values below each square indicate the mean and the standard deviation of oil saturation, respectively. Shades were also assigned to the oil saturation values. Thus, the gray scale on the right hand side shows the range of values and their corresponding lightness or darkness. In all the images shown in Fig. 8 darker shades indicate lower oil saturation. For instance, black means zero oil saturation. This stage was finished after 4 hours of oil injection when we opened the lateral ports of the core holder to reach the maximum possible oil saturation. The corresponding images for the wide fracture system are shown in Fig. 9.

Water-Oil Displacement. Images for the third stage of the experiment (water displacing oil) were also obtained. Due to problems with one of the pumps, we could not analyze properly the rest of the data for the case of thin fracture system. Images for the third stage of the second experiment (water displacing oil in the wide fracture system) were also obtained. The most interesting sets are shown in Fig. 10 and

Fig. 11 corresponding to 16 hours (7.2 PV) and 17 hours (7.7 PV) of water injection, respectively. One can see that the displacement of oil is completed almost perfectly.

Analysis of Experimental Results

Due to the fact that the experiments ended differently, we were able to compare the results up to the second stage of both experiments. This section shows the comparison graphically.

Three-dimensional (3-D) reconstructions of the sets of water saturation images were calculated. All these images were obtained using a color map for values above 50% of water or oil saturation, respectively. These 3-D images show more clearly how the fluids actually flow through the fractured porous media. This work was intended to show the differences between the flow patterns for different fracture thicknesses. The reconstructed images are shown in **Fig. 12**. Each image corresponds to a specific time. Thus, the first image on the left-hand side of **Fig. 12** is compared to the first one of the right hand side of the same **Fig. 12**. The second image on the left hand side was computed for the same time as the second image on the right hand side, and so on. There, we can see that the water front for the wider fracture system moves at almost at the same speed as the water front for the narrow fracture system does. For instance, one can see that after 1.5 hours of water injection, the water front of the narrow fracture system seems to be ahead of the front for the wide fracture system; however, after two hours of water injection water has filled up both systems.

Similarly, the 3-D oil injection history was also reconstructed. Since the oil saturations were low for both cases, it is a bit harder to see the differences. It is possible to see that the thinner fracture system has higher oil saturation close to the fracture. This is shown in **Fig.13**. One can see that the narrow fracture system has lighter colors close to the injection surface. We can also see that for the case of the wide fracture system, the oil does not penetrate. Furthermore, oil flows almost completely through the fracture. This is shown in all the images on the right hand side of **Fig. 13** by the almost white horizontal line that corresponds to the fracture. Remember that for the case of oil injection stages, darker means lower oil saturation.

Numerical Simulation Results

Numerical simulations of the experiments using a commercial reservoir simulator were conducted to study fracture relative permeability and matrix/fracture interaction, to match previous experimental results, and to provide experimental-numerical based suggestions how to simulate multiphase flow in fractured porous media.

A cubic Cartesian gridding proportional to the cores was designed in such a way that we had three different “boxes”. Two of them simulate the top and bottom blocks with matrix rock properties; i.e., matrix capillary pressure curves, matrix relative permeability curves, matrix absolute permeability, porosity, and the other set of blocks simulate the horizontal fracture with fracture properties; i.e., large absolute permeability, fracture capillary pressure curves, and fracture

relative permeability curves. It was assumed that relative permeability and capillary pressure in the matrix are constant.

For the water-air cases, we used the curves presented by Persoff et. al. (1991) for matrix relative permeability and the ones measured by Sanyal (1972) for matrix capillary pressure. For the oil-water cases, we used the matrix capillary pressure curves presented by Sanyal (1972). Unfortunately, no relative permeability curves for oil-water system in Boise sandstone have been reported, so we followed the procedure presented by Purcell(1949) to obtain the matrix relative permeability curves.

Fracture relative permeability and capillary pressure curves were obtained by history matching the experiments. Sensitivity analysis of parameters such as fracture relative permeability, capillary pressure in the fracture, and fracture width have been completed. We considered each parameter independently.

First, we studied capillary pressure in the fracture under the assumption that it is a linear function of water saturation. Different cases for capillary pressure curves and relative permeability curves for the fracture are shown in **Fig. 14**. It is important to note that for fracture relative permeability curves, the lower the slope of the straight line, the higher the resistance to flow through the fracture. Starting with a high capillary pressure in the fracture, we observed that the matrix front matched well, but the breakthrough time was faster compared to our experimental results. We decreased the slope of the straight line capillary pressure in the fracture up to a point in which the breakthrough time as well as the matrix front matched the experiments. For the sake of completeness, we also studied extreme cases that are often used in real practice such as very low and no capillary pressure in the fracture, although neither one resembled experimental data. The no capillary pressure case showed us that the blocks worked independently, so the capillary continuity that we had seen in the experiments did not occur in the numerical studies.

After obtaining the proper description of capillary pressure in the fracture, we continued by studying the effect of fracture relative permeability curves. The assumption of fracture relative permeability equal to phase saturation is often used in numerical simulation. This assumption suggests, no resistance, ideal flow of fluids in the fractures, such that inside the fracture the phases can move past each other without hindrance. However, if relative permeabilities with a slope less than 1.0 are used, the effective total mobility is reduced. Pan and Wang (1996) discussed that higher resistance in the fractures must be used, such as 0.75 and 0.6 slope straight lines instead. Our results show that the best matches are achieved when an 0.6 slope is used. This indicates that the presence of a small amount of one phase interferes significantly with the flow of the other phase.

In order to obtain the best match, we also examined the heterogeneities present in the systems, especially in the wide fracture system. We ran different heterogeneous cases until we obtained better matches. We achieved this goal by assigning higher porosity (14%) to the bottom block and lower porosity (13%) to the top block. The results for this case are shown in

Fig. 15 which proves that, capillary pressure in the fracture can not be neglected and the heterogeneity must also be considered.

Similarly, we followed the same analysis for the system with no spacer in between the blocks. An interesting observation was that neither the capillary pressure in the fracture nor the relative permeability curves affected the results. That led us to the conclusion that the fracture is so thin that there is almost perfect continuity, and it acts like a solid block. **Fig. 15** also shows the comparison of the experimental results with the numerical simulation results for the system with no spacer. The fracture system without a spacer showed a more stable front and faster breakthrough than the other, as we had seen in the experimental results.

Similarly, we followed the same procedure for the oil injection stages. The matches for the thin fracture system were not very good, but the results for the wide fracture system, where we had seen in the experiments that most of the oil flowed through the fracture, are better. In this case, the assumption of zero capillary pressure in the fracture worked properly. These results are shown in **Fig. 16**.

Discussion and Recommendations

An apparatus that can be used to obtain detailed measurements of pressure, rate and saturation distribution was built and tested. It consists of a fractured core in an epoxy core holder, with six different locations for pressure measurements. Phase distribution in the matrix and inside the fracture were also determined by means of a (CT) scanner.

Multiphase flow runs were performed, and data were obtained. The experimental results were matched well by numerical simulation. Numerical simulations were also used to estimate the influence of variables such as fracture relative permeability, matrix/fracture capillary pressure, and fracture width. The results show that capillary pressure is a dominant parameter in this type of water displacement. Knowledge of the capillary pressure function is then critical.

The combined experimental and simulation study resulted in a much better understanding of the physical processes that occur when two or three phases flow in a fractured system, compared to previous reported studies (Guzman and Aziz (1993), Hughes (1995)).

This work allowed us to find areas with lower permeability and porosity, and then use them in the numerical simulations to obtain the best matches as shown in **Fig. 15**, for instance. All the CT images and the three-dimensional reconstructions obtained from them made the understanding of multiphase flow and the comparison with simulation results much easier, as shown in **Fig. 15** and **Fig. 16**.

Several authors (Kazemi and Merrill (1979), Beckner (1990), Gilman et al. (1994)) have assumed that fracture capillary pressures are negligible. Others have shown experimentally that capillary continuity becomes important when gravity provides a driving force (Horie et al. (1988), Firoozabadi and Hauge (1990), Labastie (1990), Firoozabadi and Markeset (1992). Kazemi (1990) stated that capillary continuity is prevalent in the vertical direction and has

suggested that, to reduce the number of equations to solve, fractured reservoir simulations should use the dual permeability formulation for the z direction and the dual porosity formulation for the x and y directions.

This work showed that capillary continuity can occur in any direction, depending on the relative strengths of the capillary and Darcy terms in the flow equations; the thin fracture systems have a more stable front and slower breakthrough compared to wide fracture systems, and that capillary pressure has more effect when the fracture is narrow. We observed that neither the capillary pressure nor the relative permeability curves in the fracture affected the results for the narrow fracture system. That led us to the conclusion that the fracture is so thin and/or each half mated so well that there is almost perfect capillary continuity, and it acts very similar to a solid block.

With this work we were able to verify that larger recoveries can be obtained when fractures are wider, and that the assumption of zero capillary pressure in the fracture is incorrect especially for air and water. The assumption of fracture relative permeabilities equal to the phase saturation was tested. This work shows that straight-line fracture relative permeability can be used, but it is not necessarily equal to the phase saturation. X-type relative permeability curves can be used for fractured system using larger flow resistances in the fractures, such as 0.75 or 0.6 sloped straight lines instead. Our results show that the best matches are achieved when a slope of 0.6 is used.

This work has shown that both capillary and viscous forces dominate the flow in fractures. Moreover, a procedure for determining the parameters involved in transmissibility and the transfer terms that appear in the flow equations in finite difference form was established.

The reader should realize that experiments presented in this work were performed only once. We assumed that our results were correct because they were comparable to Hughes (1995) results, and we could reproduce the experiments very accurately by numerical simulations. However a repeatability test should be performed on these experiments to achieve a higher certainty of the conclusions presented here.

Conclusions

The results of the combined experimental and the simulation study show:

1. Areas with lower permeability and porosity can be identified and used in numerical simulation.
2. Thin fractures systems showed more stable fronts and slower breakthrough compared to wide fracture systems.
3. Capillary pressure has more effect when the fractures are narrow.
4. Larger recoveries can be obtained when the fractures are wider.
5. Assuming zero capillary pressure in the fracture is incorrect.
6. X-type relative permeability curves can be used for fractured systems using high resistance through fractures, but relative permeability is not equal to phase saturation.

Nomenclature

CT = CT scanner number

S = saturation

ϕ = porosity

Subscripts

a = air

aw = air and water saturated

cd = dry core

cw = 100% water saturated

ow = oil and water saturated

w = water

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TABLE 1. RATES AND TIMING FOR THE CORE WITH NARROW FRACTURE

<u>Fluid Injected</u>	<u>Time (hrs)</u>	<u>Flow Rate (cm³/min)</u>	<u>Ports</u>
Water	0.0 - 4.0	2.0	Top and bottom closed, lateral open.
Water	4.0 - 5.0	2.0	Top and bottom open, lateral closed.
Water	5.0 - 19.8	0.5	Top and bottom open, lateral closed.
Water	19.8 - 23.0	2.0	Top and bottom closed, lateral open.
Decane	23.0 - 27.0	2.0	Top and bottom closed, lateral open.
Decane	27.0 - 28.0	2.0	Top and bottom open, lateral closed.
Decane	28.0 - 44.7	1.0	Top and bottom open, lateral closed.
Water	44.7 - 46.0	9.99	Top and bottom closed, lateral open.
	46		Shut down everything

TABLE 2. RATES AND TIMING FOR THE CORE WITH WIDE FRACTURE

<u>Fluid Injected</u>	<u>Time (hrs)</u>	<u>Flow Rate (cm³/min)</u>	<u>Ports</u>
Water	0.0 - 5.0	2.0	Top and bottom closed, lateral open.
	5.0 - 21.0	0.0	Top and bottom closed, lateral closed.
Water	21.0 - 22.5	2.0	Top and bottom open, lateral closed.
Water	22.5 - 25.6	2.0	Top open, bottom closed, and lateral closed.
	25.6 - 45.0	0.0	Top and bottom closed, lateral closed.
Decane	45.0 - 48.0	2.0	Top and bottom closed, lateral open.
Decane	48.0 - 49.2	4.0	Top and bottom closed, lateral open.
Decane	49.2 - 51.0	4.0	Top and bottom open, lateral closed.
	51.0 - 52.0	0.0	Top and bottom closed, lateral closed.
Water	52.0 - 54.3	4.0	Top and bottom closed, lateral open.
Water	54.3 - 68.8	0.5	Top and bottom closed, lateral open.
Water	68.8 - 71.0	4.0	Top and bottom open, lateral closed.
	71		Shut down everything

Fig. 5- Porosity distribution for the experimental system.

Fig. 6- CT Saturation images for the thin fracture system after 1 hr 30 min of water injection (0.67 PV.)

Fig. 7- CT Saturation images for the wide fracture system after 1 hr 30 min of water injection (0.67 PV.)

Fig. 8- CT Saturation images for the narrow fracture system after 2 hr 30 min of oil injection (1.13 PV.)

Fig. 9- CT Saturation images for the wide fracture system after 2 hr 30 min of oil injection (1.13 PV.)

Fig. 10- CT Saturation images for the wide fracture system after 16 hours of water injection (7.2 PV.)

Fig. 11- CT Saturation images for the wide fracture system after 17 hours of water injection (7.7 PV.)

Fig. 12- 3-D Reconstruction for both systems for water injection.

Fig. 13- 3-D Reconstruction for both systems for oil injection.

Fig. 14- Capillary pressure curves and relative permeability curves for the fracture.

Fig. 15- Comparison between experiments and numerical simulations for narrow and wide fracture systems for different PV of water injection.

Fig. 16- Comparison between experiments and numerical simulations for narrow and wide fracture systems for different PV of water injection.

1.3 IMBIBITION STUDIES OF LOW-PERMEABILITY POROUS MEDIA

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Imbibition Studies of Low-Permeability Porous Media

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Abstract

A systematic investigation of capillary pressure, relative permeability, and fluid flow characteristics within diatomite (a high porosity, low permeability, siliceous rock) is reported. Using an X-ray computerized tomography (CT) scanner, and a specially constructed imbibition cell, we study spontaneous cocurrent water imbibition into diatomite samples at various initial water saturations. Air-water and oil-water systems are used. Despite a marked difference in rock properties between diatomite and sandstone, including permeability and porosity, we find similar trends in saturation profiles and dimensionless weight gain versus time functions. Diatomite is roughly 100 times less permeable than sandstone, yet it imbibes water at rates rivaling sandstone. Importantly, the spontaneous imbibition data when combined with CT-scan images provides a means to determine dynamic relative permeability and capillary pressure functions.

Introduction:

Spontaneous imbibition is perhaps the most important phenomenon in oil recovery from fractured reservoirs. In imbibition, capillary suction draws wetting liquid into the rock matrix. In fractured systems, the rate of mass transfer between the rock matrix and fractures usually determines the oil production^{1,2}. Imbibition is also essential in evaluation of rock wettability³. Because of strong capillary forces, the smallest pore bodies next to the fracture are usually invaded first. The rate of imbibition is a function of porous media and fluid properties such as absolute and relative permeability, viscosity, interfacial tension, and wettability⁴. Most experimental work on imbibition behavior has concentrated on the scaling aspects of the process in order to estimate oil

recovery from reservoir matrix blocks that have shapes, and sizes different from those of laboratory core samples⁴⁻⁶.

Two different approaches are used, typically, to model imbibition behavior. The first approach employs a sugar-cube type matrix-fracture system proposed by Warren and Root¹ where the communication occurs at the matrix-fracture interface². The second approach is based on a representative elemental volume averaged fracture-matrix system⁶⁻⁸. Both approaches use empirically determined mass transfer functions.

Kazemi et al.² presented numerical and analytical solutions of oil recovery using empirical, exponential transfer functions based on the data given by Aronofsky et al.⁹ and Mattax and Kyte¹⁰. They proposed a shape factor that included the effect of size, shape, and boundary conditions of the matrix. Later, this shape factor was generalized by Ma et al.¹¹ to account for the effect of viscosity ratio, sample shape, and boundary conditions. The following equation was proposed:

$$t_D = t \sqrt{\frac{k}{\phi} \frac{\sigma}{\mu_s L_c^2}} \quad (1)$$

where

$$\mu_s = \sqrt{\mu_w \mu_{nw}} \quad (2)$$

The above scaling equation was used by Zhang et al.⁴ who report that ultimate oil recovery on a pore volume basis by spontaneous imbibition in Berea sandstone cores is approximately constant for systems with differing lengths, viscosity ratios, and boundary conditions.

To date, much of the focus on imbibition has centered on carbonaceous rocks as a result of the importance of the North Sea Chalks¹²⁻¹⁵, the West Texas Carbonates, and the Middle Eastern Limestones. An important, yet relatively unstudied, low permeability, reservoir rock for which imbibition is believed to be an important recovery mechanism is diatomite¹⁶. Imbibition phenomena are also important during steam injection into diatomite because steam condenses upon injection into a cool reservoir, and the hot condensate imbibes into the formation.

Diatomites are high porosity (>50%), low permeability (0.1-10md)¹⁷ rocks of a hydrous form of silica or opal composed of the remains of microscopic shells of diatoms, which are single-celled aquatic plankton¹⁸. Diatomite reservoir rock is assumed to be moderately to strongly water wet¹⁶. Estimates of the original oil in place (OOIP) for diatomite reservoirs in California range from 10 to 15 billion barrels¹⁹ and primary recovery has been estimated to be about 5% of the OOIP²⁰. Primary recovery is low despite hydraulically fractured wells that improve injectivity and productivity. Because of the high porosity, large initial oil saturation (35 to 70%), and large OOIP, the target for potential production is high.

Multiphase flow in diatomite is assumed to be dominated by capillary forces, but we lack a good understanding of fluid flow and capillary pressure behavior in the rock matrix. Compounding problems, there are only a few reported capillary pressure curves¹⁶ and little information on the extent and rate of imbibition²¹. The mechanisms of oil displacement and trapping are unclear, but assumed to be similar to those in sandstone. However, rock morphology is very different²².

This study presents basic core analysis data and experimental results of spontaneous water imbibition in diatomite and sandstone. A state of the art experimental imbibition cell is employed along with computed tomography (CT) scanning to quantify the saturation distribution along the core. It is found that the imbibition data scales according to the equation proposed by Ma et al.¹¹. The major focus of this article is estimation of relative permeability and capillary pressure from the experimental saturation data using a nonlinear history-matching technique.

Experimental Details

Liquids

For water-air imbibition experiments, de-aerated water and air were used, whereas for water-oil imbibition, n-decane and de-aerated water were used. Properties of the experimental fluids are given in Table 1.

Porous Media

Diatomite cores were cut in a direction parallel to the bedding plane from a block of outcrop diatomite sample (Grefco Quarry, Lompoc, CA). This sample is fairly homogeneous, but there are regions of higher density and correspondingly lower porosity. Figure 1 displays porosity images of the cores used in this study. The procedure to obtain these images will be presented, shortly. The average porosity of the samples was slightly greater than 65% and porosity varied by about 2 to 4%. The liquid permeability varied from 6 to 9 md. Table 2 lists exact values for each core. It should be noted that because of the brittle nature of the diatomite, the cores were machined instead of drilled using conventional core bits as described by Schembre et al.²¹. The lengths (3.45 in. or 8.763 cm) and diameters (0.95 in. or 2.413 cm) of the cores were all close to constant. Cores were potted with epoxy in 1 inch (2.54 cm)

Plexiglas tubes. Both ends are left open to enable cocurrent imbibition.

Experimental Setup

The experimental setup is designed specifically for CT measurements. It consists of three main parts: (i) a potted diatomite core inside a (ii) water jacket and (iii) a data acquisition system based on a personal computer, as shown in Figure 2. Two end caps hold the imbibition cell in position within the cylindrical water jacket. The end caps are machined with spider-web-shaped fluid distribution grooves where the end cap contacts the core. The entire assembly is placed inside the CT gantry. Fluid is circulated through the jacket to maintain a constant temperature up to 90°C (194°F) using a heating circulator bath. The experiments reported here were conducted at 20 °C (68 °F). The main function of the water jacket is to reduce effects of beam hardening. Beam hardening is the change of overall X-ray attenuation with distance into the object. It occurs when a polychromatic X-ray beam passes through a material that preferentially absorbs lower energy photons. The remaining beam becomes more and more chromatic at higher energy levels and the beam becomes "harder". The effects of this phenomenon are usually more pronounced at the boundaries where there is a high density contrast (i.e. air and core).

Three polyethylene lines (input, output, and bypass) allow the imbibing fluid to enter the imbibition cell and the produced fluid to exit the cell. The bypass line is used to flush any non-wetting fluid from the lines, thus allowing the imbibing fluid to fill the inlet end plate and easily contact the core face. The amount of water imbibed is found by measuring the weight loss of the water reservoir using an electronic balance connected to a personal computer. The CT-Scanner used in this study is a fourth generation Picker™ 1200-SX scanner with 1200 fixed detectors. It allows rapid scanning on a single vertical volume section in the center of the core as a function of time. The acquisition time of one CT image is 3 seconds, whereas the processing time is around 40 seconds. The total time of measurement is selected such that it is short enough to capture accurately the position of the imbibition front, yet it is large enough to provide necessary X-ray energy. Table 3 gives CT scan parameters used in this study.

Imbibition Measurements

For water-air imbibition experiments, the core samples were dried in a vacuum oven at approximately 70°C (158°F) for 24 hours before starting the experiments. After pressure testing for leaks, the imbibition cell is placed into the water jacket either horizontally or vertically and the jacket is filled with water. All experiments reported here are in the horizontal mode. Next, the setup is placed inside the gantry and positioned such that the main scanning plane is at the center of the core sample and there is nothing around the setup (i.e. the patient couch is not in the scanning plane). A reference dry CT image is obtained using the parameters given in Table 3.

The image covers the entire length of the core (e.g., Figure 1) rather than conventional circular slices.

Operating in this fashion, rather than conventionally, allows us to track accurately the position of the imbibition front. Additionally, saturation measurement with CT is much cheaper and faster than conventional techniques such as X-ray absorption^{14,23}. A final advantage is that continuous two dimensional porosity and saturation maps are generated that can be used to identify heterogeneity and non-uniform displacement. Hence, no three-dimensional reconstruction is necessary.

CT scans are taken during this period at prescribed time intervals. After a day of water imbibition into the initially air-filled core, several pore volumes of de-aerated water are injected using a positive displacement pump to ensure complete saturation. A reference 100% water saturated CT image is then captured. The core plug is then de-saturated to an irreducible water saturation by pumping n-decane at a rate of 1 ml/min. Another CT image is taken to observe the water saturation distribution at the end of this stage. Water imbibition is initiated and progress of imbibition monitored by CT. Three oil-water diatomite imbibition experiments with nonzero initial water saturations are reported here.

In order to observe the effect of zero initial water saturation, another experiment was conducted. This time the air-filled core is imbibed with oil to 100% oil saturation, and then imbibed with water following the same experimental procedure for at least 14 days.

The porosity, ϕ , and saturation, S_w , distributions in water-air imbibition experiments were calculated using the following set of equations:

$$\phi = \frac{CT_{wet} - CT_{dry}}{CT_{water} - CT_{air}} \quad (3)$$

$$S_w = \frac{CT_{obj} - CT_{dry}}{CT_{wet} - CT_{dry}} \quad (4)$$

where CT is the CT number in Hounsfield units. The subscripts dry and wet refer CT numbers for air-filled and water-filled cores, respectively, whereas water and air indicate the CT numbers of pure water and air. The subscript obj describes any CT number during the spontaneous imbibition process.

A slightly different approach is taken for analysis of the water-oil imbibition experiments because it is hard to obtain 100% oil and 100% water saturated images within the same experiment. For experiments with nonzero initial water saturation, the 100% water saturated image is used as the reference image in the following equation:

$$S_o = \frac{CT_{obj} - CT_{wet}}{\phi(CT_{oil} - CT_{water})} \quad (5)$$

where ϕ is the independently measured porosity for a given voxel. However, for experiments with no initial water, the reference image is 100% oil saturated and thus the saturation calculation equation has the following form:

$$S_w = \frac{CT_{obj} - CT_{oil}}{\phi(CT_{water} - CT_{oil})} \quad (6)$$

These equations are comparable and compatible with Eq. 4 in terms of accuracy, because the images obtained here are flat and exhibit minimum beam hardening and nondetectable positioning effects.

Results and Discussion

Water-Air Imbibition Experiments

A total of four experiments with core properties as reported in Table 2 and the porosity field shown in Fig. 1 were conducted. The permeability varies from 6.2 to 8.6 md., and the porosity, as derived from CT, varies from 67 to 71%. The differences in permeability of the samples, although small, affect the duration of imbibition. Using a 95% water saturation level as a benchmark, an 18 hour difference is observed between completion of the fastest and slowest imbibition experiments.

To confirm that results for diatomite reduce to a single curve when scaled by porosity and permeability, dimensionless weight gain versus dimensionless time, as given by Eq. 1, is plotted in Figure 3. Good agreement is achieved. This demonstrates that water-air imbibition in these diatomite samples can be modeled using a single set of capillary pressure and relative permeability curves. For reference, Fig. 3 also plots the results of imbibition in a roughly 800 md Berea sandstone.

CT derived water saturation images at differing times for each experiment are given in Figure 4. Common to all tests, flow is initially spherical about the inlet region. It was noticed that the duration and the extent of this regime is a function of permeability. That is, if the permeability of the system is small, the spherical flow lasts longer. The most probable reason for the occurrence of this flow regime was the inlet port where a 1/8 inch line was connected to the spider-web-shaped grooved end cap. Water contacted the core from a single point rather than the whole bottom surface of the core plug. We minimized the duration and occurrence of this initial spherical flow regime by placing filter paper between the core and the end cap, but this did not eliminate this problem completely. Localized wettability alteration from epoxy fumes has also been suggested, but this effect, if present, was probably most significant near the sides of the samples.

After the end of the spherical flow period, the front becomes flat and one dimensional displacement occurs. The

water saturation gradient across the displacement front is high and the front expands and diffuses somewhat as it advances along the core. There are essentially three regions in the core. In the swept zone, the saturation of the displacing fluid (water) is high, and very close to residual nonwetting phase saturation. The displacement front is a transition zone that separates swept and unswept regions. Downstream of the transition zone, the saturation of the displaced fluid (air) is at its initial value.

Despite slight spreading of the front, the shape of the transition zone changes little throughout the displacement until the front approaches the end of the porous medium. At this point, the transition zone disappears as the end of the core fills with water. The time for arrival of the leading edge of the front at the end of the core corresponds to the change in the slope of the dimensionless weight gain curve as shown in Fig. 4.

Handy proposed that imbibition could be described by the solution of either a diffusion-like equation or a frontal-advance equation²⁴. In the former, the diffusion coefficient is proportional to the partial derivative of capillary pressure with respect to water saturation and computed displacement fronts are significantly diffuse. Whereas in the latter, the velocity of the imbibed phase is proportional to the gradient of capillary pressure with respect to distance, and fronts are assumed to be sharp. The experimental findings as observed in Fig. 4 suggest that the correct idea for imbibition into dry, low permeability porous media that are relatively short is more similar to frontal advance. A relatively sharp front, rather than a diffuse one, is observed in all experiments regardless of the level of heterogeneity present.

Heterogeneity clearly affects the flatness of the front as observed in cores 2 and 5. A sharp front suggests that during spontaneous imbibition, pores of all sizes fill simultaneously. This implies that large pores are well interconnected to small pores. This observation is consistent with the complex, small-diameter pore network of diatomite^{22,25}. Thus, it is hard to fill the very small pores selectively, leaving relatively large pores unfilled.

A comparative imbibition experiment was conducted with a Berea sandstone core to verify development of similar sharp fronts. The sample had a porosity slightly above 22% and the permeability to water was around 800 md. Following the same experimental procedure, weight gain and CT images were recorded throughout the experiment. Relatively sharp fronts similar to those detected in diatomite experiments were observed as shown in Fig. 5. These observations are in agreement with other recent experiments in Berea sandstone where the position and shape of a water imbibition front were tracked²⁶. The primary difference between diatomite and sandstone results is that the water saturations are considerably smaller at identical dimensionless time as observed in Fig. 4. The water saturation at the end of the spontaneous imbibition in the sandstone was 65%, compared to about 95% obtained in diatomite experiments. This suggests that the complex structure and low permeability of diatomite actually aid the

spontaneous imbibition of water via capillary pressure into a plug filled initially with air.

Modeling Imbibition Experiments

The CT images and weight gain data from spontaneous imbibition experiments contain much useful information about relative permeability and capillary pressure. We use a nonlinear history-matching technique to model these experiments and extract this information. The following sections describe the methodology used to model diatomite imbibition experiments. These techniques are also applicable to analysis of the oil-water experiments to come.

The nonlinear history-matching performed here is standard. A reservoir simulator (Eclipse, 1995) is coupled with an optimization code. We apply global optimization based on simulated annealing to minimize the following objective function

$$\sum_{i=1}^N \sum_{j=1}^M \left(s_{ij}^{\text{experimental}} - s_{ij}^{\text{model}} \right)^2 \quad (7)$$

where s_{ij} are saturation values along the core. An initial guess of relative permeability and capillary pressure data at specified values of saturation is provided and Eq. (7) is minimized. More detail on the optimization methodology can be found elsewhere²⁷.

The grid used for water-air imbibition is one dimensional with circular cross section and consists of 122 blocks in the horizontal direction. Two grid blocks at each end were assigned high porosity and permeability to account for the water reservoir and the spider-web-shaped end plugs. Constant porosity and permeability along with varying porosity and permeability fields were considered.

Figure 6 compares CT derived experimental saturation histories with numerically estimated values for experiment 4 using the average porosity and permeability of each core. Solid lines are experimental data, while dashed lines represent the model data. For consistency with Fig. 3, lines are labeled in units of dimensionless time. It should be noted that capillary pressure and relative permeability curves were estimated simultaneously. It can be observed that the one dimensional numerical model fails to describe experimental results at the inlet and outlet ends of the core. The inlet end mismatch is explained by the aforementioned spherical, nonlinear, flow observed during the early times of the experiments. A close analysis of Fig. 6 reveals that this flow regime is effective for dimensionless distances less than 0.10. After this point, the match between the numerical model and the experimental results gets better, showing that the one dimensional flow assumption is valid.

Figure 7 gives the relative permeability curves obtained using the history-matching procedure. Solid symbols refer to data obtained assuming that porosity and permeability are

homogeneous. The computed relative permeability curves feature typical water-wet rock characteristics such as the crossover water saturation is greater than 50% and the end-point water relative permeability is less than end-point gas relative permeability. The estimated capillary pressure curve shown in Fig. 8 is also of the same magnitude, yet somewhat lower, than our limited experimental data. Measured data for the air/water system were obtained with a Ruska porous plate cell (model no. 1081-802) in the imbibition mode of capillary pressure measurement. The maximum capillary pressure measurable with this set up is roughly 480 kPa (4.8 atm).

We attempted to improve the analysis by incorporating heterogeneity during history matching. First, the local porosity obtained from the CT measurements was used rather than average values. Second, the following empirical permeability versus porosity relationship obtained from the average data given in Table 2 is used to estimate the local permeability from porosity:

$$k = 1.161 * 10^{1.286\phi} \quad (8)$$

Using the above equation for each CT voxel, a permeability map is generated from measured porosity information. Porosities and permeabilities are down sized to 1-D by arithmetically averaging the voxel data corresponding to each of the 120 grid blocks. Consequently, the history matching is repeated. The sum of the squares residual was less than the homogeneous run and the matches improved. Figure 9 compares the model and experimental saturation histories for the heterogeneous case. Again, solid lines are experimental data, while dashed lines represent the model data. A slight improvement in the match apart from the inlet and outlet matches can be observed. The relative permeability and capillary pressure curves shown in Figs. 7 and 8 are more or less the same as the homogeneous case. Values obtained by incorporating heterogeneity are given as open, unfilled symbols.

Water-Oil Imbibition Experiments

A total of four experiments were conducted using the same diatomite cores utilized in water-air imbibition experiments. One experiment was conducted without initial water saturation, whereas the others were conducted by starting at an irreducible water saturation. An additional experiment was conducted utilizing core 3 to check the repeatability of the results.

Similar to the water-air experiments, the data collected during the experiments consisted of weight gain measurements and CT scans. In general, the duration of the experiments were much longer compared to water-air imbibition experiments. This is expected because the water-n-decane interfacial tension (51.4 mN/m) is lower than the water-air (72 mN/m) interfacial tension, oil is more viscous than air and much harder to displace, and with initial water saturation the imbibition

capillary forces should be less. For the zero initial water saturation case, an initial spherical flow period followed by a strong, relatively sharp displacement front similar to the fronts observed in water-air imbibition experiments was noticed, as shown in Fig. 10. A stabilized zone was also present with an extent that is somewhat larger than the zone identified in the water-air imbibition experiments. The displacement was not as effective as the water-air case. Note that the water saturation at dimensionless time equal to 5000 is around 73% compared to an almost 95% water saturation for water-air imbibition.

For experiments with a nonzero initial water saturation, a sharp, clear front is not observed as illustrated in Fig. 11. In this case, water saturation increases almost uniformly throughout the entire core. In this sense, the water-oil imbibition results are diffuse and diffusion-like as compared to the water-air case. The oil saturation reached at the end of spontaneous imbibition experiments varied between 20% to 27% as reported in Table 4. This corresponds to oil recoveries between 18% to 22%. These are in agreement with 8% to 20% oil recovery reported in non-fractured diatomite core plugs by Wendel et al.¹⁶ Figure 12 presents dimensionless imbibition performance curves for all oil-water experiments. The shape of curves for runs with initial water saturation are all similar, within experimental error, indicating that a single set of relative permeability and capillary pressure curves can represent spontaneous imbibition for these experiments. Note, that the core without initial water saturation displays significantly better imbibition performance. Figure 12 also shows that 18 to 22% recovery can be obtained within reasonable time frames.

A comparative water-n-decane imbibition experiment was conducted using the same Berea sandstone core as in the water-air imbibition experiment. Again, weight loss and CT images were recorded throughout the experiment. Figure 13 presents n-decane saturation images obtained at various times. Similar to the experiments in diatomite, water imbibes into the core without the development of a front despite different initial water saturation and residual oil saturation values as given in Table 4. The dimensionless weight gain curve for the sandstone is plotted on Fig. 10. It shows that the imbibition performance of the sandstone is somewhat below diatomite because the sandstone curve lies to the right of the diatomite curves.

Modeling Water-Oil Imbibition Experiments

The same grid system and procedure used for the water-air case is used to history match the water-oil imbibition experiments. Runs 3 and 4 are the furthest apart of the experiments with initial water saturation, as shown in Fig. 12. Hence, we choose to history match these results to test the sensitivity of our procedure to input data. Figure 14 compares CT derived experimental saturation histories with numerically estimated profiles. Figure 14(a) presents experimental and simulated saturation profiles for run 3 and Fig. 14(b) gives the same information for run 4. Again, capillary and relative

permeability curves were estimated simultaneously for each experiment. Heterogeneity was considered by assigning averaged porosity and permeability data to individual grids as described in conjunction with water-air imbibition. Reasonable agreement in timing and saturation values is observed indicating the reproduction of the essential physics of water imbibition into partially water saturated cores.

Figure 15 gives the relative permeability data obtained for each match. Good agreement is seen in shape, endpoint values, and the crossover point for oil and water curves. As in the water-air case, the relative permeability curves have the characteristics of strongly water-wet media.

The capillary pressure curves found for these experiments are given in Fig. 16. As in the case of oil-water relative permeability, good agreement is found between the two curves. Interestingly, the shape of these curves and the magnitude of the capillary pressure differs substantially from that found for the water-air case. In Fig. 8, the history-match, air-water capillary pressure curve increases little as S_w decreases, whereas in the oil-water system shown in Fig. 16, the capillary pressure increases substantially between water saturations of 0.65 and 0.45. Additionally, the magnitude of P_c found at low water saturations in the oil-water system exceeds that of P_c in the air-water system. For example, at S_w equal to 0.5 capillary pressure in the air-water system is roughly 200 kPa while in the oil-water system it is about 1600 kPa. Previously measured curves for oil and water have displayed maximum capillary pressures of roughly 600 kPa¹⁶. We expected the oil-water values to be lower than air-water values because the surface tension of n-decane is 51 mN/m, while that of water is 72 mN/m.

Conclusions

Spontaneous imbibition experiments were performed on core plugs machined from an outcrop diatomite block. Displacement of gas (air) or oil (n-decane) by water was considered. The experiments were limited to spontaneous cocurrent imbibition and were monitored with a CT scanner to observe local saturation evolution. For comparison, experiments were conducted using Berea sandstone. It was observed that both water-air and water-oil imbibition results in diatomite can be correlated with a single dimensionless function. Moreover, it was noticed that imbibition fronts in the absence of initial water saturation are sharp suggesting that during spontaneous imbibition pores of all sizes fill simultaneously. In turn, this implies that large pores are interconnected to small pores in the complex network structure of diatomite.

Relative permeability and capillary pressure information is extracted from the experimental data. The evolution of local water saturation was correctly reproduced in all cases. For both water-air and water-oil systems, relative permeability curves display strongly water-wetting tendencies. Capillary pressure curves for the water-air case agree with available experimental data.

Nomenclature

A	= cross-sectional area, L^2
CT_{air}	= CT value of air, Hounsfield;
CT_{dry}	= CT value for the dry core, Hounsfield
CT_{obj}	= CT value of the image being processed, Hounsfield
CT_{oil}	= CT value of oil, Hounsfield
CT_{wet}	= CT value for a fully water saturated core, Hounsfield
CT_{water}	= CT value of water, Hounsfield
g	= gravitational acceleration, L/t^2
h	= height, L
J	= objective function
k	= permeability, L^2
m	= mass of water imbibed, M
P_c	= capillary pressure, $ML^{-1}T^{-2}$
S	= saturation, dimensionless
t	= time, T
x	= distance, L
ρ	= density, mL^{-3}
μ	= viscosity, $ML^{-1}T^{-1}$
Φ	= potential, $ML^{-1}T^{-2}$
ϕ	= porosity, dimensionless
σ	= interfacial tension, MT^{-2}

subscripts:

r	= relative
w	= wetting
nw	= nonwetting

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Table 1: Properties of experimental fluids.

	ρ (kg/m ³)	μ (mPa - s)	σ liquid- air (mN/m)	CT # (Hounsfield)
water	1000	1	72	0
decane	730	0.84	51.4	-283
air	1.2	0.0182	-	-1000

Table 3. CT scan operating parameters.

CT scan time, sec	3
Current, ma	65
Voltage, kV	140
Voxel size, mm x mm x mm	0.5 x 0.5 x 10
Matrix size, voxels	512 x 512

Table 2: Diatomite permeability and porosity values.

Core	Permeability, md	Average porosity
1	6.2	0.67
2	6.6	0.68
3	8.6	0.71
4	6.65	0.71

Table 4. Spontaneous imbibition end-point data, water-oil system.

	S_w initial, %	S_{or} , %	Recovery, %
1	0.0	23.9	73.36
2	55.79	21.74	18.51
3	56.05	25.7	22.21
4	58.70	26.6	20.52
Berea	37.0	28.0	35.0

Fig. 1.—Porosity images of diatomite cores 1 (top) through five (bottom).

Fig. 3.—Dimensionless weight gain versus dimensionless time, water/air imbibition.

Fig. 2.—Schematic of the experimental setup.

Fig. 4.—Water saturation distribution during water imbibition into an initially air-filled core. The top left series of images corresponds to core 2, top right core 3, bottom left core 4, and bottom right core 5. In each series, the images correspond to

dimensionless times of 5 (top), 15, 40, 65, 500, and 3000+.

Fig. 5.—Water saturation distribution along Berea sandstone during water imbibition into an air-filled core. Images correspond to dimensionless times of 200 (top), 500, 5000, 12,000+ (bottom).

matching water/air imbibition results.

Fig. 8.—Capillary pressure functions obtained from history matching water/air imbibition results compared to measured imbibition capillary pressure.

Fig. 6.—Comparison of experimental and model water saturation profiles at various dimensionless times for water/air imbibition, run 4. Homogeneous porosity and permeability is assumed.

Fig. 9.—Comparison of experimental and model water saturation profiles at various dimensionless times for water/air imbibition, run 4. Heterogeneous porosity and permeability fields are honored.

Fig. 7.—Relative permeability functions obtained from history

Fig. 10.—Water saturation distribution along diatomite during water imbibition into an oil-filled core. Images correspond to dimensionless times of 0 (top), 5, 500, 1000, 5,000 (bottom).

Fig. 12.—Dimensionless weight gain versus dimensionless time, water/oil imbibition.

Fig. 11.—Oil saturation distribution along diatomite during water imbibition into a partially water saturated core. Images correspond to dimensionless times of 0 (top), 5, 50, 500, 5000 (bottom).

Fig. 13.—Oil saturation distribution along Berea sandstone during water imbibition into a partially water saturated core. Images correspond to dimensionless times of 5 (top), 50, 500, 5000.

Fig. 14.—Water saturation profiles at various dimensionless times for water/oil imbibition into diatomite cores with initial water saturation (a) results from run 3 and (b) results from run 4.

Fig. 15.—Relative permeability functions obtained from history matching water/oil imbibition results.

Fig. 16.—Capillary pressure functions obtained from history matching water/oil imbibition results compared.

1.4 SPONTANEOUS IMBIBITION CHARACTERISTICS OF DIATOMITE

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Abstract

A systematic investigation of fluid flow characteristics within diatomite (a high porosity, low permeability, siliceous rock) is reported. Using an X-ray computerized tomography (CT) scanner, and a novel, CT-compatible imbibition cell, we study spontaneous cocurrent water imbibition into diatomite samples. Air-water and oil-water systems are used and the initial water saturation is variable. Mercury porosimetry and a scanning electron microscope (SEM) are employed to describe diatomite pore structure and the rock framework. Diatomite exhibits a fine pore structure and significant pore-level roughness relative to sandstone thereby aiding the flow of imbibing water. Despite a marked difference in permeability and porosity as compared to sandstone, we find similar trends in saturation profiles and dimensionless weight gain versus time functions. Although diatomite is roughly 100 times less permeable than sandstone, capillary forces result in a strong imbibition potential for water such that imbibition rates rival and surpass those for sandstone

Introduction

Water imbibition is fundamental to both waterflood and steamdrive performance in low permeability reservoir rocks such as diatomite and chalk. In imbibition, wetting fluid is drawn into rock by capillary suction and the non-wetting fluid is spontaneously expelled. The rate and the extent of imbibition depend primarily on the capillary pressure versus water saturation relationship for the rock, the initial water saturation of the rock, the relative permeability curves, and the viscosity of the wetting and nonwetting phases. In turn, the capillary pressure characteristics of a rock depend on factors such as permeability, pore structure, the pore-throat to pore-body size ratio, wettability, and the interfacial tension between the resident and the imbibing phase.

On the macroscopic scale, capillary imbibition forces determine, in part, how rapidly and easily water is injected into a low permeability formation via a hydraulically fractured injector. In naturally fractured systems with a high degree of interconnectedness, imbibition forces must be strong for a waterflood to be successful. If not, water will propagate through the fracture network from injector to producer with little imbibition into the matrix and the waterflood will fail. Capillary phenomena are equally important during steam injection into diatomite or lower permeability sandstones. Steam injection, especially at short times, is accompanied by condensation and flow of the resulting hot water away from the injector. Likewise, for steam vapor to enter the matrix of a low permeability rock, a substantial capillary entry pressure must be overcome.

Because of the importance of imbibition on oil recovery, it has been studied widely. However, our knowledge of hydrocarbon recovery is less than complete and it is not yet possible to generalize imbibition performance across rock types. Thus, it seems helpful to

undertake a systematic study of fluid transport in diatomite; it appears that capillary driven flow is relevant to reservoir flow processes. For diatomite, a clear understanding of the relationship between pore structure, reservoir architecture, rock wettability, and oil recovery remains to be developed. This is a monumental task and beyond the scope of this paper. Nevertheless, we have begun by examining flow in clean, well-characterized rock samples. Our main focus is the design of a novel CT monitorable imbibition cell and the study of multidimensional spontaneous imbibition of water. We examine water-air and water-oil systems. Results are best understood by considering jointly rock characteristics such as pore-size distribution, the wetting-phase occupancy of pore space, and permeability.

Before describing our pore-level and core-level rock characterization procedures, experimental imbibition apparatus, and results, we briefly discuss the mineralogy and depositional environment of diatomite. Then theory regarding imbibition is reviewed to put experimental results into context and simplify the discussion to follow.

Diatomite

Diatomite is a hydrous form of silica or opal composed of the depositional, consolidated skeletal remains of colonies of unicellular aquatic plankton (Stosur and David 1971). Rock color and texture bear resemblance to chalk, but diatomite contains virtually no carbonaceous material. The mineral composition is primarily biogenic silica, detritus, and shale. Depending on depth and composition, the silica may exist as Opal-A, Opal-CT, or Brown Shale (Schwartz 1988). Generally, amorphous silica is found at the shallowest depths. Diatomite is characterized by permeabilities ranging from about 0.1 to 10 mD that result from a complex small diameter pore network. Curiously, this low permeability occurs with a very high porosity that varies from 35 to 65%. The permeability versus porosity relationship indicates that this rock is different from sandstone in pore-level characteristics. Mechanically, it is described as brittle and friable (Wendel et al. 1988; Chase and Dietrich 1989). This aspect poses a problem when coring samples for laboratory analysis.

In the San Joaquin Valley, California, diatomite is the uppermost productive member of the Monterey formation. Initial oil saturations vary from 30 to 65% and total oil accumulations in diatomite are estimated to be at least 10 to 12 billion barrels original oil in place (Ilderton et al. 1996). Oil-bearing diatomite layers are interbedded among shale and mudstone layers (Schwartz 1988). Individual layers vary in thickness from a few centimeters to several meters and the gross thickness of these layers can exceed 330 m (1000 ft). The interbedding of diatomaceous rock and shale resulted from cyclic/seasonal deposition of diatoms, mud, and silt and the subsequent consolidation of diatom fragments and grains of mud/silt (Schwartz 1988). Thus, the quality of diatomite varies from layer to layer and field to field. In some cases, diatomite has a matrix that is almost entirely biogenic silica containing very little clay and is probably water wet. In others, clay content can be relatively high and might contribute to mixed-wettability of the rock. As described later, we choose to begin with and characterize relatively clean diatomite.

In addition to natural fractures that may be cemented or uncemented, wells in diatomite are hydraulically fractured to improve well productivity and injectivity. Induced fractures are massive with heights on the order of 100 m and total lengths of roughly the same magnitude (Ilderton et al. 1996). Such fractures are used for water (Patzek 1992) and steam injection (Kovscek et al. 1996 a and b), as well as production.

Imbibition

Spontaneous imbibition is perhaps the most important phenomenon in oil recovery from fractured reservoirs. For such reservoirs, the rate of mass transfer between the rock matrix and fractures usually determines the oil production (Warren and Root 1963). Imbibition is also essential in evaluation of the rock wettability (Jadhunandan and Morrow 1991; Morrow et al. 1994). For spontaneous imbibition, the main driving force is capillary suction. Because of strong capillary forces, the smallest pore bodies that are next to the interface are always invaded first. Usually, the displacement takes place at small but finite capillary numbers (c.f., (Sahimi 1995)).

One of the earliest investigations of the dynamics of imbibition is provided by Handy (1960) who examined the limit of capillary forces dominating over buoyancy and viscous forces. He notes that imbibition could be described by either a diffusion-like equation or a frontal-advance equation, depending on assumptions. The main difference between the two is that the diffusion-like equation predicts that the smallest pores fill first and the larger pores fill later. The frontal advance equation assumes that pores of all size fill simultaneously because large pores are well connected to small pores, and *vice versa*. Both developments predict that the mass of water imbibed depends linearly on the square root of time. Since, the end of imbibition was quite abrupt, which is contrary to the expected result for a diffusive type process, Handy asserted that the frontal advance assumption more nearly described the true process. Indeed, recent experiments where the position and shape of a co-current water imbibition front in an initially dry Berea sandstone were tracked accurately using X-ray computed tomography (CT) scanning indicate relatively steep and sharp fronts (Garg et al. 1996).

Upon assuming that water imbibes in a piston-like manner, gas viscosity is negligible, capillary forces outweigh gravity forces, and the spatial gradient of capillary pressure is linear, the mass of imbibed water is given by (Handy 1960)

$$m = \rho_w A \left(\frac{2 P_c k \phi S_w}{\mu_w} \right)^{1/2} t^{1/2} \quad (1)$$

In Eq. (1), m is the mass of water imbibed, ρ_w is the density of water, A is the cross-sectional area, P_c is the capillary pressure, k is the permeability to water, ϕ is the porosity, S_w the aqueous phase saturation, μ_w is the wetting-phase viscosity, and t is the time. The linear relation between the mass imbibed and the square root of time is apparent and the slope is proportional to the square root of $P_c k S_w / \mu_w$. Thus, the larger is the product of capillary pressure and permeability, the more rapid is spontaneous imbibition.

A great deal of experimental work on imbibition has concentrated on the scaling aspects of the process in order to estimate oil recovery from reservoir matrix blocks that have shapes and sizes different from those of laboratory core samples (Morrow et al. 1994; Zhang et al. 1996). Kazemi et al. (1989) presented numerical and analytical solutions of oil recovery using empirical, exponential transfer functions based on the data given by Aronofsky et al. (1958) and Mattax and Kyte (1962). They proposed a shape factor that included the effect of size, shape, and boundary conditions of the matrix. More recently, this shape factor was generalized by Ma et al. (1995) to account for the effect of viscosity ratio, sample shape, and boundary conditions. They proposed a dimensionless time, t_D ,

$$t_D = t \sqrt{\frac{k}{\phi}} \frac{\sigma}{\mu_s L_c^2} \quad (2)$$

where k is absolute permeability, σ is interfacial tension, and L_c is a characteristic length. The geometric average of phase viscosity and the characteristic length are given by the following equations:

$$\mu_s = \sqrt{\mu_w \mu_{nw}} \quad (3)$$

$$L_c = \sqrt{\frac{V_b}{\sum_{i=1}^n \frac{A_i}{l_{Ai}}}} \quad (4)$$

where V_b is the bulk volume of the rock sample, A_i is the area open to imbibition at the i^{th} direction, and l_{Ai} is the distance traveled by the imbibition front from the open surface to the no-flow boundary. The above scaling equation was used by Zhang et al. (1996) and they report that ultimate oil recovery on a pore volume basis by spontaneous imbibition in Berea sandstone cores is approximately constant for systems with differing lengths, viscosity ratios, and boundary conditions.

Additional work in sandstones focused on closed form modeling of counter-current imbibition in single and multidimensions to displace oil and air (Reis and Cil 1993; Cil and Reis 1996). The basic assumption in the models is that water imbibes in a piston-like manner. Dimensionless time is taken as the time required for the imbibition front to reach the center of a matrix block.

Imbibition in chalk rock systems has also been examined. Cuiec et al. (1994) found experimentally that oil recovery by imbibition is rapid and efficient for strongly water-wetting chalk. Reduction in interfacial tension (IFT) decreased the rate of imbibition, but ultimate oil recovery increased with lower IFT. This was attributed to the mobilization of oil ganglia. Milter and Øxnevad (1996) evaluated the imbibition potential of different chalk facies and linked imbibition performance to the rock framework and pore system. Chalk with significant interconnected pore-wall roughness spontaneously imbibed large amounts of water. It was hypothesized that surface recessions allowed the rock to maintain water wettability by retaining water along pore walls.

Experimental Apparatus and Method

Our objectives are to obtain the change of average water saturation with time, the rate of imbibition, and determine fluid displacement patterns for water imbibition into air-filled and partially oil-saturated cores using CT scanning. We examine water-wet diatomite and, for reference, Berea sandstone cores.

An imbibition cell was designed specifically for use in the CT scanner. Rather than scan in a conventional mode where X-ray CT data is collected in cylindrical volume sections that are normal to the central axis of the core, the entire length of the core is scanned as illustrated in Fig. 1. Because we scan along the direction of flow, a single slice provides us with a picture of the progress and saturation pattern of imbibition. Nonlinearity in flow behavior and the effect of heterogeneities are thereby easily gauged. In conventional CT-scanning many rapid scans are required to determine accurately the position and shape of displacement fronts. This change in scan orientation required redesigning the coreholder assembly such that beam hardening and x-

shaped imaging artifacts caused by asymmetry in the shape of the scanning plane are minimized.

An exploded view of the imbibition cell is shown in Fig. 2. The cell is constructed from acrylic tubes. There are two separate chambers: the main chamber is the core holder and the second is a water jacket of circular cross section that surrounds the core. Cores are potted inside the acrylic tube with epoxy (Epoxy 907, Miller-Stephenson Chemical Co.). There is no fluid exchange between the two chambers and the outer water-filled chamber allows for some measure of temperature control. The core holder may be placed in either a horizontal or vertical position. The core holder has two end caps, shown schematically at the top and bottom of the core, for fluids to flow in and out of the core holder. The endcap faces that meet the core contain spider-web-shaped channels to distribute water evenly. The inlet cap is connected to a water supply tank through a plastic tube. The inlet also contains a bypass line so that the endcap can be filled quickly with water. The weight of imbibed fluid is measured directly by means of a balance and is recorded on a computer as shown in Fig. 1.

The circular, symmetric shape of the water jacket and the large mass density of water minimize beam hardening effects and x-shaped artifacts. Beam hardening refers to the differential adsorption of longer wavelength X-rays and leads to shadows around the periphery of CT images. Artifacts can also be introduced if the cross-sectional shape of the object being scanned is not symmetric about its central axis. The imbibition cell has an L-shaped mounting bracket for bolting of the apparatus to a precision positioning system (Compumotor RP240, Parker-Hannifin Corp). Hence, images obtained are flat and do not exhibit detectable positioning errors.

Our CT-scanner is a Picker™ 1200 SX X-ray scanner with 1200 fixed detectors. The voxel dimension is (0.5 mm by 0.5 mm by 5 mm), the tube current is 65 mA, and the energy level of the radiation is 140 keV. The porosity and aqueous-phase saturation fields are measured on a single vertical volume section in the center of the core as a function of time. The acquisition time of one image is 3 seconds while the processing time is around 40 seconds. The total time of measurement is short enough to capture accurately the position of the front and construct the saturation profiles along the core.

The experimental procedure depends slightly upon whether water/air or water/oil imbibition is performed. The core is exposed to house vacuum and a temperature of 50 °C for at least ten hours to ensure a “dry” core at the beginning of the experiment. Filter paper is placed at the bottom of the core, the end caps placed on the core holder, and the core holder is placed inside the water jacket and leveled. The purpose of the filter paper is to obtain a uniform distribution of water at the bottom rock face. For water/air experiments, the core is placed in a vertical position and water is introduced at the bottom of the core. For water/oil experiments, the core is placed in a horizontal position such that the only driving force for oil production is capillary pressure. Since, the core holder is surrounded by water during the CT experiments, a leak test is performed by applying a slight gas pressure and checking that the pressure in the core holder is maintained for a period of time. The water jacket is then filled with water and a dry image of the core is taken to obtain the reference dry core CT values.

Water-Air Experiments

For water/air imbibition, water is introduced to the open bottom face of the core and the progress of imbibition is monitored with frequent CT scans. Since imbibition is spontaneous, care is taken to maintain the water level even with the bottom of the core to minimize forced imbibition. The Bond number ($=(\rho_w - \rho_{air})gk/\sigma$) for water/air displacement in diatomite is of order 10^{-9} using representative values for water and air density, permeability, and water-air surface tension (Tables 1 and 2). Hence, buoyancy forces are negligible relative to capillary forces. This indicates that the vertical position of the core has no effect on water-air imbibition results.

Once the imbibition begins, the weight of the water reservoir is recorded every 10 seconds. The amount of water imbibed is computed directly from the change of weight measured in the tank containing the water as illustrated in Fig. 1. Scanning is performed every 50 seconds for about the first 30 minutes of imbibition. Afterward, the intervals between images are longer since the change of saturation in the core becomes slower with time. Data acquisition ceases when changes in weight are no longer observed. De-aerated water is pumped through the core after imbibition to ensure that the core is completely filled with water.

Raw CT-number data is converted to a map of porosity and the saturation profiles are obtained by subtracting images of raw CT data. Thus, in the case of the porosity

$$\phi = \frac{CT_{wet} - CT_{dry}}{CT_{water} - CT_{air}} \quad (5)$$

where CT denotes the CT value for a voxel. The subscript wet refers to a fully water saturated core, obtained from a last scan performed after 12 hours or more of water imbibition and dry refers to an air-filled core, obtained from scanning the core before the imbibition process starts. The subscript water refers to the CT value for the water phase, and likewise air refers to the value of the gas phase. It is important to note that CT_{wet} corresponds to 100% water saturation. The CT numbers for the water and air phases are listed in Table 1 along with other properties of the experimental fluids.

To construct saturation profiles we use

$$S_w = \frac{CT_{obj} - CT_{dry}}{CT_{wet} - CT_{dry}} \quad (6)$$

where CT_{obj} is the CT value of the image being processed.

Water-Oil Experiments

The oil phase employed is n-decane and its properties are shown in Table 1. For water/oil imbibition, the dry core is prepared and filled with water as above. Then oil is injected at a low rate using a liquid chromatography pump forcing water to drain. One experiment is conducted with no initial water for more direct comparison with the water/air case. The uptake of water by the core is again computed from the change in weight of the water reservoir. In the water/oil system, the rate of change of the saturation is slower than in the water/air case. Thus, CT scans are conducted at less frequent intervals and experiments take much longer to complete.

A slightly different approach is taken for analysis of the water-oil imbibition experiments because it is hard to obtain 100% oil and 100% water saturated reference images within the same experiment. For experiments with nonzero initial water saturation, the 100% water saturated image is used as the reference image in the following equation

$$S_o = \frac{CT_{obj} - CT_{wet}}{\phi(CT_{oil} - CT_{water})} \quad (7)$$

where ϕ is the independently measured porosity for a given voxel and oil is the CT number of the pure oil phase. However, for experiments with no initial water, the reference image is 100% oil saturated, and thus, the saturation calculation equation has the following form

$$S_w = \frac{CT_{obj} - CT_{sat}}{\phi(CT_{water} - CT_{oil})} \quad (8)$$

where the subscript sat refers to a fully oil saturated core. These equations are comparable and compatible with Eq. 6 in terms of accuracy, because the images obtained here are flat and exhibit minimum beam hardening and undetectable positioning effects.

Rock Sample Characterization

Diatomite samples were obtained from the Grefco Quarry (Lompoc, CA). They had no initial oil or water saturation. The rock is consolidated, almost pure white indicating little silt or mud, and some evidence of bedding planes is visible to the naked eye. The samples are from outcrop or near surface formations; hence, silica is in the amorphous or opal-A state. We characterize the rock with CT scanning to measure porosity, mercury porosimetry to establish pore-size distribution and drainage capillary pressure curves, and with scanning electron microscopy to view pore-scale features and roughness.

The largest scale at which we studied diatomite was a block roughly 30 cm by 30 cm by 12 cm. The dry block was placed in the CT scanner and the X-ray attenuation of vertical cross sections measured at roughly 2 cm intervals along the length of the block. The contrast in CT number across the sample was not large, only one hundred or so Hounsfields, indicating that material properties do not vary drastically. For most purposes, the rock sample was relatively homogeneous at this scale.

Next the rock was cut in a direction parallel to the bedding plane and shaped into cylindrical cores with diameters of 2.5 cm and lengths of 9.5 cm. The diatomite block could not be cored using conventional drilling and cutting methods because of its brittle and fragile nature. A piece of the rock is cut by band saw to approximate dimensions. Then it is shaped manually by fixing two circular 1 inch patterns to each end of the rough-cut core. These pieces are used as guides in the shaping process with a file. Final shaping is achieved with sandpaper. Cores are then potted into acrylic tubes using quick-set epoxy.

Dry CT scans of these cores were taken. Then the cores were saturated with water and porosity was calculated using Eq. 5. Figure 3 shows porosity maps of four such cores. The gray shading of Fig. 3 is the porosity scale and it has a notably narrow range. The average porosity of each of the samples used here was around 65-71%. The porosity images of cores 3 and 4 were then converted to pixel-based porosity distributions to demonstrate the variation in porosity.

Figure 4 presents the frequency of pixels with a given porosity. The standard deviation of each distribution is about 1% and the distributions have an apparent log-normal shape. Point to point porosity in each of these samples varies by roughly 4 to 6% about the mean.

The absolute air permeability of several samples was measured using a permeameter (Ruska Instrument Co. model 1101-801) at three different gas flow rates. The air permeability versus flow rate data was then used to find the absolute permeability of the samples after correcting for the Klinkenberg effect. The average permeability of the samples was 7.0 mD. For reference, Berea sandstone (Cleveland Quarries, Amherst, OH) is also listed. The measured permeability data is listed in Table 2.

Next, we performed mercury injection porosimetry to characterize pore-throat and pore-body size distributions and capillary entry pressure for rock samples. For this purpose, we used a porometer (Ruska Instrument Co. model 1051-801). The experiments were performed using the guidelines listed in the operating manual of the porometer and two runs were performed. The apparatus was vacuum evacuated before the start of each run. Then mercury was injected into the sample at different pressures and the pressure versus volume of mercury injected into the sample recorded. The maximum injection pressure possible with the porometer was 850 psig, and it was assumed that this was sufficient to saturate the sample with mercury.

From the mercury intrusion data, the dimensionless capillary pressure curve in Fig. 5a was generated. The capillary pressure data does not exhibit a plateau once the initial capillary entry pressure is overcome. This indicates that progressively smaller pore space is entered as the mercury injection pressure is increased and the rock contains a variety of pore sizes. The mercury intrusion data set was also used to generate pore-size distributions (Ritter and Drake 1945) resulting in the volume-based frequency of pore-throat sizes shown in Fig. 5b. It is true that this type of approach does not take into account the connectivity of the rock and is an oversimplification of the porous medium, but it still has the capacity to give us an idea about the size distribution. Since capillary entry pressure is controlled by pore throats, this method is more accurate for pore-throat size than it is for pore-body size. To guide the eyes, a log-normal curve is plotted in Fig. 5b. Different symbols denote two runs with different rock samples. Good reproducibility is found. The mean pore-throat size from this analysis is around 15 μm .

Next, we examined several different samples at the pore-level using scanning electron microscopy (SEM). The samples were sputtered with a palladium-gold (Pd-Au) coating to make the surfaces conducting. However, the thickness of the coating was quite thin such that all the surface features of the samples are retained. Figure 6 compares pore-level features of Berea sandstone and diatomite and partially explains the low permeability of diatomite relative to sandstone. At 100X magnification, pore structure, sand grains, and possible flow paths in the sandstone are evident in Fig. 6a. Sand grains are about 200 μm in size. To obtain the same level of detail about grain size in diatomite, a magnification of 1190X was required as shown in Fig. 6b. Regions of diatomite that could be acting as pore bodies, range in size from 50 μm to 10 μm . The features that might be acting as pore throats have size in the range of 5 μm . In both sandstone and diatomite samples, pore corners appear to be sharp indicating that both should imbibe water well.

Another observation in this SEM study is that for most part there are few complete intact diatoms. The size of diatom fragments is quite varied and the definition of pores from Fig. 6b would incorporate many fragments into a single pore. Consequently, the magnitude of pore-wall roughness is large relative to throat or body size leading to ample pathways in addition to

pore corners for wetting liquid during imbibition. By comparison, pore walls in the sandstone are formed by single grains and pore-wall roughness is confined to relatively small recessions on grain surfaces. Wetting liquid should flow primarily in pore corners and not along surface roughness.

The SEM analysis also revealed regions of inhomogeneity with characteristic dimension that greatly exceeds grain size. We identified one such inhomogeneity at a magnification level of 25X in Fig. 7a. The length of the void is around 2.5 mm. The same feature seen at a magnification of 500X in Fig. 7b shows that its width is around 20 μm and the depth may be of the same order. The depth of the feature reveals the highly layered nature of the rock. It is seen that in spite of this layered structure there is significant porosity within the layers. The frequency of such features encountered in these samples was not large.

Results, Water-Air Imbibition

First, we present and summarize weight gain as a function of time for water imbibition into dry cores and then the corresponding CT-images are given. Recall that cores are in a vertical position in these experiments. Data were converted into the weight of water imbibed as a function of time in order to measure the rate of change of water saturation in the core.

Figure 8 shows dimensionless weight gain, w_D , versus the dimensionless square root of time for the Berea sandstone imbibition experiment. Different sets of symbols indicate different experiments. Time is computed according to Eq. (2) and weight is normalized by the product of pore volume and water density. In this fashion, w_D is proportional to the pore volume of nonwetting phase ejected from the core. Discrepancy in the early time response is probably explained by the filter paper at the bottom of the core. If the filter paper does not saturate rapidly and uniformly at the beginning of the imbibition process ($t_d = 0$) imbibition is retarded initially. However, all experimental runs show a good reproducibility and later time response is linear when plotted versus the square root of time. Spontaneous imbibition ends when the water front reaches the end of the core. This time is practically the same for all three cases. These initial experiments with Berea sandstone confirmed that our experimental approach was adequate to collect the data required.

Imbibition in diatomite was considered next. Figure 8 also shows weight gain obtained for the diatomite during two different experiments. Here, there is only slight deviation from linearity in the initial stages of imbibition. For later times, the response is linear with respect to the square root of time. Interestingly, the water imbibition process is quite rapid despite the low permeability of diatomite. These diatomite samples imbibe water at rates that rival sandstone in an absolute sense and exceed sandstone in a nondimensional sense.

CT images were also taken during the air/water imbibition process and the raw CT data were converted into maps of aqueous phase saturation as a function of time using Eq. (6). Strong capillary forces are evident in the saturation images for imbibition in diatomite displayed in Fig. 9. In the images, black indicates no water and white indicates fully water saturated. Nondimensional time is given below each image. The first saturation map shows that water enters the core in the center, but it spreads laterally quite rapidly. The water front is sharp and the displacement is piston like. As the front progresses down the core it diffuses somewhat and it reaches the end of the core at about $t_D = 25$ (3600 s). These saturation images correspond to the lower diatomite curve in Fig. 8 that deviated from linearity at short time. Figure 9 shows

that the saturation front was not piston like until roughly $t_D = 0.62$. Hence, the deviation from linearity results from the non-uniform displacement front.

A comparison of the absolute time necessary for spontaneous imbibition to be completed in each type of rock, confirms the weight gain results. Diatomite takes about three times longer than Berea sandstone. It is important to recall that the pore volume of the diatomite is about five times the pore volume of the Berea sandstone, whereas its permeability is more than an order of magnitude less.

Results, Water-Oil Imbibition

Four experiments were conducted using the diatomite cores shown in Fig. 3. One experiment was conducted without initial water saturation, whereas the others were conducted by starting at an irreducible water saturation. In these experiments, the core is placed in a horizontal position and oil recovery is by capillary imbibition forces only.

Similar to the water-air experiments, the data collected during the experiments consisted of weight gain measurements and CT scans. In general, the duration of the experiments was much longer compared to water-air imbibition experiments. This is expected because the water-n-decane interfacial tension (51.4 mN/m) is lower than the water-air (72 mN/m) interfacial tension; oil is more viscous than air and much harder to displace. Further, with initial water saturation, the imbibition capillary forces should be less. Figure 10 gives the weight gain as a function of the square root of time.

For the zero initial water saturation case, an initial, short spherical flow period followed by a strong, relatively sharp displacement front similar to the fronts observed in water-air imbibition experiments was noticed, as shown in the saturation profiles of Fig. 11. Dark shading corresponds to low water saturation. A stabilized zone was also present with an extent that is somewhat larger than the zone identified in the water-air imbibition experiments. The displacement was not as effective as the water-air case. Note that the water saturation at dimensionless time equal to 500 is around 73% compared to an almost 95% water saturation for water-air imbibition.

For experiments with a nonzero initial water saturation, a sharp, clear front is not observed as illustrated by the CT-derived aqueous-phase saturation profiles shown in Fig. 12. Here, water is introduced on the right side of the images, but water saturation increases almost uniformly throughout the entire core. In this sense, the water-oil imbibition results are diffuse as compared to the water-air case. The oil saturation reached at the end of spontaneous imbibition experiments varied between 20% to 27% as reported in Table 3. This corresponds to oil recoveries between 36% to 73% of the original oil in the cores. These figures are in agreement with other data for non-fractured diatomite core plugs (Wendel et al. 1988). Figure 10 presents dimensionless imbibition performance curves for all water-oil experiments. The shape of curves for runs with initial water saturation are all similar, within experimental error, indicating that a single set of relative permeability and capillary pressure curves can represent spontaneous imbibition for these experiments. Note, that the core without initial water saturation displays significantly better imbibition performance.

A comparative water-n-decane imbibition experiment was conducted using the same Berea sandstone core as in the water-air imbibition experiment. Again, the core is positioned horizontally and weight loss and CT images were recorded throughout the experiment. Figure

13 presents n-decane saturation images obtained at various times. Similar to the experiments in diatomite, water imbibes into the core without the development of a front despite different initial water saturation and residual oil saturation values as given in Table 3. The dimensionless weight gain curve for the sandstone is plotted in Fig. 10. It shows that the imbibition performance of the sandstone is somewhat below diatomite because the sandstone curve lies to the right of the diatomite curves.

Discussion

The simple water displacing air experiments provide some insight into the pore-filling processes and the capillary pressure characteristics of diatomite, whereas the water displacing oil experiments illustrate the difficulty in oil recovery from diatomite. Taken together, the contrast in results between water-air and water-oil systems provides insight into flow processes in diatomite.

The simple water displacing air experiments teach us about the imbibition capillary pressure characteristics of diatomite. In Eq. (1), the rate of imbibition is proportional to the square root of $P_c k S_w$. Denote this quantity as the imbibition potential, IP , and take the ratio of diatomite IP to sandstone IP :

$$\frac{IP_d}{IP_s} = \frac{(P_c k S_w)_d}{(P_c k S_w)_s} \quad (9)$$

where the subscripts d and s refer to diatomite and sandstone, respectively. Next, we take average values of the slope of weight gain versus the square root of time in Fig. 8 and dimensionalize the results. The ratio given in Eq. (9) is 2.4 indicating the strong tendency of diatomite to imbibe water. From Fig. 9, we judge that the water saturation upstream of the saturation front is about 1. So, the ratio of saturations in Eq. (9) is about 1. Next, we replace P_c with the appropriate Leverett J-function (Leverett 1941):

$$P_c = \sigma \left(\frac{\phi}{k} \right)^{1/2} J(S_w) \quad (10)$$

for water wet rocks. In Eq. (10), σ is the air-water interfacial tension and $J(S_w)$ is the Leverett J-function. Upon some rearrangement

$$\frac{IP_d}{IP_s} = \left(\frac{k_d}{k_s} \right)^{1/2} \left(\frac{\phi_d}{\phi_s} \right)^{1/2} \left(\frac{J_d(S_w)}{J_s(S_w)} \right) > 1 \quad (11)$$

The first term in parentheses on the right is clearly less than 1 because diatomite is less permeable than sandstone. With typical sandstone (100 to 1000 md) and diatomite (0.1 to 10 md) permeabilities, k_d/k_s might range from 0.1 to 10^{-4} . Table 2 shows that the diatomite samples are roughly 4 times as porous as the sandstone. For Eq. (11) to be greater than 1, the ratio of Leverett J functions for diatomite and sand, J_d/J_s , must be greater than 1. This explains, in part, why the relatively impermeable diatomite imbibes strongly when the rock is initially filled with air. Likewise, it explains why diatomite recoveries lay to the left of the sandstone curves in Fig. 8.

In Figure 10, dimensionless water-oil imbibition results for diatomite also lay to the left of the curve for sandstone. These results again indicate that the Leverett J function for diatomite exceeds that of sandstone. Thus, employing Leverett J-functions for sands or sandstones to diatomite and merely rescaling by $(\phi/k)^{1/2}$ to estimate capillary pressure results in significant error for capillary dominated recovery.

Another interesting aspect of the water-air experiments is the very low trapped gas saturation. For instance, Fig. 9 at $t_D = 25$ (3600 s) shows water saturation in excess of 95% shortly after breakthrough. The strong capillary forces and the small pore throat to body aspect ratio of diatomite suggests much snap-off and trapped gas. However, for snap-off to occur, pore corners and crevices must fill with wetting liquid and sufficient liquid for snap-off must accumulate at pore throats before the pore is filled completely by the advancing imbibition front. We speculate that trapped gas saturation is low because the advancing front fills pores with water at least as rapidly as pore corners fill with water. Indeed, recent pore-level network modeling of imbibition shows that in the absence of flow in pore corners the displacement pattern is a flat front with little or no trapping of the nonwetting phase (Blunt and Scher 1995). In dynamic models of imbibition, snap-off is suppressed as the capillary number increases (Mogensen and Stenby, 1998). Pores fill with wetting liquid by frontal advance in less time than it takes for corners to swell with water and for snap-off to occur. Additionally, the low residual gas saturation indicates that the frequency of dead-end pores is not high.

During spontaneous imbibition into air- or oil-filled diatomite where the initial oil saturation is zero, pores of all sizes fill simultaneously as indicated by the CT-derived water saturation images. The size of the frontal zone is small and the fraction of residual nonwetting phase recovered large as compared to sandstone. In turn, this implies that large pores are well connected to small pores and significant flow pathways for wetting fluid exist despite low absolute permeability. On the other hand, inhomogeneous regions, such as those illustrated in Fig. 7, do not appear to play a significant role in the flow of imbibing water. This suggests that under conditions of spontaneous imbibition in diatomite such regions are not well connected and conducting of wetting fluid.

Pore shape and pore-level roughness are also important in determining spontaneous imbibition characteristics. A significantly reduced capacity to imbibe would be found if pores were smoother and had circular cross-section as opposed to angular and rough cross sections (Milter and Øxnevad 1996). Thus, the comparison of pore structure in Fig. 6 explains partially why diatomite imbibes, in a dimensionless sense, more rapidly than sandstone. These observations are consistent with the complex, small-diameter pore network of diatomite noted above and elsewhere (Fassihi et al. 1982). It is hard to fill the very small pores selectively, leaving relatively large pores unfilled.

The oil-water imbibition results in Fig. 10 display a square-root of time character, but it is not as strong as that found for the air-water results. In all water-oil cases, the initial response is nonlinear with respect to the square root of time over a significant period. Oil is dense and viscous compared to air and resists being set into motion by the imbibing water; thus, the rate of weight gain is slow initially until capillarity dominates the displacement. Examining the curve for diatomite without initial water, the slope of the w_D versus $t_D^{1/2}$ curve steadily increases until becoming linear at approximately $t_D^{1/2} = 9$. Thereafter, linearity with respect to $t_D^{1/2}$ is maintained. The weight gain curves for cases with initial water appear to display three flow regimes. For diatomite with initial water saturation, the initial period lasts up to about $t_D^{1/2}$ equal to 9, whereas for sandstone the duration is to $t_D^{1/2}$ equal to 15. In both diatomite and

sandstone, a sensibly linear region follows that corresponds to capillarity dominated imbibition. Next, beginning at about $t_D^{1/2}$ equal to 21 for diatomite and $t_D^{1/2}$ equal to 22 for sandstone, the rate of weight gain decreases signaling a slow approach to final saturation distributions.

Table 3 summarizes results for water-oil imbibition and displays an expected result for water-wet rocks. When the core initially contains water, the larger the initial water saturation, as in case 4 for diatomite, the larger the residual oil saturation and the less the ultimate recovery of the initial oil in place. With increased initial water saturation comes lesser imbibition forces, and the increased likelihood of snap-off and creation of residual oil. Despite low permeability, these diatomite samples imbibe water prolifically because the pore structure and pore roughness provide many water pathways.

Future Work

We believe that the imbibition cell described here provides us with an important new tool for exploring fluid flow in low-permeability porous media during imbibition. A portion of our future work will concentrate on understanding how fluid flow in reservoir samples of diatomite differs from the results presented here for outcrop rock. Reservoir rock can contain significantly more clay and adsorbed organic material that possibly alters imbibition performance. Importantly, we are pursuing a matching procedure to extract capillary pressure and relative permeability information from spontaneous imbibition experiments.

Conclusions

An experimental apparatus and method to collect data and CT images during spontaneous imbibition of water into low-permeability porous media was designed and tested. The novel coreholder permits imaging of the entire length of a core with a single scan incurring little or no CT artifacts. Several tests were performed on Berea sandstone and diatomite samples machined from a block of outcrop material. These tests confirmed the repeatability of the method.

CT imaging of the imbibition process permits observation of the advance of the water front into the cores and explains the observed trends in weight gain as a function of time. Porous media with little initial water saturation show a homogeneous and piston-like water front during the imbibition process, whereas at large initial water saturation no discernable front is found. In impermeable diatomite, capillary forces result in a strong imbibition potential for water such that imbibition rates rival, in an absolute sense, and surpass, in a dimensionless sense, those for sandstone.

From spontaneous imbibition tests, we conclude that although the pore structure of diatomite is complex, flow pathways are well connected as evidenced by rapid imbibition of water into air-filled pores and low residual phase saturations following imbibition. Secondly, the Leverett J-function for diatomite exceeds that for sands and sandstones. Employing Leverett J-functions for sands or sandstones to diatomite and merely rescaling by appropriate values of porosity and permeability results in significant error. Finally, for water-oil imbibition where the initial water saturation is nonzero, oil production scales with the square root of time over a relatively narrow range of time, as compared to water-air imbibition.

Nomenclature

A = cross-sectional area
 CT = CT number
 $J(S_w)$ = Leverett J-function, dimensionless
 k = permeability
 L_c = characteristic length
 m = mass of water imbibed
 ρ_w = water density
 S_w = water saturation
 P_c = capillary pressure
 t = time
 ϕ = porosity, dimensionless
 σ = interfacial tension
 μ = viscosity
 V = volume
 w = weight gain

Subscripts

air = CT value of air phase
b = bulk
d = diatomite or dimensionless
dry = CT value for the dry core
nw = nonwetting
obj = CT value of the image being processed
oil = CT value of oil phase
s = sandstone
sat = CT value for a fully oil-saturated core
w = wetting
water = CT value of water phase
wet = CT value for a fully water-saturated core

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Figure 1: Schematic of the imbibition cell and experimental setup.

Figure 2: Exploded view of the imbibition cell.

Figure 3. Porosity images of diatomite cores 1 (top) through 4 (bottom).

(a)

(b)

Figures 4(a) and 4(b).

(a)

(b)

Figure 5: (a) Drainage dimensionless capillary pressure curve for diatomite and (b) normalized pore-size distribution from mercury injection.

(a)

(b)

Figures 6(a) and 6(b).

(a)

(b)

Figures 7(a) and 7(b).

Figure 8: Dimensionless weight gain versus the square root of dimensionless time, water-air imbibition.

Figure 9.

Figure 10: Dimensionless weight gain versus dimensionless square root of time, water-oil imbibition.

water
saturation

1

0

Figure 11: Water saturation distribution along diatomite during water imbibition into an oil-filled core.
Images correspond to dimensionless times of 0 (top), 5, 50, 100, 500 (bottom).

Figure 12: Oil saturation distribution along diatomite during water imbibition into a partially water saturated core.
Images correspond to dimensionless times of 0 (top), 5, 50, 500, 5000 (bottom).

Figure 13: Oil saturation distribution along Berea sandstone during water imbibition into a partially water saturated core.

Images correspond to dimensionless times of 5(top), 50, 500, 5000.

1.5 STATISTICAL NETWORK THEORY OF SILICA DEPOSITION AND DISSOLUTION IN DIATOMITE

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Statistical Network Theory of Silica Deposition and Dissolution in Diatomite

by

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ABSTRACT

One area of concern accompanying steam injection into siliceous diatomite is the evolution of permeability and porosity as silica dissolves in hot steam condensate and is redistributed on rock surfaces as condensate cools. We employ a network model to gauge how evolving pore topology affects permeability and porosity. The model is anchored to real rock through measured pore-throat and pore-body size distributions. Macroscopic trends are correlated in terms of network connectivity and pore body to throat aspect ratio.

When porosity change from 0 to 10% is examined closely, it is found that permeability decrease correlates with porosity decrease according to a power-law expression with power-law exponents in the range of 8 to 9. Such a power-law dependence has been measured previously by experiment¹. These values confirm the strong dependence of permeability on deposited silica. The dissolution process displays an equally strong dependence on porosity change, and again, permeability increases most dramatically for dissolution in networks with small values of connectivity and large values of the pore-body to pore-throat aspect ratio. Power-law expressions adequately describe permeability increase as a function of porosity for a moderate range of dissolution.

INTRODUCTION

Steam drive with careful control of injection pressure can be an efficient technique for heavy and light oil recovery from tight, siliceous reservoirs such as the South Belridge and Cymric diatomite (Kern Co, CA).²⁻⁷ These reservoirs have permeability ranging from 0.01 md to 10 md, and porosity that ranges from 38 to 65%^{8, 9}. The low permeability results from extremely small-sized pore throats usually, 1 to 10 μm in diameter.

Steam injection into diatomite is not without complications. One factor is the solubility of diatomite in hot water. Diatomite is a hydrous form of silica or opal composed of microscopic shells of diatoms which are the remains of single-celled microscopic aquatic plankton. During steam injection, diatomite, which is principally silica (SiO_2) containing very little clay or silt,⁸ can dissolve in hot aqueous condensate. This dissolved silica is carried forward by the flowing condensate. As the condensate travels away from the injector, it cools and the dissolved silica precipitates. Experiments¹ with injection of hot, silica-laden water into cooled cores suggest that silica deposition in diatomite leads to plugging. Later, the deposited silica may again dissolve if the local temperature increases. This process causes permeability redistribution within the reservoir, which is already relatively impermeable. Similarly, silica dissolution might increase permeability around the injection wells, thereby improving injectivity.

Silica deposition could also be a concern for waterflood operations. In the San Joaquin Valley, diatomaceous reservoirs underlie permeable sands containing heavy oil.⁸ This heavy oil is recovered by steam injection. Produced hot condensate from the heavy-oil recovery operations could be reinjected in the diatomites. These waters are frequently laden with silica that might precipitate when contacting the cooler diatomaceous reservoirs.

Prior work on the effects of dissolution and precipitation of solids within porous media, as they relate to the recovery of oil, has focused mainly on acidizing sandstones and carbonates.¹⁰ Here, a quantity of acid is injected into the porous medium with the intent of dissolving a portion of the rock matrix around a well, thereby improving the flow of oil into the well. In early studies,^{11,12} the porous medium was modeled as a bundle of capillary tubes and the change in the pore-size distribution computed as a function of time. It was predicted that the dissolution process takes place almost uniformly, and the permeability ratio (current upon initial permeability) increases with the square of the porosity ratio.¹¹

On the other hand, acidizing carbonate rocks results in selective dissolution and the formation of channels or "wormholes". To understand wormhole growth and the onset of a rapid increase in permeability, Hoefner and Fogler¹³ used a regular two-dimensional network with a representative distribution of pore sizes to simulate flow, reaction, and dissolution of the solid matrix. As the dissolution rate relative to the advection rate (*i.e.*, the Damkohler number) decreased, channeling became more prevalent. Experiments in 4-inch-long carbonate cores provided qualitative confirmation of network predictions.

Daccord *et al.*¹⁴ presented a "phase diagram" for classifying the different reactive flow regimes by defining and plotting a dimensionless kinetic number versus Peclet number (*i.e.*, the characteristic time for diffusion upon that for convection). Forced convection, diffusion, and surface reaction were the only phenomena considered. For low values of the Peclet number, they found that processes are limited by convection of reactants, not reaction rates.

More recently, Bekri *et al.*¹⁵ proposed a scheme whereby flow is computed in a given fixed geometry (*e.g.*, a plane channel with a grooved wall or cubic arrays of spheres). Then convective-diffusive equations are solved in order to obtain the flux of dissolved material and the evolution of the fluid-solid interface. For the channel geometries examined, normalized permeability versus porosity plots display power-law relationships with power-law exponents of roughly 3. Over most parameter ranges, these exponents were independent of transport and kinetic rates. Permeability evolved in the arrays of cubic spheres in a semi-log fashion. In the case where both the convection and dissolution rates are large, dissolution on the upstream side of the sphere is computed to be more rapid than on the downstream portion. This leads sphere shapes to evolve asymmetrically. For reaction-limited cases, dissolution of the solid surface occurs uniformly. When convection is dominant and convection rates large, wormholing occurs along the preferential paths of the convecting fluid. Investigations on a Menger sponge showed that similar conclusions apply to more complex geometries.

For the flow and capture of particulates from suspension, and the ensuing permeability decrease of a porous medium, Sharma and Yortsos¹⁶ propose an effective medium theory incorporating population balances for the various size particles and pore throats. Particulates such as clays or emulsions may have dimensions greater or smaller than pore dimensions. Particles that exceed pore throat dimensions are trapped by the pore throat, while particles that are significantly smaller than pore throats deposit uniformly over pore bodies. They illustrate the model using deep-bed filtration and fines migration examples.^{17,18} As expected, pore-size distribution and coordination number are important parameters for the theory.

Motivated by deposition and clogging of porous media by salts precipitated during the reinjection of water into geothermal fields, Salles *et al.*¹⁹ studied pore geometries and

configurations similar to Ref. 15 using a random-walk method. Note that the work of Salles *et al.* preceded that of Bekri *et al.* For arrays of cubic spheres, permeability versus porosity decreased in a roughly semi-logarithmic fashion as solids deposited. For low flow rates, deposits are equally distributed on upstream and downstream sides of the sphere. At very large flow rates, most deposition occurs on the upstream side of the sphere.

Grain consolidation and cementing is also a deposition process, and permeability and porosity decrease accordingly. Such diagenetic processes have been examined using percolation models. For example, Wong *et al.*²⁰ employed two- and three-dimensional networks of pore throats and bodies. Throats were modeled as cylindrical tubes. At random, a throat was selected and its radius reduced by a constant factor and the new permeability (network conductivity) and porosity calculated. Then, another throat was selected at random and the process repeated. Network permeability was calculated to decrease with porosity in a power-law fashion. The power-law exponent depends on the dimensionality of the network and the factor used to reduce throat radii. In two dimensions the power-law exponent ranged from roughly 2.6 to 5.4, whereas in three dimensions it varied between 2.5 and 4.7. They also concluded that network permeability depends on the distribution of tube radii. These results were in qualitative agreement with measurements using fused glass beads. Also in work related to grain consolidation, single-phase permeability as a function of the degree of cementation of North Sea sandstones was predicted using a physically based modification of the Carmen-Kozeny equation²¹. It was found that pore-bridging clays more strongly reduce permeability than do pore-lining or pore-filling clays. A detailed understanding of pore-size distribution and its variation and skewness are necessary to the theory.

Permeability-damage experiments in diatomite have also displayed a power-law dependence for reduction in permeability as the initial porosity fills with deposited solids. However, for diatomite, the experimental data fit a power-law with an exponent of 9^{-1} . This is much larger than any power-law exponent reported above. The small grain size and small pore throat sizes, as evidenced by low permeability, apparently require only a small amount of solid to be precipitated to reduce effectively fluid mobility. The dissolution process was not studied, but, presumably, the increase in permeability as diatomite dissolves could be similarly drastic. In closing this introductory review, we note that there is much additional work on chemical dissolution, migration of fines, flow of stable emulsions in porous media, and plugging of catalyst pellets. Sahimi *et al.*²² provide an extensive review up to 1990.

To interpret changes in permeability and porosity of low-permeability siliceous rocks undergoing diagenesis and to explain the large experimentally determined power-law dependence, we develop a simple network model and incorporate evolving pore shapes. The porous medium is represented by a network of known properties, and percolation theory used to predict the effects of dissolution and deposition on permeability and porosity of the porous medium. We consider only the flow of liquid water. Changes in permeability are correlated with changes in conductivity of the network, and the amount of silica deposited is related to available porosity of the lattice at any stage of deposition or dissolution. Our networks are anchored to real porous media through pore-throat and pore-body distributions as well as pore body to throat aspect ratios garnered from outcrop diatomite samples. Before proceeding to describe the network model and the determination of model parameters, it is useful to establish briefly the relative ordering of reaction and transport times for silica in diatomite.

REACTION AND TRANSPORT SCALING

We are most interested in the evolution of permeability and porosity relatively deep in the formation and not the near wellbore region. For representative flow, reaction, and diffusion rates, a local equilibrium assumption for silica is appropriate under expected conditions. To achieve local equilibrium, the fluid convection rate must be slow and reaction rates large enough that the aqueous phase at a given temperature saturates with dissolved silica. The Damkohler number, Da , the ratio of characteristic fluid convection to reaction times, resulting from a one-dimensional silica mass balance with constant aqueous phase velocity, and first-order reaction kinetics is

$$Da = \frac{l\kappa}{u} \quad (1)$$

where l is a characteristic length, u is the interstitial velocity, and κ is the first-order rate constant.²³ The transition from kinetic-limited to convection-limited cases occurs when Da is of order 1. When Da is 10, local equilibrium is achieved.²⁴ For the case of interest, the reservoirs are shallow, impermeable, and mechanically weak; hence, injection rates are low.^{6, 7} The magnitude of u is expected to vary between 1 E-6 and 1 E-5 m/s (0.1 to 1 m/d). For deposition in diatomite, Koh *et al.*¹ report rate constants ranging between 8.6 E-5 and 1.7E-3 s⁻¹ for temperatures of 40 °C and 200 °C, respectively. Using Eq. (1), these numerical values, and a characteristic length of 1 m, Da ranges from order 10 to 1000, indicating that local equilibrium should be achieved in distances less than 1 m from the injection point. In a more extensive study that examines both dissolution and deposition of silica via a coupled advection and reaction model employing a variety of reaction-rate expressions, it is predicted that local chemical equilibrium is achieved in as little as one day at a distance of 1.5 m from the liquid injection point.²³

The dimensionless group to assess whether equilibration is retarded by mass-transfer resistance between dissolved silica at a pore wall and bulk fluid flowing through the pore space is the Sherwood number,²⁵

$$Sh = \frac{Kl_{MT}^2}{D} \quad (2)$$

where K is a mass transfer coefficient and D is the diffusivity. Hence, the characteristic time for diffusion is l_{MT}^2/D , whereas the characteristic time for mass-transfer is $1/K$. Also note that the proper characteristic length for this mass-transfer problem, l_{MT} , is the reciprocal of interfacial area per unit volume, A_v . Local equilibrium is maintained so long as the characteristic time for advection, l/u , is much greater than the characteristic times for mass-transfer or diffusion.

The problem of silica mass-transfer resistance in diatomite is unstudied. However, correlations developed for mass-transfer resistance in packed beds with porosities of 40 to 45% vary from 1 for stagnant liquid to 1000 for large flow rates.²⁶ The magnitude of the diffusivity for solutes in liquid solvents^{25,26} is roughly 1 E-9 m²/s, while BET isotherms²⁷ of diatomite surface area measure A_v to be about 180 m²/m³. These values put $1/K$ between 3.1 E+4 and 31 s for Sh equal to 1 and 1000, respectively. Similarly, the characteristic time for diffusion is 3.1 E+4 s, whereas the characteristic time for convection lies between 1 E+5 and 1 E+6 s for the values of u given above and the l of 1 m identified by the Damkohler number analysis. In short, convection times are at least an order of magnitude longer than the characteristic times for diffusion and mass transfer. Local equilibrium of silica in aqueous solution should prevail inside

the reservoir. These conclusions are supported by related studies of chemical reaction in porous media. For instance, deZabala *et al.*²⁸ developed a complex nonequilibrium theory that accounts for kinetic and mass-transfer resistances during alkaline floods. They found that even in laboratory scale corefloods at high flow velocities, local equilibrium prevailed.

The relative rates of heat transfer also need to be examined to decide the conditions for local equilibrium. An ordering analysis conducted elsewhere of the reservoir heating during a steam-injection pilot found that thermal diffusivity (system thermal conductivity upon system volumetric heat capacity) is at least an order of magnitude less than hydraulic diffusivity (total phase mobility upon system compressibility).^{6,7} These results indicate that the primary macroscopic mode for heat transfer during steam injection within diatomite reservoir layers is convection as opposed to conduction. Conduction remains important for heat transfer between swept and unswept volumes. As shown above, the characteristic time for reaction was substantially less than that for convection. It follows that reaction rates are sufficiently fast to equilibrate the silica concentration in solution as the aqueous-phase temperature changes.

Further, Yortsos and Satik²⁹ developed a pore network model for steam injection in porous media. Separate but coupled heat balances were solved for the solid matrix and fluids inside the pore space. For low injection rates and representative values of a dimensionless heat transfer-condensation group, they found that the solid is at or very near to thermal equilibrium with liquid in the network pore space.

The picture that emerges for silica dissolution/deposition in diatomite is that a condition of local chemical and thermal equilibrium prevails in the formation away from injection regions. The silica solution concentration is thus set by the temperature and pH of the solution. Dissolution or deposition takes place relatively uniformly. This is supported by previous observations of uniform permeability change in diatomite with respect to time¹ as opposed to sudden large increases that occur during the formation of wormholes or sudden large decreases when plugging occurs simultaneously in pores of all sizes. Because flow rates are low and ample time exists for reaction and diffusion, it appears unlikely that asymmetric evolution of pore shape on either side of a pore throat or constriction happens to an appreciable extent.

NETWORK APPROXIMATION

This section describes our network approximation of porous media, the rationale for our choice, and how pore throats and bodies are represented. Next, incorporation of silica dissolution and precipitation into first a single-pore model and then the network is discussed. A Bethe network is chosen to model the pore space and used to generate permeability versus porosity relationships for silica redistribution in tight siliceous rocks. It is an endlessly branching network characterized by the coordination number, Z , and distribution of pore sizes. An advantage of the Bethe lattice is that analytical solutions are available for network conductivity³⁰. Bethe lattices have been used to understand two- and three-phase relative permeabilities in porous media,³¹⁻³⁴ among other applications.

Despite its lack of reconnections and the observation that an endlessly branching network bears little apparent resemblance to three-dimensional pore structure, a Bethe network can approximate some properties of a porous rock. The conductivity functions for a Bethe network will be described in some detail shortly. First, however, consider typical network conductivity results for Bethe networks and a cubic lattice. Figure 1 presents normalized network

conductivity (conductivity is analogous to permeability) versus the open fraction of throats (*i.e.*, bonds) for Bethe networks with Z equal to 4, 5, and 6 and a simple cubic network where Z equals 6. Cubic lattice results are from Ref. 35. The solid lines give the Bethe network result, while the symbols are for various sized cubic networks. Figure 1 demonstrates that the Bethe network for Z equal to 5 is an excellent estimator of the conductivity of this three-dimensional network. At low values of f , the Bethe network ($Z = 5$) does underestimate conductivity slightly. This results from the lack of reconnections in the Bethe network. In the cubic lattice, some loops or reconnections in the flow field are maintained even though the fraction of open throats becomes small and approaches the percolation threshold. Hence, the simple cubic network exhibits slightly greater conductivity as it approaches the percolation threshold.

Moreover, the Bethe network results for conductivity when Z equals 4 and 6 decrease and increase, respectively, relative to the cubic lattice results as expected. For a fixed value of f , greater connectivity leads to greater conductivity, and vice versa. These observations are the reason that Bethe lattices with coordination numbers from 4 to 6 were originally proposed as an approximation to rock pore space.^{33,36,37} The Bethe network is a convenient tool for the problem here of estimating the increase or decrease of the single-phase permeability of a porous rock provided that the fraction of open throats does not approach too closely the percolation threshold. Although networks more complicated than Bethe lattices are certainly possible, we have little knowledge of diatomite pore structure beyond pore-size distribution and some notion of grain size. Thus, a more complicated network is unwarranted at this stage.

Each pore within the network has some conductivity, g , and network conductivity is governed by a distribution $G(g)$. The effective conductivity G_{eff} for such a network is given by the first derivative of a generating function $C(x)$ at the origin,

$$G_{eff} = -ZC'(0) \quad (3)$$

where Z is the coordination number of the Bethe lattice and the prime on C indicates differentiation with respect to the variable x which parameterizes the network³⁰.

The Laplace transform of the generating function satisfies the following recursive integral equation:

$$\int_0^\infty e^{-tx} C(x) dx = \frac{1-f}{t} + f \int_0^\infty G(g) \left\{ \frac{1+t}{g} + \frac{g^2}{(t+g)^2} \int_0^\infty \exp\left(\frac{x-g}{g+t}\right) C(x)^{Z-1} dx \right\} dg \quad (4)$$

subject to the boundary condition, $C(0) = 1$. For single-phase flow and in the absence of silica diagenesis, all the throats of the network are available and f equals 1.0. This remains true for dissolution. But in the case of deposition, some throats will clog, and hence f decreases. Various series approximations to Eq. (4) have been developed, and we use those of Heiba *et al.*^{33, 37}

Equation (4) requires a functional form for $G(g)$ that represents correctly the porous medium under investigation. Pores are characterized by a dimensionless throat radius r_t and their frequency of occurrence $\alpha(r_t)$. Throat radius, r_t , equals R_t / R_c , where R_c is a universal characteristic dimension for the pore-size distribution. Since f is the fraction of pores that are

available for flow, $G(g)$ can be represented, as³⁸

$$G(g) = (X_i + \frac{1 - X_A}{f})\delta(g) + \frac{X_A}{1 - X_i}\alpha(r_t)\frac{dr_t}{dg} \quad (5)$$

where $g(r_t)$ relates the size, r_t , of a pore to its individual conductivity. The Dirac delta function, δ , indicates that the clogged and inaccessible pores do not contribute to conductivity. The accessibility of the network, X_A , accounts for the random distribution of throats in the network that causes some nonclogged pores to be unavailable for flow because they are cut off from the main mass of open pores by nonclogged pores. The value of X_i depends on the fraction of throats that are clogged and the connectivity, Z , of the network. The term X_i defines the fraction of the original throats that are not allowed to flow because they have been clogged during deposition. For dissolution and single-phase flow, X_A equals 1 and X_i equals 0.

For a Bethe lattice, the accessibility function of the network has been found analytically³⁹. It is related to the fraction of open throats by

$$X_A(f) = \begin{cases} f(1 - (f^*/f)^{\frac{2Z-2}{Z-2}}) & ; f \geq f_p \\ 0 & ; f < f_p \end{cases} \quad (6)$$

where f_p is the percolation threshold, *i.e.*, the minimum value of the allowable fraction of throats below which no flow across the lattice occurs and f^* is the root of the following equation:

$$f^*(1 - f^*)^{Z-2} - f(1 - f)^{Z-2} = 0 \quad (7)$$

such that the root $f^* \rightarrow 0$ as $f \rightarrow 0$ and $f \rightarrow 1$. In our calculations to follow, we use the series approximation to Eq. (7) provided by Larson and Davis³⁶ as illustrated by Heiba³⁷.

The effective pore volume of the network is

$$V_{eff} = \int_0^\infty \alpha(r_t)v(r_t)dr_t \quad (8)$$

where v is the volume of an individual pore. Evolving pore volume is described shortly.

The network model equations, Eqs. (3) to (8), are anchored to a real porous medium by a two-step process. (i) The form of $\alpha(r_t)$ is based on measured pore-size distributions for diatomite. Likewise, actual body-size distribution information is included. (ii) The functional forms of $g(r_t)$ and $v(r_t)$ are developed analytically, using a single-pore model suggestive of observed pore structure. These steps are described in detail in the following sections.

POROUS MEDIUM CHARACTERIZATION

A real porous medium is composed of pores of varying size with each pore characterized by a large main body and a narrow throat. In turn, the porous medium is characterized by the sizes of bodies and throats and their distributions. For the size distributions, we use a bivariate gamma function,

$$\alpha = \frac{r^{a-1}e^{-br}b^a}{\Gamma(a)} \quad (9)$$

where a and b are parameters that characterize the distribution. The use of this function for diatomite body and throat distributions is supported by experimental results and scanning electron microscope (SEM) images⁴⁰. Based on Eq. (9), we generate body- and throat-size distributions as shown in Fig. 2 for diatomite. These distributions closely resemble our experimental results⁴⁰ where the average throat radius is 6 μm and the body radius is about 40 μm , and it follows that the average body to throat aspect ratio is 20/3. While developing our single-pore model to follow, we also use the distribution $\alpha(r_t, r_b)$ where r_b equals R_b/R_c , to average over all possible body sizes that can be associated with a given throat size. This is the constrained pore-size distribution, also shown in Fig. 2 as a dashed line marked with solid circles. Physically, a throat must be connected to a body that is greater than or equal to throat size:

$$\alpha_b(r_t, r_b) = \begin{cases} 0 & ; r_b < r_t \\ \alpha_b(r_b) / \int_{r_t}^{\infty} \alpha_b(r_b) dr_b & ; r_b \geq r_t \end{cases} \quad (10)$$

The properties we calculate from the network model are effective hydraulic conductivity and available network volume, which are related to permeability and porosity, respectively. The dimensionless effective hydraulic conductivity of a porous medium is inversely proportional to pressure drop, $|\Delta p|$, and scales as

$$G_{eff} \propto \frac{q}{A\phi} \frac{L\mu}{|\Delta p|R_c^2} \quad (11)$$

where $q/A\phi$ is the interstitial fluid velocity, μ is the fluid viscosity, and L is the length over which $|\Delta p|$ is measured. Substituting Darcy's law for single-phase flow for q/A in Eq. (11) and solving for permeability, we obtain

$$k \propto \phi R_c^2 G_{eff} \quad (12)$$

where k is permeability and porosity is proportional to available network volume. Thus, the permeability change at any stage based on these equations is related to original permeability and pore volume, k_0 and $V_{eff,0}$ by the following relation:

$$\frac{k}{k_0} = \frac{V_{eff}}{V_{eff,0}} \left(\frac{R_c}{R_{c,0}} \right)^2 \frac{G_{eff}}{G_{eff,0}} \quad (13)$$

Similarly, porosity at any stage of diagenesis can be related to original porosity and network parameters as

$$\frac{\phi}{\phi_0} = \frac{V_{eff}}{V_{eff,0}} \quad (14)$$

Equations (13) and (14) relate network parameters to the porous medium at any stage in the diagenetic process. Thus, given the mass of silica dissolved or precipitated and a model for how that mass distributes inside pores, we obtain k and ϕ .

SINGLE-PORE MODEL

To proceed, pore conductivity and volume need to be incorporated into the network. Every pore in a porous medium has at least two distinct parameters, throat size and body size. We use sinusoidal pores as shown in Fig. 3a that are constricted according to the following dimensionless equation:

$$\lambda(z) = 1 - \left(\frac{1 - \lambda_t}{2} \right) \left(1 + \cos \frac{2\pi z}{\xi R_b} \right) \quad (15)$$

where z is the distance along the axis, $\lambda = R(z) / R_b$, R_b is a characteristic pore-body dimension, $\lambda_t = R_t / R_b$ is the pore throat to pore body aspect ratio R_t is pore-throat radius, $\xi = L / R_b$ and L is the pore wavelength. Sinusoidal pore shapes have been used frequently to understand transport phenomena in porous media (*e.g.*, Refs. 41-43). Based on this representation of shape for a single pore and restricting the cross section to be circular, we find the single-phase conductivity and volume of a single pore analytically. We note that pore corners do not have a major effect on single-phase conductivity calculations:

Single Pore Conductivity

Reynolds number for fluid flow in pores varies between 4 E-3 and 4 E-4 based on the interstitial flow rates given previously, a pore dimension of 40 μm , and the viscosity (7.5 E-5 Pa-s) and density of liquid water⁴⁴ (870 kg/m^3) at 200°C. Hence, an assumption of creeping flow is appropriate^{41,45} as is an approximation that the single-phase flow obeys the Hagen-Poiseuille relation locally. In a sinusoidally constricted pore, the following equation results:

$$\frac{|\Delta p| \pi R_b^4}{8qL\mu} = \frac{(9\lambda_t^2 - 6\lambda_t + 5)}{\sqrt{\lambda_t^7}} \quad (16)$$

where q is the volumetric flow rate through the pore. The dimensionless hydraulic conductivity, g , for a single pore related to the pressure drop is

$$g = \frac{qL\mu}{|\Delta p|} \frac{8}{\pi R_c^4} \quad (17)$$

These two equations are combined to yield

$$g(r_t, \lambda_t) = \frac{8r_t^4}{\sqrt{\lambda_t} (9\lambda_t^2 - 6\lambda_t + 5)} \quad (18)$$

Single-Pore Volume

Assuming circular cross section, we find the dimensionless volume of a single pore per unit wavelength to be

$$v(\lambda_t, r_b) = \frac{\pi}{8} r_b^2 (3\lambda_t^2 + 2\lambda_t + 3) \quad (19)$$

The single-pore model, Eqs. (18) and (19), takes into account both body-and throat sizes. But the expressions for generating conductivity and volume distributions require that the single-pore equations be in terms of a single characteristic dimension. We select the dimensionless throat radius because the Bethe lattice effective conductivity, Eq. (3), is for the bond-

percolation problem, and bonds are analogous to throats in a porous medium. Thus, Eqs. (18) and (19) are modified to yield a form containing throat size only while retaining some embedded body-size information.

Equations (18) and (19) are averaged over all possible body sizes associated with a given throat size. If $\alpha_b(r_t, r_b)$ is the frequency distribution of the dimensionless body radius, r_b , associated with a given dimensionless throat radius, r_t , then the effective conductivity and available volume associated with a throat size are approximated as

$$\langle g(r_t) \rangle = \int_{r_t}^{\infty} \alpha_b(r_t, r_b) g(r_t, r_b) dr_b \quad (20)$$

$$\langle v(r_t) \rangle = \int_{r_t}^{\infty} \alpha_b(r_t, r_b) v(r_t, r_b) dr_b \quad (21)$$

PORE-EVOLUTION MODELS

There are two distinct pore-evolution mechanisms that occur in diatomite during steam drive. One is dissolution of the rock matrix and the other is precipitation of rock minerals. The following subsections describe the models that we choose to approximate these aspects of diagenesis. As our understanding of these processes in diatomite improves, these models will naturally improve.

Dissolution

Silica dissolution has been studied widely in regard to alkaline waterflooding. For instance, dissolution kinetics have been examined^{46,47} as well as the migration of alkali⁴⁸. The evolution of grain sizes and shapes during dissolution has also been studied via glass beadpacks⁴⁹. On the basis of observation of grain shapes and from assessing surface chemical potentials, it was suggested that silica dissolution was most likely to occur uniformly along pore walls⁴⁹. Following these results and based upon the scaling arguments presented above, we dissolve pore walls uniformly. Each pore in the network suffers the same amount of dissolution as shown schematically in Fig. 3b.

We denote the dissolved thickness by t in dimensionless form, where normalization is by the characteristic radius R_c . Based on this thickness, the dimensions of a pore after dissolution can be related to its original dimensions by

$$r_{t,new} = r_{t,old} + t \quad (22)$$

$$r_{b,new} = r_{b,old} + t \quad (23)$$

These single-pore equations are used to model the effect of dissolution at the network level. The body- and throat size distributions are still given by the gamma distributions of Eq. (9). The distribution parameter a is related to average size by

$$\langle r \rangle = a - 1 \quad (24)$$

The new average radius can be related to the old average radius and the thickness by

$$\langle r \rangle_{new} = \langle r \rangle_{old} + t \quad (25)$$

Note that Eqs. (22), (23), and (25) are in fact approximations, Uniform dissolution most likely occurs normal to the pore wall. However, Eq. (24) is exact in as much as an average pore size for a porous medium can be defined.

Based on Eqs. (24) and (25), the parameters for generating the body- or throat size distributions are updated as

$$a_{new} = a_{old} + t \quad (26)$$

Since dissolution of the network does not block any pore, the pores at any stage of dissolution are always accessible to flow and X_A for dissolution will always be 1.0.

Based on this approach, different stages of dissolution of the rock matrix are denoted by different values of dimensionless thickness t . Then, the pore-size distributions are updated and effective network conductivity and pore volume recalculated. Finally, the permeability and porosity change associated with a particular t are found.

Deposition

We use two different deposition models as given in Fig. 4 to explore clogging of pores. The first model, Fig. 4a, considers deposition to be a pore-lining process and the second considers deposition to occur only at throats. Deposition at pore throats is indicated on Fig. 4b by dark gray shading. In both models, deposition takes place in a series of steps. Each step in the deposition sequence depends on the network condition in the previous step. At any stage of deposition, the condition of the network is determined by the condition of the throats constituting the network. Even though the radii change as deposition proceeds, we characterize all throats of the network at any stage of deposition by their original radii, *i.e.*, before the process of deposition starts. The extent of deposition is given by the largest blocked throat with initial size r_{ti} . The current size of this throat is zero. This implies that the frequency of a throat with current radius r_t is given by $\alpha_t(r_t + r_{ti})$ when r_t has been clogged. We also define X_i as

$$X_i = \int_0^{r_{ti}} \alpha_t(r_t) dr \quad (27)$$

This X_i defines the fraction of the original throats that are not allowed to the main flow due to deposition. The fraction of throats that are allowed to flow is, $1 - X_i$. All of these allowed pores are not accessible to flow due to random connectivity of the network. Based on the current value of $1 - X_i$, we estimate the current accessible fraction of allowable pores of the network by using Eq. (6). The fraction of completely blocked pores X_{bl} ³⁸

$$X_{bl} = \int_0^{X_i} \frac{X_A}{1 - X_i} dX_i \quad (28)$$

Determination of these parameters at any stage of deposition tells us the condition of our network. A deposition step of Δr_{ti} implies that the largest blocked throat in the network had a radius between r_{ti} and $r_{ti} + \Delta r_{ti}$ prior to the start of deposition. First, we completely block accessible throats with current radius between 0 and Δr_{ti} . That is, pores with an original radius

between r_{ii} and $r_{ii} + \Delta r_{ii}$ fill with precipitate. We call this amount of precipitate deposited volume V_r . Second, we deposit a layer of thickness Δr_{ii} in all currently accessible pores with current radius between Δr_{ii} and infinity. We call this deposited volume V_r .

Volume Determination

To find the available pore volume at the stage $r_{ii} + \Delta r_{ii}$ of deposition, we calculate V_1 , V_2 and subtract their total from network volume at start of the deposition step. Thus,

$$V_{eff}(r_{ii} + \Delta r_{ii}) = V_{eff}(r_{ii}) - (V_1 + V_2) \quad (29)$$

The deposition calculations start at r_{ii} equal to 0, and we successively subtract from the base volume as calculated in Eq. (8) with the original pore-size distribution. Formulas for V_1 and V_2 depend on where silica deposition occurs.

In calculating V_1 and V_2 for deposition occurring only at pore throats, we hold the body-size distribution constant and change only the throat-size distribution. Thus, the current body sizes are the same as the original body sizes. The volumes are calculated as,

$$V_1 = \frac{X_A}{1 - X_i} \int_0^{\Delta r_{ii}} \alpha_t(r_t + r_{ii}) \langle v(r_t) \rangle dr_t \quad (30)$$

$$V_2 = \frac{X_A}{1 - X_i} \left[\int_{\Delta r_{ii}}^{\infty} \alpha_t(r_t + r_{ii}) \langle v(r_t) \rangle dr_t - \int_0^{\infty} \alpha_t(r_t + r_{ii} + \Delta r_{ii}) \langle v(r_t) \rangle dr_t \right] \quad (31)$$

To find the volume change associated with a uniform deposition, we consider a single-pore model and calculate the volume that is deposited by a uniform layer of thickness t :

$$v_{dep}(r_b, r_t, t) = \pi(r_b + r_t - t) \quad (32)$$

We then average V_{dep} over all possible body sizes to get volume change associated with a throat for a given thickness. The expression is

$$\langle v_{dep}(r_t, t) \rangle = \int_{r_t}^{\infty} \alpha_b(r_b, r_t) v_{dep}(r_b, r_t, t) dr_b \quad (33)$$

We use this equation to calculate the deposition occurring in all throats with current radii ranging from 0 to ∞ . Thus, for the uniform deposition case, the volume deposited is given by

$$V_1 + V_2 = \frac{X_A}{1 - X_i} \int_0^{\infty} \alpha_t(r_t + r_{ii}) \langle v_{dep}(r_t, t) \rangle dr_t \quad (34)$$

Conductivity Calculations

The effective conductivity of the network is found by considering the three kinds of throats in the network at any deposition stage r_{ii} . First, some throats have original size less than r_{ii} and are currently blocked completely and so do not contribute towards network conductivity. Second, some throats have original size greater than r_{ii} are accessible to flow and, hence, contribute towards network conductivity. Last, there are throats that have original size between 0 to ∞ but are not accessible to flow due to random connectivity of the network and, thus, do not contribute to flow. The conductivity distribution of the network is then computed according to Eq. (5).

RESULTS

The models for deposition and dissolution were translated into the following algorithm to gauge the effect of dissolution and precipitation on permeability. A base case with given network connectivity, Z , average aspect ratio, λ_t , pore-body distribution, and pore-throat distribution is constructed. Next, we calculate the effective conductivity and available network volume of the base case. We then update the pore-body and pore-throat distributions according to the rules given above for a specified amount of silica precipitated or dissolved. The effective conductivity and available volume of the network are recalculated. Finally, Eqs. (13) and (14) are used to relate the conductivity and available volume changes to permeability and porosity changes. Connectivity and aspect ratio are varied independently for new base cases, and the process repeated.

In the case of uniform dissolution, Fig. 5 represents the permeability versus porosity relationship for a body-to-throat aspect ratio of 20/3 and different values of coordination number, Z . The plot has two distinct trends. First, the permeability increases rather dramatically for all values of Z . An order of magnitude increase in permeability is predicted if the porosity increases by 50%. Second, for a fixed value of porosity change, permeability increases most for small Z . For Z equal to 3, the permeability has increased 40 times when ϕ/ϕ_0 reaches 1.5. As Z becomes larger, the effect is less. It is reasonable that permeability increases most sharply and more severely for rocks with low connectivity as compared to rocks with high connectivity. The role of an individual pore throat on permeability is accentuated in networks with small connectivity.

Figure 6 displays similar dissolution calculations for the case of Z equal to 5 and variable pore aspect ratio. The most dramatic increases in permeability occur for the most constricted pores. When pore throats are small relative to pore bodies, a small amount of dissolution dramatically increases permeability. Note that the least constricted pores display the smallest amount of permeability increase relative to the base case.

Deposition results are similar to dissolution, except that the trends are reversed. Permeability decreases with porosity. Figure 7 shows the permeability versus porosity relationship for an R_b/R_t of 20/3 and various connectivities. Deposition occurs at throats only. For all values of Z , the decrease in permeability is very marked near ϕ/ϕ_0 equal to 1 and becomes less steep as the network approaches blockage. Again, the most dramatic effects are seen in the networks with the least connectivity. Figure 8 displays the permeability versus porosity relationship for a Z of 5 and various pore body to throat size ratios. A small amount of silica deposited in a narrow throat in a poorly connected network can substantially decrease network conductivity, and hence permeability.

For completeness, Figs. 9 and 10 present results for uniform deposition. Figure 9 applies to a fixed aspect ratio of 20/3 with variable Z , whereas Fig. 10 shows results for fixed connectivity and various R_b/R_t . Trends are similar to Figs. 7 and 8, but the trend of decreasing permeability with silica deposition is not as strong. Because deposition is pore lining rather than throat blocking, it takes a greater change in porosity to effect the same change in permeability. As expected, the most dramatic decreases in permeability are found for small Z and large λ_t . Finally, careful comparison of Figs. 5 and 6 with Figs. 9 and 10 reveals that dissolution and uniform deposition results, respectively, approach ϕ/ϕ_0 equal to 1 with identical slope.

DISCUSSION

The range of porosity variation given in Figs. 5 to 10 is extreme and not expected in the field far away from the wellbore. Likewise, experiments have probed a much narrower range than given above. If we limit the porosity changes to 10%, which are likely the maximum for the field case¹, and examine permeability versus porosity on a log-log plot, a power-law relationship is found for dissolution and both modes of deposition.

This point is illustrated graphically in Fig. 11 for the case of deposition at pore throats and a limited porosity variation of 10%. The pore throat to body aspect ratio is fixed at the experimentally observed value 20/3 and the network connectivity varied between 3 and 13. The power-law exponent increases with decreasing Z as indicated by the slopes of the lines. Figure 11 also presents the experimental results of Koh *et al.*¹ who measured permeability of diatomite plugs that were flushed with hot (204 °C), silica saturated laden water while the diatomite plug was maintained at 49 °C to induce silica deposition. Experimentally, they found that permeability reduction correlated with porosity decrease according to a power-law relationship with a best-fit power-law exponent of 9. This experimental result is quite close to the network result for Z equal to 6 and lies between network results for Z equal to 5 and 7 in Fig 11. Recall that the conductivity of Bethe networks with Z from 4 to 6, such as the results shown in Fig. 1, corresponds to the conductivity of three-dimensional networks of pore space. The result given in Fig. 11, therefore, provides a check on the validity of the network model. If the model had not reproduced the experimentally observed trend with coordination numbers from about 4 to 6, it would have been necessarily discarded.

In the case of pore-lining silica deposition, our network model exhibits an exponent of 9 when Z is between 3 and 4 and R_b/R_t is 20/3. Comparing throat-deposition and pore-lining deposition network results, it is hard to distinguish between the two mechanisms because no measurements of coordination number for diatomite have been made, to our knowledge. Thus, we do not know if diatomite pore structure is well or poorly connected. However, for network parameters chosen because they correspond to measured pore dimensions and size distributions and, likewise, limited to networks with coordination numbers from about 4 to 6, we reproduce the experimentally observed trend.

It is interesting to note that the power-law exponents resulting from the experiments and the network model are rather extreme compared to previous studies where exponents from about 2 to 5 were prevalent. Since, porous medium permeability correlates strongly with pore-throat size, and since the pore body to throat aspect ratio is large ($\lambda_t \approx 20/3$), small amounts of deposited solids, as gauged by porosity decrease, effectively plug pores and block the movement of fluids. On the other hand, diatomite permeability should increase substantially with moderate dissolution.

The power-law form exhibited in Fig. 11 for porosity change limited to 10% suggests that the network results within this range can be correlated according to

$$\frac{k}{k_0} = \left(\frac{\phi}{\phi_0} \right)^\gamma \quad (35)$$

where Figs. 5 to 10 show that the power-law exponent γ is a function of the network parameters Z and $\lambda_t = R_b / R_t$. Inspection of the results yields the following correlating form for γ :

$$\gamma = B(\lambda_t) + \frac{E(\lambda_t)}{Z - F(\lambda_t)} \quad (36)$$

Note that λ is correlated in terms of measurable rock quantities. The coefficients B, E, and F are functions of λ_t . We obtain them by fitting polynomial expressions to network results. Table 1 details the exact functional forms for each of these coefficients. An error analysis²³ between network results, Eqs. (35) and (36), and the information in Table 1, demonstrates that the power-law correlation is accurate to within 3%. The largest deviations are found at low values of porosity change. Deviations likely result from small errors in the polynomial representation of coefficients B, D, and E.

The results listed in Table 1 provide us with a means of understanding how the power-law exponent shifts as diatomite quality varies. For example, as the average initial pore-body to pore-throat aspect ratio increases, we now have an estimate of the magnitude of increase in λ , and vice versa. This moves us closer to our goal of mechanistic simulation of the steam-drive process in diatomite. Additionally, the relations obtained here could be incorporated into a continuum model representing silica redistribution in diatomite even if local equilibrium of silica is not assumed. Recognize that we have redistributed known amounts of silica and computed how the redistribution affects porous medium conductivity. A kinetic model could provide the amount of silica dissolved or deposited and then the results reported here could be used to estimate the ensuing change in permeability.

TABLE 1: Coefficients in expression for power-law exponent.

Para- meters	Dissolution	Uniform Deposition	Deposition at Throats
$B(\lambda_t) =$	$0.0201 \lambda_t^2 + 0.645 \lambda_t + 1.8633$	$0.0003 \lambda_t^2 + 0.147 \lambda_t + 2.827$	$-0.0311 \lambda_t^2 + 0.5699 \lambda_t + 2.8841$
$E(\lambda_t) =$	$-0.0168 \lambda_t^3 + 0.3126 \lambda_t^2 + 0.0525 \lambda_t + 2.7042$	$-0.0059 \lambda_t^3 + 0.08 \lambda_t^2 + 0.614 \lambda_t + 5.2683$	$0.0128 \lambda_t^3 - 0.3715 \lambda_t^2 - 3.876 \lambda_t + 2.9961$
$F(\lambda_t) =$	$-0.0279 \lambda_t + 0.5886$	$-0.0299 \lambda_t + 1.7014$	$-0.0185 \lambda_t + 1.6878$

Results from this initial study are promising and several extensions are suggested. First, all results obtained here were for a pristine network. That is, silica deposition followed by subsequent redissolution was not examined. This is the expected trend in the field and the network model needs to be developed further to simulate this cycle. Second, agreement between network-suggested trends and those found experimentally for the case of deposition appears to be quite good. In order to perform a similar check for the dissolution case, experiments should be conducted. Last, the model and calculations presented here are for single-phase flow of water. To be applicable to the simultaneous flow of water and oil in diatomite, the model should be extended to multiple phases. Because phase conductivity must be computed when the fraction of the pore space occupied by a phase is low, it is likely that this task requires a true three-dimensional network.

CONCLUSIONS

Beginning with a simple network representation of diatomite and knowledge of pore sizes, we are able to model permeability evolution as a function of porosity change. For moderate increases or decreases in porosity, we find a power-law relationship describing permeability alteration. Such a trend for silica deposition has been determined experimentally elsewhere. Network-predicted power-law exponents for deposition are about 9 and in agreement with this prior work. The power-law feature holds promise that permeability change during steam drive, or waterflood, in diatomite can be incorporated into reservoir simulators.

Specifically, we find that the porosity-permeability relationship for diatomite can be modeled as a simple power-law relation only for moderate ranges of deposition and dissolution. At low ranges of deposition, it is difficult to distinguish between deposition only at pore throats and pore-lining deposition of silica for expected pore body to throat aspect ratios and connectivity. Also, it is found that permeability change is most dramatic for networks with very small pore throats connected to relatively large bodies. This last aspect largely explains the extreme magnitude of power-law exponents observed for diatomite.

NOMENCLATURE

a, b	=	parameters in gamma distribution
A	=	cross-sectional area
A_v	=	interfacial area per unit volume
B, E, F	=	coefficients for data reduction
$C(x)$	=	network generating function
D	=	diffusivity
Da	=	Damkohler number
f	=	fraction of pore throats available for flow
f^*	=	root in Eq. (7)
g	=	conductivity of a pore segment
G	=	network conductivity distribution
k	=	permeability
l	=	characteristic length
L	=	pore length
p	=	pressure
q	=	volumetric flow rate
r	=	dimensionless pore radius
R	=	radius

Sh	=	Sherwood number
t	=	dimensionless thickness of deposited layer
u	=	interstitial velocity
v	=	volume of a pore segment
V	=	network volume
X_A	=	network accessibility
X_i	=	fraction of throats not allowed to flow
z	=	distance in axial direction
Z	=	coordination number

Greek Symbols

α	=	pore-size distribution function
γ	=	power-law exponent
Γ	=	gamma function
Δ	=	difference operator
δ	=	Dirac delta function
k	=	first-order reaction rate constant
K	=	mass-transfer coefficient
λ	=	radius nondimensionalized by body size
μ	=	viscosity
ξ	=	pore length nondimensionalized by body size
ϕ	=	porosity

Subscripts

b	=	body
bl	=	completely blocked pores
c	=	characteristic pore dimension
eff	=	effective for flow
p	=	percolation threshold
t	=	throat
0	=	denotes properties before dissolution or precipitation
$1, 2$	=	volumes deposited in precipitation model

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FIG. 1 Comparison of Bethe network conductivity to a cubic lattice.

FIG. 2 Body size, throat size, and constrained body size distributions.

(a)

(b)

FIG. 3 Single-pore model (a) and uniform dissolution model (b).

FIG. 4 Uniform deposition model (a) and deposition at pore throats only (b).

FIG. 5 Dissolution case, permeability versus porosity relation for different coordination numbers.

FIG. 6. Dissolution case, permeability versus porosity relation for different aspect ratios.

FIG. 7 Throat deposition case, permeability versus porosity relation for different coordination numbers.

FIG. 8 Throat deposition case, permeability versus porosity relation for different aspect ratios.

FIG. 9 Uniform deposition case, permeability versus porosity relation for different coordination numbers.

FIG. 10 Uniform deposition case, permeability versus porosity relation for different aspect ratios.

FIG. 11 Straight-line fits for permeability versus porosity relations. Axes are logarithmic.

PROJECT 2: IN-SITU COMBUSTION

To evaluate the effect of different reservoir parameters on the in-situ combustion process. This project includes the study of the kinetics of the reactions.

2.1 INSITU COMBUSTION

(Louis Castanier)

During the period of performance of the contract, in-situ combustion research was initially focused towards the effect of metallic additives on the reactions of combustion. Equipment failures lead to a complete rebuilding of the experimental set up which is still ongoing. As a result, no significant new experimental data were obtained. The following is a summary of why we continue to research in-situ combustion and an example of application of this method to unusual (e.g., offshore, arctic) or difficult reservoirs. We are currently studying the feasibility of a combination of solvent and in-situ combustion techniques and will continue this work in the future.

NEED FOR IN-SITU COMBUSTION

The traditional recovery techniques for heavy oils can be classified in two categories: thermal methods, which aim at decreasing the crude viscosity by increasing temperature and non-thermal methods, which in heavy oil reservoirs generally suffer from the high mobility contrast between the oil and injected fluids. By far the most widely used recovery method for heavy oils is steam injection, either cyclic or as a drive. Steam injection, however, is not applicable to many reservoirs. Among the parameters affecting the success of steam injection are:

- Depth: If the reservoir depth is over 3,000 ft. (1,000 m) the heat losses in the injection well will make the steam injection inefficient. Insulated tubing can be used as a partial remedy, but it is expensive and difficult to install properly in the field. Downhole steam generators have been tried without success in the last twenty years. They suffer from reliability problems.
- Pressure: At high pressure, the latent heat of water drastically decreases and high pressure also means high injection temperature causing less efficient heat transport and added heat losses.
- Permeability: Clays present in the reservoir may swell in presence of the fresh water produced by steam condensation. This can reduce injectivity and impair the process.
- Oil saturation and porosity: These two parameters are very important. If not enough oil is in place, the energy used to heat the reservoir will be more than the energy obtained by the oil produced. Steam injection is generally impossible in reservoir where the product of porosity by oil saturation is less than 0.1.

From the above discussion, it is clear that despite its usefulness, steam injection is limited to a select group of reservoirs. Let us now turn our attention to in-situ combustion. In-situ combustion is not limited by reservoir depth. It is also applicable where wellbore heat transfer would severely limit steam injection. Two possible cases are offshore fields and the presence of permafrost where steam is not applicable but combustion would be feasible. Combustion is more energy efficient and less polluting than steam. It has been field proven in a number of countries. Combustion, however, suffers from poor sweep efficiency caused by the mobility contrast between the injected air (or oxygen) and the reservoir fluids. It also requires careful engineering and is less flexible and forgiving of errors than steam.

SOLVENT USE

Solvents can be used to reduce the oil viscosity and facilitate production. The solvents can be gases used in miscible or non-miscible mode, or liquids at reservoir conditions. Liquid solvents are usually expensive and the price of the oil recovered low. Consequently, the economics of liquid solvent injection are usually not favorable. Gases are used in miscible or non-miscible modes. Miscibility of gases with heavy oils requires very high pressures. Again the economics of miscible flooding of heavy oils are not good. Non-miscible gas injection has been tried with some success, especially in a cyclic fashion. Some of the technical problems involved with solvent injection arise from precipitation of asphaltene or other compounds during the mixing of the oil and the solvent. This phenomenon is well known and described in a number of publications. It leads to reduction in permeability, especially near the wells, and can, in some cases be very severe and cause technical failure of the process. We expect this problem to be acute in the case of heavy oils.

PROPOSED WORK FOR DIFFICULT RESERVOIRS

None of the existing enhanced oil recovery methods is applicable to difficult or unusual heavy oil reservoirs such as deep, high pressure, offshore, or arctic fields. The rest of this section proposes a combination of several of the existing techniques that is designed to solve this problem. The economic aspects have been taken into account in our preliminary study and we feel that the proposed improved oil recovery ideas can be field applied with existing technology.

METHOD DESCRIPTION

The following description should be considered as a possible outline to be followed. Although the concepts will remain the same, local technical and economic conditions will dictate numerous variations and changes in methods. Cyclic oil recovery has advantages, both technical and in terms of economics. It can also be easily optimized in a given oil reservoir. We propose cyclic injection of solvents, either gas or liquid, followed by in-situ combustion of a small part of the reservoir to increase the temperature near wellbore, and also to clean the wellbore region of the residues left by the solvents. Alternate slugs of solvent and air will be injected and production will occur after each solvent slug injection and after each combustion. The process can be repeated until the economic limit is reached. One important fact to note is that both solvent injection and in-situ combustion have been proven to be effective in a variety of reservoirs, the combination of the two methods has, to our knowledge, never been tried.

First Cycle: Solvent Injection

A slug of solvent (liquid or a gas) should be injected. The amount to be injected needs to be determined by reservoir simulation or from calculations. We expect that in most cases, the amount to be injected will be a small fraction of the pore volume of the reservoir considered. By mixing with the oil, the solvent will decrease the viscosity of the mixture as compared to the viscosity of the original oil. This viscosity reduction can be evaluated if the solvent is a liquid by using Eq. 1 from Burger, *et al.*, page 41. The improvement in oil production can be estimated using an approach similar to the one used by Boberg for steam cycling and taking into account only the change of viscosity caused by the solvent. Production after the injection

of the first slug of solvent can thus be easily calculated. If the solvent is a gas, Eq. 1 does not apply, instead, the dissolved gases will induce swelling of the oil and also viscosity reduction as the gas saturated oil will be less viscous than the original oil. Of course, these parameters are best established by laboratory measurements on the crude/solvent system to be studied. Numerical simulation can now predict the expected improvement in production from the solvent injection part of the process. The VAPEX process in Canada is a good example of application of such methods to heavy oils.

At the end of the first solvent cycle, one can expect some damage to the near wellbore area. Most significant will be the precipitation and/or deposition of heavy hydrocarbons such as asphaltenes or paraffins. The produced oil is expected to be upgraded by the solvent cycle.

First Cycle: Combustion

Unlike the classic well-to-well in-situ combustion, we will only try to improve near wellbore conditions by burning the solid residues left after the solvent cycle. The amount of air (or oxygen) to be injected can be easily determined by laboratory experiments and numerical simulation or simple calculations. The benefits of using combustion at this stage are expected to include:

- Productivity improvement through removal of the solids precipitated from the solvent cycle.
- Possible deactivation of the near wellbore clays caused by the high temperature of the combustion
- Reduced viscosity of the oil due to temperature increase.
- Effect of CO_2 production causing swelling and viscosity reduction of the oil.
- Pressure maintenance.

One must note that combustion burns only the heaviest part of the crude, and hence the upgrading already observed during the solvent cycle should be maintained, and even improved. A soak period will probably be necessary at the end of the air/oxygen injection to ensure that all of the oxygen injected has reacted in the reservoir and will not be produced during the subsequent production cycle. The recovery predictions at this stage of the process will be difficult and probably require a fine grid thermal/compositional simulation. It is, however, possible to obtain some estimates based on simple correlations similar to the well-known steam cyclic models.

Repeat Cycles:

The process can be repeated by alternating solvent injection and near well in-situ combustion as long as the economics warrant it. Optimization of slug sizes and duration needs to be made to ensure maximum economic efficiency. Reservoir conditions will, of course, dictate these parameters.

DISCUSSION OF POSSIBLE PROBLEMS: PRELIMINARY WORK REQUIRED:

During the solvent injection, the main problem will be the efficient use of a limited amount of expensive injected fluid. Reservoir heterogeneities and density and mobility differences between solvent and reservoir oil will cause sweep efficiency problems. This in turn can cause concentration gradients and hence large viscosity differences among the fluids to be produced. As in any improved recovery technique, proper reservoir definition is needed. The fact that the

process is cyclic in nature should ensure no waste of solvent. The other problems to be encountered in the solvent part of the process are well-known and described in the literature. The type of work required prior to field application involves PVT studies and creation in the laboratory of various oil/solvent mixtures. Deposition of asphaltenes or other residues on rocks similar to reservoir rocks should also be studied. The following describes some simple tests that can be used as screening techniques for the method.

FEASIBILITY STUDY

Heavy crudes, such as Cold Lake or Hamaca, will be mixed with liquid solvents and viscosities of the mixtures will be measured. Filtration of the mixtures should determine whether or not solid precipitation or formation damage will occur. This first step has been done with Cold Lake Bitumen and has shown both significant viscosity reduction and the presence of solid precipitate. This work is still preliminary and will be continued. During the second step, the mixtures will be injected into a sandpack of permeability and pore structure close to the field sands. Permeability versus injected volume of the mixtures will be recorded. When (and if) the ratio of the permeability, before and after passage of the mixtures reach a given value, the sand pack will be subjected to an in-situ combustion tube run. Fuel concentration and composition as well as an estimate of the air requirements will be obtained. At this stage, simple calculations will give estimates on the production in the field.

Economic evaluation should start at this level. Laboratory tests for gas in solution may be more difficult to implement. The literature is poor in PVT data for heavy oils and more complex experiments may be required. A significant research effort is being made on the "foamy oil" problem, and data from those experiments can probably be used in our project. The database of heavy oil with CO_2 projects should also be used.

OPTIMIZATION AND PRACTICAL APPLICATION

Although reservoir and economic conditions will dictate the directions to be followed for optimization of the process, some general comments are warranted at this point. It would appear that cycle size should increase if the reservoir is thick. One would expect the combustion area to move toward the top of the reservoir in this case. Gravity drainage will be an important part of the recovery mechanism and should be taken into account in the implementation of any field test. Position of the injection/production intervals should be calculated to optimize the process. Numerical simulations will probably be needed to fully optimize the recovery.

As previously mentioned, a soak period will be needed at the end of the combustion portion of each cycle. Its duration can be determined by kinetics studies giving the reactivity of the oil with air. As temperatures near wellbore at the end of the air injection will be high, there should be no problem of oxygen backflush into the well as it will react with the oil. Careful monitoring of the produced gases will be needed for added safety.

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2.2 IN-SITU COMBUSTION USING WATER SOLUBLE METALLIC ADDITIVES

(U. Diwan and L. Castanier)

2.2.1 INTRODUCTION

Crude oils are often grouped into three categories based on specific gravity range:

1. Heavy Oil (10° - 20° API)
2. Intermediate Oil (20° - 30° API)
3. Light Oil (greater than 30° API)

Heavy oils consist mainly of high density naphthenes, aromatics and heteroatoms that are poor in alkanes, while light oils consist mainly of alkanes (Boduszynski, 1987; Boduszynski, 1988). Bitumen or tar are extremely dense hydrocarbons (about 10° API or less), and are non-volatile liquids, semi-solids or solids. The deposits are often referred to as oil sands or tar sands.

2.2.2 IN-SITU COMBUSTION

In-situ combustion is a thermal recovery technique in which a part of the heavy oil in place is burned to generate heat. This heat brings about a reduction in viscosity of the crude oil to get improved mobility and hence oil production rate and recovery. In a laboratory the process of ignition is initiated by using electric heaters while a stream of air is injected into a combustion tube to initiate and sustain combustion. Pure oxygen may also be used, but for economy sake air is popular. The fuel that is burned is the unrecoverable carbon rich residue of that is left on the reservoir matrix behind the steam front as a result of steam distillation, thermal cracking and some catalytic cracking. The heat that is generated partially distills the crude oil. The lighter ends are distilled off, and they condense in the cooler regions ahead of the combustion front along with the vaporized connate water and water produced as a combustion byproduct. The region ahead of the combustion front is heated by conduction, by convection of combustion gases, and by the condensation of volatiles (light ends) and steam. The oil ahead of the combustion front is displaced toward the production well by gas drive provided by the combustion gases, by hot water and steam drive, and by miscible drive provided by the condensed light hydrocarbons (Alexander *et al.*, 1962; Holt, 1992)

A typical combustion front moves through the reservoir matrix by consuming the fuel as it moves ahead, thereby leaving no oil behind the burning front. As the combustion front approaches the volume element, the temperature of the element rises and water and light ends are vaporized. These vapors are carried in the gas stream and condense in the cooler regions ahead of the combustion front. The water vapors condense to form a water bank (E), following a bank of light hydrocarbons (F). A steam plateau (D) comprises of the steam-liquid, two-phase region. As the temperature in the volume element exceeds 350°C the oil undergoes thermal cracking to form a volatile fraction and a low volatility heavy residue (represented by C). The volatile fraction gets carried in the gas stream and the heavy residue constitutes the fuel which gets burned in the combustion zone (B). The heat generated in the combustion zone gets transported ahead of the front by conduction and convection by the vapors and liquids. The combustion zone is often only a few inches in thickness and has a temperature in the range 350°

- 650°C. As the combustion front moves past this volume element it leaves behind a zone of clean sand (A) which serves as a preheater for the incoming air.

2.2.2.1 Applicability, Merits and Demerits of In-situ Combustion

In-situ combustion is applicable to a wide range of reservoir fluid characteristics. The absence of well bore heat losses in the injection well allows in-situ combustion to be carried out in deeper reservoirs having thinner, tighter sand sections which are not amenable to steam injection. The oil that is produced is also lighter than the oil in place as a result of cracking and distillation.

This technique is amongst the most energy efficient of improved oil recovery methods available today for heavy oils. However, one major constraint that limits its practical application is the amount of fuel formation in the matrix. If a sufficient amount of fuel is not deposited, as is often the case for light oils, the combustion front will not sustain itself. Conversely, if the quantity of fuel deposited is large, as is often the case with very heavy oils, the rate of advance of the front tends to be slower, with an uneconomically high demand for compressed air to sustain combustion (Alexander *et al.*, 1962).

As a result it would be desirable to find substances that alter the reaction kinetics of oil oxidation during in-situ combustion. There are a number of factors affecting the rates of these reactions, among which are the composition and concentration of the catalyst, surface of the catalyst available for reaction temperature. Combustion tube studies with metallic additives (Baena *et al.*, 1990; Castanier *et al.*, 1992; Holt, 1992) have shown that the addition of water soluble metallic salts can change the reaction kinetics of combustion. These indicate an increased fuel deposition in runs which carried salts of iron or tin. It has not been established for certain how the presence of these substances affect the fuel deposition mechanism, but it may be due to the reduction in the temperature required for cracking reactions.

Kinetic tube studies with metallic additives (de los Rios *et al.*, 1988; Shallcross, 1989) indicate that aqueous solutions of certain metallic salts like zinc, iron and tin increased the fuel concentrations. All these studies indicate that the overall oxidation mechanism of crude oils in porous media is the result of an overlap of several reactions that occur at different ranges of temperature. These have been classified as low temperature, medium temperature and high temperature reactions.

In a properly designed combustion process there should be minimal amount of low temperature oxidation (Agrawal and Sidqi, 1996). Therefore, the presence of metallic additives, which affect reaction kinetics, would affect the overall performance of the combustion process. It is believed that low temperature oxidation reactions affect fuel formation, therefore the alteration of this reaction would affect fuel deposition characteristics.

Earlier work done by De Los Rios (1987), has shown, on a quantitative basis, that the use of metallic additives affects the nature of the fuel formed; this, in turn, will affect the heat of combustion, the air-oil ratio, the air requirements, the front velocity and the oil recovery rate.

2.2.2.2 Metallic Additives in In-situ Combustion

Studies on the effect of metallic additives (salts) on in-situ combustion date back to the early nineteen seventies. It was observed that reservoirs having mineral contents with high metallic content in the rock matrix had increased fuel deposition (Burger and Sahuquet, 1972). Early attempts at understanding the mechanics of oil oxidation reactions was done through kinetic tube experiments in the presence of metallic additives.

A brief description to explain how the oxidation reactions take place is given on the following page.

2.2.2.3 Kinetic Studies with Metals and Metallic Additives

Several studies have been performed to determine the influence of metals and metallic additives on the oxidation characteristics of crude oils. The work performed and observations made during these experiments are described below.

1. Of particular importance to the current study are the kinetic studies performed by De Los Rios (1988) and Shallcross (1989) at Stanford University. De Los Rios performed kinetic experiments on Huntington Beach Oil and various metallic additives. Shallcross performed kinetic experiments on Huntington Beach and Venezuelan oils with various metallic additives. Both studies analyzed oxygen consumption data by decoupling the total oxygen consumption data into three parts. This was done to represent the oxygen consumed by the three competing groups of reactions. They found that metallic additives iron and tin increased oxygen consumption. They also reported that iron and tin increased the reaction rates for the oxidation reactions. They found that zinc did not produce the effects reported for iron and tin. From the results of these studies it was expected that iron and tin may be useful agents in catalyzing combustion reactions (Mamora, 1993).
2. Burger and Sahuquet (1981) report on two kinetic runs using 2000 ppm copper performed on a 27°API oil. The activation energy for the LTO was found to be lowered due to the copper additive (Holt, 1992).
3. Fassihi *et al.* (1981) report on kinetic experiments with 2000 ppm copper says that the activation energy of the LTO was unaffected, but its reaction rate was increased. Fassihi also reported that the activation energy for the high temperature portion of the reaction was reduced (Baena *et al.*, 1990; Mamora, 1993).

Reactions taking place:

1. Low temperature oxidation (LTO)	Result in formation of peroxides, hydroperoxides, aldehydes, ketones, carboxylic acids and alcohols (Burger and Sahuquet, 1972). Alexander <i>et al.</i> (1962) and Al-Saadon (1970) showed that fuel availability was increased when LTO of crude oil took place. Dabbous and Fulton (1974) corroborated this and also found that LTO causes a substantial decline in recoverable oil from the distillation and cracking zones, an increase in the fuel deposition and marked changes in the fuel characteristics and coked sand properties. Weijdemer (1968) and Fassihi (1981) found that the order for the LTO reaction, in terms of the partial pressure of oxygen, was close to unity.
2. Medium temperature oxidation (MTO)	Involve distillation, visbreaking and coking of the fuel along with partial oxidation of the products formed (Bardon and Gadelle, 1977; Fassihi, 1981; Sidqi <i>et al.</i> , 1985). The amount of coking and atomic H/C ratio of the fuel were found to decrease with increasing coking temperature (Bousaid, 1967). Increasing pressure was found to increase the amount of hydrocarbon residue formed; but the fuel deposited has a lesser hydrogen content (Sidqi <i>et al.</i> , 1985).
3. High temperature oxidation (HTO)	Involve oxidation of the cracked hydrocarbon residue. HTO takes place according to the following equation:
$CH_n + \left(\frac{2m+1}{2m+2} + \frac{n}{4} \right) O_2 \rightarrow \left(\frac{m}{m+1} \right) CO_2 + \left(\frac{1}{m+1} \right) CO + \left(\frac{n}{2} \right) H_2O$ <p>m: ratio of carbon dioxide to carbon monoxide formed upon oxidation of the fuel n: atomic hydrogen to carbon ratio of fuel burned The reaction rate has been found to be first order with respect to the fuel concentration and was found to have an order of 0.5 to 1.0 with respect to the oxygen partial pressure (Dabbous and Fulton, 1974).</p>	

To recap:

- It has been found by previous researchers that the presence of certain metals can affect the kinetic parameters used to model combustion reactions.
- These kinetic experiments can give estimates of activation energies and reaction rates for combustion reactions.

2.2.2.4 Tube Studies with Metallic Additives

Observations that the physical and chemical properties of the core material can affect the combustion performance in tube runs was made by several researchers. Some of them are mentioned below.

1. Vossoughi *et al.*, 1982, Hardy *et al.*, 1972, Alexander *et al.*, 1962 observed that, for a particular oil, more fuel can be expected from a native core than from a clean sand pack. This is believed to result from both physical and chemical processes. Fine grained material

in native cores provides more surface area for combustion reactions while metals in natural minerals act as catalysts (Baena *et al.*, 1990; Bardon and Gadelle, 1977).

2. Baena *et al.*, 1990 report on experiments with 22°API Huntington Beach Oil. They observed that tin and iron additives increased oxygen utilization efficiency, burning front velocity, and fuel concentration. The zinc additive showed lesser effect (Baena *et al.*, 1990).
3. Castanier *et al.*, 1992 report on thirteen metallic additive combustion tube runs. The runs made with 10°API Hamaca oil show that tin and iron increased the fuel deposition, oxygen utilization efficiency, and front velocity while zinc was less effective (Bardon and Gadelle, 1977).

2.2.3 PRESENT STUDY

The present study shall involve a series of combustion tube experiments performed with the Wilmington Oil (~15°API) and a couple of aqueous metallic additives.

2.2.4 WORK COMPLETED

- Thermocouple testing
- Calibration of mass flow meter
- Calibration of mass flow controller
- Calibration of carbon monoxide gas analyzer
- Calibration of carbon dioxide gas analyzer

2.1.5 FUTURE WORK

- Calibration of oxygen analyzer
- Checking apparatus for leakages
- Checking data logger
- Transducer calibration to measure pressure drop along tube
- Conduct a trial run using no packing, just air
- Observe temperature profiles
- Pack the combustion tube with the crude oil
- Perform runs

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PROJECT 3: STEAM WITH ADDITIVES

To develop and understand the mechanisms of the process using commercially available surfactants for reduction of gravity override and channeling of steam.

3.1 FOAM FLOW IN HETEROGENEOUS POROUS MEDIA: EFFECT OF CROSS FLOW

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Foam Flow in Heterogeneous Porous Media: Effect of Cross flow

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Abstract

Previous experimental studies of foam generation and transport were conducted, mainly, in one-dimensional and homogeneous porous media. However, the field situation is primarily heterogeneous and multidimensional. To begin to bridge this gap, we have studied foam formation and propagation in an annularly heterogeneous porous medium. The experimental system was constructed by centering a 0.050 m diameter cylindrical Fontainebleau sandstone core inside an 0.089 m acrylic tube and packing clean Ottawa sand in the annular region. The sandstone permeability is roughly $0.1 \mu\text{m}^2$ while the unconsolidated sand permeability is $6.7 \mu\text{m}^2$. Experiments with and without cross flow between the two porous media were conducted. To prevent cross flow, the cylindrical face of the sandstone was encased in a heat-shrink Teflon sleeve and the annular region packed with sand as before. Nitrogen is the gas phase and an alpha olefin sulfonate (AOS 1416) in brine is the foamer. The aqueous phase saturation distribution is garnered using X-ray computed tomography (CT).

Results from this study are striking. When the heterogeneous layers are in capillary communication and cross flow is allowed, foam fronts move at identical rates in each porous medium as quantified by the CT-

scan images. Desaturation by foam is efficient and typically complete in about 1 PV of gas injection. When cross flow is prohibited, foam partially plugs the high permeability sand and diverts flow into the low permeability sandstone. The foam front moves through the low permeability region faster than in the high permeability region.

Introduction

Foam is applied broadly as a mobility-control and profile modification agent for flow in porous media. Foams are usually formed by nonwetting gases such as steam, carbon dioxide (CO_2), or nitrogen (N_2) dispersed within a continuous surfactant-laden liquid phase. Typical applications include aqueous foams for improving steam-drive¹⁻³ and CO_2 -flood performance,⁴ gelled-foams for plugging high permeability channels,⁵ foams for prevention or delay of gas or water coning,⁶ and surfactant-alternating-gas processes for clean up of ground-water aquifers.^{7,8} All of these methods have been tested in both the laboratory and the field.

An unfoamed gas displays low viscosity relative to water or crude oil and is thereby very mobile in porous media. However, dispersing the gas phase within a surfactant solution where the surfactant stabilizes the gas/liquid interface can substantially reduce gas mobility in porous media. Mobility is reduced because pore-

spanning liquid films (foam lamellae) and lenses block some of the flow channels. Additionally, flowing lamellae encounter significant drag because of the presence of pore walls and constrictions. One aspect of foams that makes them attractive is that a relatively small amount of surfactant chemical can affect the flow properties of a very large volume of gas. The volume fraction of gas in a foam frequently exceeds 80 percent and stable foams up to 99 percent volume fraction are not uncommon. Recent reviews of foam flow phenomena and mechanisms are given in Refs. 9-11.

Laboratory studies on foam generation and transport have aided greatly in formulating and improving both our microscopic and macroscopic understanding of foam flow in porous media. They have focused, for the most part, on one-dimensional and homogeneous porous media. These studies, however, leave gaps in our knowledge of foam behavior because the field situation is primarily heterogeneous and multidimensional.

While much work has been conducted in homogeneous systems, the literature on flow in stratified systems is sparse. Notable experiments in stratified systems include Casteel and Djabbarah who performed steam and CO₂ displacements with foaming agents in two parallel porous media.¹² Robin studied foam generation and transport in layered beadpacks that simulated reservoir strata.¹³ In these experiments he surmised that foam blocked the high permeability layer. Llave et al observed that foam can divert gas flow from high permeability layers to low permeability layers when the layers are isolated.¹⁴ Yaghoobi and Heller studied CO₂ foam in short composite cores composed of sand and sandstone.¹⁵ They report diversion of CO₂ to the low permeability section and delay in CO₂ breakthrough from the high permeability section.

More recently, Hirasaki et al. performed foam displacements in layered porous media to study the removal of organic liquids from groundwater aquifers.⁷ Gas was injected at a fixed pressure gradient rather than a specified rate. By dyeing the various fluids, they observed displacement patterns directly. They found that injection of gas slugs into a porous medium containing surfactant resulted in foam generation and selective mobility reduction in the high permeability layers. In turn, recovery of the organic liquid was greatly enhanced.

For the most part, the effect of flow among parallel layers in capillary communication has not been investigated. We denote this as cross flow. To bridge this gap, we perform foam flow experiments in an axially symmetric, cylindrical, heterogeneous porous

medium. A sandstone core fills the center of the heterogeneous pack and a uniform sand fills the annular region between the core and the pressure vessel wall. By the use or absence of a heat-shrink Teflon jacket around the sandstone, fluid communication, or cross-flow, is prohibited or allowed. The sandstone is about two orders of magnitude less permeable than the sand to provide a strong permeability and capillary pressure contrast. We interpret the experiments in terms of the evolution of in-situ water saturation as a function of time.

In the following sections we first discuss the construction and characterization of the porous medium, characterization of our foamer solution, and then experimental determination of water saturation via X-ray CT scanning. CT provides accurate resolution of the progress of displacement fronts as well as the in-situ displacement efficiency. Next, we outline the experimental program and give results. A discussion and conclusions complete the paper.

Experimental Apparatus and Procedures

The centerpiece of the experimental program is a heterogeneous porous medium. It is constructed by centering a 0.050 m diameter and 0.368 m long Fontainebleau sandstone core inside a long, cylindrical, acrylic core holder. The inner diameter of the acrylic tube is 0.089 m and total length is 0.65 m. The annular space between the sandstone and the coreholder wall is packed with Ottawa sand. Similarly, the remaining 0.282 m of coreholder is packed with sand. Hence, the tube contains a heterogeneous portion consisting of sandstone and sand, and a homogeneous portion filled with sand. For experiments where we seek to prevent exchange of fluid between the sandstone and the sand, a heat-shrink Teflon tube is fit around the sandstone. In this case, only the circular faces at the beginning and end of the Fontainebleau core are left open to flow. When cross-flow is not prevented, we make no special preparation of the sandstone core.

Figure 1 is a reconstructed image of the porosity field measured using the CT scanner. Further details on the scanner and imaging methods will be given later, but the sandstone and sand portions of the porous medium are labeled and apparent in Fig. 1. Average porosity values obtained with CT along both the sand and sandstone portions of the core are displayed in Fig. 2. Each component of the heterogeneous porous medium is relatively homogeneous with an average sandstone porosity of 0.14 and an average sand porosity of 0.32. The Ottawa sand permeability was measured at 6.7 μm^2 and the Fontainebleau sandstone permeability is 0.1 μm^2 . Both values are permeability to brine. Thus, the

contrast in permeability between the sand and sandstone is 67 to 1.

The coreholder is designed so that fluids can be injected from either end. Machined endcaps distribute injected fluids across the cross-sectional area of the coreholder. The endcaps are cylindrical, fit inside the coreholder, and seal against the acrylic wall using a double o-ring assembly. Endcaps are held in place by a flange and bolts. Thus, a relatively uniform compression can be applied to the sand face. When gas is injected into the sand-filled region, foam can generate before reaching the heterogeneous zone. On the other hand, when gas is injected directly into the heterogeneous end, foam generation in an initially liquid-filled heterogeneous porous medium can be observed. Unfortunately, coreholder design precluded the installation of pressure taps to measure the pressure at the inlet face and the in-situ pressure profile.

Nitrogen (N_2) is injected into the porous medium saturated with foamer solution at a rate of 3 SCCM (standard cubic centimeters per minute) using a Matheson (Montgomeryville, PA) Model 8141 0-10 SCCM mass flow controller. Nominally, the superficial gas velocity is 0.67 m/day (8.04×10^{-6} m/s) relative to the outlet pressure of 101325 Pa. Foamer solution was not injected simultaneously with the gas because such experiments in one-dimensional, homogeneous sandstones can lead to pressure drops on the order of 2000 kPa/m (100 psi/ft) (c.f. Ref. ¹⁶), and we did not wish to over-pressurize the experimental system. Previous one-dimensional experiments using gas only injection and a similar surfactant and sand resulted in moderate pressure drops.¹⁷

Through the combination of communicating and noncommunicating heterogeneities and injection into homogeneous or heterogeneous portions of the core, 4 different experiments are possible:

1. Injection across the heterogeneous side, noncommunicating sand and sandstone.
2. Injection across the heterogeneous side, communicating sand and sandstone.
3. Injection across the homogenous sand, noncommunicating sand and sandstone.
4. Injection across the homogenous sand, communicating sand and sandstone.

The surfactant is an alpha olefin sulfonate (AOS C14-C16) supplied by Shell Chemical (Houston, TX). A concentration of 0.1% by weight active surfactant in a 0.5 wt% NaCl brine was chosen via a two-pronged procedure. First, the pressure drop across a small (0.01

m diameter and 0.25 m long) chromatography column packed with the Ottawa sand was measured for increasing surfactant concentrations for a variety of brines. Nitrogen was injected at a rate of 10 SCCM while the surfactant solution injection rate was 0.34 cm^3/min . A backpressure of 696 kPa (100 psi) was maintained at the sandpack outlet. Next, the interfacial tension between air and surfactant solution was measured for increasing concentration of surfactant using the pendant drop method. A broad range of surfactant concentrations was used in order to identify the critical micelle concentration (CMC). Figure 3 presents both the pressure drop and surface tension data collected. For the 0.5 wt% brine, strong foaming action, as witnessed by pressure drop, is found once the CMC is exceeded. Further, no benefit is found when surfactant concentration is increased above 0.1 wt% in either the pressure-drop or surface-tension data. Hence, this 0.1 wt% surfactant concentration was chosen.

An experiment begins by flushing the porous medium with gas (CO_2 or N_2) until it is free of liquid. A dry scan is made. If needed, the core is refilled with CO_2 , and saturated with brine to 100% water saturation, because the CO_2 is soluble in the brine. At the minimum, 10 pore volumes of surfactant solution are injected to saturate the solid and aqueous phases with surfactant. A wet scan of the porous medium is then obtained. Once the porosity field has been determined, CT is used to confirm that 100 % liquid saturation is obtained. If it has not, more foamer solution is injected to dissolve the CO_2 . Once an S_w of 100% is obtained, N_2 is injected at a constant rate. The progress of foam generation is seen directly in the CT data collected. Gas is injected until foam breakthrough at the outlet occurs and is continued for an additional 0.5 to 1. PV.

Porosity and aqueous-phase saturation fields are measured on 18 volume sections equally distributed along the core using a fourth generation (1200 fixed detectors) Picker™ 1200 SX X-ray scanner. Typically, the voxel dimension is (0.25mm by 0.25mm by 4mm). The acquisition time of one image is 7 seconds while the processing time is around 40 seconds. The total time of measurement is short enough to capture accurately the position of the front and construct saturation profiles along the core. Porosity and saturation measurements require so-called "dry" and "wet" scans to obtain CT numbers for the medium free of liquid and then fully saturated with liquid. Porosity is computed by a ratio between the dry and wet counting (CT numbers) from the known air attenuation. Saturation is determined in

the same way after garnering CT numbers during the displacement. The measurement accuracy depends on the different parameters chosen for the X-ray emitter such as voltage and intensity, filters, and reproducibility of the positioning system moving the coreholder apparatus into and out of the gantry. Measurement error of saturation is less than 6%^{18,19}.

Results

To provide a baseline, gas was injected at 3 SCCM into surfactant-free cores. Figure 4 illustrates gas flow in the absence of foam when heterogeneities communicate. Injection is from the heterogeneous face. Saturation values are obtained by averaging the saturation for each voxel in a cross section for the sandstone and sand regions, respectively. Time is reported nondimensionally in pore volumes of injected gas and distance is measured from the inlet. Figure 4a shows the saturation profiles for the sandstone, while Fig. 4b the profiles for the sand. Straight lines connect the individual data points. Gas displacement of water is poor. Strong displacement fronts are not witnessed and even after 4 PV of gas injection, water saturation remains high. In this case, gas breakthrough occurs at 0.05 PV and the total production at 4.5 PV is about 0.1 PV. In the case of noncommunicating heterogeneities, gas breakthrough occurs at 0.11 PV and total water production at 3 PV is only 0.17 PV.

Next, foam displacements in all four of the configurations were completed. First injection through the heterogeneous face of the core is discussed and then injection through the homogeneous face. The extent of foam generation and reduction in gas mobility is quantified through the saturation profiles determined by CT and the length of time to gas breakthrough. A breakthrough time approaching or exceeding 1 PV indicates efficient desaturation of the core and by implication strong foam generation.

Noncommunicating layers, heterogeneous face injection. In this experiment, the heat-shrink wrapped Fontainebleau sandstone core cannot exchange fluid with the sand except through the inflow and outflow ends. N_2 is injected from the heterogeneous region and across both sand and sandstone. Figure 5 displays typical saturation fields across the cross section of the core holder at various locations. The elapsed time is 0.23 PV of gas injection. In the figure, a shading of black represents a water saturation, S_w , of 1 while white indicates S_w equal to 0. In the last section, taken at x/L equal to 0.25 (0.16 m), it is evident that gas is present in the low permeability sandstone, but has not yet reached

this location in the sand. The light shading of the first two cross sections in Fig. 5 indicates that an effective displacement is occurring.

As mentioned in the discussion of the experimental setup, collection of in-situ pressure drop data is precluded by the design of the core holder. However, gas pressure drop over the entire core did not exceed 1.4×10^5 Pa (20 psi) in this or any of the subsequent experiments employing surfactant.

Figure 6 presents the averaged aqueous-phase saturation profiles measured at various times. Figure 6a shows the saturation profiles for the sandstone, while Fig. 6b the profiles for the sand. The 18 data points for the sand region correspond to the number of volume sections analyzed.

The first point to note is that desaturation of both the sand and sandstone by foamed gas is efficient. Saturation fronts are relatively steep and sharp in both porous media. The entire porous medium is saturated with foamer solution initially, and S_w downstream of the front is everywhere 1. In the case of the sandstone, S_w immediately following the passage of the foam front is about 0.3 whereas in the sand it is roughly 0.15. That is, following desaturation by foam, each layer is within a few saturation units of its irreducible saturation. Foam has effectively diverted a portion of the injected gas into the sandstone.

Secondly, the position of the saturation fronts as a function of time tells an interesting story. When t equals 0.12 PV the leading edge of the foam front in both the high and low permeability media is at a dimensionless position of 0.14. At the succeeding time of 0.23 PV the front in the sandstone is at x/L equal to 0.3, while in the sand it is slightly behind at x/L equal to 0.26. At a t of 0.35 PV, it appears that the foam front has just exited the sandstone. The leading edge of the foam front in the sand at this same time significantly lags behind at a position of 0.35. Examination of the saturation profiles at 0.46 PV confirms this trend. The saturation at x/L equal to 0.45 in the sandstone clearly shows that foam has pushed its way through the sandstone core and exited into the sand. On the other hand, the foam front pushing through the annular sand region is only positioned at x/L equal to 0.4. It is apparent that the foam front in the low permeability layer moves more quickly than in the high permeability layer. This fact is also evidenced in the raw saturation data in Fig. 5. In the discussion, we will rationalize this behavior in terms of the effect of capillary pressure on foam texture.

At times of 0.46, 0.54, and 0.70 PV the effect of the foam fronts moving at different speeds in each media

can still be seen in the saturation profiles. The foamed gas exiting the sandstone results in fronts positioned in the neighborhood of x/L from roughly 0.75 to 0.85. A second trailing front is also apparent. For instance, at 0.70 PV this trailing front is positioned at roughly 0.45. This second front results from the foamed gas and water exiting the annular region packed with sand as well as water that was not displaced in the completely sand-filled region by the first foam front. By 1.3 PV, the average aqueous-phase saturation in the sand is about 0.10 and desaturation is complete.

Communicating layers, heterogeneous face injection.

The configuration is similar to the first case, except that the heat-shrink Teflon is absent from the sandstone. The sandstone and the sand are thereby in capillary communication and free to exchange fluid along the length of the sandstone. Figure 7 shows typical saturation fields derived from CT data. Again a white shading indicates no water and black indicates fully water saturated. The saturation images for dimensionless distances of 0.29 and 0.35 (0.19 and 0.225 m) show that the water saturation distribution is fairly low and uniform, but not equal in each porous medium. From the saturation distributions at x/L equal to 0.40 and 0.45 (26 and 29.5 cm), it is obvious that the foam front lies between these positions. It also appears that the foam in both porous media moves at roughly the same velocity. This point becomes clear when saturation profiles along the length of the core are plotted.

Figure 8 presents the results of the displacement in a fashion similar to that in Fig. 6. Sandstone saturation profiles are given in Fig. 8a and sand profiles in Fig. 8b. Again, strong and steep displacement fronts in both the sandstone and sand are witnessed. Foam effectively desaturates each layer. In contrast to Fig. 6, the displacement fronts in each porous medium move at the same rate. At times of 0.25, 0.33, and 0.41 PV, the front position is 0.3, 0.4, and 0.45, respectively. At the shortest time of 0.08 PV, displacement fronts do not coincide exactly. This is likely due to differing rates of net foam generation near the beginning of each region.

Displacement fronts that move with identical velocity in communicating heterogeneous zones is also the expected result for the propagation of unfoamed gas²⁰ if the effect of gravity is minimal. The striking result is the degree of displacement in the low permeability sandstone.

Noncommunicating layers, homogeneous face injection. In this the third case, gas is injected into the saturated porous medium from the side completely filled with sand. Flow across the cylindrical face of the sandstone is again prohibited by the Teflon jacket barrier.

Figure 9 shows the saturation profiles. In the initial homogeneous section, gas enters the sand, a foam is generated, and displacement of foamer solution from the sand is quite efficient. Foam generation is rapid as a strong displacement front is seen even at 0.15 PV. At roughly 0.54 PV, the foamed gas first enters the sandstone. Again, the foamed gas moves more quickly through the low permeability sandstone than it does through the more permeable sand. The leading edge of the displacement front is at x/L equal to 0.7 in the sandstone and at 0.6 in the sand after 0.7 PV of total gas injection. At 0.85 PV of injection, front positions are roughly 0.85 and 0.75 in the sandstone and sand, respectively. Consistent with this observation, gas breakthrough occurs from the sandstone first.

Initially, foam displacement in the sandstone is not as efficient as in the earlier two cases. Examining the saturation profiles at 0.7, 0.85, and 1.1 PV, we find that the height of the displacement front grows from 0.3 to over 0.6 saturation units.

Communicating layers, homogeneous face injection.

Figure 10 reports the saturation profiles obtained during gas injection into the homogeneous sand-filled side with capillary communication between the sandstone and sand. Displacement from both the sand and the sandstone is excellent at all time levels. As in the second case where cross flow occurs, saturation fronts move at the same speed in each layer. It is seen by comparing Figs. 10a and 10b that front positions at times of 0.63, 0.74, and 0.88 PV are equal and correspond to x/L equal to 0.65, 0.75, 0.85, respectively.

Discussion

Prior to discussing the flow behavior measured here, it is helpful to consider briefly the integrity of the teflon jacket. If stretched, gas might channel between the sandstone and the jacket. Fortunately, there is no evidence of such channeling. In Fig. 6 (a) it is apparent that the foam front exits the sandstone section at roughly 0.46 PV. Careful examination of Fig. 6(b) reveals no gas saturation downstream of the sandstone prior to 0.46 PV. If channeling between the jacket and sandstone had occurred, gas would have appeared quickly in the sand

section downstream of the sandstone, and it is likely that the strong displacement fronts found in the sandstone would not have developed as gas channeled around the sandstone.

The foam flow behavior discovered in each of these cases can be rationalized by considering foam texture (i.e., bubble size) and the effect that porous medium capillary pressure plays in setting foam texture. It is well established that finely textured foams, that is foams with small average bubble size, present a larger resistance to flow than do coarsely textured foams.^{21,22} In turn, bubble size is dictated by pore-level foam generation and coalescence events.¹¹ Foam generation is largely a mechanical process and independent of surfactant formulation and concentration. On the other hand, foam collapse depends strongly on the foamer solution. Foam lamellae are stable provided that surfactant can stabilize the gas-liquid interface against suction capillary pressure that seeks to thin foam films. Hence, it is expected that the rate of foam coalescence increases with porous medium capillary pressure. In turn, the foam becomes more coarse. The characteristic value of capillary pressure that a porous medium approaches during strong foam flow is referred to as the limiting capillary pressure, and it is set principally by surfactant type and concentration.²³

In the first case reported in Fig. 7, each porous medium accepts whatever portion of the injected gas it desires. Because the sand and sandstone are isolated in the sense that cross flow is not allowed, foam is generated and assumes a texture that is independent of events occurring in the adjacent layers. We surmise that in the low permeability sandstone foam is not as finely textured as in the sand. One explanation is that sandstone capillary pressure is larger for a given saturation due to lower medium permeability. This statement follows from applying the Leverett J-function²⁴ to both media, using the appropriate surface tension from Fig. 4, and choosing aqueous-phase saturations of 0.35 for the sandstone and 0.18 for the sand (i.e., approximate saturations in the swept zone prior to foam exiting the sandstone). Thus, it appears that capillary pressure is about 5 times larger in the sandstone than the sand. Since capillary pressure is greater in the sandstone, a greater suction pressure is exerted on foam lamellae inducing foam coalescence. This in turn leads to a more coarsely textured foam that is more mobile than its finely textured counterpart in the sand.

Another explanation is that foam in the low permeability sandstone is simply not generated to any extent. Friedmann et al ²⁵, for example, describe a

weak foam that they attribute to the "leave-behind" mechanism. In foam generation only by leave-behind, very few, if any, flowing foam bubbles are ever generated and gas mobility is reduced by a factor of about 10 or less ^{25,26}. In this manner, foam generates in the sand region reducing gas mobility while there is very little foam in the low permeability sandstone and, hence, little mobility reduction. A problem with this explanation, however, is that rapid and strong foam generation should have ensued at the permeability discontinuity between the sandstone and sand when the weak foam arrived at the end of the sandstone region. This foam generation would have plugged the outlet of the sandstone, but no evidence of such plugging is found.

With either explanation, it is clear that gas mobility in the sand has been reduced substantially, foam is more finely textured in the sand than it is in the sandstone, and water displacement from both regions is more efficient than gas displacement without surfactant. Hence, foamed gas progresses more rapidly through the low permeability porous media. In either instance, gas mobility has been lowered substantially and water displacement is efficient.

In the situation summarized in Fig. 8, the porous media communicate with each other across the long cylindrical interface of the sandstone. Since the porous media communicate, gas at the foam front minimizes its flow resistance via foam bubble texture. When the flow resistance in the sandstone increases, foamed gas diverts into the sand, and vice versa. Thus, foam propagates at an equal rate in each layer because the saturation fronts are bound together by the necessity to maintain the minimum flow resistance. Bubble textures in sand and sandstone are not expected to be identical, nor are capillary pressures. However, conditions yield nearly identical gas mobility in both the sand and sandstone.²⁰ In this case, the mobility is low and promotes effective desaturation. In the cases where the heterogeneous section is at the end of the core, similar explanations follow.

When cross flow is prohibited between the sand and the sandstone, each porous medium sets foam texture and, hence, gas mobility independently. Capillary pressure in the low permeability sandstone is higher than in the sand, coalescence ensues, and foam in the sandstone is more mobile than in the sections filled with sand. If cross flow occurs, gas mobility is balanced because high flow resistance forces gas to divert into the adjacent layer. Foam propagates at equal rates in each layer.

Interestingly, the transient experimental results

shown here bear striking qualitative similarity to previous simulation of foam behavior in heterogeneous porous media reported elsewhere.²⁰ We note that the preceding qualitative arguments bear strong resemblance to the quantitative results reported in that study.

Conclusions

An experimental study of foam generation and propagation in heterogeneous porous media using Fontainebleau sandstone and Ottawa sand was performed. The contrast in permeability between high and low permeability homogeneous zones was 67 to 1. Despite this drastic permeability contrast and despite the fact that high permeability zones typically lead to gas channeling, foamed gas is diverted to low permeability channels in these experiments. In this regard, the foam generated in these experiments can be regarded as strong. Foam effectively desaturates both the high permeability and low permeability portions of the experimental setup and desaturation is complete following roughly 1 PV of gas injection. This result is found for systems that both permit and prohibit cross flow.

Foam in heterogeneous systems appears to be self regulating in that gas mobilities in each porous medium equalize when layers communicate and nearly equalize in noncommunicating systems. Similar efficient diversion and desaturation is likely in heterogeneous, layered field situations provided that foam is tolerant of the presence of oil. Also, the capillary pressure of each layer must be less than the critical capillary pressure for foam coalescence.

In general, it is observed that when permeability heterogeneities communicate and there is fluid cross flow that foam displacement fronts move at equal velocity in each zone. When cross flow is prohibited by an impermeable barrier, rapid foam propagation and desaturation of the low permeability zone is witnessed.

Nomenclature

L = length of core holder, L, m

S_w = aqueous phase saturation, dimensionless

x = axial position, L, m

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Figure 1: Reconstruction of porosity distribution in heterogeneous porous medium.

Figure 2: Average sand and sandstone porosity along the length of the heterogeneous core.

Figure 3: Pressure drop and surface tension for foamer solution as a function of surfactant concentration. Brine concentration is 0.5 wt%.

(a)

(b)

Figure 4: Transient aqueous phase saturation profiles for gas-only displacement with cross flow: (a) sandstone and (b) sand regions.

Figure 5: Water saturation profiles in heterogeneous core section, $t=0.23$ PV. Cross flow is not permitted.

(a)

(b)

Figure 6: Transient aqueous phase saturation profiles for displacement without cross flow: (a) sandstone and (b) sand regions.

Figure 7: Water saturation profiles in the heterogeneous core section, $t = 0.49$ PV. Cross flow is permitted.

(a)

(b)

Figure 8 Transient aqueous phase saturation profiles for displacement with cross flow: (a) sandstone and (b) sand regions.

(a)

(b)

Figure 9: Transient aqueous phase saturation profiles for displacement without cross flow: (a) sandstone and (b) sand regions.

(a)

(b)

Figure 10: Transient aqueous phase saturation profiles for displacement with cross flow: (a) sandstone and (b) sand regions.

3.2 MECHANISTIC FOAM FLOW SIMULATION IN HETEROGENEOUS AND MULTIDIMENSIONAL POROUS MEDIA

(Anthony R. Kavscek, Tadeusz W. Patzek, and Clayton J. Radke)

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Mechanistic Foam Flow Simulation in Heterogeneous and Multidimensional Porous Media

(Anthony R. Kavscek, Tadeusz W. Patzek, and Clayton J. Radke)

Gases typically display large flow mobilities in porous media relative to oil or water, thereby impairing their effectiveness as displacing fluids. Foamed gas, though, is a promising agent for achieving mobility control in porous media. Because reservoir-scale simulation is a vital component of the engineering and economic evaluation of any enhanced oil recovery (EOR) or aquifer remediation project, efficient application of foam as a displacement fluid requires a predictive numerical model. Unfortunately, no such model is currently available for foam injection in the field where flow is multidimensional and the porous medium is heterogeneous.

We have incorporated a conservation equation for the number density of foam bubbles into a fully implicit, three-dimensional, compositional, and thermal reservoir simulator and created a fully functional, mechanistic foam simulator. Because foam mobility is a strong function of bubble texture, the bubble population balance is necessary to make accurate predictions of foam-flow behavior. Foam generation and destruction are included through rate expressions that depend on saturations and surfactant concentration. Gas relative permeability and effective viscosity are modified according to the texture of foam bubbles. In this paper, we explore foam flow in radial, layered, and heterogeneous porous media. Simulations in radial geometries indicate that foam can be formed deep within rock formations, but that the rate of propagation is slow. Foam proves effective in controlling gas mobility in layered porous media. Significant flow diversion and sweep improvement by foam are predicted, regardless of whether the layers are communicating or isolated.

Introduction

Field application of foam is a technically viable enhanced oil recovery (EOR) process as demonstrated by steam-foam field studies¹⁻⁴ and recent nonthermal application of foam in the Norwegian sector of the North Sea^{5, 6}. Foam can mitigate gravity override of oil-rich zones and/or selective channeling through high permeability streaks, thereby improving volumetric displacement efficiency.

For example, Patzek and Koinis¹ report that two different pilot studies in the Kern River Field (Kern Co., California) showed major incremental oil-recovery response after about two years of foam injection. They report increased production of 5.5 to 14% of the original oil in place (OOIP) over a five year period. Djabbarah *et al.*² report improved vertical and areal sweep efficiency in the South Belridge Field (Kern Co., California) and an incremental oil production of 183,000 B from two, contiguous, ten-acre, inverted, nine-spot patterns following

a foam injection period of about 1 year. Friedmann *et al.*³ report more even injection profiles with steam foam, *in-situ* foam generation, and foam propagation in rock formations.

Efficient application and evaluation of candidates for foam EOR processes, though, requires a predictive numerical model of foam displacement. A mechanistic model would also expedite scale-up of the process from the laboratory to the field. No mechanistic, field-scale model for foam displacement is currently in use.

The population-balance method for modeling foam in porous media^{7, 8} is mechanistic and incorporates foam into reservoir simulators in a manner that is analogous to energy and species mass balances^{7, 8}. Accordingly, a separate conservation equation is written for the concentration of foam bubbles. This simply adds another component to a standard n-component compositional simulator.

Until recently, the population balance method has only been used to model steady-state results in glass beadpacks⁷ and Berea sandstone⁹ or to predict transient flow^{10, 11}, but not both. Previously, we presented the results of an extensive experimental and simulation study of transient and steady foam flow in one-dimensional porous media^{12, 13}. This initial work detailed the development of a mechanistic model for foam displacement that was easily implemented, fit simply into the framework of current reservoir simulators, and employed a minimum of physically meaningful parameters. Propagation of foam fronts within

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Boise sandstone was tracked experimentally and simulated successfully under a variety of injection modes and initial conditions^{13, 14}.

This paper extends our foam displacement model to multidimensional, compositional, and nonisothermal reservoir simulation. For numerical stability and to accommodate the long time steps necessary for successful reservoir-scale simulation, a fully implicit backward-differencing scheme is used. The simulator employs saturation and surfactant concentration dependent rate expressions for lamella formation and destruction. Lamella mobilization is similarly included.

Our objectives are to show that not only is population-balance-based simulation of foam displacement possible in multidimensional heterogeneous porous media, but also highly instructive in regard to the physics of foam displacement. We consider geometrically simple, isothermal, oil-free systems. This allows easy comparison with our previous experimental and simulation results^{12, 13}. Numerous verification exercises are performed to discover the role foam plays in gas displacement through zones of contrasting permeability and to highlight the interplay of foam-bubble texture and gas mobility.

Foam in Porous Media. Foam microstructure in porous media is unique¹⁵. Accordingly, to model gas mobility it is important to understand foamed-gas microstructure¹². In water-wet porous media, the wetting surfactant solution remains continuous, and the gas phase is dispersed. Aqueous liquid completely occupies the smallest pore channels where it is held by strong capillary forces, coats pore walls in the gas-filled regions, and composes the lamellae separating individual gas bubbles. Only minimal amounts of liquid transport as lamellae. Most of the aqueous phase is carried through the small, completely liquid-filled channels. Gas bubbles flow through the largest, least resistive pore space while significant stationary bubbles reside in the intermediate-sized pore channels where the local pressure gradient is insufficient to sustain mobilized lamellae.

Foam reduces gas mobility in two manners. First, stationary or trapped foam blocks a large number of channels that otherwise carry gas. Gas tracer studies^{11, 16} show that the fraction of gas trapped within a foam at steady state in sandstones is quite large and lies between 85 and 99%. Second, bubble trains within the flowing fraction encounter significant drag because of the presence of pore walls and constrictions, and because the gas/liquid interfacial area of a flowing bubble is constantly altered by viscous and capillary forces^{17, 18}. Hence, foam mobility depends strongly on the fraction of gas trapped and on the texture or number density of foam bubbles.

Bubble trains are in a constant state of rearrangement by foam generation and destruction mechanisms¹⁵. Individual foam bubbles are molded and shaped by pore-level making and breaking processes that depend strongly on the porous medium^{9, 15}. To account for foam texture in a mechanistic sense, foam generation and coalescence must be tracked directly. Additionally, bubble trains halt when the local pressure gradient is insufficient to keep them mobilized, and other trains then begin to flow. Bubble trains exist only on a time-averaged sense. More detailed summaries of the pore-

level distribution of foam, and the mechanisms controlling texture are given in refs. 12 and 15.

Modeling Foam Displacement

A variety of empirical and theoretical methods for modeling foam displacement are available in the literature. These range from population-balance methods^{7, 8, 10, 11, 13, 19, 20} to percolation models²¹⁻²⁵ and from applying so-called fractional flow theories^{26, 27} to semi-empirical alteration of gas-phase mobilities^{1, 28-34}. Of these four methods, only the population balance method and network or percolation models arise from first principles.

Population-Balance Method. The power of the population-balance method lies in addressing directly the evolution of foam texture and, in turn, reduction in gas mobility. Gas mobility is assessed from the concentration or texture of bubbles. Further, the method is mechanistic in that well-documented pore-level events are portrayed in foam generation, coalescence, and constitutive relations. Most importantly, the population balance provides a general framework where all the relevant physics of foam generation and transport may be expressed.

We chose the population-balance method because of its generality and because of the similarity of the equations to the usual mass and energy balances that comprise compositional reservoir simulation. Only a brief summary of the method is given here as considerable details of our implementation are available in the literature¹²⁻¹⁴.

The requisite material balance on chemical species i during multiphase flow in porous media is written as

$$\frac{\partial}{\partial t} \left[\phi \sum_j (S_j C_{i,j} + \Gamma_{i,j}) \right] + \sum_j \nabla \cdot \bar{F}_{i,j} = \sum_j q_{i,j} \quad , \quad (1)$$

where ϕ is the porosity, S is the saturation of phase j , C is the molar concentration of species i in phase j , Γ is the adsorption of species i from phase j in units of moles per void volume, $\bar{F}$ is the vector of combined convective and diffusive flux of species i in phase j , and q is a rate of generation of i in phase j per unit volume of porous medium. To obtain the total mass of species i , we sum over all phases j .

In the foam bubble population balance, $S n_f$ replaces $S_j C_{i,j}$ where n_f is the number concentration or number density of foam bubbles per unit volume of flowing gas and S_f is the saturation of flowing gas. Hence, the first term of the time derivative is the rate at which flowing-foam texture becomes finer or coarser per unit rock volume. Since foam partitions into flowing and stationary portions, Γ becomes $S n_t$ where S_t and n_t are the saturation of the stationary gas and the texture of the trapped foam per unit volume of trapped gas, respectively. Thus, the second term of the time derivative gives the net rate at which bubbles trap. Trapped and flowing foam saturation sum to the overall gas saturation, $S_g = S_f + S_t$. The second term on the left of Eq. (1) tracks the convection of foam bubbles where the flux, $\bar{F}$, is given by $u_f n_f$, and u_f is the Darcy velocity of the flowing foam. We neglect dispersion. Finally, q becomes the net rate of generation of foam bubbles. Within the above

framework, foam is a component of the gas phase and the physics of foam generation and transport become amenable to standard reservoir simulation practice.

The net rate of foam generation:

$$q_f = \phi S_g \left[k_1 |\vec{v}_w \parallel \vec{v}_f|^{1/3} - k_{-1} |\vec{v}_f| n_f \right] \quad (2)$$

is written per unit volume of gas. In the simulations to follow, we do not inject pregenerated foam and so we do not require a source/sink term for bubbles. Interstitial velocities, *i.e.*, $\vec{v}_i = \vec{u}_i / \phi S_i$, are local vector quantities that depend on the local saturation and total potential gradient, including gravity and capillary pressure. Foam generation is taken as a power-law expression that is proportional to the magnitude of the flux of surfactant solution multiplied by the 1/3 power of the magnitude of the interstitial gas velocity. The liquid-velocity dependence originates from the net imposed liquid flow through pores occupied by both gas and liquid, whereas the gas-velocity dependence arises from the time for a newly formed lens to exit a pore³⁵. Snap-off is sensibly independent of surfactant properties consistent with its mechanical origin¹⁵. The proportionality constant reflects the number of foam germination sites. Intuitively, the number of snap-off sites falls with decreasing liquid saturation. However, k_1 is taken as a constant here. The generation rate expression does vary implicitly with liquid saturation through the gas and liquid velocities.

To prevent coalescence of newly formed gas bubbles, a surfactant must stabilize the gas/liquid interface. Foam lamellae form given sufficient suction capillary pressure and a stabilizing surfactant. However, too high of a suction-capillary pressure will overcome the stabilizing influence of surfactant and collapse a lamella¹⁵. A flowing lamella is vulnerable to breakage in termination sites as it flows into a divergent pore space where it is stretched rapidly. If sufficient time does not elapse for surfactant solution to flow into a lamella and heal it, coalescence ensues³⁶.

Equation (2) shows that foam lamellae are destroyed in proportion to the magnitude of their interstitial flux, $v_f n_f$, into such termination sites. The coalescence rate constant, k_{-1} , varies strongly with the local capillary pressure and surfactant formulation. It is given by

$$k_{-1} = k_{-1}^o \left(\frac{P_c}{P_c^* - P_c} \right)^2, \quad (3)$$

where the scaling factor, k_{-1}^o , is taken as a constant and P_c^* is the limiting capillary pressure for foam coalescence³⁷.

The "limiting capillary pressure," P_c^* , as identified by Khatib *et al.*³⁷, refers to the characteristic value of capillary pressure that a porous medium approaches during strong foam flow. It is set primarily by surfactant formulation and concentration. Highly concentrated foamer solutions and robust surfactants lead to a high P_c^* . In situations where surfactant transport is transient, we expect P_c^* to vary locally with surfactant concentration. The experimental work of Aronson *et al.*³⁸ suggests the following functional form for P_c^* versus surfactant concentration of robust foamer solutions:

$$P_c^* = P_{c, \max}^* \left(\frac{C_s}{C_s^o} \right) \quad (4)$$

where $P_{c, \max}^*$ is a limiting value for P_c^* and C_s^o is a reference surfactant concentration for strong net foam generation.

In the simulations of heterogeneous porous media to follow, we assume that P_c^* is independent of absolute permeability. Foam-lamella coalescence is determined mainly by the rupture capillary pressure of isolated lamellae which, in turn is set by the concentration and type of surfactant, and not the nature the porous medium³⁶. Equations (3) and (4) correctly predict that at high capillary pressures or for ineffective foamer solutions k_{-1} is quite high^{37, 38}. The foam coalescence rate approaches infinity as the porous medium capillary pressure approaches P_c^* . We also assume geometric similarity between layers of differing permeability. Thus, for a uniform liquid saturation in the heterogeneous medium, foam is more vulnerable to breakage in the low permeability zones because P_c scales inversely as the square root of the absolute permeability according to the Leverett J-function³⁹.

In addition to bubble kinetic expressions, the mass balance statements for chemical species demand constitutive relationships for the convection of foam and wetting liquid phases. Darcy's law is retained, including standard multiphase relative permeability functions. However, for flowing foam, we replace the gas viscosity with an effective viscosity relation for foam. Since flowing gas bubbles lay down thin lubricating films of wetting liquid on pore walls, they do not exhibit a Newtonian viscosity. We adopt an effective viscosity relation that increases foam effective viscosity as texture increases, but is also shear thinning

$$\mu_f = \mu_g + \frac{\alpha n_f}{|\vec{v}_f|^{1/3}} \quad (5)$$

where α is a constant of proportionality dependent mainly upon the surfactant system. In the limit of no flowing foam we recover the gas viscosity. This relation is consistent with the classical result of Bretherton⁴⁰ for slow bubble flow in capillary tubes (see also, refs. 17, 18, 41, 42).

Finally, stationary foam blocks large portions of the cross-sectional area available for gas flow and, thus must be accounted for to determine gas flux. Since the portion of gas that actually flows partitions selectively into the largest, least resistive flow channels, we adopt a "Stone-type" relative permeability model⁴³ that, along with effective viscosity, specifies gas-phase flow resistance. Because wetting aqueous liquid flows in the smallest pore space, its relative permeability is unaffected by the presence of flowing and stationary foam in accordance with the experimental results of refs. 44-48. Since flowing foam partitions selectively into the largest pore space, the relative permeability of the nonwetting flowing gas is a function of only S_f . Consequently, gas mobility is much reduced in comparison to an unfoamed gas propagating through a porous medium, because the fraction of gas flowing at any instant is quite small^{11, 16}.

Compositional Foam Simulator

Our starting point for multidimensional foam simulation is M³NOTS (Multiphase Multicomponent Nonisothermal Organics Transport Simulator), a nonisothermal, n-component, compositional simulator capable of handling three-phase flow in response to viscous, gravity, and capillary forces⁴². It is a compositional extension of TOUGH2^{43, 44} and uses the integral finite difference method (IFDM) to discretize the flow domain (*cf.*, ref. 52).

Spatial gradients are calculated in a manner identical to the classic block-centered finite difference method. Flow mobilities are upstream weighted except for the absolute permeability between blocks of differing permeability. These are based on harmonic weighting.

A cubic equation of state represents the thermodynamic properties of the gas phase, which for nitrogen gas at the temperatures and pressures simulated here reduces to an ideal gas. The method of corresponding states describes the oil phase. The International Steam Tables provide the properties of the aqueous phase⁴⁵. The simulator has been used successfully to model the deposition and clean up of petroleum hydrocarbons from soils and groundwater⁴².

We treat foam bubbles as a nonchemical component of the gas phase. Thus, the additional transport equation for foam-bubble texture described above is added to the mass balances for water, gas, and *n* organic components. The discretized foam-bubble equation is fully implicit with upstream weighting of the gas-phase mobility consistent with all other chemical species. Surfactant is simply an organic component that is highly soluble in water. In each grid block, the magnitude of the vectors representing the interstitial gas and liquid velocities are used to compute foam generation and coalescence rates from Eq. (2). The magnitude of each velocity is obtained by first summing the flow of each phase into and out of a grid block in the three orthogonal directions. Then the arithmetic average for each direction is calculated and the magnitude of the resultant vector used to calculate foam generation and coalescence rates. The gas velocity is similarly computed for the shear-thinning portion of the foam effective viscosity.

Numerical values of the population balance parameters are determined from steady-state measurements in one-dimensional linear flow. Steady-state flow trends, saturation, and pressure drop profiles are matched. These can all be obtained within one experimental run. The suite of foam displacement parameters do not need to be adjusted to accommodate different types of transient injection or initial conditions. Parameter values used here are taken from ref. 13 and apply specifically to very strong foams in the absence of oil and under isothermal conditions.

Numerical Model Results

Because there are numerous initial conditions, modes of injection, and multidimensional geometries of interest, we present the results from several carefully chosen illustrative examples. First, we compare simulator predictions against experimental results for the simultaneous injection of nitrogen and foamer solution into a linear core presaturated with surfactant solution. Second, foam flow in heterogeneous, noncommunicating and communicating linear layers is considered. The layers are again assumed to be presaturated with surfactant. Next, we simulate the one-

dimensional radial flow of foam and consider two different initial conditions. In one case, the porous medium is initially free of surfactant, while in the following case, it is initially saturated with surfactant solution. Here, we focus on the evolution of gas mobility as foam flows outward radially. Finally, we present simulations of surfactant and gas coinjection into a 2.5-acre five-spot pattern that is initially free of surfactant. To avoid confusion between foam formation, surfactant propagation and adsorption, foam-oil interaction, and partitioning of surfactant into the oil phase, we choose a porous medium that does not adsorb surfactant and never contains any oil. In all simulations, the rock is initially filled with the aqueous phase, $S_w = 1$. Nitrogen and foamer solution are coinjected simultaneously. Thus, we focus attention on foam formation, coalescence, transport, and reduction of gas mobility.

Linear Core. In the first example, nitrogen is injected continuously into a linear core of length 0.60 m at a rate of 0.43 m/day relative to the exit pressure of 4.8 MPa. Foamer solution is also injected continuously at 0.046 m/day to give a quality or gas fractional flow of 0.90 at the core exit. The medium is initially filled with surfactant solution. These flow rates and initial conditions correspond exactly to our previous experiments conducted in a 1.3-d Boise sandstone with a length of 0.60 m^{12, 13}. The foamer is a saline solution (0.83 wt% NaCl) with 0.83 wt% active Bioterg AS-40, a C₁₄₋₁₆ alpha olefin sulfonate, available from Stepan Chemical Company.

Fig. 1—Experimental and model transient aqueous-phase saturation profiles for 1-d displacement.

Figures 1 and 2 display the transient experimental and simulated saturation and pressure profiles, respectively. Figure 3 displays the foam texture profiles. Theoretical results are represented by solid lines. Dashed lines simply connect the individual data points. Elapsed time is given as pore volumes of total fluid injected, that is, as the ratio of total volumetric flow rate at exit conditions multiplied by time and divided by the core void volume.

Steep saturation fronts are measured and predicted at all time levels in Fig. 1 whereby aqueous-phase saturation upstream of the front is roughly 30% and downstream it is 100%. Model fronts are somewhat steeper and sharper than those measured experimentally, but the theoretical saturation profiles track experimental results very well. From the

saturation profiles it is apparent that foam moves through the core in a piston-like fashion. Note that the simulation displays little numerical dispersion.

Even though nitrogen and surfactant solution are injected separately, rapid foam generation and liquid desaturation occur at the core inlet. A region of net foam generation near the inlet is clearly evident in the transient pressure profiles of **Fig. 2**. Both the experiments and calculations show that pressure gradients near the inlet are shallow, indicating that flow resistance is small and foam textures are coarse consistent with the injection of unfoamed gas. Steep pressure gradients are found downstream of the inlet region. These steep pressure gradients confirm the existence of a strong foam piston-like front moving through the core.

Fig. 2—Experimental and model transient pressure profiles for 1-d displacement.

Fig. 3—Model transient flowing-foam texture profiles for 1-d displacement.

Figure 3 reports the predicted foam texture as a function of dimensionless distance and time. We find a coarsely textured foam near the inlet. Beyond the first fifth of the core, foam texture becomes very fine and nearly constant at each time level until it rises and peaks immediately before the front. High pressure gradients and fine foam textures are seen where liquid saturation is low and *vice versa*. No method currently exists to measure *in situ* foam texture directly. However, the predicted effluent bubble textures do match the bubble size of foams exiting a similar Berea sandstone^{9, 19}.

One interesting feature of **Fig. 3** is the elevation of foam texture near the foam front above that in steady-state and also that immediately upstream of the foam front. This feature is more pronounced here than in our previous modeling work^{12, 13}. The difference occurs because of the switch from an IMPES (implicit pressure explicit saturation) method with explicit upstream weighting of fluid mobilities to a fully implicit scheme. Physically, foam texture is fine at the foam front because the aqueous-phase saturation increases from roughly 0.30 to 1. For high aqueous-phase saturation, Eq. (3) gives a very low foam coalescence rate. At the same time, interstitial liquid and gas rates are high resulting in a large rate of net foam generation. Setting Eq. (2) to zero and solving for the value of the local equilibrium foam texture indeed shows that texture can be quite high at the foam front. Because this intensive foam generation is confined to a very small region, pressure gradients at the foam front are affected negligibly, as displayed in **Fig. 2**. Further, as this result appears to be physical, we make no attempt to suppress it.

Gas compressibility effects are also found in **Fig. 3**. At steady state, the foam texture decreases along the latter portion of the of the core. As they flow downstream, the small compressible foam bubbles find themselves out of equilibrium with the lower pressure. Consequently, bubbles expand increasing their velocity. This increased velocity triggers increased foam coalescence and a more coarsely textured foam. Gas compressibility similarly accounts for foam textures finer than the steady-state texture upstream of the foam fronts at time levels of 0.65 and 0.80 PV.

In addition to good agreement with experiment, the model results in **Figs. 1 to 3** agree quite well with our previous calculations¹³. Again, the foam displacement parameters employed here and in the previous work are identical. In the remaining simulations, we assume that the fraction of gas flowing in the presence of foam is a constant equal to 0.10. This shortens the computation time required for multidimensional calculations by decreasing the stiffness of the equations. The impact of the increase in foam texture at the foam front on gas mobility is also moderated.

Heterogeneous Noncommunicating Linear Layers. In this section, we consider the case of two linear layers with different permeabilities and without cross flow. Both layers are initially filled with foamer solution as in the previous case. This geometry applies to a reservoir with continuous impermeable shale breaks and to parallel core experiments in the laboratory. The high permeability layer is assigned a permeability of 1.3 d which is identical to the permeability of the Boise sandstone cores used in our laboratory^{12, 13}. The permeability of the second layer is made a factor of 10 smaller, 0.13 d. Each layer is assumed to be geometrically similar and is given the same porosity, Leverett J-function³⁹, and relative permeability functions. Initially, both layers are saturated with aqueous surfactant solution. Superficial velocities maintained in these simulations are the same as in the linear corefloods portrayed in **Figs. 1 - 3**. The system length is set at 0.60 m to allow direct comparison with these corefloods. Continuity of pressure is maintained at the inlet and outlet. Otherwise, each layer accepts whatever portion of the injected fluids it desires.

It is useful to begin by considering the effect that foam has on reappportioning the production from each layer. **Figure 4** displays as solid lines the fraction of the original water displaced from each layer as a function of the time. The small amounts of surfactant solution injected with the gas are not included in the produced volume. Time is again given nondimensionally by the total pore volumes injected. Also, injection of nitrogen at 0.48 m/day in the absence of surfactant is shown with dashed lines as a reference case. For the unfoamed gas injection, little liquid is produced from the low permeability layer. Although the displacement of water from the high permeability layer is initially rapid, gas quickly moves through the 1.3-d layer and the production rate declines after only 0.2 PV. Nitrogen is very mobile relative to water making it an exceptionally poor displacement fluid. Foaming the nitrogen has a dramatic effect. Production from both layers is maintained for about a pore volume of injection indicating that foam provides efficient displacement in both the high and low permeability layers. Production stops at 1 PV because the displacement is essentially complete in 1 PV.

Fig. 4—Fraction of the initial water displaced by foamed gas and unfoamed gas as a function of time from two isolated layers.

The improvement of diversion with foam is seen quite strikingly in **Fig. 5** which gives the simulated saturation profiles in each layer as a function of time. In the high-permeability-layer saturation profile shown in **Fig. 5a**, the foam front initially moves more quickly than in the low permeability layer as illustrated in **Fig 5b**. However, at 0.44 PV the foam displacement fronts in each layer are positioned at approximately x/L equal to 0.55. By examining the displacement fronts at 0.66 PV, we find that the front in the low permeability layer is actually ahead of the front in the high permeability layer. Foam breakthrough occurs first in the low permeability layer. Again, these are very efficient displacements because we began with the porous medium saturated with surfactant solution and use strong foam displacement parameters characteristic of AOS 1416 in oil-free porous media.

Another interesting feature of **Fig. 5** is the steady-state aqueous-phase saturation in each layer. Because the layers are isolated, the strong foam generated in each layer causes the capillary pressure of each layer to approach P_c^* . The aqueous-phase saturation at steady state in each layer is thus

set by P_c^* , and the steady-state saturations are related by P_c^* through the Leverett J-function³⁷. Hence, the 0.13-d layer only desaturates to an S_w of about 0.38 before the limiting capillary pressure is approached, whereas in the 1.3-d layer, the S_w at steady state is 0.30.

(5a)

(5b)

Fig. 5—Transient aqueous-phase saturation profiles for displacement from two isolated layers.

Saturation profiles in **Fig. 5** are best understood by considering the foam texture in each layer. A finely textured foam forms in the high permeability layer, as portrayed in **Fig. 6a** leading to substantial flow resistance. Conversely, the foam that is generated in the lower permeability layer shown in **Fig. 6b**, is over an order of magnitude coarser. Accordingly, the low permeability layer presents an overall flow resistance comparable with that of the high permeability layer. Roughly half of the entire gas flow is diverted to the 0.13-d layer.

Figure 7 presents the companion pressure-drop information for simultaneous injection of nitrogen and foamer solution into isolated layers of differing permeability. Pressure gradients build quickly in both layers consistent with the rapid foam generation displayed in **Fig. 6**. Interestingly, the total system pressure drop is only 2/3 of that found in the one-dimensional linear flow of **Fig. 2** at these same superficial velocities. Because the foam texture in both layers of **Fig. 6** is substantially less than that

predicted in **Fig. 3** for one-dimensional flow, flow resistance and pressure drop are significantly less.

Comparison of **Fig. 7** with the saturation and bubble texture profiles in **Figs. 5 and 6** shows that saturation, bubble texture, and pressure fronts track exactly as they did in one-dimensional linear flow. Where foam texture is large, S_w is low, pressure gradients are large, and *vice versa*.

(6a)

(6b)

Fig. 6—Transient flow-foam textures for displacement from two isolated layers.

Heterogeneous Communicating Linear Layers. The geometry, initial conditions, and flow rates employed here are identical to those for the noncommunicating linear layer case. However, cross-flow between the layers is allowed. **Figure 8** contrasts production of the original aqueous-phase fluid from each layer when foam is both present (solid lines) and absent (dashed lines). Again, we find that foam induces significant production from the low permeability zone compared to gas injection. Displacement in both layers is quite efficient.

Figure 9 shows that sharp saturation fronts propagate at equal rates in both layers. Since the layers are communicating, gas at the foam front minimizes its flow resistance. For example, when the local flow resistance in the 1.3-d layer rises, some portion the foamed gas diverts into the 0.13-d layer, and *vice versa* yielding equal propagation rates in each layer. Saturation fronts in each layer are, thus, bound together by the necessity to maintain

the minimum flow resistance. Likewise, this is true for unfoamed gas. The striking feature of **Fig. 9** is the efficiency of displacement in each layer.

(7a)

(7b)

Fig. 7—Transient pressure profiles for displacement from two isolated layers.

Fig. 8— Fraction of the initial water displaced by foamed gas and unfoamed gas as a function of time from two isolated layers.

Prior to foam breakthrough, S_w upstream of the saturation front in the low permeability, 0.13-d layer is larger in **Fig. 9b** than it is in **Fig. 5b** for noncommunicating layers. During foam propagation, each layer attempts to come to the S_w

corresponding to the limiting capillary pressure. Because there is cross-flow and capillary connection between the layers, water is drawn into the low permeability layer maintaining S_w at slightly higher levels than in the noncommunicating layers of **Fig. 5b**.

(9a)

(9b)

Fig. 9—Transient aqueous-phase saturation profiles for displacement from two communicating layers.

Foam breakthrough occurs just after 0.66 PV. After breakthrough, the aqueous-phase saturation in the high permeability layer remains constant at about 0.27. In the low permeability layer, however, S_w slowly increases over time. At 20 PV the average aqueous phase saturation downstream of the inlet is 0.58. The lower permeability layer slowly refills with water in an attempt to come into capillary equilibrium with the high permeability layer where the capillary pressure is much lower. Equilibrium is achieved when S_w reaches roughly 0.87 everywhere in the low permeability layer.

Refilling of the 0.13-d layer with foamer solution has a dramatic effect on the foam texture over time, as shown in **Fig. 10b**. Prior to foam breakthrough, foam textures are comparable with those found in the previous cases. At 0.22 PV the bubble density in the low permeability layer averages about 30 mm^{-3} in the foam-filled region. After breakthrough and as the layer refills with water in order to reduce its capillary pressure, the rate of foam coalescence

decreases with decreasing capillary pressure as indicated by Eq. (3). Consequently, the net rate of foam generation increases according to Eq. (2) as does the flowing bubble texture. The average bubble texture between 0.66 PV and 10 PV increases by a factor of nearly 3.5. Increasing textures indicate increasing flow resistance. As flow resistance increases, the gas flow rate and also the foam coalescence rate decrease exacerbating the growth in foam texture. The texture in the high permeability layer shown in **Fig. 10a** is relatively coarse in obedience with the large foam coalescence rates caused by the high capillary pressure there and the relatively high gas flow rates. The texture becomes coarser with time because the high permeability layer carries increasingly more gas that in turn increases foam coalescence in the 1.3-d layer. In the meantime, the lower permeability layer fills with foam. The refilling effect is unlikely to be encountered in practical application of foam since it only occurs after many PV of foam injection.

(10a)

(10b)

Fig. 10—Transient flow-foam textures for displacement from two communicating layers.

The pressure drop profiles shown in **Fig. 11** contain several additional interesting features. First, note the magnitude of the pressure drops. The maximum pressure drop displayed is roughly 200 kPa (22 psi), whereas the identical flow rate conditions in **Fig. 2** yielded a steady state pressure drop of a little more than 1600 kPa (230 psi). The flow resistance in the 1.3-d layer is small because foam is

coarsely textured there and because the gas superficial velocity is large exploiting the shear-thinning foam rheology. This commands a smaller net flow resistance than that found in the linear one-dimensional and noncommunicating layer cases. Second, the pressure drop declines in time as the foam coarsens in the high permeability layer. The system pressure drop at 0.66 PV is nearly 190 kPa while at 20 PV it has declined to 140 kPa.

(11a)

(11b)

Fig. 11—Transient pressure profiles for displacement from two communicating layers.

Radial Flow. Next, we consider simultaneous injection of nitrogen and foamer solution into a radial, one-dimensional, homogeneous porous medium that is 1 m thick with a radius, R , of 71.5 m and a permeability of 1.3 d. The medium is initially saturated with surfactant-free brine. Volumetric injection rates are $0.165 \text{ m}^3/\text{day}$ of surfactant solution and $3.14 \text{ m}^3/\text{day}$ of nitrogen relative to the 4.8 MPa backpressure to give a gas fractional flow of 0.95.

Figure 12 displays the radial aqueous-phase saturation profiles as a function of time. Dimensionless radial distance is simply $r' = (r - r_{\text{well}})/R$. At short times (e.g., 0.1 PV), two saturation fronts exist. The front that has propagated farthest into the medium corresponds to unfoamed gas. For the 0.1 PV saturation profile, this first front is at roughly $r' = 0.7$. Little liquid is displaced by this front because gas mobility is high in the absence of foam. The trailing front is quite steep,

sharp, and indicates efficient displacement. It arises because of foam generation and propagation. Behind these strong foam fronts, S_w is only about 5 saturation units above connate saturation. The velocity of both fronts slows as they move outward radially, as expected. Further, foam front propagation is slow because it can move only as quickly as surfactant propagates.

Fig. 12—Transient aqueous-phase saturation profiles for the radial flow of foam and surfactant.

Fig. 13—Transient flowing-foam texture profiles superimposed upon surfactant concentration profiles for the radial flow of foam.

This point is better illustrated in **Fig. 13** which presents the transient bubble concentration profiles superimposed over the surfactant concentration profiles. Foam texture is presented with a dashed line and bubble concentrations are located on the right y-axis. Surfactant concentration in weight percent is given with a solid line and values are found on the left y-axis. Foam texture is fine close to the injector, but texture falls off quickly as foam moves out radially and the gas and liquid velocities fall off as $1/r$. Foam texture declines abruptly where the surfactant concentration falls to zero, because P_c^* approaches zero in the absence of surfactant and the foam coalescence rate approaches infinity. Ahead of the surfactant front, a continuous channel of unfoamed gas exists. It is interesting to compare the bubble profiles in **Fig. 13** with the previous three cases, **Figs. 3, 7, and 10**. In the earlier examples, foam textures immediately behind the foam fronts are elevated

above the steady-state value and above the textures found farther upstream. Since there was ample surfactant at the front due to presaturation, stable foam films were generated; since the capillary pressure at the front is relatively low, the foam coalescence rate is low. In the radial case shown in **Fig. 13**, there is no elevation in the foam texture immediately behind the foam front. Here, the foam coalescence rate is relatively high. The surfactant concentration decreases to zero across the foam front leading to low P_c^* at the front according to Eq. (4) and to large foam coalescence rates according to Eq. (3).

Figure 14 shows the radial pressure profiles as a function of time on a semi-logarithmic scale. Pressure drop initially builds quickly in time as foam generates and fills the region around the injection well. The rate of pressure increase declines with time as the foam propagation rate slows in outward radial flow. Pressure gradients near the injection point are shallow just as they are for linear flow in **Fig. 2**. Because the radial grid is relatively coarse (72 grid blocks) compared to the radial distance spanned (*i.e.*, 72 m), the change in pressure gradient near r' equal to 0.02 is abrupt. Little foam is present in the first grid block making flow resistance small. Away from the inlet region, the pressure gradient declines as $1/r$ similar to a Newtonian fluid. Apparent Newtonian behavior is maintained because foam texture falls off as foam flows in the r -direction as shown in **Fig. 13**.

Tremendous pressure drops are predicted in **Fig. 14**, consistent with the foam-displacement parameters used to match our linear core floods (**Fig. 2**). These are incredibly strong foams reflecting the high limiting capillary pressures of AOS 1416 at concentrations around 1 wt% in the absence of oil. Practical field implementation of foam requires careful selection of the foaming agent and concentration⁵⁴. Recall, that **Figs. 6 and 11** predict that pressure drops should decrease in heterogeneous environments.

Patzek and Koinis¹ reported mobility reduction factors (MRF = gas mobility / foam mobility) inferred from the Kern River steam-foam pilots, that decreased steadily with increasing distance from the injection well¹. **Figure 15** shows this same trend for our foam simulations. The predicted MRF decreases with increasing radial distance from the injector. Although the foam displacement simulated here is shear thinning at constant texture, we find that the MRF must decrease consonant with the decreasing foam textures of **Fig. 13**. In radial flow, the decrease of foam texture with increased distance has a greater effect on gas mobility than shear-thinning in Eq. (5). Again, high MRF is predicted because we employ parameters and initial conditions in the foam displacement simulator that give strong, efficient foams. In the steam-foam field tests, the gas fractional flow was very high and gravity override was significant leading to dry foams that were very vulnerable to coalescence forces and hence much more coarse in texture. Additionally, heat losses that cause steam condensation, surfactant losses due to adsorption and precipitation, and foam coalescence due to the presence of oil were significant. All of these factors lessen the impact that foam has on gas mobility.

Fig. 14—Transient pressure profiles for the radial flow of foam and surfactant.

Fig. 15—Transient mobility reduction factors for the transient flow of foam and surfactant.

Fig. 16—Volume of foam in place versus total fluid injection (compare to Fig. 13 of ref 1).

Another interesting comparison to the results reported by Patzek and Koinis¹ is given in **Fig. 16**. Radial foam growth rate is displayed by plotting *in-situ* foam volume in PV as a function of the total cumulative fluid injection, relative the system backpressure of 4.8 MPa. Results garnered from the present case are shown as well as those from a radial simulation at identical injection rates and conditions except that the medium has been preflushed with surfactant. Both curves reveal that the PV of foam in place increases linearly with the total injected PV. Qualitatively, this trend agrees with that observed in steam-foam field studies where foam propagated in proportion to the injected PV of surfactant solution. The slope of the presaturated case demonstrates how efficient foam displacement might be in the absence of rock adsorption and foam-oil interaction under conditions of coinjection of gas and foamer solution.

5 Spot. The final case is simultaneous injection into one quarter of a confined 5-spot pattern. Hence, we simulate diverging/converging flow. The formation is assumed to be 20 m thick, the injector to producer spacing is 72 m, and the formation is not dipping. These dimensions correspond roughly to the conditions of the Mecca steam-foam pilot so

that we may continue our qualitative comparisons to documented field results. The simulation assumes that the formation is homogeneous with a permeability of 1.3 d, is initially filled with brine, and is bounded by impermeable layers. Injection occurs across the bottom 1/8 of the formation, whereas the producer is completed across the entire interval and is maintained at a pressure of 4.8 MPa. We present saturation, bubble texture, and surfactant concentration profiles in the vertical cross section between injector and producer. Grid spacing is 2 m in the horizontal direction and 1 m in the vertical direction.

To provide contrast with the highly efficient foam displacement to follow, we first ran simulations of unfoamed gas injection. Gas saturation contours in the vertical cross section are presented in **Fig. 17** at 50, 100, 200 and 300 d. The gray-scale shading indicates the gas saturation. Unshaded portions of the graph refer to an S_g of zero, and progressively darker shading corresponds to larger S_g . Without foam, areas contacted by gas are poorly swept. Buoyancy quickly drives injected gas to the top of the formation, a gas tongue forms, and gas breakthrough at the producer occurs quite quickly.

Fig. 17—Gas saturation profiles for unfoamed gas injection into a confined 5 spot.

Fig. 18—Gas saturation profiles for the simultaneous injection of gas and foamer solution into a confined 5 spot.

After breakthrough, little desaturation occurs because pressure gradients are low and buoyancy prevents gas from contacting areas along the lower horizontal boundary. This is classical gravity override.

With simultaneous injection of N_2 and 0.83 wt% foamer solution, foam generates where surfactant and gas are present, and the results are dramatically different. **Figures 18, 19, and 20** present S_g , n_r , and C_s profiles respectively, in the vertical cross section at times of 50, 100, 200, and 300 days. The gas fractional flow is identical to the radial case, 0.95. Injection rates for N_2 and aqueous solution are 15.5 and 0.85 m^3/d , respectively. In **Fig. 18**, the gas saturation contours indicate that both a strong displacement by foam is occurring and a weak displacement by the unfoamed gas ahead of the foam front in a fashion similar to the radial displacement in **Fig. 12**. Near the injector, the high gas saturation region associated with the foamed gas assumes a semi-spherical shape. The contours at later times in **Fig. 18** illustrate that spherical growth and efficient displacement continue. The darkly shaded region immediately below the upper impermeable boundary indicates a tongue of unfoamed gas that forms due to gravity override. Although

not depicted, areal sweep is also good where foam is present.

Figure 19 illustrates foam texture as a function of time. The foamed regions correspond exactly with zones of high gas saturation. The bubble textures associated with black shading are 100 mm^{-3} , and the light-gray shading at the foam front is roughly 20 mm^{-3} . Interestingly, and in agreement with **Fig. 13**, the most finely textured foams are found adjacent to the well bore where gas and liquid flow velocities are largest.

The most provocative result of this simulation is found in **Fig. 20**: surfactant is actually lifted in the formation above its injection point. Black shading indicates a concentration of 0.83 wt%. Foamed gas effectively desaturates the zone around the injector. Although the aqueous-phase relative permeability function is unchanged in the presence of foam, the low S_w results in low relative permeability and highly resistive flow for the aqueous phase. The flow of surfactant-laden water is rerouted and surfactant is pushed upward in the formation. In this example, gravity override has been effectively negated.

Fig. 19—Foam texture profiles for the simultaneous injection of gas and foamer solution into a confined 5 spot.

Fig. 20—Surfactant concentration profiles for the simultaneous injection of gas and foamer solution into a confined 5 spot.

When foam reaches the upper boundary of the layer, displacement continues from left to right in the horizontal direction in a piston-like fashion that expels the resident liquid phase. Propagation is slow until the flow begins to converge.

Discussion

Recently, Rossen and coworkers^{26,27} presented a fractional-flow theory for foam displacement in porous media. Their approach is notable since they consider gas diversion by foam among layers of differing permeability. Beginning with the steady-state experimental observations that aqueous-phase relative permeability is unchanged from the foam-free case^{44-48, 55} and that aqueous-phase saturation is virtually constant^{9, 56} they use Darcy's law, as illustrated by Khatib *et al.*³⁷ and Persoff *et al.*⁵⁶, to obtain a fractional-flow theory for gas mobility in the presence of foam. This method does not explicitly account for the role that foam texture plays in reducing gas mobility. Additionally, the method is not readily applied to two- and three-dimensional flow. It does address, however, radial flow, diversion among isolated layers of differing permeability, and layers in capillary equilibrium.

Our simulations of layered porous media presaturated with surfactant solution reveal that significant flow diversion and production from low permeability layers occurs regardless of whether the layers communicate or not. For practical applications, the extent of diversion into low permeability layers predicted by our population-balance model is quite different than the prediction of the fractional flow theory of Rossen *et al.*^{26, 27}. Because the fractional-flow model sets the capillary pressure in each layer equal to P_c^* at all times, it predicts strong foams in the low permeability layer and diversion into the high permeability layer. This asymptotic behavior is seen in **Fig 10b** as finely textured foam evolves in the low permeability layer because of the low coalescence rate at high water saturation and low P_c there. However, this behavior occurs only after more than 1 PV of foam has been injected, an occurrence unlikely to happen in the field. We should caution here that the foam textures predicted after many pore volumes of injection in **Fig. 10b** are exceptionally fine. When bubbles become so closely spaced, we expect foam generation by snap-off to cease as the close spacing of the bubbles prevents sufficient liquid accumulation for snap-off⁷.

For radial flow, our population balance method predicts that foam texture and, consequently, MRF falls with increasing distance from the injection well in both steady and unsteady flow consistent with field observations of gas mobility¹. The fractional-flow model for foam, though, predicts that MRF is independent of radial distance. In the fractional flow model, all of the effects of foam on gas mobility are inferred from the wetting liquid mobility which is nearly constant for foam flow at the limiting capillary pressure. Since we explicitly account for the coarsening of foam texture as foam flows radially and the effect that texture has on gas mobility, we are able to obtain trends qualitatively similar to those observed in the field.

The case of simultaneous foamer and gas injection into a 5-spot geometry also permits some comparison with field trends. Firstly, we are able to simulate the propagation of

foam far into the reservoir and improved vertical sweep^{1, 4}. Secondly, there is some evidence for spherical growth of the foam zone, such as that shown in **Fig. 19**, in the Kern River steam-foam pilots¹. We predict a constant growth rate of the foam zone, just as was found in the field. Likewise, temperature observations collected at the pilots indicate foam zones with roughly spherical shape.

The calculations presented in this paper represent only a small fraction of the interesting cases possible. Since we specified that all porous media were initially free of oil, we discovered the effect that foam might have if strong foam generation occurred *in situ*. We have not included coalescence terms for the interaction of foam with oil. Although our simulator is fully capable of modeling steam injection, we have not simulated such cases. Additionally, there is speculation of a minimum pressure gradient required to propagate foams under field conditions^{3, 21-24}. We have not simulated foam including such a mobilization pressure gradient.

Only the effects of strong foam were simulated here. By simulating a surfactant system with a smaller P_c^* , it is possible to simulate weak foams that can display even more interesting diversion behavior. For example, if P_c^* is less than the capillary entry pressure of a porous medium, foam will not form³⁷. Hence, stable foam may be generated in high permeability layers where the capillary entry pressure is slightly lower than P_c^* but not at all in low permeability layers. Flow resistance in the high-permeability layer will thus be significant and will divert substantial gas flow into the foam-free low-permeability layer. Further, gravitational effects and the interplay with heterogeneity should be considered more closely. Gravity might cause the top of a reservoir to be so dry that only very weak foams subject to rapid coalescence can form or the rock may be so dry that no foam formation is possible. Finally, we need to simulate steam foams for which condensation is important.

Hence, we caution that the results shown here are not general. Foam displacement in porous media depends strongly on bubble texture which is influenced through the limiting capillary pressure by foamer formulation including the type of surfactant, surfactant concentration, the concentration and type of ions in solution, as well as the temperature. In all cases presented here, displacements begin with the formation full of water. High water saturation and low capillary pressure are conducive to foam formation. Different initial and injection conditions might change the effectiveness of foam as a displacement agent. Likewise, our knowledge of foam trapping is not sufficient to predict whether trapping occurs to the same degree, and in the same fashion, in high and low permeability rocks, even if geometrically similar.

Summary

We have shown that it is practical to model foam displacement mechanistically in multidimensions. Beginning with an n-component compositional simulator, the bubble population-balance equations are successfully incorporated within the simulator's fully implicit framework. The mechanistic population-balance approach allows us to insert the physics of foam displacement directly into a reservoir simulator. Foam is treated as a nonchemical

"component" of the gas phase and the evolution of foam texture is modeled explicitly through pore-level foam generation and coalescence equations. As foam mechanisms become better understood, this framework allows for their inclusion.

For both noncommunicating and communicating linear heterogeneous layers, foamed gas efficiently diverts to low permeability layers when the layers are initially saturated with surfactant solution in the absence of gravity. For communicating layers, the foam propagation rate is equal in both layers. In this instance, foam dramatically evens out injection profiles.

For one-dimensional radial flow, we find that foam pressure drop scales as $1/r$ similar to a Newtonian fluid. The gas mobility reduction factor for radial foam flow falls off as foam moves outward radially from the injector because the foam coarsens. This decline in mobility reduction factor in radial flow is consistent with previous field observations of steam-foam propagation¹.

For simultaneous injection of gas and foamer solution into a confined 5-spot pattern, we clearly see two displacement fronts. Unfoamed gas moves upward through the formation due to buoyancy and is ineffective in displacement. The second front tracks with surfactant propagation and the generation of foam. The strong foam desaturation front takes a semi-spherical shape for short times with the origin of the sphere at the injector. Importantly, it is found that gravity override can be effectively negated.

These predictions are a result of the direct approach taken to model foam displacement. Since gas mobility in the presence of foam depends strongly on foam texture, it is necessary to account for foam-bubble evolution to model gas mobility generally and correctly.

Nomenclature

$\bar{C}$	=	concentration
$\bar{F}$	=	component vector flux,
k	=	rate constant
K	=	permeability
L	=	length of linear porous medium
MRF	=	mobility reduction factor
n	=	number density of foam
p	=	pressure
P_C	=	capillary pressure
PV	=	pore volume (injected or in place)
q	=	generation rate and source/sink term
r	=	radial distance
R	=	radial extent of porous medium
S	=	phase saturation
t	=	time
u	=	Darcy velocity
v	=	interstitial velocity

Greek Letters

α	=	proportionality constant for foam effective viscosity
$\nabla \cdot$	=	divergence operator
ϕ	=	porosity
μ	=	viscosity
Γ	=	adsorption isotherm

Subscripts

l	=	generation rate constant
$-l$	=	coalescence rate constant
f	=	flowing foam
g	=	gas phase
i	=	phase (<i>i.e.</i> , aqueous, gas, or oil)
j	=	chemical species
s	=	surfactant
t	=	stationary foam
w	=	water or wetting phase
wc	=	connate water saturation
$well$	=	denotes well radius

Superscripts

o	=	denotes reference value
$*$	=	value corresponds to the limiting capillary pressure
$'$	=	denotes normalized radial distance

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3.3 PORE-LEVEL VISUALIZATION OF OIL/FOAM INTERACTIONS IN A SILICON MICROMODEL

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Pore-Level Visualization of Oil/Foam Interactions in a Silicon Micromodel

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Abstract

This paper presents some cases of pore-level visualization of foam with oil. Many laboratory investigations have been carried out in the absence of oil, but comparatively few have been carried out with oil present. For field application, where the residual oil saturation may vary depending on the recovery method applied, any effect of the oil on foam stability becomes a crucial matter.

Silicon micromodels were used in this study. The micromodels are two-dimensional replicas of the flow path of Berea sandstone etched on a silicon wafer to a prescribed depth, using fabrication techniques from the computer industry. After flooding the models to connate water and residual oil saturations, a surfactant flood was followed by gas injection to generate foam. Starting with low concentrations of surfactant the concentrations were gradually increased. Visual observations were recorded on videotape.

Two different surfactants were used, a fluoro-surfactant (for generating a non-aqueous oil foam) and an alpha-olefin sulfonate. Oseberg (Norway) crude was the oleic phase in the first set of experiments, and kerosene in the next two sets. While the fluoro-surfactant created a strong static gas-blocking foam in the presence of oil, the alpha-olefin sulfonate (AOS) foamer did not.

The fluoro-surfactant foam gave the oil-tolerant behavior expected from its non-entering, non-spreading characteristics. The AOS on the other hand, did not behave in accordance with its bulk observations, and its behavior was seen to be controlled by formation of emulsions. Generation sites for both foam and the emulsions were controlled by pore geometry and local saturations. For the foam, no obvious link could be

found between the number of films observed and the strength of gas blockage. Many other observations were made; including snap-off, emulsion formation and breakdown sequences, foam lamella formation and breakdown sequences, and static emulsion and foam in different configurations within the model.

Introduction

Primary and secondary production of an oil field often recovers less than half of the original oil in place. To recover additional oil, it is necessary to apply enhanced oil recovery (EOR) techniques such as miscible or immiscible gas displacement (CO_2 , hydrocarbon gases, etc.). However, major problems occur in these EOR methods because the displacing agent has high mobility and low density compared with that of reservoir fluids. Fingering (channeling) and gravity override reduce the sweep efficiency, contribute to early breakthrough of injected fluid, and thus reduce the amount of oil recovered. The use of surfactant stabilized foams to counteract these problems was suggested decades ago. Foam is advantageous compared with a simple fluid of the same nominal mobility because the foam, which has an apparent viscosity greater than its components, lowers the gas mobility in the swept or higher permeability parts of the formation. This diverts at least some of the displacing gas to other parts of the formation that were previously unswept, thus recovering additional oil. Because foam mobility is reduced more in the higher permeability zones, improvement in both vertical and areal sweep efficiency is achieved.

This work focuses on stability of foams in porous media when oil is present. For field applications, any effect of the oil on foam stability becomes a crucial matter. Two differing approaches have been used in an attempt to scale up foam properties to the reservoir. They represent differing descriptions of the physics of foam. One approach is purely empirical and relies on calibrating a flexible simulator, based on laboratory data in conventional cores, to represent the most critical foam effects. The empirical foam model implemented in STARSTM, a pseudo-compositional reservoir simulator, is probably the best example of this approach (Coombe and Mohammadi¹ and Surguchev *et al.*²). The other approach attempts to extract information about the physical flow

mechanisms observed at the pore level, and constructs mathematical relationships between experimental parameters. One such model was developed at the University of California at Berkeley. The physical mechanisms in this model rely on observations of foam flow in etched-glass micromodels, especially from a comprehensive study by Radke and Chambers.³

Currently accepted theories on foam flow are largely built on the basis of pore model observations by Mast,⁴ Owete *et al.*,⁵ Huh *et al.*,⁶ and Schramm and Novosad.⁷ Most micromodels were made by etching a flow pattern into a glass plate and are two-dimensional representations of the pore structure. With glass micromodels, the reaction kinetics of acid etching makes it necessary to enlarge the pore sizes of typical sandstone pores, usually by a factor of 5 to 50 compared to their natural size. Obviously, this is a serious limitation for their use in studying processes that depend critically on capillary forces, and involve thin films as seen in foam flow.

The silicon micromodel on the other hand, replicates a pore structure of any design on a silicon wafer, adapting fabrication techniques from the computer industry. The pore cavities are created photochemically, a process which is not rate-limited. The micromodels used in this study offer 1:1 scaling of typical sandstone pores in two dimensions.

Literature Survey

Many laboratory investigations of foam flooding have been carried out in the absence of oil, and comparatively few have been carried out with oil present. When research began to include oil, it was seen that the effect of oil on foam stability was indeed complicated. Oil saturation could alter foam stability and its propagation mechanisms to a great extent. The nature of the foam, foam injection procedures, reservoir wettabilities and pore geometry were seen to affect the oil-foam interactions.

A foam is a dispersion of gas in liquid, usually with a surface active agent present. Foams are not thermodynamically stable and ultimately decay into their constituent phases, but can be mechanically stable for long periods of time. When a foam exists inside a confining medium, the dimensions of this confining medium relative to the average bubble size, determine the texture and properties of the foam. Holm⁸ found liquid to flow as a continuous phase, but gas flow to occur by a sometimes very slow process of continually breaking and reforming of liquid films. This explains why gas flow may be essentially blocked by foam in pores, while the permeability to liquid is merely a function of its saturation. The ability of foam to reduce gas mobility has led to its application in a number of processes, including gas injection, steam flooding, and well treatment techniques.

The use of surfactant stabilized foams to counteract problems such as fingering and gravity override was suggested several decades ago. Marsden⁹ reviewed previous findings, surmising that apparent gas viscosity was a function of the gas

flow rate, flow history and surfactant concentration, and that 'breaking and reforming' processes are a dominant mode of foam propagation in micromodels. Sanchez and Schechter¹⁰ suggested that gas permeability was a function of the wetting phase film thickness.

Owete *et al.*¹¹ suggested that air was propagated by displacement of lamellae in long bubbles flowing and extending across several pore lengths while the liquid flowed through the network of films. They also observed snap-off occurring at pore constrictions and the resultant bubble 'break and reform' process. Radke and Ransohoff¹² categorized the various mechanisms of foam generation within glass bead packs as 'snap-off', 'lamella leave behind', and 'lamella division', with snap-off being the primary mechanism for strong foam generation.

Roof¹³ studied the specific pore geometry necessary for snap-off mechanisms. Roof noted that the front is always at least seven pore radii from the throat of the constriction before snap-off can occur. Falls *et al.*¹⁴ and Kovscek *et al.*¹⁵ proposed that at steady state conditions, the rates of generation and collapse of lamellae must be equal.

Castanier and Hanssen¹⁶ noticed in silicon micromodels the sliding of lamellae (bubble train flow) over liquid films covering the grains.

Oils are among the earliest chemicals to be used for foam inhibition and foam breaking, as discussed in the literature by Harkins,¹⁷ Robinson,¹⁸ and Ross and McBain.¹⁹ Some theories for the mechanisms of anti-foaming action have emerged from this work. In general matters are not simple. Apparently oils can destabilize foams by several mechanisms, and more than one mechanism may be operating in a given case.

Some possibilities for the mechanisms of foam destabilization by a given oil phase include: (1) surfactant partition in the oil, (2) surfactant adsorption by the foam lamellae, (3) alteration of the wettability of the solid phase, (4) oil spreading on foam lamellae, and (5) oil emulsification.

Nikolov *et al.*,²⁰ studying bulk oil-foam interactions, reported three distinct films during three phase foam thinning: 'foam films', or water films between air bubbles; 'emulsion films', or water films between oil droplets; and 'pseudo-emulsion films' where water film resides between air and oil droplets. Pseudo-emulsion films, droplet sizes and numbers may all contribute to foam destabilization.

Kuhlman²¹ and Manlowe,²² and Radke³ observed in micromodels that oils were destroying foams. Work done by Novosad and Ionescu²³ Jensen and Friedmann,²⁴ Kristiansen and Holt,²⁵ and Hutchinson *et al.*²⁶ suggested that foam sensitivity to oil is also manifested as an increased difficulty of forming and propagating foams through porous media containing oil. Novosad and Ionescu²³ found lower mobility reduction factors in foam floods in cores containing residual oil, compared to the same floods in oil free. The lower mobility reduction factors are due to some kind of foam destruction by the oil.

Lau and O'Brien²⁷ through the use of sand packs and both spreading and non-spreading oils, showed an increase in foam

generation time, a reduction in foam propagation rate, and an increased foam destruction rate with spreading oils. This finding is contrary to Manlowe²² and Kovscek *et al.*,¹⁵ who argue that there is no correlation between oil spreading and foam stability. Instead, they state that pseudo-emulsion film collapse is the general destabilization mechanism.

Hanssen and Dalland,²⁸ in a model system (glass bead pack) of a gas blocking foam in an oil reservoir, observed clear trends in mobility reduction with varying concentrations of oil and surfactant. They concluded that the change in interfacial tension between gas and the aqueous phase, upon equilibration with oil, appeared to be a good predictor of gas blockage efficiency for the systems studied.

Oil interactions with foam have been studied by Dalland and Hanssen.²⁹ Conducting experiments with flowing and gas blocking foams, using a set of conventional and fluorinated surfactants and various oils, they found varying oil tolerance and sensitivities. In all cases the combination of non-spreading and non-entering oil created a strong gas-blocking foam, or a low mobility flowing foam. The 'Lamella Number', L , calculated from interfacial tensions was found to be of no predictive power. The results indicated that the mechanisms of oil-foam interactions are the same for gas blocking and for flowing foams in porous media³⁰

Castanier and Hanssen¹⁶ worked on oil-foam interactions using a fluorosurfactant foam and an AOS foam. The fluorosurfactant foam gives the oil tolerant behavior expected from its non-entering, non-tolerant characteristics. AOS foam by contrast, is controlled by formation of oil-water emulsions which appear to prevent the formation of strong foam until a large excess of foaming agent is present.

Because petroleum reservoirs are often of intermediate, mixed or oil wettability, some researchers have included reservoir wettabilities in their studies. Huh *et al.*⁶ found that micromodel wettability was altered to intermediate or oil wet as a result of crude oil saturation. The effects of various phase affinities to grain surfaces was studied by Sanchez and Hazlett³⁰ as well. An indication of surface reaction, from hydrophobic to hydrophilic, was detected after the addition of surfactant. This suggests wettability alteration. Kuhlman²¹ obtained similar results, and said that the deleterious effect of oil on foam was due to the high concentration of light hydrocarbons in the oil, and the oil wetness of the medium itself.

Castanier and Hornbrook³¹ noted that wettability alteration was due to the surfactant injection procedure. They found that with surfactant slug injection, oil appeared to be the wetting phase, and with foam injection, surfactant appeared to wet the medium.

Schramm and Novosod⁷ conducted oleophilic microvisual cell experiments with several foams. They found that the foams were significantly less stable in the presence of crude oil and oleophilic solid surfaces, compared with some crude oils and hydrophilic surfaces. Kristiansen and Holt³² studied foams

flowing in cores under North Sea reservoir conditions that were either partially or completely oil wetted. Any degree of oil-wetness reduced the effectiveness of the flowing foam.

The surfactants may adsorb onto the solid surfaces and alter the wettability. In the microvisual experiments of Schramm and Novosod,⁷ some of the foaming systems changed the wettability to water wet, in which case the foam sensitivities to oil reverted back to the water wet cases. This result is consistent with the coreflood tests of Sanchez and Hazlett³³ in which foaming surfactants caused wettability reversal in oil wet porous media. This reversal was the reason that stable foams could be generated and propagated in their oil wet cases.

Experimental Apparatus

All the experiments were done at constant pressure. A basic line diagram of the apparatus is shown in Fig. 1.

Surface tensions were measured using a ring tensiometer and interfacial tensions were determined by a spinning drop tensiometer. More details on the equipment can be found in Sagar *et al.*³⁴ The micromodels used in our experiments replicate the pore structure of Berea sandstone on a silicon wafer. The fabrication technique is a direct adaptation of the etching techniques used by the computer industry. The pore cavities are created photochemically, a process which is not rate-limited, such as acid etching. These high resolution silicon micromodels offer 1:1 scaling of the Berea sandstone pores and throats. A detailed description of the fabrication and construction process of the models used can be found in Castanier and Hornbrook,³¹ Woody *et al.*³⁵ and Lolomari.³⁶

A description of the pore structure of the model along with permeability calculations can be found in Woody *et al.*³⁸ Scanning electron microscopy (SEM) photographs of the model can be seen in Fig. 2. The characteristics of the models are outlined below:

1. 500 micron patterns repeated 10,000 times on a 5 cm square silicon wafer
2. Etch depth 15 microns.
3. Porosity 35%
4. Permeability: 1 millidarcy
5. Pore Sizes : 1-150 mm
6. Throats : 0.5-10 mm

Experimental Procedure and Experiments Performed

Before beginning the experiments, the model was first cleaned with water followed by methanol to remove traces of foam or solvent. All the experiments were performed at ambient temperatures using the following generic procedure:

1. CO₂ flood
2. Water/brine flood to displace all CO₂.

3. Oil flood to connate water saturation.
4. Water/brine flood to residual oil saturation.
5. Surfactant flood.
6. Gas injection to generate foam
7. Surfactant flood (higher concentration)
8. Gas injection to generate foam

Steps 5 and 6 were repeated (as shown in 7 and 8) for higher concentrations of surfactant. The outlet was at atmospheric pressure and the inlet pressure was measured. All fluids entering the micromodel were filtered to 0.22mm. A dye (benzene-azo-2-naphthol) was used to color the surfactant for better viewing. Note that interfacial properties were not affected by the dye. All the experiments were recorded on videotape and can be made available for viewing on request. Table 1 shows the experiments conducted.

For Experiments 2 and 3, brine was used as the aqueous phase instead of water. The brine formulation is based on North Sea water.

Results

The results of the three sets of experiments conducted are described in this section. Each set of experiments is discussed separately, and is accompanied by images captured during the course of the experiment.

Experiment 1

Experiment 1 was the only experiment in which crude oil (Oseberg, a Norwegian crude) was used. The surfactant was an alpha olefin sulfonate, AOS 1618, and the gas used for foam generation was carbon dioxide. The aqueous (wetting) phase was water.

The pressure difference applied to the micromodel was usually 35 psig. Only two concentrations of the AOS surfactant were used in this experiment. The experiment had to be discontinued due to excessive clogging and asphaltene deposition.

The flow through the micromodel was extremely slow as compared to the flow mentioned by Hornbrook *et al.*³³ and Woody *et al.*,³⁷ even though the etch depth for the new models was greater than those used by Hornbrook³³ and Woody.³⁷ This slow flow was caused by overheating the glass during the bonding process which reduced the model permeability. This defect has since been corrected, in more recent micromodels.

After injection of the surfactant and more often during gas injection, stable crude oil emulsions were observed in the model (Fig. 3). These emulsions were seen at concentrations of 0.01% by weight and over and were observed to be stable in the micro- model. This emulsion formation occurred extremely rapidly when gas was injected.

An emulsion forming sequence with the crude oil was observed where crude oil coated water bubbles started forming in a narrow throat and were seen being pushed into a pore where these bubbles conjoined to form an emulsion (Fig. 4).

A bubble coalescence sequence was also observed where carbon dioxide bubbles coated with crude oil were generated from a very narrow throat and were pushed into a pore where they coalesced to form a larger carbon dioxide bubble (Fig. 5).

Emulsion breakdown sequences were also seen to occur in the pores. A very visible crude oil emulsion broke down in a matter of seconds and was captured in the stills shown (Fig. 6).

Emulsions sizes with crude oil varied from 10µm to 60µm in size. The emulsions were usually composed of water surrounded by oil films. The color of the water seemed to be darker (brownish) after contact with the crude oil. This shows some, still unexplained, mass transfer mechanism between the oil and water.

As can be seen from the above observations, the Oseberg crude oil clearly formed emulsions in the presence of the AOS surfactant, indicating the non-tolerant behavior of the Oseberg crude to this surfactant. No foam lamella were observed during the experiment. The displacements were according to the theory of immiscible displacement: Displacement of water by gas caused viscous fingering, and displacement of water by oil was a stable front propagation.

Experiment 2

In Experiment 2, the surfactant used was the same alpha olefin sulfonate (AOS 1618). The wetting phase was brine and the non-wetting phase (oil) was kerosene. Kerosene emulsions were observed at as low as 0.001% concentration of surfactant (Fig. 7). During carbon dioxide flooding the gas phase was discontinuous and did not fill the model completely. The emulsions were very mobile at low concentrations of surfactant (< 1%) .

The diameters of the emulsions were smallest at lower concentrations and were seen to increase with higher concentrations. The stability of the emulsions also increased at higher concentrations. Certain areas in the model which had less oil were seen to have thinner emulsions (Fig. 8). At increased surfactant concentrations the oil saturations in the model had decreased, and the emulsions were still widely present and more stable.

The emulsion bubble size also was seen to depend on the pore geometry (Fig. 8). Different sizes of emulsions seem to co-exist but varied in size with the size being the smallest in the throats and the smaller pores.

Emulsions started at 0.001% surfactant concentration. The emulsions at this stage flowed very rapidly through the narrow throats until they clogged the flow path after gathering in large pores. Emulsion sizes, as mentioned earlier, increased with surfactant concentration. Although the pore geometry had some effect, emulsion sizes after 0.1% surfactant flood varied from a few microns to about 60µm. Emulsion composition after the 1% surfactant flood was very similar to that seen at 0.1%. The emulsion were hardly seen to move, while at lower concentrations, most of the narrow throats had been blocked by the smaller emulsions. Two emulsion forming and breaking sequences were also observed. No

foam was observed during the experiment.

Experiment 3

In Experiment 3, we formed an oil-foam. The surfactant was a fluorocarbon diluted in kerosene. The wetting phase was brine and the non-wetting phase was kerosene. Carbon dioxide was used to generate the foam in-situ. The exploitation of the gas blocking properties of foam, by using it as a gas coning barrier, is an old idea. An attractive means of achieving a favorable barrier geometry is by using low density non-aqueous foams. This is the primary reason for our focus on the oil foam studied in this experiment.

Flow through this model was extremely fast as compared to Experiments 1 and 2. After the water flood and the oil flood, water was seen residing in the narrow throats indicating that the model was water wet. Before foam was generated in the system, the carbon dioxide was discontinuous. Stable foam lamellae were seen by 0.01% surfactant concentration. A snap off sequence of a carbon dioxide bubble coated with surfactant was seen (Segar³⁶).

In certain areas, at 0.1% concentration of the fluoro-surfactant, a lamella formation and breakdown sequence was captured in a five frame sequence.

Foam was observed as lamellae throughout the model at the differing concentrations of the surfactant (Fig. 9). Certain pore geometries were seen to be more conducive to forming stable lamellae than others. Many different lamellae configurations were seen. Stable lenses were also observed to co-exist with the lamellae.

In one configuration, a water droplet in a foam lamella did not seem to affect the lamella stability (Fig. 10). Also, in a certain pore geometry, thin stable lamellae were seen as well as thicker lamellae.

No significant increase in the number of foam lamellae were seen from 0.1% to 1.0% surfactant concentration, in spite of a noticeably higher gas blockage as seen from the flow rate reduction at higher concentration.

Interfacial Tension Measurements

Interfacial tensions between the surfactant solutions and the oil (kerosene) were measured for Experiments 2 and 3 and are listed below. No significant difference in the interfacial tension was observed because of the presence of an ethyl-oil soluble dye, benzene-azo-2-naphthol used as dye.

Conclusions

The silicon micromodels used in these experiments give us a realistic feel for the pore level mechanisms which would occur in reality. The three sets of experiments conducted in this study, which included different oils and foamers, lead us to the conclusions outlined below.

The AOS Surfactant

With the AOS surfactant, with differing oils present, no foam formation was seen at the pore level. However emulsions were observed which varied in size and stability with surfactant concentration. The pore and throat sizes also seemed to influence the emulsion size.

Most of the emulsions seen were composed of water covered with oil films. With the crude oil the water became discolored to brown. Very little oil was required to form these emulsions as could be seen at higher concentrations of surfactant after the model had been flooded with surfactant a few times resulting in low oil saturation. It can be concluded that the emulsion formation with the AOS surfactant could be the reason for the delayed foam forming observed by Kristiansen and Holt.^{25,32} These have, in the past, often been ascribed to so-called "aging" or "incubation" effects. According to a mechanism suggested by Lau and O'Brien,²⁷ the emulsion scavenges all surfactant and makes it impossible to generate foam initially. Only when excess surfactant solution becomes available, through reinjection or rearrangement, can foam generation take place. This concept, however, does not apply to our case, where excess surfactant is always present.

The ability to form water-in-oil emulsions is governed by the balance of forces between those tending to keep the oil in large drops (or continuous oil areas) and those favoring formation of small emulsified oil droplets. It is well known, from anti-foam theory, that small emulsion droplets make better anti-foam agents. An attempt to develop a theory for foam destabilization by emulsified oil droplets has been published by Schramm and Novosad.⁷ However, their lamella number fails to correlate oil tolerance data for fluorosurfactant foams; likewise the lack of emulsions formed with the fluorosurfactant. The presence of oil in the system resulted in emulsions for the AOS surfactant, and no foam was observed. Whether foam would actually form on continuous surfactant and gas flooding is yet to be seen in the micromodel and would be an interesting area for further research. Currently no theory explains this phenomenon of emulsion with foamers which seems different from the conventional "aging" of cores, which is essentially a liquid/solid adsorption process.

The Fluorosurfactant

For the oil-foam with the fluorosurfactant, a strong foam was generated both in the oil saturated and the oil free zones. Lamellae were observed both in oil saturated and oil free areas, indicating oil tolerant behavior, as expected, from the negative entering coefficient and the bulk behavior. Generation sites for the foam lamellae were seen to be controlled by pore geometry and local saturation. The foam lamellae were stabilized by the surfactant and therefore an

increase in foamer concentration increased the lifetime of the lamellae, and thereby the strength of the foam.

No obvious link could be found with the number of films observed and strength of gas blockage. Although lenses were observed to be present in the system, no lens-lamella transition was observed. We can conclude from these observations that this oil-foam can be used for achieving a favorable barrier geometry for gas coning problems.

Further Research

Further research suggestions include studying the behavior of emulsions and how they effect foam properties in porous media with an emphasis on foam destabilization by emulsified foam droplets.

Using different thin sections (other than Berea) would help in studying flow behavior in other pore geometries. The effect of "aging" as the reason for the delayed foam forming (if any) with the AOS surfactant also deserves more investigation. As mentioned before, whether foam would form under continuous surfactant and gas injection would be an interesting area to be looked into to supplement this work. The new micromodels are currently under construction at the 'Center for Integrated Systems' at Stanford.

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Figure 1. Line diagram of the Experimental Apparatus.

Figure 2(a). SEM of Model A (old).

Figure 2(b). SEM of Model A (old).

EXPERIMENT 1

Figure 3 Crude oil emulsions seen in the large pores in the model after 0.01% surfactant flood.

EXPERIMENT 1 **Figure 4(a)** **Figure 4(b)**

EXPERIMENT 1

Figure 4((c)

Figure 4(d) Emulsion formation sequence.

EXPERIMENT 1

Figure 5(a)

Figure 5(b) Bubble coalescence.

EXPERIMENT 1

Figure 6(a)

Figure 6(b)

EXPERIMENT 1

Figure 6(c) Emulsion breaking sequence.

EXPERIMENT 2

Figure 7 Emulsions as seen after 0.01% surfactant flood.

Figure 8 Very stable emulsions seen at the end of the experiment (after 1% surfactant flood).

EXPERIMENT 3

Figure 9 Four stable lamellae seen after 0.1% FC740 surfactant flood.

Figure 10 A drop of water in a lamella. It doesn't seem to affect the lamella stability.

3.4 A NUMERICAL ANALYSIS OF THE SINGLE-WELL STEAM ASSISTED GRAVITY DRAINAGE PROCESS (SW-SAGD)

(K.T. Elliot and A.R. Kavscek)

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A Numerical Analysis of the Single-Well Steam Assisted Gravity Drainage Process (SW-SAGD)

By

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Abstract

Steam assisted gravity drainage (SAGD) is an effective method to produce heavy oil and bitumen. In a typical SAGD approach, steam is injected into a horizontal well located directly above a horizontal producer. A steam chamber grows around the injection well and helps displace heated oil toward the production well. Single-well (SW) SAGD attempts to create a similar process using only one horizontal well. This may include steam injection from the toe of the horizontal well with production at the heel. To improve early-time response of SW-SAGD, it is necessary to heat the near-wellbore area to reduce oil viscosity and allow gravity drainage to take place. Ideally heating should occur with minimal circulation or bypassing of steam.

Since project economics are sensitive to early production response, we have investigated early-time processes to improve reservoir heating. A numerical simulation study was performed to gauge combinations of cyclic steam injection and steam circulation prior to SAGD in an effort to better understand and improve early-time performance. Results from this study, include cumulative recoveries, temperature distributions, and production rates. Variances are displayed within the methods. It is found that cyclic steaming of the reservoir prior to SAGD offers the most favorable option for heating the near-wellbore area to create conditions that improve initial SAGD response. Additionally, a sensitivity analysis was performed with regard to reservoir height, oil viscosity, horizontal to vertical permeability anisotropy, and dead versus live oil. More favorable reservoir conditions such as low viscosity, thick oil zones, and solution gas, improved reservoir response. Under unfavorable conditions, response was limited and could prove to be uneconomical in actual field cases.

Introduction

Steam assisted gravity drainage maximizes the role of gravity forces during steam flooding of heavy oils. Generally, a pair of horizontal wells is used. As steam enters the reservoir, it heats the reservoir fluids and surrounding rock. Hot oil and condensed water drain through the force of gravity to a production well at the bottom of the formation. In conventional SAGD, steam is injected through a horizontal well placed directly above a horizontal producer. Thus, a steam chamber forms around the injection well. In SW-SAGD, a horizontal well is completed such that it assumes the role of both injector and producer¹. In a typical case, steam is injected at the toe of the well, while hot reservoir fluids are produced at the heel of the well. Recovery mechanisms are envisioned to be similar to conventional SAGD. Advantages of SW-SAGD might include cost savings in drilling and completion and utility in relatively thin reservoirs where it is not possible to drill two vertically spaced horizontal wells. However, the process is technically challenging.

In a reservoir where cold oil is very viscous and will not flow easily, initial production rates via any gravity drainage process are very low. In a strict definition of SAGD, steam only enters the reservoir to fill void space left by produced oil. If the oil is cold, we must heat it to reduce viscosity. Therefore, initial heating of the area around the wellbore is required so that SAGD can take place. After SAGD is initiated, a steam chamber will grow upward to the top of the reservoir and then begin to extend horizontally². At the

steam-chamber boundary, steam condenses as heat is transferred to the oil. Condensed water and hot oil flow along the steam chamber to the production well².

Joshi found experimentally that under various injection/production well configurations, the steam chamber grows to cover a majority of the reservoir and the recovery efficiency is very good in all cases³. Therefore, we expect that early-time production results from SW-SAGD may vary from the conventional approach, but at late times, we expect similar recovery efficiencies. Additionally, Oballa and Buchanan⁴ simulated various scenarios to evaluate the difference between cyclic steam injection and SAGD. They focused, partially, on the interactions between the reservoir, the well completion, and the recovery of oil. It was concluded that the drainage process may be feasible provided that a proper operating strategy is identified.

Field tests of SW-SAGD are not extensively documented in the literature. Falk et al overview the completion strategy and some typical results for a project in the Cactus Lake Field, Alberta Canada¹. A roughly 850 m long well was installed in a region with 12 to 16 m of net pay to produce 12 °API gravity oil. The reservoir is a clean, unconsolidated, sand with 3400 md permeability. Apparently, no attempts were made to preheat the reservoir before initiation of SW-SAGD. Steam was injected at the toe of the well and oil produced at the heel. Oil production response to steam was slow and gradually increased to more than 100 m³/d. The cumulative steam-oil ratio was between 1 and 1.5 for the roughly one-half year of reported data.

McCormack et al also describe operating experience with nineteen SW-SAGD installations⁵. Performance for approximately two years of production was mixed. Of their seven pilot projects, five were either suspended or converted to other production techniques because of poor production. Positive results were seen in fields with relatively high reservoir pressure, relatively low oil viscosity, significant primary production by heavy-oil solution gas drive, and/or insignificant bottom-water drive. Poor results were seen in fields with high initial oil viscosity, strong bottom-water drive, and/or sand production problems. Although the authors note that the production mechanism is not clearly understood, they suspect that it is a mixture of gravity drainage, increased primary recovery because of near-wellbore heating via conduction, and hot water induced drive/drainage⁵.

Problem Definition

This paper is grouped into two general topics: (i) a screening study of early time operating performance of SW-SAGD and (ii) a sensitivity analysis of the effect of reservoir and well completion parameters. Understanding the operating conditions to improve initial performance relates directly to understanding methods of heating the near-wellbore area at early-time. A central idea realized through our research is that the near-well region must be heated rapidly and efficiently for significant early-time response. The sensitivity analysis helps us understand reservoir properties, fluid conditions, and well completion strategies that make the process an appropriate production technique.

To gain an understanding of early-time performance, we build and compare various computer simulations. The processes examined include cyclic steaming, steam circulation within the well, and an extreme pressure differential between the injection and production sections of the well. Each initial operating period was followed by SAGD; that is, continuous steam injection and oil production with little net injection of steam. For the sensitivity analysis, we compared a base case against runs in which we varied oil viscosity and gas content, reservoir height, permeability anisotropy, and finally the well completion. Computer Modeling Group's (CMG) STARS thermal simulator was used to perform all of the work.

Model Description

The base case is STARS example sthrw009.dat released with Version 98.01⁶. Reservoir and fluid properties represent a typical Alberta reservoir. The operating conditions and well completion were modified to develop additional cases.

The grid system and dimensions are illustrated in Fig. 1. Figure 1a displays the cross section along the length of the well and Fig. 1b a cross section perpendicular to the well. The grid system is Cartesian with local grid refinement immediately around the 800 m long well. An element of symmetry, with one boundary lying along the wellbore, is used to represent the reservoir volume. We assume that wells will be developed in multiple patterns and thus all boundaries are no flux. The single horizontal well is represented using a discretized wellbore model that accurately represents fluid and heat transfer in the well. Additionally, the well is broken into two sections as if the well contained a packer. Each section is equal in length and they lie directly end to end. This gives us freedom to explore various completion strategies and operating conditions. Our initial simulation work suggested that this inject at the toe and produce at the heel strategy had greater probability of success than attempting to develop countercurrent steam and oil flow along the entire length of the well. Table 1 lists the exact dimensions of the reservoir model, grid-block information, and reservoir properties. Initially, the average reservoir pressure is 2,654 kPa (380 psi), the pressure distribution is hydrostatic, and the initial reservoir temperature is 16 °C.

The water-oil relative permeability and gas-liquid relative permeability functions are displayed graphically in Figs. 2 and 3, respectively. Table 1 also lists the initial fluid saturations of the reservoir. The homogeneous porosity is 33%. The horizontal permeability, k_h , is 3400 md, whereas the vertical permeability, k_v , is 680 md. Hence, the ratio $k_h:k_v$ is about 5 to 1.

A live, black-oil model is used. The initial oil phase is made up of 90 % by mole oil component and 10 % by mole gas component for a solution gas-oil ratio (GOR) of about 28 SCF/STB. The effect of solution GOR on oil recovery is explored in the sensitivity analysis. The oil viscosity versus temperature relationship is given in Table 1, and the oil viscosity at the initial reservoir temperature is 4.043 Pa-s (4043 cP). An increase of oil temperature to 100 °C decreases the oil viscosity to 0.030 Pa-s (30 cP).

Table 2 lists the operating constraints for the four base cases created to explore a range of early-time procedures. Briefly the cases represent SAGD operating conditions, extreme pressure differential conditions where steam is injected near the fracturing or parting pressure of the formation, cyclic steam injection, and steam circulation through the well. Arbitrarily, 100 d was chosen as the duration of attempts to heat the near-well region. In all cases, SAGD conditions follow this initial period. Each of the constraints will be discussed in more detail below.

Early-Time Performance Study

To heat the near-wellbore area and improve the initial production response of SAGD, we combined the operating conditions displayed in Table 2 into the seven cases: (i) SAGD operation conditions from the start, (ii) extreme pressure differential between injector and producer sections for 100 days followed by SAGD, (iii) steam circulation for 100 days followed by SAGD, (iv) circulate for 100 days followed by 100 days of an extreme pressure differential followed by SAGD, (v) cycle once followed by SAGD, (vi) cycle twice

followed by SAGD, and (vii) cycle three times followed by SAGD. In each case except the first, an initial preheating phase precedes SAGD.

Results of this screening study are displayed in Figs. 4 to 6. Figures 4 and 5 display recovery factor histories for the first year of production and for 10 years of production, respectively. Recovery factor refers to the fraction of oil produced from the entire simulation volume. The cumulative steam oil ratio (CSOR) versus time curves for each case are displayed in Fig. 6. It is obvious from the curves that it is possible to improve initial production response. In general, cyclic steaming as applied in cases 5 to 7 leads to the most rapid oil recovery. However, the late time recovery performances shown in Fig. 5 display similar behavior for all cases. Recovery factor ranges from 19-22% after 3650 days of injection and all curves increase at similar rates.

Case 1 represents a base case in which SAGD was initiated from the beginning without preheating. This case produced the lowest percent recovery curve. In Case 2, we increase the injection rate constraint which thereby increases the pressure differential between the injection and production wells; hence, the conditions are referred to as extreme. The pressurization of the system improves production somewhat relative to the base case.

The "Circulate" phase in Cases 3 and 4 is a modified form of steam circulation in the well. Steam is injected so as to maintain the initial reservoir pressure. We did not simulate true steam circulation where steam that exits the tubing may only flow in the well before it is produced. A true circulating case in which the near-wellbore area is heated only by conduction would be inefficient, and the other techniques that we explore present better options. Circulation here is similar to the SAGD case: steam may replace oil volume in the reservoir when oil is produced. Hence, our "circulating" condition is somewhat of a misnomer.

The cumulative steam oil ratio displayed in Fig. 6 is standard: the cumulative steam volume injected as cold water equivalent volume divided by the cumulative hydrocarbon production. The CSOR varies substantially among the studies during the initial period because significantly different production and injection schemes were used. At late time, however, the CSOR for all cases averages roughly 3.0. Note, however, that the cyclic cases perform somewhat better, with regard to CSOR, in initial and late-time response.

Case Studies

Three of the cases above will be examined in greater detail to explain the differences in oil recovery and heating of the reservoir volume. The reference continuous SW-SAGD, extreme pressure differential, and one cycle prior to SW-SAGD cases will be discussed. Production rates, well pressures, and temperature profiles around the well will be examined.

Case 1, Continuous SAGD. In Case 1 we immediately operate at SAGD conditions and do not include a preheat phase. Figure 7 displays the injection and production curves for Case 1. The darkest curve in Fig. 7 represents the oil production rate. As expected, the initial oil rate is low, but increases with time as a steam chamber slowly develops and more oil is heated. Oil production peaks at roughly 80 m³/day.

Note that our "SAGD" case is actually a combination of SAGD and pressure draw-down. Production well conditions are such that reservoir pressure must decline. It is clear from the similarity between the steam injection rate and water production rate in Fig. 7 that steam short-circuits from the injection region to the production region and the contact time is short between steam and reservoir. Recall that we represent the horizontal well with two separate sections placed end to end. The pressure differential between the regions and the

proximity of "injection" and "production" perforations causes steam to travel quickly between them. Albeit inefficiently, a steam chamber is created within the reservoir as heated oil drains to the production region and steam migrates up to fill the void space. Optimizing the effect of spacing between injection and production sections represents another interesting problem to be addressed later in this report.

Figure 8 displays bottom-hole pressure curves for injection and production in Case 1. A large pressure differential of about 3000 kPa exists initially between the two sections of the well. Over time, the reservoir pressure decreases because we produce more fluids than we inject. This also causes the injection pressure to decrease. Figure 9 displays a temperature profile at 100 days for Case 1. Light shading corresponds to high temperature, and dark shading to low temperature. At late times, a large steam chamber grows in the middle region of the system. At 100 days, however, the steam chamber is just beginning to grow above the area between the injection and production sections. It is important to maximize the amount of net heat injection into the reservoir at early times to maximize the size of the heated volume surrounding the wellbore.

Case 2, Extreme Pressure Differential Prior to SAGD. In the extreme pressure differential case, the injection rate constraint is increased and this increases the pressure differential between the injection and production wells. Figure 10 displays the bottom-hole pressure histories. For the first 100 days, steam is injected at roughly 7000 kPa forcing steam into the formation and increasing the average reservoir pressure. Figure 11 displays the production response for the extreme period in the first 100 days followed by SAGD. Observing the oil rate in the first 100 days and comparing to Fig. 7, we see that the oil rate ramps up faster than Case 1. This is logical because Case 2 is an accelerated version of SAGD.

Figure 10 also indicates that a very high injection bottom-hole pressure is obtained between 0 and 100 days. High pressure results because the water production rate is substantially less than the steam injection rate, as shown in Fig. 11. Under the given conditions a limited amount of steam short-circuits, and an appreciable amount of steam enters the reservoir and increases the reservoir pressure. Pressure does not exceed the critical pressure where the formation parts or fractures.

If we view the oil production rate in Fig. 11 during and after the extreme period, it is obvious that we have improved response. Direct comparison of Cases 2 and 1 is somewhat misleading. Injection conditions have led to high reservoir pressure at the beginning of SAGD, causing significant production through pressure depletion in addition to gravity drainage of heated reservoir fluids. A better comparison is the temperature profile along the length of the well displayed in Fig. 12. The profile represents a relative time similar to the Case 1 profile, 100 days after SAGD inception. Again, light shading is high temperature and dark shading is low temperature. The profile for Case 2 is much more favorable. There is a larger heated area with a larger steam chamber. The steam chamber forms in the middle of the well because pressure drawdown is large; thus, the steam flux is largest in this location.

Case 5, One Cycle Prior to SAGD. Our cyclic case is very similar to typical cyclic conditions common in many thermal recovery operations. We inject steam along the entire well for 50 days, let it soak for 10 days, then produce along the entire length of the well for 120 days. The injection and production profiles in Fig. 13 summarize this cycle of steam injection, shut in, production followed by SAGD.

Figure 14 shows that the bottom-hole pressure increases to about 8000 kPa during the cyclic phase, but still remains within a feasible range. Because the oil is very viscous, this energy is rapidly depleted from the reservoir when production begins. From the oil production rate after the cycling period in Fig. 14, it is again obvious that SAGD response is

improved. The slow increase of production rate found in Case 1, Fig. 7, is not evident here. The minimum production rate at roughly 200 days occurs because reservoir pressure is depleted somewhat following the cyclic period, as shown by the plot of well bottom hole pressure in Fig. 14. Again, the maximum oil production rate is about 80 m³/day. In this case, the reservoir pressure at SAGD inception is similar to that in Case 1. We conclude that SAGD performs better because the near-wellbore area is heated.

Figure 15 displays the temperature profile at 120 days following SAGD inception. The temperature distribution is more uniform along the entire wellbore. Also, a large steam chamber is growing in the area immediately above the injection/production section. Additionally, the shading indicates that the entire horizontal length of the well has been heated somewhat. Hence, the rapid production response displayed in Fig. 14. Also, by comparing Figs. 15 and 12, it is apparent that the heated zone is larger than case 2 above.

Contrary to the extreme pressure differential and SAGD cases where short-circuiting caused much of the steam to exit the reservoir immediately, the cyclic case is more efficient. All of the injected steam enters the reservoir and heats the near-wellbore area. One consequence of this is the uniform temperature distribution along the entire wellbore. Because of the increased thermal efficiency of the cyclic process, it appears that this procedure is the most appealing method of initiating SAGD.

Discussion of Early-Time Analysis

The problem of improving early-time performance of SW-SAGD transforms, essentially, into a problem of heating rapidly the near-wellbore area to create conditions that allow gravity drainage of oil. More specifically, for a steam chamber to grow, oil viscosity must be low enough so that fluid drains creating voidage for steam to fill. After the conditions necessary for gravity drainage of oil have been initiated by preheating, the SW-SAGD process allows for continuous steam chamber growth and oil production. Comparing the various simulation results, cyclic steam injection appears to be the most efficient method of heating the near-wellbore area. The problem of optimizing the early-time cyclic procedure is studied further in our companion paper⁷.

An important general observation is that regardless of the process, early-time procedures should be carried out to maximize steam injection and heat delivery to the reservoir. The goal of any early-time procedure should be to heat the near-wellbore area as uniformly as possible. This goal is easier to achieve when operating at a maximum steam temperature. Later in the SAGD process, pressure can and should be reduced to a target operating pressure which optimizes efficiency and production rate.

The late time performance for all of the cases is favorable regardless of the early-time process. This confirms Joshi's finding that a steam chamber will grow in the reservoir. Favorable recovery factors are obtainable regardless of the injection/ production configuration³.

There are obvious factors that will improve or inhibit SW-SAGD performance. For example, lower viscosity will certainly improve response, as will larger rock permeability and system compressibility. The actual variance in performance due to differing reservoir parameters is an interesting problem that we begin to address using a sensitivity analysis in the next section.

Sensitivity Analysis

We performed a sensitivity analysis of various reservoir parameters to gain a better understanding of their effect on production performance. Oil viscosity and gas content, reservoir thickness, and horizontal to vertical permeability anisotropy were studied.

The sensitivity analysis base case varies slightly from the base case used in the early-time analysis. Table 3 displays the operating conditions for the sensitivity analysis base case. The base case consists of two steam injection cycles followed by SAGD operating conditions. We adjusted the operating conditions in an attempt to reduce steam short-circuiting, reduce steam-oil ratio, and to improve efficiency. During the initial cyclic period, we reduced the maximum reservoir pressure to 8,000 kPa from 10,000 kPa while keeping the rest of the operating conditions the same. The steam injection temperature was 296 °C, which corresponds to a steam pressure of 8004 kPa. During SAGD, operating conditions were chosen so that the process operated near the original reservoir pressure of 2654 kPa. Maximum injection pressure was set slightly above initial reservoir pressure at 3610 kPa; minimum production pressure was set slightly below initial reservoir pressure at 2230 kPa. The steam injection temperature was reduced to 244 °C from 295 °C. A steam temperature of 244 °C corresponds to a steam pressure of 3610 kPa. Recall that initial reservoir conditions were chosen to approximate field conditions⁴. The rate constraints remained the same at 300 m³/d maximum liquid production and 200 m³/d maximum steam injection rate.

Between the injection and production sections of the horizontal well, a 30 m long unperforated section was added. This is an attempt to force steam to penetrate substantially far into the reservoir and reduce the amount of steam short-circuiting.

Sensitivity Cases

The various cases created in the sensitivity analysis include: (i) a base case with the properties in Table 1 and operating conditions shown in Table 3, (ii) dead oil to examine the effect of solution gas, (iii) an increase in pay thickness to 28 m, (iv) a reduction in pay thickness to 4 m, (v) permeability anisotropy $k_h:k_v$ of 1, (vi) $k_h:k_v$ equal to 2, (vii) $k_h:k_v$ equal to 10, (viii) oil viscosity of 20 Pa-s at initial reservoir temperature (20,000 cP), and (iv) oil viscosity of 40 Pa-s (40,000 cP) at initial reservoir temperature. The viscosity versus temperature relation for the base case is given in Table 1 and those for the 20 and 40 Pa-s cases are listed in Table 3.

All calculations in the sensitivity cases include a separation of 30 m between injector and producer sections of the well to achieve better steam distribution in the reservoir. Two periods of cyclic steam injection precede continuous injection and production. In both periods, steam is injected for 50 days at a rate of 300 m³/day with a maximum allowed pressure of 8000 kPa. The soak period is 10 days and the length of production is a relatively short 40 days to avoid large heat withdrawal from the reservoir. Continuous SAGD operating conditions follow. All of the various sensitivity runs follow this procedure.

The recovery factor vs. time and cumulative steam-oil ratio (CSOR) versus time curves are presented in Figs. 16 and 17, respectively. Significant variation is found in production response between most of the cases. Briefly, production is highest when the reservoir permeability is isotropic, and is worst when the initial oil viscosity is high or the oil-saturated thickness is small. The steam-oil ratio is largest for the short formation thickness of 4 m. The next largest CSOR is for the viscous 40 Pa-s initial oil case. The smallest CSOR is found for the case with a pay zone thickness of 28 m and for cases where $k_h:k_v$ approaches 1. Note that recovery factors are less here, Fig. 16, compared to the early-time study, Fig. 5, because the steam injection pressure has been reduced substantially as seen by comparing Tables 2 and 3.

Due to space restrictions, it is impossible to display the instantaneous production rates, bottomhole pressure, and temperature profiles for all of the sensitivity cases. However, a brief discussion of the results of each of the sensitivity cases will be given.

The base case oil viscosity is 4.043 Pa-s at initial reservoir conditions. Two additional cases were created by increasing the initial viscosity to 20 Pa-s and then 40 Pa-s. The recovery factors at 3650 days (10 years) for these two less favorable cases were 4.5% and 2.7%, respectively, as compared to 10.6% for the base case. The cumulative steam-oil ratios were also unfavorable. The CSOR were 3.7 and 4.8 after 10 years for the 20 and 40 Pa-s cases, respectively. The problem is quite simple: it is hard for the viscous oil to drain and therefore it cannot be replaced by steam. In the base case, the average oil production rate is 75 m³/day over the 3650 days of production and excluding the startup period. For the 20 Pa-s case the average oil production rate is 30 m³/day while for the 40 Pa-s case it is 20 m³/day. As the viscosity increases the average oil production rate decreases, it becomes difficult to meet the steam injection target, and the size of the steam chamber is smaller compared to the less viscous cases.

The base horizontal to vertical permeability ratio, $k_h:k_v$, is 5. The vertical permeability was adjusted to create three more cases with ratios of 1, 2, and 10. In all cases, the horizontal permeability remained fixed at 3400 mD. Larger ratios present less favorable conditions for upward steam migration and oil drainage. As $k_h:k_v$ increases, cumulative recovery at a given time decreases and the CSOR increases, as shown in Figs. 15 and 16.

The base case oil composition contained 10% gas by mole which translates to a solution gas oil ratio of about 28 SCF/STB. An additional case that models a dead oil was created by reducing the solution gas to 1% by mole. The two effects of solution gas in regard to the oil phase are to increase oil-phase compressibility and reduce the oil-phase viscosity. Oil phase viscosity is computed as a mole fraction weighted sum of the oil and gas component viscosities⁶. Recovery factor and CSOR were less favorable with the dead oil at 9.2 % and 3.0 after 10 years as compared to 10.4% and 2.8 for the base case. Likewise, the average oil production rate is 64 m³/day over the ten year period as compared to 80 m³/day for the base case.

Discussion of Sensitivity Analysis

Taken together, the results from the sensitivity analysis suggest that application of SW-SAGD to exceptionally viscous oils will be difficult. When the initial viscosity is greater than 10 Pa-s, oil drainage becomes very slow and it is difficult to form a large steam chamber. Likewise, the application of SW-SAGD to thin oil zones with thickness of roughly 4 m does not appear to be feasible. A steam chamber of significant height must develop for efficient oil drainage. On the other hand, the 19.6 and 28 m thick cases showed significant recovery and little difference was found between these two cases.

The presence of solution gas also aids recovery somewhat. Even though the oil is viscous and the amount of gas low in both cases, volumetric expansion of the oil is aided by solution gas and cumulative recovery increases by 10% relative to the dead oil case at 3650 days.

As expected, oil recovery improves as the permeability anisotropy decreases. Both steam injection and oil drainage are aided as the vertical permeability increases relative to the horizontal. Resistance to flow in the vertical direction decreases relative to horizontal. Additionally, less short circuiting occurs as k_v increases relative to k_h .

Injector to Producer Spacing

In all of the runs in the sensitivity analysis, 30 m of unperforated well separated injection and production regions. Separating the regions further increases the length of time required for steam to flow from the injection to the production section and thereby improves

growth of the steam chamber. However, separation between injection and production regions delays oil production.

To quantify the effect of separating injection and production regions, we performed the following highly approximate analysis. The Darcy velocity of a particle or volume of steam (u_v) flowing upward above the horizontal injection section is given by

$$u_v = -\frac{k_v k_{rs}}{\mu_s} \Delta \rho g \quad (1)$$

where k_v is the vertical permeability, k_{rs} is the relative permeability to steam, μ_s is the steam viscosity, $\Delta \rho$ is the density difference between the oil and steam, and g is the acceleration due to gravity. The Darcy velocity in the horizontal direction (u_h) between the injection and production regions is given by

$$u_h = -\frac{k_h k_{rs}}{\mu_s} \frac{dp}{dx} \quad (2)$$

where k_h is the horizontal permeability and dp/dx is the pressure gradient in the horizontal direction.

The characteristic times required for a particle to travel to the top of the reservoir (t_v) and from the injection section to the production section (t_h) are given by

$$t_v = \frac{h}{u_v} \quad (3)$$

and

$$t_h = \frac{L}{u_h} \quad (4)$$

where h is the height of the reservoir and L is the distance between injection and production regions. If the $t_v:t_h$ ratio is larger than one, then the time required for a particle to travel to the top of the reservoir is larger than the time required to travel to the production section. Therefore, we expect a large amount of steam short-circuiting. Conversely, if the $t_v:t_h$ is close to 1, we would expect limited steam short-circuiting, better steam flow through the reservoir, and increased steam chamber growth.

Substituting the Darcy velocity equations into the characteristic time equations and simplifying allows us to calculate the ratio in terms of available parameters.

$$\frac{t_v}{t_h} = \frac{h}{L} \frac{k_h}{k_v} \frac{\frac{dp}{dx}}{\Delta \rho g} \quad (5)$$

We can approximate dp/dx as,

$$\frac{dp}{dx} \cong \frac{\Delta p}{L} = \frac{P_{inj} - P_{prod}}{L} \quad (6)$$

The subscripts inj and prod refer to injection and production pressure. Substituting this into Eq. 5 yields

$$\frac{t_v}{t_h} = \frac{h}{L^2} \frac{k_h}{k_v} \frac{(P_{inj} - P_{prod})}{\Delta \rho g} \quad (7)$$

For the base case of the sensitivity analysis and using the properties displayed in Table 4, we calculate a $t_v:t_h$ value of 12. Therefore, we do not expect steam to have a strong tendency to flow upward. Although this analysis is just an approximation, it does help us to estimate if short-circuiting will be an immediate production problem. A more thorough analysis would consider variations in temperature, pressure, density, and viscosity as well as a more complete description of the pressure distribution.

To gauge the effect of greater distance between the injection and production regions, additional simulations were conducted using the properties of the base case for the sensitivity analysis, Table 3. As previously, two cycles of steam stimulation preceded continuous steam injection. Separation sizes of 60, 90, and 120 m were examined. These distances correspond to $t_v:t_h$ of 3.0, 1.3, and 0.75, respectively. In all cases, the length of the injection and production regions is the same only the distance between the two changes.

Recovery factor and CSOR histories for these calculations are given in Figs. 18 and 19. The general trend over 3650 days is that recovery decreases as the injection to production region spacing increases. The recovery factor for the 30 m separation distance is 10.4 % after 3650 days whereas that for the 90 m separation is 9.7%. On the other hand, the CSOR decreases only slightly as the separation size increases. This indicates that recovery efficiency has not increased greatly with the addition of unperforated well between the injection and production regions.

Throughout, we have specified equal lengths of injection and production regions of the well for simplicity. In practice, this is both unnecessary and unlikely to be the case. It is probable that the injection region could be shortened considerably thereby devoting a greater portion of the well to production. As steam is much less viscous and dense than the oil, steam mobility is large. Thus, it should be possible to distribute steam in the reservoir and develop a steam chamber over the horizontal well with a much smaller perforated section provided that steam short circuiting from injector to producer regions is minor. We do not attempt such an analysis here.

Conclusions

Here it is shown that to improve early-time performance of SW-SAGD, it is necessary to heat the near-wellbore region rapidly and uniformly to reduce oil viscosity and promote gravity drainage. Cyclic steaming, as a predecessor to SW-SAGD, represents the most thermally efficient early-time heating method. Uniform heating along the length of the wellbore appears achievable with cyclic steam injection. Immediately placing a cold well on SAGD hinders the early-time heating process and initial production response in this case will be low. Regardless of the early-time process, it should be performed to provide maximum heat delivery to the reservoir. Additionally, despite different initial procedures, the oil production rates after several years of steam injection are all very similar.

The sensitivity analysis performed here indicates that SW-SAGD is most applicable to heavy oils with initial viscosity below 10 Pa-s (10,000 cP). Additionally, the reservoir must be sufficiently thick to allow significant vertical steam chamber growth. Recovery from thin oil zones is not significant. The sensitivity analysis also indicates that the presence of relatively small amounts of solution gas aids the recovery process by reducing oil-phase viscosity and enhances volumetric expansion of the oil on heating.

A simplified analysis was completed to examine the effect of increasing the distance between regions of the well dedicated to injection and production. The cumulative steam-oil

ratio did not decrease substantially with increased separation indicating that this strategy does not enhance steam chamber growth or recovery efficiency.

Nomenclature

CWE	cold water equivalent
g	acceleration due to gravity
h	net reservoir height
k	permeability
k_r	relative permeability
L	distance between injection and production regions
p	pressure
t	characteristic time
u	Darcy velocity
x	distance

Greek

μ	viscosity
ρ	phase density

Subscripts

h	horizontal
s	steam
v	vertical

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Table 1: Grid, rock, and oil property description.

Grid System		Rock Properties	
total number of blocks	5,568	porosity	33%
x-dimension	1,400 m	k_h	3,400 mD
y-dimension	80 m	k_v	800 mD
z-dimension	19.6 m		
well length	800 m		
Reservoir Properties		Oil Properties	
initial pressure	2,654 kPa	components	water, oil, and gas
initial temperature	16 °C	initial composition	90 % (mole) oil
initial S_o	85%		10% (mole) gas
initial S_w	15%	viscosity (mPa-s) versus temperature (K) function	$\mu = 1.74 \times 10^{-6} \exp\left(\frac{6232.74}{T}\right)$

Table 2: Operating conditions for early-time performance study.

Property	Operating Condition			
	SAGD	Extreme	Cyclic	Circulating
steam temperature (°C)	295	295	295	295
maximum injection rate (CWE m ³ /day)	200	600	300	300
maximum injection pressure (kPa)	10,000	10,000	10,000	10,000
maximum production rate (m ³ /day)	300	600	300	300
minimum production pressure (kPa)	500	500	500	500

Table 3: Description of sensitivity analysis.

Property	Operating Condition	
	Initial Cyclic	SAGD
steam temperature (°C)	295	244
maximum injection rate (CWE m ³ /day)	300	300
maximum injection pressure (kPa)	8000	8000
maximum production rate (m ³ /day)	300	300
minimum production pressure (kPa)	2,230	500
Oil Viscosity		
20,000 mPa-s case (μ (mPa-s), T(K))	$\mu = 8.61 \times 10^{-6} \exp\left(\frac{6232.74}{T}\right)$	
40,000 mPa-s case (μ (mPa-s), T(K))	$\mu = 1.72 \times 10^{-5} \exp\left(\frac{6232.74}{T}\right)$	

Table 4: Parameters used to calculate t_v/t_h .

Parameter	Value
h	19.6 m
L	30 m
k _h	3400 mD
k _v	800 mD
p _{inj}	3230 kPa
p _{prod}	2230 kPa
oil density	950 kg/m ³
steam density	42.6 kg/m ³
g	9.8 m/s ²

Figure 1: Schematic of grid system: (a) parallel to well bore and (b) perpendicular to wellbore.

Figure 2: Water-oil relative permeability function.

Figure 3: Gas-liquid relative permeability function.

Figure 4: Recovery factor for the first year of production.

Figure 5: Recovery factor for 10 years of production.

Figure 6: Cumulative steam oil ratio for 10 years of production.

Figure 7: Steam injection and production history, SAGD from start-Case 1.

Figure 8: Bottomhole pressure, SAGD from start-Case 1.

Figure 9: Temperature profile at 100 days, SAGD from start-Case 1.

Figure 10: Bottom-hole pressure, extreme pressure differential then SAGD-Case 2.

Figure 11: Steam injection and production history, extreme pressure differential then SAGD-Case 2.

Figure 12: Temperature profile at 200 days, extreme pressure differential then SAGD-Case 2.

Figure 13: Steam injection and production history, 1 cycle then SAGD-Case 5.

Figure 14: Bottomhole pressure, 1 cycle then SAGD-Case 5.

Figure 15: Temperature profile at 200 days, 1 cycle then SAGD-Case 5.

Figure 16: Recovery factor versus time, sensitivity analysis.

Figure 17: Cumulative steam oil ratio, sensitivity analysis.

Figure 18: Recovery factor versus time, effect of injector to producer section spacing.

Figure 19: Cumulative steam-oil ratio, effect of injector to producer section spacing.

3.5 AN ANALYTICAL MODEL FOR SIMULATING HEAVY-OIL RECOVERY BY CYCLIC STEAM INJECTION USING HORIZONTAL WELLS

(U. Diwan and A.R. Kovscek)

Abstract

In this investigation, existing analytical models for cyclic steam injection and oil recovery are reviewed and a new model applicable to horizontal wells is proposed. A new flow equation for oil production during cyclic steaming of horizontal wells is developed. The model accounts for the gravity-drainage of oil along the steam-oil interface and through the steam zone. Oil viscosity, effective permeability, geometry of the heated zone, porosity, mobile oil saturation, and thermal diffusivity of the reservoir influence the flow rate of oil in the model. The change in reservoir temperature with time is also modeled, and it results in the expected decline in oil production rate during the production cycle as the reservoir cools. Wherever appropriate, correlations are incorporated to minimize data requirements. A limited comparison to numerical simulation results agrees well indicating that essential physics are successfully captured.

Cyclic steaming appears to be a systematic method for heating a cold reservoir provided that a relatively uniform distribution of steam is obtained along the horizontal well during injection. A sensitivity analysis shows that the process is robust over the range of expected physical parameters.

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Nomenclature

A	= Cross sectional area	$\text{ft}^2 (\text{m}^2)$
A_L	= Dimensionless factor for linear flow	
C_w	= Specific heat of water	$\text{Btu/lb.}^\circ\text{F} (\text{kJ/kg}^\circ\text{K})$
f_{HD}	= Dimensionless factor that accounts for horizontal heat losses	
f_{VD}	= Dimensionless factor that accounts for vertical heat losses	
f_{PD}	= Dimensionless factor that accounts for heat lost by produced fluids	
f_{sdh}	= Steam quality at down-hole conditions (mass fraction)	
F_l	= Ratio of heat lost to heat injected	
g	= Gravitational acceleration	$32.17 \text{ ft/s}^2 \text{ or } 9.81 \text{ m/s}^2$
h	= Formation thickness	$\text{ft} (\text{m})$
h_{st}	= Steam zone thickness	$\text{ft} (\text{m})$
$h_w(T)$	= Enthalpy of liquid water	$\text{Btu/lb} (\text{kJ/kg})$
H_{inj}	= Amount of heat injected during a cycle	$\text{Btu} (\text{kJ})$
H_{lost}	= Amount of heat remaining in the reservoir at the end of a cycle	$\text{Btu} (\text{kJ})$
K_R	= Reservoir thermal conductivity	$\text{Btu/ft.D.}^\circ\text{F} (\text{kJ/m.d.}^\circ\text{K})$
k_o	= Effective permeability of oil	Darcy
k_{st}	= Effective permeability of steam	Darcy
k_w	= Effective permeability of water	Darcy
k_{ro}	= Relative permeability to oil	fraction
k_{rw}	= Relative permeability to water	fraction
ΔH_{lvdh}	= Latent heat of vaporization at down-hole conditions	$\text{Btu/lb} (\text{kJ/kg})$
L	= Length of horizontal well	$\text{ft} (\text{m})$
M_o	= Volumetric oil heat capacity	$\text{Btu/ft}^3.^\circ\text{F} (\text{kJ/m}^3.^\circ\text{K})$
M_w	= Volumetric water heat capacity	$\text{Btu/ft}^3.^\circ\text{F} (\text{kJ/m}^3.^\circ\text{K})$
M_{SH}	= Volumetric shale (overburden / underburden) heat capacity	$\text{Btu/ft}^3.^\circ\text{F} (\text{kJ/m}^3.^\circ\text{K})$
M_T	= Total volumetric heat capacity	$\text{Btu/ft}^3.^\circ\text{F} (\text{kJ/m}^3.^\circ\text{K})$
M_σ	= Volumetric rock heat capacity (M_{rr})	$\text{Btu/ft}^3.^\circ\text{F} (\text{kJ/m}^3.^\circ\text{K})$
N_p	= Cumulative oil production	$\text{bbl} (\text{m}^3)$
OIP	= Volume of oil in place	$\text{bbl} (\text{m}^3)$
p_s	= Down-hole steam pressure	psia, (kPa)
p_{wf}	= Bottom hole flowing pressure	psia, (kPa)
q_o	= Oil production rate	$\text{BPD} (\text{m}^3/\text{d})$
q_w	= Water production rate	$\text{BPD} (\text{m}^3/\text{d})$
$\overline{Q}_i$	= Amount of heat injected per unit mass of steam	$\text{Btu/lb} (\text{kJ/kg})$
Q_i	= Cumulative heat injected	$\text{Btu/lb} (\text{kJ/kg})$

Q_{MAX}	= Maximum heat entering in the reservoir in one cycle	Btu (kJ)
Q_l	= Heat loss to reservoir	Btu (kJ)
w_s	= Steam injection rate (cold water volume)	BPD (m^3/d)
Q_p	= Rate of heat withdrawal from reservoir with produced fluids	Btu/D (kJ/d)
r_w	= Well radius	ft (m)
R_h	= Heated zone horizontal range	ft (m)
R_x	= Distance along hot oil zone interface	ft (m)
S_{ors}	= Residual oil saturation in the presence of steam	
S_{orw}	= Residual oil saturation in the presence of water	
S_{wi}	= Initial water saturation	
$\bar{S}_w$	= Average water saturation	
t_{DH}	= Dimensionless time for horizontal heat loss	
t_{DV}	= Dimensionless time for vertical heat loss	
t_{inj}	= Injection period	days
t_{prod}	= Production time	days
t_{soak}	= Soak time	days
T_{avg}	= Average temperature of heated zone	$^{\circ}F$ ($^{\circ}K$)
T_R	= Original reservoir temperature	$^{\circ}F$ ($^{\circ}K$)
T_S	= Down-hole steam temperature	$^{\circ}F$ ($^{\circ}K$)
T_{SOAK}	= Temperature at the end of soaking period	$^{\circ}F$ ($^{\circ}K$)
T_{STD}	= Temperature at standard conditions	$^{\circ}F$ ($^{\circ}K$)
u	= Velocity of the steam-oil interface growth downwards	ft/sec (cm/sec)
v	= Darcy velocity	ft/sec (cm/sec)
V_s	= Steam zone volume	ft^3 (m^3)
w_i	= Total injection rate	BPD (m^3/d)
WIP	= Volume of water in place	bbl (m^3)
W_p	= Cumulative water production	BPD (m^3/d)
α	= Reservoir thermal diffusivity	ft^2/D (m^2/d)
α_{SH}	= Shale thermal diffusivity	ft^2/D (m^2/d)
Δp	= Pressure drawdown	psi (kPa)
Δt	= Time step size	days
ϕ	= Porosity	
ϕ	= Potential	ft^2/sec^2 (m^2/sec^2)
μ_o	= Oil viscosity	cp
μ_{st}	= Steam viscosity	cp
μ_w	= Water viscosity	cp
ν	= Kinematic oil viscosity	cs
ν_{avg}	= Kinematic oil viscosity at average temperature	cs
ν_s	= Kinematic viscosity of oil at steam temperature	cs
ν_{ws}	= Kinematic viscosity of water at steam temperature	cs
ρ_o	= Oil density	lb/ ft^3 (gm/ml)
ρ_{oSTD}	= Oil density at standard conditions	lb/ ft^3 (gm/ml)
ρ_{st}	= Steam density	lb/ ft^3 (gm/ml)
ρ_w	= Water density	lb/ ft^3 (gm/ml)
θ	= Angle between steam-oil interface and reservoir bed	degrees (radians)
ξ	= Hot oil zone thickness	cm

Chapter 1

1. Introduction

Shell discovered the process of steam stimulation by accident in Venezuela during production of heavy crude by steamflooding the Mene Grande field near the eastern shore of Lake Maracaibo (Butler et al., 1980). During the flood, a breakthrough of steam to the ground surface occurred and, in order to reduce the steam pressure in the reservoir, the injection well was allowed to flow back. Copious quantities of oil were produced; from this accidental discovery in 1959 came the cyclic steam stimulation process, which also goes by the name of *steam soak* and *huff and puff*. Since then, there have been several mathematical models describing the phenomenon. These range from complex numerical simulators to simple analytical expressions.

This work concerns the application of horizontal wells to thermal oil recovery. It consists of an analytical model developed to calculate oil recovery and reservoir heating during cyclic steam injection. It holds for heavy-oil, pressure-depleted reservoirs where the main driving force for production is gravity. Our objective is to present a relatively simple model taking into account gravity-drainage along the sides of the steam-oil interface, the pressure draw down driving force and the drainage of oil through the steam zone. A brief overview of the analytic and semi-analytic models for response to steam injection follows next.

1.1 Horizontal Wells

Horizontal wells are applied increasingly in steam injection projects for recovering heavy-oil (Basham et al., 1998). In the well-known steam assisted gravity-drainage (SAGD) process, a horizontal injector is located above a horizontal producer (Butler et al., 1980). The producer below collects and drains away the mobilized oil and water (condensed steam). Often the injector contains tubing for delivering steam to the well toe, while the annulus directs the steam to the formation and produces the excess for circulation. Recently, there has been interest in heavy-oil reservoirs in the application of a single dual-stream horizontal well, where the annulus assumes the role of the producer, and the tubing, the injector. Cyclic steaming using a single horizontal well could be considered a variant of SW-SAGD and should be useful for efficient initial heating of the reservoir volume.

The performance of such wells may be predicted from empirical correlations, simple analytical models, or thermal reservoir simulation. Empirical correlations can be extremely useful for correlating data within a field and for predicting performance of new wells in that and similar fields. However, use of such correlations for situations much different from those that led to their development might be subject to large errors. On the other hand, one can use a compositional or black-oil thermal model to predict the performance of cyclic steam operations. Thermal models are based on the fundamental laws of conservation of mass and heat. Fluid flow is related to pressure gradient through the concept of relative permeability. In addition, a thermal model is sensitive to rock properties, fluid properties and geological features. Much of this information is often unknown and must be estimated from limited data and experience in similar situations. Furthermore, because of the complexity of the SAGD recovery process, the equations of a thermal simulator could be difficult and expensive to solve depending upon the exact scenario. Simply, an analytical model of cyclic steam injection might be useful to expose the first-order mechanisms of reservoir heating and oil production.

1.2 Cyclic Steam Injection

There are two important reasons to study cyclic steam injection with horizontal wells. Firstly, the thermal efficiency of cyclic operation is high, and it is thus attractive. Secondly, cyclic steaming to promote effective initial reservoir heating might precede continuous steam injection

during a SW-SAGD recovery process (Elliott and Kovscek, 1999). The steam is usually injected at a fixed rate and known wellhead quality for a short period of time. After some heat loss in the wellbore, the steam enters the reservoir. Uniform steam distribution along the well is an important factor in the success of cyclic operations with a horizontal well (Mendonza, 1998). The bottom-hole quality and pressure may be predicted from a wellbore model of the type discussed by Fontanilla and Aziz (1982). After injection for a specified period of time, the well is shut-in and the steam is allowed to "soak" into the reservoir for another specified period. To complete the cycle, the well is produced until the oil production rate reaches a minimum rate. This cyclic process is repeated until the recovery per cycle drops below an economic limit. Generally, the length of cycles increases as recovery matures. Given the bottom-hole conditions during the production cycle, the wellhead conditions may be predicted.

In principle, the reservoir simulator yields the most accurate answer, but it is only possible to generalize after many different simulations (Aziz and Gontijo, 1984). Further, the reservoir simulator is sensitive to data that are often not known or unreliable. It is natural to develop simpler analytical models (Boberg and Lantz, 1966) that account for the important mechanisms involved in this process from which we may draw general conclusions about performance. This indeed has been the case and several models and correlations of varying degree of complexity are available in the literature for cyclic operation of vertical wells and continuous steam injection in dual horizontal wells (Butler et al., 1980).

1.3 Existing Models

Briefly, let us touch upon the models that have been created so far. Marx and Langenheim (1959) describe a method for estimating thermal invasion rates, cumulative heated area, and theoretical economic limits for sustained hot-fluid injection at a constant rate into an idealized reservoir. Full allowance is made for non-productive reservoir heat losses. In all cases, the heat conduction losses to the overburden and the underburden impose an economic limit upon the size of the area which can be swept out from any one injection point, for a given set of reservoir conditions, at any given heat injection rate.

Jones (1977) presented a simple cyclic steam model for heavy-oil, pressure-depleted, gravity-drainage reservoirs. The Boberg and Lantz (1966) procedure was used as the basis for the reservoir shape and temperature calculations versus time. Here, the only driving force assumed is gravity, and hence, the model tends to calculate lower initial oil rates than observed in the field. For matching, certain empirical parameters are employed.

Van Lookeren(1977) has presented calculation methods for linear and radial steam flow in reservoirs. He assumes immediate gravity overlay of the steam zone and presented analytical expressions to describe interface locus. The steam zone shape is governed by factors A_{LD} and A_{RD} which are dimensionless parameters that characterize the degree of steam override for linear and radial flow, respectively. A simplistic formulation is given to calculate the average steam zone thickness.

Myhill and Stegemeier (1978) presented a model for steam drive correlation and prediction. Assuming a piston like displacement, they modified Mandl and Volek's (1969) method to calculate the steam zone volume. It identifies a critical time beyond which the zone downstream of the advancing front is heated by the water moving through the condensation front. Also, a thermal efficiency of the steam zone is calculated as a function of the dimensionless time and the ratio of the latent heat to the total energy injected.

Butler *et al.* (1979) presented theoretical equations for gravity-drainage of heavy-oils during in-situ steam heating. The method described consists of an expanding steam zone as a result of steam injection and production of oil via the mechanism of gravity-drainage along the interface of the steam chamber. The oil is produced through a horizontal well located at the bottom of the steam chamber. Oil flow rate is derived starting from Darcy's law. Heat transfer takes into account the thermal diffusivity of the reservoir and it is proportional to the square root of the driving force. In the case of an infinite reservoir, an analytical dimensionless expression describing the position of the interface is derived. When the outer boundary of the reservoir is considered, the position of the interface and the oil rate are calculated numerically. A simple second order polynomial is found to fit dimensionless oil production versus time. An equation describing the growth of the steam chamber is also presented. The method is limited to gravity-drainage and linear flow of heavy-oil from horizontal wells.

Jones (1981) presented a steam drive model that is basically a combination of Van Lookeren's (1977) and Myhill and Stegemeier's (1978) methods. It is limited to continuous steam drive and it uses empirical factors to match calculated rates with measured values.

Vogel (1982) considered heat calculations for steam floods. Similar to Van Lookeren (1977), this model works on the basic assumption of instantaneous steam rise to the top of the reservoir. After this happens, the steam chamber grows downward at a very low velocity. Heat losses to the adjacent formations are calculated by solving the heat conduction problem from an infinite plane. The model characterizes two main driving forces affecting oil production: gravity-drainage and steam drag. In his conclusions, Vogel says that, above a certain limit, injection rates have little influence on oil production.

Finally, Aziz *et al.* (1984) presented a model that considers the flow potential to be a combination of pressure drop and gravity forces. The flow equation for oil and water production derived is based on the method illustrated by Butler *et al.* (1979). They solve a combined Darcy flow and a heat conduction problem. The structure of the model is based on Jones' (1981) method.

Chapter 2

2. Model Development

Cyclic steam injection, commonly referred to as "Huff-n-Puff" involves steam injection into a pressure-depleted reservoir, followed by soaking, and finally the oil production period. As stated earlier, oil production is governed by the gravity driving force. The model developed in the following section incorporates gravity as a prime driving force for oil flow towards the well and thereby predicts the oil production flow rate per unit length of the horizontal well in the production period. Heat losses from the heated zone to the overburden and the adjacent unheated oil-bearing formation are also included.

2.1 Introduction

Steam introduced at the bottom of the formation through a horizontal well, displaces the oil and rises to the top of the formation where it is trapped if an impermeable cap rock exists. We assume that the steam zone adopts a triangular shape in cross section as shown in Fig.1. The steam injection rate around the well remains constant over the entire injection interval. Therefore, the steam injection pressure generally remains constant over the injection process if the reservoir is pressure-depleted. Steam heats up the colder oil sand near the condensation surface and oil drains along the condensation surface by a combination of gravity and pressure difference into the production well. Steam condensate also drains toward the well. In addition, oil drains through the steam chamber into the production well. The mechanisms involved in oil production during cyclic steam injection are diverse and intricate. Reduction of oil viscosity as a

result of an increase in the temperature greatly improves the production response. Gravity-drainage and pressure drawdown are the major mechanisms of oil production in the case of cyclic steaming.

The model is divided into three periods: the injection period, the soaking period and the production period. Each will be described in detail. In what follows, correlations and equations are given in field units ($^{\circ}\text{F}$, psi, ft, BTU, BPD, etc), unless otherwise noted.

2.2 Injection and Soak Periods

During the injection interval, heat losses from the steam zone to the reservoir are considered negligible although heat losses to the overburden must be considered. The oil sand near the wellbore is assumed to be at steam temperature T_s , the saturated steam temperature at the sand face injection pressure. Pressure fall-off away from the well during injection is neglected during this analysis. In the soaking period, heat is lost to the overburden and the reservoir. The temperature of the steam zone thus decreases while soaking

There are certain important variables such as the steam zone volume and the zone horizontal range that need to be addressed. The steam zone volume, V_s , during injection is calculated using the approach of Myhill and Stegemeier (1978) as

$$V_s = \frac{Q_i E_{h,s}}{M_T \Delta T} \quad (1)$$

where, $E_{h,s}$ is the thermal efficiency, ΔT is the temperature rise of the steam zone above the initial reservoir temperature (assumed to be same as the temperature rise at down-hole condition, ΔT_i), and M_T is the total volumetric heat capacity of the reservoir. Heat capacity is the sum of the rock and fluid contributions and is written

$$M_T = (1 - \phi)M_{rr} + \sum_{\beta=w,o,g} \phi S_{\beta} M_{\beta} \quad (2)$$

where, ϕ is the porosity, S is the phase saturation, and the subscript β refers to the individual phases. The quantity Q_i is the cumulative heat injected including any remaining heat from the previous cycles. The heat injection rate is given by

$$\dot{Q}_i = w_i (C_w \Delta T + f_{sdh} \Delta H_{vdh}) \quad (3)$$

where, w_i is the mass rate of steam injection in the reservoir, C_w is the average specific heat of water over the temperature range corresponding to ΔT , f_{sdh} and ΔH_{vdh} are the steam quality and the latent heat of vaporization both at down-hole conditions, respectively.

The thermal efficiency is calculated from the function given by Myhill and Stegemeier (1978) as

$$E_{h,s} = \frac{1}{t_D} \left\{ G(t_D) + (1 - f_{hv}) \frac{U(t_D - t_{cD})}{\sqrt{\pi}} \left[2\sqrt{t_D} - 2(1 - f_{hv})\sqrt{t_D - t_{cD}} - \int_0^{t_{cD}} \frac{e^x \operatorname{erfc}(\sqrt{x})}{\sqrt{t_D - x}} dx - \sqrt{\pi} G(t_D) \right] \right\} \quad (4a)$$

In the expression above, the heat losses to both, the overburden as well as the underburden are included. In the model proposed, the triangular cross sectional shape of the steam chamber is such that heat loss occurs only to the overburden. Thus, $E_{h,s}$ needs to be modified to be consistent with our model. Now, $E_{h,s}$ is defined as the ratio of the heat remaining in the zone to the total heat injected. Thus, the quantity $(1 - E_{h,s})$ is the ratio of the heat lost to the total heat injected

$$F_l = 1 - E_{h,s} \quad (4b)$$

Neglecting loss to the underburden, heat losses are approximately one-half of the value predicted by Eq.4(b). Therefore, the modified thermal efficiency is

$$(E_{h,s})_{mod} = 1 - \frac{1}{2} F_l = \frac{1}{2} (1.0 + E_{h,s}) \quad (4c)$$

In Eq.(4a), $G(t_D)$ is

$$G(t_D) = 2\sqrt{\frac{t_D}{\pi}} - 1 + e^{t_D} \operatorname{erfc}(\sqrt{t_D}) \quad (5)$$

The symbols t_D and t_{cD} represent dimensionless times given by

$$t_D = 4 \left(\frac{M_{SH}}{M_T} \right)^2 \frac{\alpha_{SH}}{h_t^2} t \quad (6)$$

where, α_{SH} is the shale thermal diffusivity, t is the time, h_t is the total reservoir thickness and M_{SH} is the shale volumetric heat capacity. The dimensionless critical time (Myhill and Stegemeier, 1978) is defined by

$$e^{t_{cD}} \operatorname{erfc}(t_{cD}) \equiv 1 - f_{hv} \quad (7)$$

The quantity f_{hv} , the fraction of heat injected in latent form, is given by

$$f_{hv} = \left(1 + \frac{C_w \Delta T}{f_{sdh} \Delta H_{vdh}} \right)^{-1} \quad (8)$$

The step-function U in Eq.(4) is defined as

$$U(x) = 0 \quad \text{for } x < 0 \quad (9a)$$

$$U(x) = 1 \quad \text{for } x > 0 \quad (9b)$$

We assume that the steam zone shape has a triangular cross section with a y-directional length equal to the length of the horizontal well, L (Fig.1). The volume is written as

$$V_s = R_h L h_{st} \quad (10a)$$

where R_h is one-half of the base of the triangular heated zone and h_{st} is the steam zone thickness also referred to as “Zone Horizontal Range”.

Rearranging Eq.(10a), it follows that,

$$R_h = \frac{V_s}{L h_{st}} \quad (10b)$$

2.3 Heat Remaining in the Reservoir

Boberg and Lantz (1966) give the average temperature of the steam zone, T_{avg} , as

$$T_{avg} = T_R + (T_s - T_R) [f_{HD} f_{VD} (1 - f_{PD}) - f_{PD}] \quad (11)$$

The dimensionless parameters f_{VD} , f_{HD} and f_{PD} are functions of time and represent horizontal loss, vertical loss and energy removed with the produced fluids, respectively. Aziz and Gontijo (1984) define them according to the following expressions.

The horizontal heat losses are expressed as (Aziz and Gontijo, 1984):

$$f_{HD} = \frac{1}{1 + 5t_{DH}} \quad (12a)$$

where,

$$t_{DH} = \frac{\alpha(t - t_{inj})}{R_h^2} \quad (12b)$$

An alternate treatment for the calculation of f_{HD} is presented in Appendix B. The quantity f_{HD} is the average unit solution for the one-dimensional heat conduction problem in the horizontal direction:

$$\alpha \frac{\partial^2 f}{\partial x^2} = \frac{\partial f}{\partial t} \quad (12c)$$

where $f = \left(\frac{T - T_o}{T_s - T_o} \right)$ is the dimensionless temperature.

The averaged solution for this equation after making use of the appropriate boundary conditions consistent with the geometry chosen for our model is:

$$f_{HD} = \frac{1}{2} + \sqrt{\frac{t_{DH}}{\pi}} \left\{ -2.0 + \exp\left(-\frac{1}{t_{DH}}\right) \right\} + \operatorname{erf}\left(\frac{1}{\sqrt{t_{DH}}}\right) \left\{ \left(\frac{2}{\sqrt{\pi}} - 1\right)t_{DH} + \frac{1}{2} \right\} \quad (12d)$$

Similarly, the vertical heat losses are expressed as (Aziz and Gontijo, 1984):

$$f_{VD} = \frac{1}{\sqrt{1 + 5t_{DV}}} \quad (13a)$$

where,

$$t_{DV} = \frac{4\alpha(t - t_{inj})}{h_t^2} \quad (13b)$$

In the equations above, α is the reservoir thermal diffusivity.

Note that for the first cycle, the initial amount of heat in the reservoir is set to zero. For all following cycles, the initial amount of energy is calculated based on steam zone volume and the average temperature at the end of the previous cycle. The average temperature at any time during the cycle is calculated using Eq.(11) (Boberg and Lantz, 1966). Since Boberg and Lantz's equation assumes a cylindrical shape for the heated zone, this equation is just an approximation for the triangular shape being considered. However, we use the approximations for f_{HD} and f_{VD} as employed by Aziz and Gontijo (1984) who assume a conical shape with triangular cross section. This is identical to the cross sectional shape that we have considered. The only difference is that they rotate the triangular shape through π radians whereas our coordinate system is Cartesian.

The term that accounts for the energy removed with produced fluids is given by

$$f_{PD} = \frac{1}{2Q_{MAX}} \int_0^t Q_p dt \quad (14)$$

where,

$$Q_p = 5.615(q_o M_o + q_w M_w)(T_{avg} - T_R) \quad (15)$$

where Q_{MAX} is the maximum heat supplied to the reservoir. It is calculated at the end of the soak period as the amount of heat injected plus the heat remaining in the reservoir from the previous cycle minus the losses to the over burden (Vogel, 1982)

$$Q_{MAX} = Q_{inj} + Q_{last} - 2R_h L K_R (T_s - T_R) \sqrt{\frac{t_{soak}}{\Pi \alpha}} \quad (16a)$$

$$Q_{inj} = 350.376 \bar{Q}_i w_s t_{inj} \quad (16b)$$

$$Q_{last} = V_s M_T (T_{avg} - T_R) \quad (16c)$$

where, L is the length of the horizontal well, K_R is the thermal conductivity of the reservoir, t_{soak} is the soak time, α is the reservoir thermal diffusivity, t_{inj} is the length of the injection cycle, Q_{last} is the heat remaining from the previous cycle, $\bar{Q}_i$ is the amount of heat injected per unit mass of steam, and w_s is the steam injection rate (cold water equivalent).

The steam pressure is calculated from the following approximate relationship (Prats, 1982):

$$p_s = \left[\frac{T_s}{115.95} \right]^{4.4545} \quad (17)$$

The volumetric heat capacities of oil and water are given as (Prats, 1982):

$$\begin{aligned} M_o &= (3.065 + 0.00355T) \sqrt{\rho_o} \\ M_w &= \rho_w C_w \end{aligned} \quad (18)$$

As the integral for f_{PD} is not easy to calculate, it is approximated by the following expression (Aziz and Gontijo, 1984),

$$\Delta f_{PD} = \frac{5.615(q_o M_o + q_w M_w)(T_{avg}^{n-1} - T_R) \Delta t}{2Q_{MAX}} \quad (19)$$

where, Δt is the time step, n is the time step number, q_o and q_w are the oil and water production rates, respectively. The average temperature of the steam zone in the ' $n-1$ 'th cycle is T_{avg}^{n-1} .

2.4 Production Period

In this period, the well is opened for oil flow from the reservoir. Presented in the following section is a model to predict oil production rates during the production period.

Butler et al. (1980) presented a series of publications related to the gravity-drainage of heavy-oil reservoirs subjected to steam injection in which the theory was directed to linear flow from horizontal wells. In this development, a similar approach is used. However, the steam zone shape has been assumed to be a prism with triangular cross section and the horizontal well lies at the bottom edge. Fig.1 shows the cross sectional shape of the zone.

It is assumed in the derivation of the flow equation that the reservoir is initially saturated with oil and water, and it is pressure-depleted. After steam flooding, the steam chamber occupies a prismatic shape. Steam distribution along the well is assumed to be uniform. Heat transfer from

the steam chamber to the neighboring oil zone is via conduction. During the injection period, the average temperature of the steam chamber is assumed to be the saturation temperature of the steam. The oil is heavy enough to permit the injected steam to go to the top of the reservoir as suggested by Van Lookeren (1977). The reservoir is assumed to be initially pressure-depleted and is only negligibly re-pressurized when steam is injected. Finally, the potential that causes flow of oil into the same well that acts as a production well is a combination of gravity forces and pressure drive.

2.4.1 Development of Flow Equation

In this approach, we assume that gravity-drainage takes place through the entire steam chamber. Thus, integration is over the entire steam chamber and proceeds angularly with the appropriate approximation for the area. Gravity and pressure drive are taken as the forces causing production. The steam zone growth is estimated in terms of the decrease in the angle of inclination of the zone with the horizontal, θ . The thickness of the steam zone is assumed to be equal to the reservoir thickness. The heated-zone geometry used to derive the flow equation is given in Fig.1. The maximum heated dimension in the horizontal direction is R_h , while the maximum heated dimension in the vertical direction is the reservoir thickness h .

According to Darcy's law, the flow rate through an incremental area, dA , normal to the direction of flow [Fig.1(b)], is written

$$dq_o = -\frac{k_o}{\mu_o}(\rho_o - \rho_s)\nabla\phi dA \quad (20)$$

where, $\nabla\phi$ is the gradient of potential in the direction of flow, μ_o is the dynamic viscosity of oil, k_o is the effective permeability to oil, and $\rho_o - \rho_s$ is the density difference between the oil and steam. According to Hubbert (1956), the potential is expressed as

$$\Delta\phi = \frac{P_s - P_{wf}}{\rho_o} + g \sin \theta \Delta h$$

Here, g is the acceleration due to gravity. The differential area depends upon distance, r , above the well:

$$dA = 2r \sin\left(\frac{1}{2}d\theta\right) \approx r d\theta \quad (21)$$

The oil-phase effective permeability, k_o is the product of the absolute permeability, k , and relative permeability, k_{ro} . With r equal to $h_{st}\text{cosec}(\theta)$ and applying the approximation suggested in Eq.(21), the differential area reduces to

$$dA = Lh_{st} \text{cosec} \theta d\theta \quad (22)$$

It follows upon substitution of Eq.(22) into Eq.(20) that the flow rate becomes

$$dq_o = -\frac{k_o}{\mu_o}(\rho_o - \rho_s) \left(\frac{P_s - P_{wf}}{\rho_o} + gh_{st} \sin(\theta) \right) h_{st} L \text{cosec} \theta d\theta \quad (23)$$

Integration over one-half of the heated area and multiplication by 2 yields

$$\frac{q_o}{L} = -\frac{2k_o}{\mu_o}(\rho_o - \rho_s) \left[144.0 \frac{p_s - p_{wf}}{\rho_o} \log \left\{ \frac{1}{\tan\left(\frac{\theta}{2}\right)} \right\} + gh_{st}(\pi - 2\theta) \right] \quad (24)$$

2.5 Property Correlations

The steam viscosity, μ_{st} , and density, ρ_{st} , are calculated by correlations given by Farouq Ali (1982).

$$\rho_{st}(\text{lb/ft}^3) = \frac{P_s^{0.9588}}{363.9}, \quad p_s \text{ in psia} \quad (25)$$

$$\mu_{st}(cp) = 10^{-4}(0.2T_s + 82), \quad T_s \text{ in } ^\circ F \quad (26)$$

The average specific heat of water is given by:

$$C_w = \frac{h_w(T_s) - h_w(T_R)}{T_s - T_R} \quad (27)$$

The water enthalpy correlation of Jones (1977) and the steam latent heat correlation of Farouq Ali (1982) are used:

$$h_w = 68 \left[\frac{T_s}{100} \right]^{1.24}, \quad T_s \text{ in } ^\circ F \quad (28a)$$

$$\Delta H_{vdh} = 94(705 - T_s)^{0.55}, \quad T_s \text{ in } ^\circ F \quad (28b)$$

where T_s is the steam saturation temperature at the given bottom-hole pressure. The oil and water densities are approximated by (Aziz, 1984):

$$\rho_o = \rho_{o_{STD}} - 0.0214(T - T_{STD}) \quad (29a)$$

$$\rho_w = 82.4 - 11 \ln \left[\frac{705 - T_{STD}}{705 - T} \right] \quad (29b)$$

where the subscript STD refers to standard conditions.

The oil density correlation holds for oils with gravity less than 20°API and temperature less than 500°F.

2.5.1 Oil and Water Viscosities

Fluid viscosities as a function of temperature are obtained from relations given by Jones (1977)

$$\mu_o(cp) = 0.001889047 e^{\frac{8956.257}{T+460}} \quad (30a)$$

$$\mu_w(cp) = 0.66 \left[\frac{T}{100} \right]^{-1.14} \quad (30b)$$

The correlation for oil viscosity is applicable to oils with specific gravity less than 30° API.

2.5.2 Fluid Saturations and Relative Permeabilities

After the soak period and before the production begins, it is assumed that the only mobile phase around the well is water.

$$\bar{S}_w = 1 - S_{orw} \quad (31)$$

Once the well is opened for production, it is assumed that the oil saturation around the well increases. Also, the water saturation is given by

$$S_w = \bar{S}_w - (\bar{S}_w - S_{wi}) \frac{W_p}{WIP} \quad (32)$$

where W_p is the water production during the cycle and WIP is the amount of mobile water in place at the beginning of the cycle (Aziz and Gontijo, 1984). The normalized water saturation expression is

$$S_w^o = \frac{S_w - S_{wi}}{1 - S_{wi} - S_{orw}} \quad (33)$$

The analytical expressions presented by Farouq Ali (1982) for the relative permeabilities suggested by Gomaa (1980) are as follows:

$$k_{rw} = -0.002167 S_w^o + 0.024167 (S_w^o)^2 \quad (34a)$$

$$k_{ro} = -0.9416 + \frac{1.0808}{S_w^o} - \frac{0.13856}{(S_w^o)^2} \quad (34b)$$

with,

$$k_{ro} = 1.0 \quad \text{if} \quad S_w^o \leq 0.2 \quad (34c)$$

2.6 Algorithm for Calculation Scheme

These equations and ideas are translated into a seven-step algorithm:

1. Initialize the model by inputting reservoir, fluid, and operational properties.
2. Calculate steam zone geometry (volume, height, and thickness), temperature, and saturations during injection and at the start of the production cycle.
3. Calculate oil and water production flow rates at small time steps within the production interval. Also, calculate the cumulative volume of fluids produced and check against original fluids in place.
4. Calculate the average temperature of the heated zone during production and at the end of the production cycle.
5. If additional steps are needed for cycle completion, go to stage 3, else proceed.
6. Calculate the amount of fluids and heat remaining in the reservoir at the termination of the cycle, and thereby calculate recovery and oil-steam ratio (OSR).
7. If a new cycle is required, repeat steps, else terminate calculations.

2.6.1 Computer Code

A C++ computer program was written to implement the proposed analytical model. The prime objective of this code is to validate the physical applicability of the model described in Chapter 2. The code outputs the oil production rate versus time. For calculating the oil production, there are many preceding steps that include calculation of thermal properties of the reservoir matrix and fluids, temperature calculations based on heat losses as well as change in rheological behavior of the fluids. The correlations used are simple and easy to apply. The program also calculates the temperature decline of the steam zone with time as well as the water production rate with time. The calculations are made for each cycle that is comprised of three periods: the “injection interval”, the “soak interval” and the “production interval”. Note that production does not begin until the production interval starts. However, the temperature decline of the heated zone begins from the soaking interval. At the end of each such cycle, the heat remaining in the reservoir is calculated based on the temperature. This heat becomes the initial heat due to steam injection for the next cycle.

2.6.2 Program Structure

The program is structured in the following order:

1. Input data
 2. Begin cycle calculations
 3. Calculate fluid properties
 4. Calculate steam zone geometry at the end of injection
 5. Calculate average temperature of steam zone at end of soak
 6. Calculate thermal properties of reservoir and fluids at each time step as a function of average temperature
 7. Calculate saturations and relative permeabilities of oil and water
 8. Calculate pressure gradient acting as driving force
 9. Calculate oil and water flow rate until the end of production
 10. Calculate cumulative fluids production
 11. Check to see if the end of production or beginning of a new cycle has been reached
- The input data is illustrated in Table 1. The complete code is provided in Appendix B.

Chapter 3

3. Validation of Model, Results and Discussions

The proposed model is appropriate for predicting the oil production from pressure-depleted reservoirs where oil gravity is less than 30° API and steam injection temperatures not more than 450°F. If the input data are not within these limits, the correlations for fluid properties and flow behavior are inaccurate.

The model was validated and results were compared with those presented by Elliot and Kavscek (1999) for cyclic steaming prior to single-well SAGD. The data and the oil viscosity correlation employed were also taken from Elliot and Kavscek (1999), as shown in Table 1. Fig.3 depicts the comparison made. It is a plot of recovery factor versus time. The recovery factor is the ratio of cumulative oil produced to the original oil in place. As observed, the recovery factor obtained using the analytical model proposed matches quite closely with the recovery factor data presented by Elliot et al (1999) at early times. At later times, it is slightly greater, but within reasonable limits.

Runs were made for an example base case and illustrated are the temperature, steam zone volume, angle of inclination of the steam zone with respect to the vertical, and the cumulative oil production versus time. A sensitivity analysis is performed in order to understand

better where cyclic steaming with a horizontal well works best and to verify the limitations of the model.

3.1 Sensitivity Analysis

The sensitivity of the model was tested for the following parameters:

1. down-hole steam quality
2. formation thickness
3. steam injection rate
4. down-hole pressure
5. soaking interval
6. production interval

3.1.1 Base Case

This case was run using the properties and input data shown in Table 2. Briefly, the oil gravity is 14 °API, reservoir permeability is 1.5 D, and the pay thickness is 80 ft. The curves for the oil and water production versus time, average temperature of the steam zone versus time and the steam zone volume versus time are given in Fig.4(a) – Fig.4(d).

Fig4(a) shows the cumulative oil production per foot of the well in STB/ft versus cumulative time in days. All production and injection information is per foot of the horizontal well as are steam zone volumes. The three regions correspond to three cycles. Each cycle has distinct flat zones, which are the injection and soak intervals. After the soak interval in a given cycle is the production interval. During steam injection and soak there is no production. The instantaneous oil rate increases rapidly at early times in the production interval, as witnessed by steep gradients in cumulative production because the oil is hot and its viscosity is low. At later times, oil production decreases as the temperature declines. The production at late times is primarily due to gravity-drainage of cool oil.

A similar explanation applies to Fig.4(b) for the cumulative water production versus time curve. Here, during steam injection and soak there is no production. The instantaneous water rate increases rapidly at early times in the production interval, accounting for condensed steam around the well. But at later times water production declines. The production at late times is primarily due to gravity-drainage of condensed water.

Fig.4(c) depicts the average heated zone temperature versus time. The temperature is at the steam saturation temperature during the injection interval. During the soaking period it decreases due to conduction heat losses to the adjacent reservoir and overburden. Since the rate of heat loss is proportional to the temperature gradient, the temperature decline is greatest at the beginning of the soak period. At the beginning of the production period, temperature decline increases rapidly because hot fluids are produced. At later times, the rate of decline decreases.

Fig.4(d) shows the volume of the heated zone versus time. The volume of the heated zone increases continuously during the injection interval. This is because the amount of injected heat increases. This trend is in agreement with Eq.16, which gives the volume of the heated zone. In the soak and production intervals, it remains constant at the value it reached at the end of the injection period. At the end of the production interval the volume of the heated zone collapses to a finite value, Eq.16(c), that accounts for the heat remaining in the hot zone. Therefore, the volume of the heated zone starts at this finite value at the beginning of the next cycle.

Fig.4(e) shows the cumulative oil-steam ratio (OSR) versus the cumulative time. During the first injection and soaking intervals, there no production and the ratio is zero. In the first

production interval, the OSR increases as oil is produced. At the beginning of the second injection interval, steam is injected and therefore the OSR decreases as no oil is produced. In the ensuing soaking interval, there is no steam injection nor any oil production. Hence, the OSR remains constant. In the production interval, as there is oil production, OSR increases and reaches a peak at the end of the production interval. As k_{rw} is very small, the water flow rate, which is directly proportional to the water relative permeability is also small.

Next, the various sensitivity runs are described.

3.1.2 Sensitivity to Steam Quality

This case was run using four different values, 50%, 70%, 85% and 95%, of down-hole steam quality at an identical mass injection rate. The results are depicted in Fig.5(a) – Fig.5(d). Arrows are drawn to indicate the trends with increasing steam quality.

In Fig.5(a), as the steam quality increases from 50% to 95%, the cumulative oil production increases. As steam quality increases at constant mass flow rate, the total heat carried by the vapor increases. Hence, there is a greater volume heated as the quality increases, thereby increasing the quantity of heated oil flowing towards the horizontal production well. An increase in the steam quality from 50% to 85% almost doubles the cumulative oil produced at the end of three cycles.

Cumulative water production is not affected much by the increase in steam quality as seen in Fig.5(b). This is because the total mass of water injected is the same for all four cases.

The temperature of heated zone shown in Fig.5(c) reaches a uniform steam saturation temperature in the injection intervals and decreases with similar trends in the soak and production intervals for all the steam quality values.

The volume of the heated zone shown in Fig.5(d) increases for increasing steam quality. As steam quality increases, the heat carried by the vapor increases, thereby causing an enhanced contacted heated volume.

3.1.3 Sensitivity to Formation Thickness

This case was run using three different values, 80 ft, 200 ft and 300 ft, of the formation thickness holding all other parameters and variables constant. The results are depicted in Fig.6(a) – Fig.6(d). Again, arrows indicate increasing pay thickness.

As the formation thickness increases, the cumulative oil and water production decreases as the volume of the reservoir contacted by the steam relative to the total volume decreases. Equation (24) predicts that gravity drainage from within the heated zone increases as the base of the heated zone, R_h , increases and θ , likewise, decreases. Hence, for identical steam volume, a short, squat heated zone produces more oil than a taller, narrower heated zone. This is shown in Fig6(a) and Fig6(b), respectively.

Figure 6(c) demonstrates that the temperature of the heated zone decreases more rapidly for smaller formation thickness. Essentially, the lesser the formation thicknesses, the larger are the heat losses to the overburden. One can visualize this by comparing it with a situation where we have a concentrated heat source in contact with a heat sink. The higher concentration heat source can be treated analogously to the heated zone with a smaller formation thickness, but with a higher energy (thermal) density. It is quite straightforward that the higher intensity heat source

emits more energy. Hence, the smaller the formation thickness, the greater are the losses and the steeper are the temperature gradients.

Fig.6(d) is self-explanatory. The larger the formation thickness, the larger the volume of the heated zone due to lesser heat losses.

3.1.4 Sensitivity to Steam Injection Rate

This case was run using three different values, 1 B/D-ft, 5 B/D-ft, and 10 B/D-ft, of the steam injection rate per foot of well. The results are depicted in Fig.7(a) – Fig.7(d).

Figure 7(a) and Fig.7(b) depict the cumulative oil and water production versus time, respectively. As the steam injection rate is increased, the oil and water production increases significantly, as greater amount of heat and water is injected and a greater volume of the reservoir is contacted, as shown in Fig.7(d).

Figure 7(c) shows the temperature of the heated zone. With increasing steam injection rates the net temperature decrease over a cycle is comparatively less. This behavior is attributed to the fact that higher injection rates lead to greater thermal energy in the reservoir volume, and this energy is slow to dissipate. Although the corresponding heated zone volumes for higher injection rates are larger, thermal energy dissipation rates are lower. The surface area for heat transfer does not increase as rapidly as the heated volume.

3.1.5 Sensitivity to Down-hole Steam Pressure

This case was run using three different values, 150 psia, 200 psia, and 300 psia, of the down-hole steam pressure. Hence, the injection temperature increases and the mass injection rate and quality are constant. The results are depicted in Fig.8(a) – Fig.8(d).

A slight increase is observed in the cumulative oil and water production as we increase down-hole steam pressure. The effects are not very large because of the following reasons. The steam quality was kept constant at 0.7. As the down-hole pressure is increased, the latent heat of vaporization decreases, thereby causing more heat to be injected as sensible heat. This fact causes the amount of heat injected to be low for that temperature. Also, the heat losses are more sensitive to the temperature than to the amount of heat injected since the temperature is high and the amount of heat injected is relatively low causing a large fraction of the injected heat to be lost quickly to the adjacent layers. Another important reason could be that higher temperatures tend to cause higher initial fluid production and a lot of heat is removed at the beginning of the cycle.

As shown in Fig.8(c), the temperature of the heated zone drops rapidly during soak and production for higher injection pressures. Larger initial fluid production results due to higher saturated steam temperatures, and larger heat loss at the beginning of each cycle occurs also. Heat losses are large because the temperature gradient is large.

3.1.6 Sensitivity to Steam Soak Interval

This case was run using four different values, 15 days, 25 days, 30 days and 50 days, of the soaking interval. Arrows indicate the lengthening soak interval. The results are depicted in Fig.9(a) – Fig.9(b).

As shown in Fig.9(a), as the soaking interval is increased, the cumulative oil production decreases. Increase in the soaking interval implies a greater time for energy dissipation. In our case, the cumulative heat losses increase due to larger soaking periods. Therefore, the temperature of the heated zone at the end of the soaking interval is predictably lesser. As a consequence the oil viscosity is larger and mobility of oil toward the horizontal production well

in the production period is reduced. Hence, cumulative oil production decreases with increased soaking time. A similar explanation can be put forth to explain the behavior of the cumulative water production depicted in Fig.9(b). However, as the viscosity of oil is not a strong function of temperature, the change in water production observed is small. Quite obviously, the production is delayed as we have longer soaking time.

3.2 Discussion

As observed, an increase in steam quality results in higher oil production, as there is more energy per unit mass of the reservoir matrix due to a greater fraction of energy injected as latent heat of vaporization. This means that a given mass of steam carries more heat due to heat in the latent form, and hence, a greater oil rate is observed. Therefore, a higher steam quality is recommended for greater oil production.

For thicker pay zones, the energy density in the reservoir rock matrix is lesser. This is because of greater volume of the reservoir rock to be contacted by steam. Also, the reservoir fluid acts as a heat sink. Therefore, too large a pay zone might lead to very low oil rates. On the other hand, if the pay thickness is too low, then the energy density within the zone is extremely large, resulting in large losses to the surrounding matrix. A way to get around this problem is to reduce the energy carried per unit mass of the steam i.e. either decreasing the mass flow rate of steam or reducing the steam quality.

Higher steam injection rates enhance heat delivery, thereby increasing oil production. Excessively high injection rates cause over heating of the reservoir matrix that results in larger heat losses, and thus causing a decrease in the thermal efficiency of the process. The optimum steam injection rate is the one that results in minimal heat losses and maximum heated volume.

Increasing down-hole steam pressure has a negligible effect on the oil production. Even though the injection temperature is higher, the excess heat gets dissipated in the soaking interval, thereby making the process thermally inefficient.

Chapter 4

4. Conclusions

A simple model for the cyclic steam process with a horizontal well has been developed. It incorporates a new flow equation for gravity drainage of heavy oil. Important factors such as pressure drive, gravity drive, and shape of the steam zone are incorporated into the model. Limited comparison of the model to numerical simulation results indicate that the model captures the essential physics of the recovery process. Cyclic steaming appears to be effective in heating the near-well area and the reservoir volume. A sensitivity analysis shows that the process is relatively robust over the range of expected physical parameters.

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Tables

Table 1: Data for validation of proposed analytical model with results of Elliot et al (1999).

Grid System

3D Cartesian System

Total Number of Blocks	5568
X-Dimension (ft)	4592.0
Y-Dimension (ft)	209.92
Z-Dimension (ft)	64.288
Well Length (ft)	2624.0

Reservoir Properties

Initial Pressure (psia)	385.132
Initial Temperature (°F)	60.8
Initial S_o (%)	85.0
Initial S_w (%)	15.0

Rock Properties

Porosity (%)	33.0
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Operating Conditions

Injection Rate (B/D)	1.57
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Table 2: Input Data

Variable	Value
Reservoir permeability (Darcy)	1.5
Reservoir porosity	0.2
Initial water saturation	0.25
Connate water saturation	0.1
Well radius (ft)	0.31
Residual oil saturation to steam	0.05
Residual oil saturation to water	0.25
Initial reservoir temperature (°F)	110
Saturated steam temperature (°F)	365
Reservoir thermal conductivity (Btu/ft D. °F)	24.0
Reservoir thermal diffusivity (ft ² /D)	0.48
Time step size (days)	1.0
Injected steam quality	0.67
Steam injection rate (B/D)	1.0
API gravity of oil (°API)	14.0
Injection pressure (psi)	150.0
Pay thickness (ft)	80.0

Figures

Fig.1(a) Schematic diagram of heated area geometry; (b) differential element of the heated area.

Fig.2 Schematic diagram of area of cross section of heated area

Fig.3 Recovery factor versus cumulative time (Comparison with results Elliot and Kavscek, 1999)

4 (a)

4 (b)

4 (c)

4(d)

4(e)

Fig.4 Base case: (a) cumulative oil production versus cumulative time, (b) cumulative water production versus cumulative time, (c) average steam zone temperature versus cumulative time, (d) steam zone volume versus cumulative time, (e) oil-steam ratio versus cumulative time.

5(a)

5(b)

5(c)

5(d)

Fig.5 Sensitivity to steam quality: (a) cumulative oil production versus cumulative time, (b) cumulative water production versus cumulative time, (c) average steam zone temperature versus cumulative time, (d) steam zone volume versus cumulative time.

6(a)

6(b)

6(c)

6(d)

Fig.6 Sensitivity formation thickness: (a) cumulative oil production versus cumulative time, (b) cumulative water production versus cumulative time, (c) average steam zone temperature versus cumulative time, (d) steam zone volume versus cumulative time.

7(a)

7(b)

7(c)

7(d)

Fig.7 Sensitivity to steam injection rate: (a) cumulative oil production versus cumulative time, (b) cumulative water production versus cumulative time, (c) average steam zone temperature versus cumulative time, (d) steam zone volume versus cumulative time

8(a)

8(b)

8(c)

8(d)

Fig.8 Sensitivity to bottom-hole pressure: (a) cumulative oil production versus cumulative time, (b) cumulative water production versus cumulative time, (c) average steam zone temperature versus cumulative time, (d) steam zone volume versus cumulative time

9(a)

9(b)

Fig.9 Sensitivity to soaking interval: (a) cumulative oil production versus cumulative time (b) cumulative water production versus cumulative time

Appendix A

Derivation for calculating f_{HD}

The heat transfer model for conduction cooling of the heated zone can be solved for the horizontal and vertical conduction mechanisms. We have solved the horizontal heat transfer mechanism problem. When conduction occurs as described, the temperature at any point within the heated geometry can be expressed as the product:

$$\begin{aligned} v &= V v_x v_z \\ V &= T_s - T_R \end{aligned} \quad (A-1)$$

where, v_x and v_z are unit solutions of component conduction problems in the x and z directions, respectively. Similarly, an integrated average temperature for the heated regions may be computed as:

$$\bar{v} = V \bar{v}_x \bar{v}_z \quad (A-2)$$

The average unit solution for $\bar{v}_x$ is obtained by solving the one-dimensional heat conduction problem in the horizontal direction:

$$\alpha \frac{\partial^2 v_x}{\partial x^2} = \frac{\partial v_x}{\partial t} \quad (A-3)$$

Initial and Boundary conditions:

1. $v_x = 1$; $t = t_{init}$; $0 < x < a$
2. $v_x = 0$; $t = t_{init}$; $|x| > a$
3. $v_x = 0$; $t \geq t_{init}$; $x \rightarrow \infty$
- 4.

Carlslaw and Jaeger have given the solution for the problem at hand with the boundary condition specified as:

$$v_x = \frac{1}{2} \left\{ \operatorname{erfc} \left(\frac{a-x}{2\sqrt{\alpha t}} \right) + \operatorname{erfc} \left(\frac{a+x}{2\sqrt{\alpha t}} \right) \right\} \quad (A-4)$$

This solution holds at any vertical distance ‘ y ’ upwards from the horizontal well as shown in the figure. To find the average $\bar{v}_x$ over the entire triangular heated area, we have,

$$\bar{v}_x = \frac{\int_0^A v_x dA}{\int_0^A dA} \quad (A-5)$$

If we take an elementary strip parallel to the x -axis, we will be integrating the given function with respect to x . The ends of this strip are bounded by the lines $x = (b/h_t)y$ and $x = -(b/h_t)y$; so that

these are the limits of integration with respect to x. Next, we integrate with respect to y from y = 0 to y = h_t. This, therefore, covers the whole area of the triangle. Thus, the integral reduces to:

$$\bar{V}_x = \frac{\int_{\frac{b}{h_t}y}^{\frac{b}{h_t}y} \int_0^{h_t} v_x dx dy}{\int_{\frac{b}{h_t}y}^{\frac{b}{h_t}y} \int_0^{h_t} dx dy} \quad (\text{A-6})$$

The final integration result is:

$$\bar{V}_x = \frac{1}{2} + \sqrt{\frac{t_{DH}}{\pi}} \left\{ -2.0 + \exp\left(-\frac{1}{t_{DH}}\right) \right\} + \operatorname{erf}\left(\frac{1}{\sqrt{t_{DH}}}\right) \left\{ \left(\frac{2}{\sqrt{\pi}} - 1\right) t_{DH} + \frac{1}{2} \right\} \quad (\text{A-7})$$

where,

$$t_{DH} = \frac{\alpha(t - t_{inj})}{R_h^2} \quad (\text{A-8})$$

R_h is the horizontal range of the extent of the steam zone and is given by,

$$R_h = h_t \cot(\theta) \quad (\text{A-9})$$

4.2 PSEUDOSTEADY STATE FLOW

(William E. Brigham)

ABSTRACT

Whenever a closed system is being produced at a constant rate, after a period of time the entire system starts to be depleted. This depletion rate is constant everywhere in the system. This state of dynamic equilibrium is usually called pseudosteady state flow, for the forms of the equations look a great deal like their steady state equivalents.

These equation forms follow the diffusivity equation. However, they can be derived without using that equation, but rather, by integrating Darcy's Law and by invoking the overall material balance equation. The forms of these equations can be examined for all three of the geometries commonly used by reservoir engineers; linear, radial and spherical. These solutions are found to be imbedded in the published transient equations for these geometries. However, these transient results are written in such a way that it is not obvious that these concepts have been included in them.

Thus the results seen in these notes are not new. But, in the form presented, they give great insight on the nature of these flow equations, and the concepts leading up to them. It is always useful to clarify concepts of this sort.

INTRODUCTION

I have written a series of notes for steady state flow, and for successions of steady states (Brigham, 1987). I'm going to deviate from that scenario to discuss transient flow in closed bounded systems. This is flow that follows the diffusivity equation; however, in discussing this type of flow behavior, the diffusivity equation will not need to be used.

PSEUDOSTEADY STATE CONCEPTS

If we have a well producing at a constant rate in a closed system, the pressure at the well drops with time; also the effect of flow at the well is to cause the pressure drop to move further into the reservoir with time. After a period of time the effect of the well is seen at the outer boundary. Since it is a closed system, with no flow at the outer boundary, the pressure gradient there must be zero. However, since the well is depleting the reservoir, the pressure must drop there with time just as it does in the remainder of the reservoir.

After a period of time in this closed system, a state of dynamic equilibrium is reached where the well begins to deplete the entire reservoir. From this time onward each part of the system will be depleted at an equal rate, thus the rate at which the pressure declines becomes the same everywhere. For now, I'll have to ask you to take this statement on faith. Later, after we have developed the equations, we will see if we can prove that it is valid.

This type of flow is often called pseudosteady state, for, as we will see later, it has many characteristics and equation forms similar to steady state flow. It has also occasionally been called quasisteady state; however, this phrase never seemed to catch on in the petroleum industry. Dake, in his book (1978) calls it semisteady state. This seems to be commonly used by the worldwide Shell organization. I don't like it because it is bad English; the prefix "semi" means "half," which is a meaningless term in this context. Craft and Hawkins (1959) on pp. 285-289 and others even refer to this type of flow as "steady state." This is clearly wrong. So I will use "pseudosteady state" when discussing constant rate depletion of closed bounded systems. Let us turn now to consideration of the implications of the depletion mechanism discussed earlier.

If we were to look at a given point in a reservoir, we can ask ourselves how much fluid is flowing past that point. It seems clear that the fluid flowing past that point is only that fluid which is being produced from points further out and which is flowing toward the well. At the outside edge the flow is zero, as we move closer to the well the flow increases, and finally, as we get to the well, the flow reaches the constant rate that was stipulated earlier.

These concepts can be put into equations, the forms of which will depend on the geometry of the system. There are three geometries that are convenient to consider; linear, radial and spherical. I'll discuss all three, starting with the cylindrical (radial) system, for that is the one most commonly seen in reservoir flow.

CYLINDRICAL (RADIAL) FLOW

Let us consider a closed cylindrical reservoir with a drainage radius, r_e , and a wellbore radius, r_w . If the flow, $q(r)$, at any distance, r , is proportional to the volume being depleted beyond r , and the flow at the well, q_w , is proportional to the entire drainage volume, we can write the following equation:

$$\frac{q(r)}{q_w} = \frac{r_e^2 - r^2}{r_e^2 - r_w^2} \quad (1)$$

which is merely a mathematical statement of the idea. Note that Eq. 1 tells us that $q(r)$ becomes zero at r_e , and q_w at r_w , as it should.

Let us now write Darcy's Law for flow at any radius, r , and substitute Eq. 1 into it.

$$q(r) = -\frac{2\pi k h r dp}{\mu dr} = q_w \left(\frac{r_e^2 - r^2}{r_e^2 - r_w^2} \right) \quad (2a)$$

or, upon rearranging:

$$-\frac{2\pi k h}{q_w \mu} dp = \left(\frac{r_e^2 - r^2}{r_e^2 - r_w^2} \right) \frac{dr}{r} \quad (2b)$$

Let us integrate Eq. 2b from the wellbore out to a general radius, r , and also out to r_e :

$$-\frac{2\pi kh}{q_w \mu} \int_{p_w}^{p(r)} dp = \frac{r_e^2}{r_e^2 - r_w^2} \int_{r_w}^r \frac{dr}{r} - \left(\frac{1}{r_e^2 - r_w^2} \right) \int_{r_w}^r r dr$$

$$-\frac{2\pi kh[p(r) - p_w]}{q_w \mu} = \frac{r_e^2 (\ln r - \ln r_w)}{(r_e^2 - r_w^2)} - \frac{(r^2 - r_w^2)}{2(r_e^2 - r_w^2)} \quad (3)$$

and, out to r_e :

$$-\frac{2\pi kh}{q_w \mu} (p_e - p_w) = \left(\frac{r_e^2}{r_e^2 - r_w^2} \right) \ln \left(\frac{r_e}{r_w} \right) - \frac{1}{2} \quad (4)$$

In general, in reservoirs, $r_e \gg r_w$; thus the ratio in front of the log term is nearly equal to 1.0, and Eq. 4 can be simplified to:

$$q_w = -\frac{2\pi kh(p_e - p_w)}{\mu [\ln(r_e/r_w) - 1/2]}$$

or,

$$= -\frac{2\pi kh(p_e - p_w)}{\mu \ln(0.607 r_e/r_w)} \quad (5)$$

A glance at either form of Eq. 5 will show why this type of flow is called pseudosteady state. The equation certainly looks a great deal like the equation for steady state radial flow. It differs only by the constant on the log term. Equation 5, however, is not very useful, for it contains the pressure drop from the outer boundary to the well; and we seldom have a convenient way to measure that outer boundary pressure. If we could, it would be much better to relate the equation to the average pressure, $\bar{p}$. This pressure could be measured easily in the field by merely shutting in the well and allowing it to build up to its average pressure. Of course, transient testing pressure buildup techniques could also be used to calculate this pressure.

The average pressure we want is the volumetric average. This means we need to integrate pressure over the entire volume and divide it by the total volume, as follows:

$$\bar{p} = -\frac{\int_{r_w}^{r_e} 2\pi r h \phi p(r) dr}{\int_{r_w}^{r_e} 2\pi r h \phi dr}$$

or,

$$\bar{p} = \frac{\int_{r_w}^{r_e} 2r p(r) dr}{(r_e^2 - r_w^2)} \quad (6)$$

Actually, we are interested in the pressure drop terms in the equation; so instead of the average pressure, let us calculate the pressure drop, $\bar{p} - p_w$, and include the flow terms by substituting Eq. 3 into Eq. 6, as follows,

$$\begin{aligned}
-\frac{2 \pi k h (\bar{p} - p_w)}{q_w \mu} &= -\frac{2 \pi k h}{q_w \mu (r_e^2 - r_w^2)} \int_{r_w}^{r_e} 2r [p(r) - p_w] dr \\
&= \frac{r_e^2}{(r_e^2 - r_w^2)^2} \int_{r_w}^{r_e} (2r \ln r - 2r \ln r_w) dr \\
&\quad - \frac{1}{2(r_e^2 - r_w^2)^2} \int_{r_w}^{r_e} (2r^3 - 2r_w^2 r) dr
\end{aligned} \tag{7}$$

When integrated, Eq. 7 becomes:

$$\begin{aligned}
-\frac{2 \pi k h (\bar{p} - p_w)}{q_w \mu} &= \frac{r_e^2}{(r_e^2 - r_w^2)^2} (r_e^2 \ln r_e - r_e^2/2 - r_w^2 \ln r_w + r_w^2/2 - r_e^2 \ln r_w + r_w^2 \ln r_w) \\
&\quad - \frac{1}{2(r_e^2 - r_w^2)^2} [r_e^4/2 - r_w^4/2 - r_w^2(r_e^2 - r_w^2)] \\
&= \frac{r_e^2}{(r_e^2 - r_w^2)^2} \left[r_e^2 \ln(r_e/r_w) - \frac{r_e^2 - r_w^2}{2} \right] \\
&\quad - \frac{1}{2(r_e^2 - r_w^2)^2} \left[\frac{(r_e^2 - r_w^2)(r_e^2 + r_w^2)}{2} - r_w^2(r_e^2 - r_w^2) \right] \\
&= \frac{r_e^4 \ln(r_e/r_w)}{(r_e^2 - r_w^2)^2} - \left(\frac{1}{r_e^2 - r_w^2} \right) \left[\frac{r_e^2}{2} + \frac{r_e^2 + r_w^2}{4} - \frac{r_w^2}{2} \right] \\
&= \frac{r_e^4 \ln(r_e/r_w)}{(r_e^2 - r_w^2)^2} - \left(\frac{1}{r_e^2 - r_w^2} \right) \left(\frac{3r_e^2}{4} - \frac{r_w^2}{4} \right)
\end{aligned} \tag{8}$$

Again, we can recognize that, $r_e^2 \gg r_w^2$, thus Eq. 8 can be simplified and rearranged to:

$$\begin{aligned}
q_w &= -\frac{2 \pi k h (\bar{p} - p_w)}{\mu [\ln(r_e/r_w) - 3/4]} \\
&= -\frac{2 \pi k h (\bar{p} - p_w)}{\mu [\ln(0.472 r_e/r_w)]}
\end{aligned} \tag{9}$$

This is the same as Eq. 6.27, p. 288 of Craft and Hawkins, and Eq. 6.12, p. 144 of Dake. It is an extremely useful equation for it gives us the flow behavior of a depleting radial system. As you can see, it looks remarkably like the steady state radial flow equation with merely an additional constant.

Considering the constant $(-3/4 \text{ or } 0.472)$, we might wish to look at this drainage cylinder in terms of its area rather than its radius. If we were to define the area, A , in acres, and r_e in feet, they are related as follows:

$$A = \frac{\pi r_e^2}{43,560}$$

or,

$$r_e = \sqrt{\frac{43,560 A}{\pi}} = 117.75 \sqrt{A} \quad (10)$$

Thus we could change Eq. 9 to:

$$q_w = -\frac{k h(\bar{p} - p_w)}{\mu \ln \left[117.75 (0.472) \sqrt{A} / r_w \right]}$$

$$q_w = \frac{k h(\bar{p} - p_w)}{\mu \ln (55.6 \sqrt{A} / r_w)} \quad (11)$$

Equation 11 is for pseudosteady state flow from a cylindrical reservoir. From the earlier description of the flow, however, it should be clear that flow toward a well from any bounded system will eventually reach pseudosteady state. Thus we should expect that other outer boundary shapes might result in an equation similar to Eq. 11. This supposition is correct. In fact, there is a commonly-used areal geometric term in transient flow called the Dietz shape factor, C_A (Dietz, 1965). In a cylindrical system, C_A is equal to 31.62. In general if we call the constant in Eq. 11, ψ , we can write the equation for pseudosteady flow as,

$$q_w = \frac{k h(\bar{p} - p_w)}{141.2 B \mu \ln (\psi \sqrt{A} / r_w)} \quad (12)$$

In oilfield units; and from the constants defined above, the equation which relates ψ to the shape factor, C_A , is:

$$\psi = 313 / \sqrt{C_A} \quad (13)$$

or,

$$q_w = -\frac{k h(\bar{p} - p_w)}{141.2 B \mu \ln \left(\frac{313}{r_w} \sqrt{\frac{A}{C_A}} \right)} \quad (14)$$

Equation 14 is extremely useful in many reservoir engineering calculations.

PROOF THAT THE RADIAL PSEUDOSTEADY STATE EQUATION FITS THE DIFFUSIVITY EQUATION

Earlier I stated that the equations developed here can be proved to fit the diffusivity equation. In radial coordinates this equation is:

$$\frac{\partial^2 p}{\partial r^2} + \frac{1}{r} \frac{\partial p}{\partial r} = \frac{\phi \mu c_t}{k} \frac{\partial p}{\partial t} \quad (15)$$

If the equations we have developed fit Eq. 15 and also fit the boundary conditions, they must be solutions to Eq. 15. The boundary conditions are:

$$-\frac{2\pi k h}{\mu} \left(r \frac{\partial p}{\partial r} \right)_{r_w} = q_w \quad (16)$$

and

$$-\frac{2\pi k h}{\mu} \left(r \frac{\partial p}{\partial r} \right)_{r_e} = 0 \quad (17)$$

The partial derivatives on the left-hand side of Eq. 15 can be evaluated by differentiating Eq. 3, as can the boundary conditions. It only remains to evaluate the time derivative on the right side of Eq. 15, to see if Eq. 3 fits Eq. 15; and also to show that Eq. 3 fits Eqs. 16 and 17.

The time derivative requires a little thought. Earlier I stated that the pressure will drop at a constant rate in all parts of the reservoir during pseudosteady flow. Looking at the equation for average pressure, we see that if the pressures at all points drop at a given rate, so does the average pressure. Thus we can write,

$$\frac{\partial p(r)}{\partial t} = \frac{d\bar{p}}{dt} = \text{Constant} \quad (18)$$

and the constant can be evaluated by performing an overall material balance, through the definition of the compressibility, as follows:

$$c_t = -\frac{1}{V} \left(\frac{\partial V}{\partial \bar{p}} \right)_T$$

or,

$$(dV)_T = -c_t V (d\bar{p})_T \quad (19)$$

Recognizing that the temperature, T , is constant, and by defining the flow rate, q_w , as the time derivative of V , we get:

$$\begin{aligned} -q_w &= \left(\frac{dV}{dt} \right) = -c_t V \left(\frac{d\bar{p}}{dt} \right) = -\pi (r_e^2 - r_w^2) \phi c_t h \left(\frac{d\bar{p}}{dt} \right) \\ &= -\pi (r_e^2 - r_w^2) \phi c_t h \frac{\partial p(r)}{\partial t} \end{aligned} \quad (20)$$

Equation 20 allows us to check whether our assumptions and equations are, in fact, a proper solution to the diffusivity equation.

To develop the terms of the diffusivity equation, we begin by rearranging Eq. 3, as follows,

$$p(r) - p_w = -\frac{q_w \mu}{2 \pi k h} \left[\frac{r_e^2 (\ln r - \ln r_w)}{(r_e^2 - r_w^2)} - \frac{(r^2 - r_w^2)}{2(r_e^2 - r_w^2)} \right] \quad (21)$$

Differentiating Eq. 21 with respect to r , as indicated in Eq. 15 we get,

$$\frac{\partial p(r)}{\partial r} = -\frac{q_w \mu}{2 \pi k h} \left[\frac{r_e^2}{(r_e^2 - r_w^2)} - \frac{r}{(r_e^2 - r_w^2)} \right] \quad (22)$$

and dividing by r , as in Eq. 15,

$$\frac{1}{r} \frac{\partial p(r)}{\partial r} = -\frac{q_w \mu}{2 \pi k h} \left[\frac{r_e^2}{(r_e^2 - r_w^2)^2} - \frac{1}{(r_e^2 - r_w^2)} \right] \quad (23)$$

We can also differentiate Eq. 22 with respect to r , to get the first term in Eq. 15,

$$\frac{\partial}{\partial r} \left[\frac{\partial p(r)}{\partial r} \right] = \frac{\partial^2 p(r)}{\partial r^2} = -\frac{q_w \mu}{2 \pi k h} \left[-\frac{r_e^2}{(r_e^2 - r_w^2)^2} - \frac{1}{(r_e^2 - r_w^2)} \right] \quad (24)$$

Rearranging Eq. 20 to evaluate the diffusivity term on the right-hand side of Eq. 15, we get,

$$\begin{aligned} \frac{\phi \mu c_t}{k} \frac{\partial p(r)}{\partial t} &= \frac{\phi \mu c_t}{k} \left[\frac{q_w}{\pi (r_e^2 - r_w^2) \phi c_t h} \right] \\ &= \frac{q_w \mu}{\pi (r_e^2 - r_w^2) k h} \end{aligned} \quad (25)$$

Now we can substitute Eqs. 23, 24, and 25 into Eq. 15, the radial diffusivity equation. The result is

$$\frac{q_w \mu}{2 \pi k h} \left[\frac{r_e^2}{(r_e^2 - r_w^2)^2} + \frac{1}{(r_e^2 - r_w^2)} \right] - \frac{q_w \mu}{2 \pi k h} \left[\frac{r_e^2}{(r_e^2 - r_w^2)^2} - \frac{1}{(r_e^2 - r_w^2)} \right] = \frac{q_w \mu}{\pi (r_e^2 - r_w^2) k h} \quad (26a)$$

Note in Eq. 26a, the terms $r_e^2 / (r_e^2 - r_w^2)^2$ cancel; while the terms $1 / (r_e^2 - r_w^2)$ are additive, thus Eq. 26a simplifies to

$$\frac{q_w \mu}{2 \pi k h} \left[\frac{2}{(r_e^2 - r_w^2)} \right] = \frac{q_w \mu}{\pi (r_e^2 - r_w^2) k h} \quad (26b)$$

The terms on both sides of Eq. 26b are identical, thus proving that the diffusivity equation is honored by Eqs. 3 and 20.

It only remains to show that the boundary conditions (Eqs. 16 and 17) are also correctly stipulated by Eq. 3. In Eq. 22, we differentiated Eq. 3 to get,

$$-\frac{2 \pi k h}{q_w \mu} \left[r \frac{\partial p(r)}{\partial r} \right] = \left[\frac{r_e^2}{(r_e^2 - r_w^2)} - \frac{r}{(r_e^2 - r_w^2)} \right] \quad (22)$$

We multiply this by r , as Eqs. 16 and 17 require, to get,

$$-\frac{2 \pi k h}{q_w \mu} \left[r^2 \frac{\partial p(r)}{\partial r} \right] = \frac{r_e^2}{(r_e^2 - r_w^2)} - \frac{r^2}{(r_e^2 - r_w^2)} \quad (27)$$

Setting $r = r_w$, as in Eq. 16, we get,

$$-\frac{2 \pi k h}{q_w \mu} \left[r^2 \frac{\partial p(r)}{\partial r} \right]_{r=r_w} = \frac{r_e^2}{(r_e^2 - r_w^2)} - \frac{r_w^2}{(r_e^2 - r_w^2)} = 1 \quad (28)$$

which is Eq. 16, as expected. Setting $r = r_e$ as in Eq. 17, we get,

$$-\frac{2 \pi k h}{q_w \mu} \left[r^2 \frac{\partial p(r)}{\partial r} \right]_{r=r_e} = \frac{r_e^2}{(r_e^2 - r_w^2)} - \frac{r_e^2}{(r_e^2 - r_w^2)} = 0 \quad (29)$$

which is Eq. 17, which again as what we expected. Thus it is clear that Eq. 3 combined with Eq. 20 is a solution to the diffusivity equation during radial pseudosteady state flow.

A thoughtful reader may wonder whether this result can be seen in Van Everdingen and Hurst's (1949) or Chatas' (1953) tables and equations for the solutions to closed systems at constant rate. It turns out that it is in their solutions, but the equations are written in such a way that this fact is hard to see. Chatas shows the long time result for the constant rate case with a closed outer boundary. This result in present day nomenclature is,

$$p_D(t_D) = \frac{1/2 + 2t_D}{r_D^2 - 1} - \frac{3r_D^4 - 4r_D^4 \ln r_D - 2r_D^2 - 1}{4(r_D^2 - 1)^2} \quad (30a)$$

where $r_D = r_e / r_w$

which, after considerable algebraic manipulation, simplifies to,

$$p_D(t_D) = \frac{2t_D}{r_D^2 - 1} + \frac{r_D^4 \ln r_D}{(r_D^2 - 1)^2} - \frac{3r_D^2 - 1}{4(r_D^2 - 1)} \quad (30b)$$

Note that the last two terms on the right side of Eq. 30b are exactly the same as Eq. 8 of these notes. But remember that Eq. 8 defines $\bar{p}_D$ which is based on $\bar{p} - p_w$, while $p_D(t_D)$ is based on $p_i - p_w$. The first term on the right of Eq. 30b, $2t_D / (r_D^2 - 1)$, defines $(p_i - \bar{p})_D$; it is the material balance term from Eq. 20.

Chatas' way of writing the equation, although correct, really didn't emphasize the physical concepts involved, while Eq. 30b shows them clearly. These same ideas will also hold true for the linear and spherical systems. I'll derive the depletion equations for these systems, but will not bother to prove they fit the diffusivity equations and the boundary conditions, as I have done here for the radial system. It seems clear that these proofs will act similarly to these radial flow results.

LINEAR FLOW

Using concepts similar to those used to derive the radial flow equation, for linear depletion at constant rate, we stipulate that the flow rate at any distance, x , is proportional to the volume beyond that point out to the total distance, L ; and the flow rate at the inner boundary is equal to the total rate, q_w . Thus we can write,

$$\frac{q(x)}{q_w} = \frac{L - x}{L} \quad (31)$$

We can now substitute this concept into Darcy's Law for linear flow, as follows,

$$q(x) = \frac{-kA}{\mu} \frac{\partial p}{\partial x} = q_w \left(\frac{L - x}{L} \right) \quad (32a)$$

We can rearrange and integrate out to x ,

$$\begin{aligned} -\frac{kA}{q_w \mu} \int_{p_w}^{p(x)} dp &= \int_0^x \frac{(L - x) dx}{L} \\ -\frac{kA}{q_w \mu} [p(x) - p_w] &= x - \frac{x^2}{2L} \end{aligned} \quad (32b)$$

and also integrate to the outer boundary at L ,

$$-\frac{kA}{q_w \mu} [p(L) - p_w] = \frac{L}{2} \quad (32c)$$

This result looks much like Darcy's Law for linear flow, as we might have expected. The only difference is the constant, $1/2$. However, it is not very useful, for it is hard to determine $p(L)$. As we did for the radial system, we will look at the average pressure instead. This is accomplished easily as follows.

$$\bar{p} - p_w = \frac{\int_0^L [p(x) - p_w] dx}{\int_0^L dx} = \frac{1}{L} \int_0^L [p(x) - p_w] dx \quad (33)$$

Substituting Eq. 32b into Eq. 33, and rearranging, we get,

$$-\frac{kA(\bar{p} - p_w)}{q_w \mu} = \frac{1}{L} \int_0^L \left(x - \frac{x^2}{2L} \right) dx \quad (34a)$$

which, when integrated, becomes

$$\begin{aligned} -\frac{kA(\bar{p} - p_w)}{q_w \mu} &= \frac{1}{L} \left[\frac{x^2}{2} - \frac{x^3}{6L} \right]_0^L \\ &= \frac{L}{3} \end{aligned} \quad (34b)$$

Note that this, too, looks a great deal like the steady state equation. The only difference is the constant, $1/3$.

We should also be interested in determining $p_i - p_w$, for this term is the one usually listed in transient flow solutions. To do this, we need to define $p_i - \bar{p}$. But this is accomplished easily by a simple material balance calculation, as we did for the radial system in Eq. 20. Using linear coordinates, the material balance results in,

$$p_i - \bar{p} = -\frac{q_w t}{\phi c_t AL} \quad (35)$$

We can define t_D as follows, for a linear system,

$$t_D = \frac{k t}{\phi \mu c_t L^2} \quad (36)$$

Substituting Eq. 36 into Eq. 35, we get

$$-\frac{kA(p_i - \bar{p})}{q_w \mu L} = t_D \quad (37)$$

We wish to define the term, $p_i - p_w$, rather than $p_i - \bar{p}$. To do this we can write,

$$-\frac{kA(p_i - p_w)}{q_w \mu L} = -\frac{kA(p_i - \bar{p})}{q_w \mu L} - \frac{kA(\bar{p} - p_w)}{q_w \mu L} \quad (38a)$$

But the two terms on the right side of Eq. 38a are defined in Eqs. 34b and 37, so the result is very simple,

$$-\frac{kA(p_i - p_w)}{q_w \mu L} = p_D = t_D + \frac{1}{3} \quad (38b)$$

The interested reader will note that Eq. 38b is exactly the same as seen in Nabor and Barham's (1964) Eq. 17 on linear flow solutions, just as we should have expected.

SPHERICAL FLOW

By now the reader should be able to predict the methods used to derive the equations for spherical pseudosteady state flow. The flow rate at any radius, r , is proportional to the volume beyond r , and the total flow rate is q_w . Thus the resulting flow rate equation is,

$$\frac{q(r)}{q_w} = \frac{r_e^3 - r^3}{r_e^3 - r_w^3} \quad (39)$$

We can now write Darcy's Law for spherical flow at a general radius,

$$q(r) = -\frac{4\pi k r^2 dp}{\mu dr} = \frac{q_w (r_e^3 - r^3)}{(r_e^3 - r_w^3)} \quad (40a)$$

and we can rearrange Eq. 40a and integrate out to a general radius, r , as follows,

$$\begin{aligned} -\frac{4\pi k [p(r) - p_w]}{q_w \mu} &= \frac{r_e^3}{(r_e^3 - r_w^3)} \int_{r_w}^r \frac{dr}{r^2} - \frac{1}{(r_e^3 - r_w^3)} \int_{r_w}^r r dr \\ &= \frac{r_e^3}{(r_e^3 - r_w^3)} \left(\frac{1}{r_w} - \frac{1}{r} \right) - \frac{r_e^2 - r_w^2}{2(r_e^3 - r_w^3)} \end{aligned} \quad (40b)$$

and also integrate to the outer boundary, r_e ,

$$-\frac{4\pi k (p_e - p_w)}{q_w \mu} = \frac{r_e^3}{(r_e^3 - r_w^3)} \left(\frac{1}{r_w} - \frac{1}{r_e} \right) - \frac{r_e^2 - r_w^2}{2(r_e^3 - r_w^3)} \quad (40c)$$

We can compare Eq. 40c with the steady state spherical flow equation from r_w to r_e . From my notes (Brigham, 1987), it is,

$$-\frac{4\pi k (p_e - p_w)}{q_w \mu} = \left(\frac{1}{r_w} - \frac{1}{r_e} \right) \quad (41)$$

Notice the similarity in the two equations. The multiplier in front of the $1/r$ terms in Eq. 40c approaches unity when r_e is quite a bit greater than r_w . And, similar to the radial and linear equations, there is an additional term, that is a function of the size of the system, which is subtracted from the steady state term. So the pressure drop for pseudosteady state is less than for steady state flow.

But, as before, we realize that $\bar{p} - p_w$ is a more useful way to look at the pressure drop terms. To evaluate $\bar{p} - p_w$, we integrate the pressure drop from r_w to r_e and divide that by the volume, as follows,

$$\begin{aligned}\bar{p} - p_w &= \frac{\int_{r_w}^{r_e} [p(r) - p_w] r^2 dr}{\int_{r_w}^{r_e} r^2 dr} \\ &= \frac{3}{(r_e^3 - r_w^3)} \int_{r_w}^{r_e} [p(r) - p_w] r^2 dr\end{aligned}\quad (42)$$

We can now substitute Eq. 40b into Eq. 42, to get,

$$\begin{aligned}-\frac{4\pi k(\bar{p} - p_w)}{q_w \mu} &= \frac{3r_e^3}{(r_e^3 - r_w^3)^2} \int_{r_w}^{r_e} \left(\frac{1}{r_w} - \frac{1}{r} \right) r^2 dr \\ &\quad - \frac{3}{2(r_e^3 - r_w^3)^2} \int_{r_w}^{r_e} (r^2 - r_w^2) r^2 dr\end{aligned}\quad (43a)$$

$$\begin{aligned}&= \frac{3r_e^3}{(r_e^3 - r_w^3)^2} \left[\frac{r^3}{3r_w} - \frac{r^2}{2} \right]_{r_w}^{r_e} - \frac{3}{2(r_e^3 - r_w^3)^2} \left[\frac{r^5}{5} - \frac{r_w^2 r^3}{3} \right]_{r_w}^{r_e} \\ &= \frac{r_e^3}{r_w(r_e^3 - r_w^3)} - \frac{3r_e^3(r_e^2 - r_w^2)}{2(r_e^3 - r_w^3)^2} \\ &\quad - \frac{3(r_e^5 - r_w^5)}{10(r_e^3 - r_w^3)^2} + \frac{r_w^2}{2(r_e^3 - r_w^3)}\end{aligned}\quad (43b)$$

We need to recognize that the dimensionless pressure term in spherical flow contains an r_w multiplier in the numerator on the left hand side of Eq. 43. Also it would be more convenient if we were to express Eq. 43b in terms of the r_e / r_w ratio, which I will call r_D . The first and fourth

terms in Eq. 43b can then be combined into a common denominator; and the second and third can also be combined to get,

$$-\frac{4\pi k r_w (\bar{p} - p_w)}{q_w \mu} = \frac{2r_D^3 + 1}{2(r_D^3 - 1)} - \frac{15 r_D^3 (r_D^2 - 1) + 3(r_D^5 - 1)}{10(r_D^3 - 1)^2} \quad (44a)$$

and after come further algebraic manipulation,

$$\begin{aligned} -\frac{4\pi k r_w (\bar{p} - p_w)}{q_w \mu} &= 1 + \frac{3}{2(r_D^3 - 1)} - \frac{18r_D^5 - 15r_D^3 - 3}{10(r_D^3 - 1)^2} \\ &= 1 + \frac{3}{2(r_D^3 - 1)} - \frac{18(r_D^5 - 1)}{10(r_D^3 - 1)^2} + \frac{15(r_D^3 - 1)}{10(r_D^3 - 1)^2} \\ &= 1 + \frac{3}{(r_D^3 - 1)} - \frac{9(r_D^5 - 1)}{5(r_D^3 - 1)^2} \end{aligned} \quad (44b)$$

As we have done before, we can define $p_i - \bar{p}$ using the depletion material balance based on the total system volume,

$$p_i - \bar{p} = \frac{-3q_w t}{4\pi \phi c_t (r_e^3 - r_w^3)} \quad (45)$$

For spherical flow, t_D is defined the same way it is for radial flow.

$$t_D = \frac{kt}{\phi \mu c_t r_w^2} \quad (46)$$

Substituting Eq. 46 into Eq. 45, and again calling the radius ratio r_D , we get,

$$-\frac{4\pi k r_w (p_i - \bar{p})}{q_w \mu} = \frac{3t_D}{(r_D^3 - 1)} \quad (47)$$

Now we can combine Eqs. 44b and 47 to define the total pressure drop for pseudosteady state spherical flow.

$$-\frac{4\pi k r_w (p_i - p_w)}{q_w \mu} = p_D = \frac{3t_D}{(r_D^3 - 1)} + 1 + \frac{3}{(r_D^3 - 1)} - \frac{9(r_D^5 - 1)}{5(r_D^3 - 1)^2} \quad (48)$$

Equation 48 has an interesting property as r_D approaches infinity. Note that it degenerates to 1.00 under this condition. That is, the dimensionless pressure drop is limited to this value. This is correct. The spherical geometry is the only case which exhibits this strange behavior.

As we have done before, it would be well to check whether this result agrees with the published equations. Chatas wrote a paper on transient behavior for spherical systems (Chatas, 1966). His equation which results in pseudosteady state flow is his Eq. 53. His long time result, in my nomenclature, is,

$$p_D = \frac{\left[(r_D - 1)^2 + 3r_D \right] \left[\frac{1}{6} (r_D - 1)^2 (2r_D + 1) + t_D \right] - \frac{1}{2} (r_D - 1)^2 \left[\frac{1}{5} (r_D - 1)^2 + r_D \right]}{(r_D - 1) \left[\frac{1}{3} (r_D - 1)^4 + 2r_D (r_D - 1)^2 + 3r_D^2 \right]} \quad (49)$$

Obviously, to compare Eqs. 48 and 49, I will need to do considerable algebraic manipulation.

First we will look at the denominator of Eq. 49.

$$\begin{aligned} (r_D - 1) \left[\frac{1}{3} (r_D - 1)^4 + 2r_D (r_D - 1)^2 + 3r_D^2 \right] \\ &= \frac{(r_D - 1)}{3} \left[(r_D - 1)^4 + 6r_D (r_D - 1)^2 + 9r_D^2 \right] \\ &= \frac{(r_D - 1)}{3} \left[(r_D - 1)^2 + 3r_D \right]^2 = \frac{(r_D - 1)}{3} (r_D^2 + r_D + 1)^2 \\ &= \frac{(r_D - 1)}{3} \frac{(r_D^3 - 1)^2}{(r_D - 1)^2} = \frac{(r_D^3 - 1)^2}{3(r_D - 1)} \end{aligned} \quad (50)$$

Note how the denominator simplifies to a great extent.

These same ideas can be used on the terms in the numerator. Since it is so long, I'll divide it up. First let's look at the multiplier on t_D . It can be simplified in a manner similar to that used above, as follows,

$$\left[(r_D - 1)^2 + 3r_D \right] t_D = \frac{(r_D^3 - 1)}{(r_D - 1)} t_D \quad (51)$$

Dividing this numerator term by the denominator we get,

$$\frac{(r_D^3 - 1)}{(r_D - 1)} \left[\frac{3(r_D - 1)}{(r_D^3 - 1)^2} \right] = \frac{3t_D}{(r_D^3 - 1)} \quad (52)$$

Note that this result is exactly the same as the first term on the right side of Eq. 48. The remainder of the terms are only functions of r_D . The first r_D term in the numerator can be divided by the

denominator to get,

$$\begin{aligned}
& \left[\frac{3(r_D-1)}{(r_D^3-1)^2} \right] \left[(r_D-1)^2 + 3r_D \right] \left[\frac{1}{6} (r_D-1)^2 (2r_D+1) \right] \\
&= \left[\frac{(r_D-1)}{2(r_D^3-1)^2} \right] \left[\frac{(r_D^3-1)}{(r_D-1)} \right] \left[(r_D-1)^2 (2r_D+1) \right] \\
&= \frac{(r_D^2-2r_D+1)(2r_D+1)}{2(r_D^3-1)} = \frac{2r_D^3-3r_D^2+1}{2(r_D^3-1)} \\
&= \frac{2(r_D^3-1)}{2(r_D^3-1)} - \frac{3(r_D^2-1)}{2(r_D^3-1)} = 1 - \frac{3(r_D^2-1)}{2(r_D^3-1)} \tag{53}
\end{aligned}$$

Note the constant, 1, in Eq. 53 is seen in Eq. 48. The ratio, $3(r_D^2-1)/2(r_D^3-1)$ is not seen.

We'll look at the remainder of the numerator in Eq. 49 to find out if everything works out to result in Eq. 48.

To evaluate the remaining numerator term in Eq. 49, we again divide by the denominator (Eq. 50) to get,

$$\begin{aligned}
& -\frac{(r_D-1)^2}{2} \left[\frac{3(r_D-1)}{(r_D^3-1)^2} \right] \left[\frac{(r_D-1)^2}{5} + r_D \right] \\
&= -\frac{3(r_D-1)^3}{10(r_D^3-1)^2} \left[(r_D-1)^2 + 5r_D \right] = -\frac{3(r_D-1)^3}{10(r_D^3-1)^2} (r_D^2 + 3r_D + 1) \\
&= -\frac{3(r_D^5 - 5r_D^3 + 5r_D^2 - 1)}{10(r_D^3-1)^2} = -\frac{3(r_D^5-1)}{10(r_D^3-1)^2} + \frac{15(r_D^3-1)}{10(r_D^3-1)^2} - \frac{15(r_D^2-1)}{10(r_D^3-1)^2} \\
&= \frac{3}{2(r_D^3-1)} - \frac{3(r_D^5-1)}{10(r_D^3-1)^2} - \frac{3(r_D^2-1)}{2(r_D^3-1)^2} \tag{54}
\end{aligned}$$

We can now combine Eqs. 53 and 54 to get the total of the r_D type terms.

$$\begin{aligned}
r_D \text{ Function} &= 1 - \frac{3(r_D^2 - 1)}{2(r_D^3 - 1)} + \frac{3}{2(r_D^3 - 1)} - \frac{3(r_D^2 - 1)}{2(r_D^3 - 1)^2} - \frac{3(r_D^5 - 1)}{10(r_D^3 - 1)^2} \\
&= 1 - \frac{3(r_D^2 - 2)}{2(r_D^3 - 1)} - \frac{3(r_D^2 - 1)}{2(r_D^3 - 1)^2} - \frac{3(r_D^5 - 1)}{10(r_D^3 - 1)^2} \\
&= 1 + \frac{3}{(r_D^3 - 1)} - \frac{3r_D^2}{2(r_D^3 - 1)} - \frac{3(r_D^2 - 1)}{2(r_D^3 - 1)^2} - \frac{3(r_D^5 - 1)}{10(r_D^3 - 1)^2} \tag{55}
\end{aligned}$$

Clearly the first two terms in Eq. 55 are the same as the first two r_D type terms in Eq. 48. It only remains to see if the remaining terms combine to become the same as the last r_D term in Eq. 48.

We can combine the third and fourth terms to get the following result for Eq. 55.

$$\begin{aligned}
r_D \text{ Function} &= 1 + \frac{3}{(r_D^3 - 1)} - \frac{3[r_D^2(r_D^3 - 1) + (r_D^2 - 1)]}{2(r_D^3 - 1)^2} - \frac{3(r_D^5 - 1)}{10(r_D^3 - 1)^2} \\
&= 1 + \frac{3}{(r_D^3 - 1)} - \frac{3(r_D^5 - r_D^2 + r_D^2 - 1)}{2(r_D^3 - 1)^2} - \frac{3(r_D^5 - 1)}{10(r_D^3 - 1)^2} \\
&= 1 + \frac{3}{(r_D^3 - 1)} - \frac{15(r_D^5 - 1)}{10(r_D^3 - 1)^2} - \frac{3(r_D^5 - 1)}{10(r_D^3 - 1)^2} \\
&= 1 + \frac{3}{(r_D^3 - 1)} - \frac{9(r_D^5 - 1)}{5(r_D^3 - 1)^2} \tag{56}
\end{aligned}$$

Note, Eq. 56 is identical to the r_D terms in Eq. 48. So Chatas' equation is the same as the one we developed here, just as we had hoped. Further it is clear that my way of expressing this equation (Eq. 48) is far simpler than Chatas' equation; Eq. 49 of these notes. Thus it gives considerable insight into the nature of the spherical depletion equation behavior.

CONCLUSIONS

It is clear from these equation developments, that the pseudosteady state conditions result in fairly simple equation forms for pressure drop with time. This statement is true for all outer boundary shapes, and for all types of inner boundary flow (linear, radial, or spherical). Further,

these equations all follow the diffusivity equation, even though this equation was not used to derive their behavior.

These equations, at first glance do not look the same as those seen in the published literature. However, after considerable algebraic manipulation, we find that they are the same. These more simplified equation forms give considerable insight on the nature to the long time behavior for all three geometries.

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4.3 WATER INFLUX, AND ITS EFFECT ON OIL RECOVERY

PART 1. AQUIFER FLOW

(William E. Brigham)

Abstract

Natural water encroachment is commonly seen in many oil and gas reservoirs. In fact, there is more water than oil produced from oil reservoirs worldwide. Thus it is clear that an understanding of reservoir/aquifer interaction can be an important aspect of reservoir management to optimize recovery of hydrocarbons. Although the mathematics of these processes are difficult, they are often amenable to analytical solution and diagnosis. Thus this will be the ultimate goal of a series of reports on this subject.

This first report deals only with aquifer behavior, so it does not address these important reservoir/aquifer issues. However, it is an important prelude to them, for the insight gained gives important clues on how to address reservoir/aquifer problems.

When looking at aquifer flow, there are two convenient inner boundary conditions to consider; constant pressure or constant flow rate. There are three outer boundary conditions; infinite, closed and constant pressure. And there are three geometries; linear, radial and spherical. Thus there are eighteen different solutions that can be analyzed.

The information in this report shows that all of these cases have certain similarities that allow them to be handled fairly easily; and, though the solutions are in the form of infinite series or integrals, the effective results can be put into very simple closed form equations. Some equation forms are for shorter times, and others are for longer times; but, remarkably, for all practical purposes, the solutions switch immediately from one to the other. The times at which they switch depend on the sizes of the systems being considered; and these, too, can be defined by simple equations. These simple equation forms provide great insight on the nature of the behavior of these systems.

Introduction

The recovery from many oil reservoirs is affected by water influx. In fact, worldwide, there is far more water produced from oil reservoirs than oil. Much of this is natural water influx. Thus an understanding of the interplay between aquifers and the oil reservoirs needs to be understood to make recovery calculations. Often this subject is treated as though only the aquifer needs to be looked at. With this view, the various inner and outer boundary conditions and geometries are addressed. This approach is useful for it gives insight into the nature of aquifer flow. For these reasons it will be discussed here in some detail. Unfortunately, it is not very useful for real reservoir problems, for typically we must define the inner boundary conditions for the oil reservoir/aquifer system.

These boundary condition dilemmas arise in two different ways. One is when trying to history match past performance of an oil reservoir/aquifer system. The other is to predict the future behavior under various operating scenarios. These problems, though difficult, are amenable to solution as will be shown in notes to be written later. In this set of notes, I'll address aquifer flow solutions.

Aquifer Flow

The equation we use for aquifer flow is the diffusivity equation, the same one we use in well testing. Also the geometries used are the same; linear, radial and spherical flow. Although these equations are well known, I'll repeat them here for later reference.

The equations are usually written in dimensionless parameters. For linear flow, we get,

$$\frac{\partial^2 p_D}{\partial x_D^2} = \frac{\partial p_D}{\partial t_D} \quad (1)$$

where the dimensionless terms used are as follows:

$$x_D = x / L \quad (2a)$$

$$\text{and } t_D = \frac{kt}{\phi \mu c_t L^2} \quad (2b)$$

where L = The length of the linear aquifer

And, as in well testing, the definition of p_D depends on the inner boundary conditions chosen. If a constant rate inner boundary is used, p_D is defined as,

$$p_D = \frac{kA(p - p_i)}{q\mu L} \quad (2c)$$

where p_i = initial aquifer pressure

A = cross sectional area of the aquifer

If a constant pressure inner boundary is used, then the definition for p_D is,

$$p_D = \frac{p - p_i}{p_w - p_i} \quad (2d)$$

where p_w = inner boundary constant pressure

Note that the subscript, w , is used at the inner boundary just as it is in well testing, even though the inner boundary is at the original boundary of the oil reservoir/aquifer system.

The dimensionless equation for radial flow is,

$$\frac{\partial^2 p_D}{\partial r_D^2} + \frac{1}{r_D} \frac{\partial p_D}{\partial r_D} = \frac{\partial p_D}{\partial t_D} \quad (3)$$

where some of the dimensionless terms are,

$$r_D = r / r_w \quad (4a)$$

$$t_D = \frac{kt}{\phi \mu c_t r_w^2} \quad (4b)$$

These equations should look familiar to well testing engineers. Note that the term, r_w , is commonly used to define the original oil reservoir/aquifer radius. The dimensionless pressure for the constant rate inner boundary of a radial system is,

$$p_D = \frac{2\pi kh(p - p_i)}{q\mu} \quad (4c)$$

and for the constant pressure inner boundary, it is Eq. 2d again.

Spherical flow is not very common in aquifers; but it can occur whenever there is an oil reservoir "bubble" surrounded on all sides and at the bottom by a very large aquifer. So this equation will also be addressed briefly, as follows,

$$\frac{\partial^2 p_D}{\partial r_D^2} + \frac{2}{r_D} \frac{\partial p_D}{\partial r_D} = \frac{\partial p_D}{\partial t_D} \quad (5)$$

In this case, r_D and t_D are defined the same as in radial flow, Eqs. 4a and 4b. Dimensionless pressure at constant rate is defined as follows,

$$p_D = \frac{4\pi kr_w(p - p_i)}{q\mu} \quad (6)$$

Note the similarity in form to Eq. 4c. For the constant pressure inner boundary, Eq. 2d is again used.

The spherical flow equation can be simplified in an interesting way. Suppose that we define a new dimensionless variable, b_D , as follows,

$$b_D = r_D p_D \quad (7)$$

When we do this, Eq. 8 simplifies to,

$$\frac{\partial^2 b_D}{\partial r_D^2} = \frac{\partial b_D}{\partial t_D} \quad (8)$$

Thus the spherical flow equation becomes identical in form to the linear equation. The boundary conditions, however, are expressed somewhat differently. We turn now to solutions of these equations, starting with the radial systems.

Radial Geometry

In general, for all aquifer geometries there are two convenient inner boundary conditions that can be used; constant pressure or constant flow rate. There are also reasonable assumptions that can be made for the outer boundary; constant pressure, closed or infinite. Thus there are six solutions which should be considered. These will be grouped together to show their differences in behavior, and the reasons for these differences.

Constant Rate Inner Boundary

We will look at the results obtained for all three outer boundary conditions for the constant rate case, compare them at short and long time, and write simplified equations for their behavior. To do all this, we will rely heavily on Chatas' tables from the Petroleum Engineer, paper in August 1953, of which the important part is duplicated and attached in Appendix A. Chatas' tables borrowed heavily from work originally done by Van Everdingen and Hurst (1949). Chatas' nomenclature is different from the nomenclature we commonly use today, so I will clarify these differences as they arise.

Infinite Aquifer

The first constant rate solution we will look at is for an infinite aquifer. The solutions are shown in Table 1 by Chatas. His nomenclature in the table refers to dimensionless time, and labels it, t . We now use t_D . The heading labeled pressure change with the symbol, $p(t)$ is $p_D(t_D)$ in present nomenclature.

A careful look at Chatas' Table 1 shows that p_D is approximately proportioned to $t_D^{1/2}$ at very small times. The reason is that the pressure has only changed significantly at points very close to the inner boundary. Thus we can treat this early time data as though flow was linear, and the equation for early time for all linear problems is,

$$p_D = \frac{2}{\sqrt{\pi}} \sqrt{t_D} \quad (9)$$

Now let us consider the longer time values of p_D . It is well known, in well testing lore, that at long time in an infinite-acting system, the following simple semi-logarithmic equation is valid.

$$p_D = \frac{1}{2} (\ln t_D + 0.80907) \quad (10)$$

For practical purposes, Eq. 10 is valid after a dimensionless time ranging from about 20 to 50, depending on the accuracy of the data.

Since the early time data approaches Eq. 9, and the late time data approaches Eq. 10, it seemed likely that we can use this information to develop simple closed form approximate equations. The short time data were fit to the following equation for t_D 's ranging from 0.0005 to 2.00,

$$p_D = 1.1237 (t_D)^{1/2} - 0.4326 t_D + 0.106 (t_D)^{3/2} \quad (11)$$

The greatest error in the fit was 0.60%, far less error than we would expect in field data.

For the long time match, I used Eq. 10 as a starting point and added an empirical time function which declines as time increases.

$$p_D = \frac{1}{2} \left[\ln t_D + 0.80907 + \frac{1.024}{(t_D + 0.40)^{0.729}} \right] \quad (12)$$

Equation 12 was found to fit Chatas' $p_D(t_D)$ data quite well for times, $0.70 \leq t_D$; with a maximum error of 0.40%.

Since the infinite aquifer solution becomes a semi-log straight line after a period of time, it can be graphed simply. Also this same graph can be used to compare this behavior with that of other outer boundary conditions [Fig.1, from Aziz and Flock (1963)]. This graph is really remarkable, for it shows that the lines for a constant pressure outer boundary look much like each other (becoming horizontal), and the lines for the no flow outer boundary also look similar in the way they rise. We'll discuss these solutions next.

Constant Pressure Outer Boundary

Consider the cases where the pressure is fixed at the outer boundary, and become constant in Fig. 1. With a little thought, we should realize that these systems approach the steady state condition, for the flow rate is constant, and the outer boundary pressure is fixed, and Darcy's Law holds, as follows,

$$p_D = \ln(r_D) \quad (13)$$

This outer boundary condition causes exponential decline when the data are graphed properly. To show this concept, I've looked at one case in detail, at $r_D = 10.0$. The exponential decline equation tells us that, if we were to graph the log of the pressure difference against time on arithmetic coordinates, we should get a straight line. For this purpose, the pressure term graphed should be $p_D(\infty) - p_D(t_D)$; and for $r_D = 10$, $p_D(\infty)$ is equal to 2.303, from Eq. 13. The results, from Chatas' Table 5, are graphed in Fig. 2. Clearly a perfect straight line is found. Note that the first point on this graph is at $t_D = 10$, and the value of $p_D(t_D)[1.651]$ is the same as in Chatas' Table 1 for the infinite system. Thus, this system can be treated as though it were infinite for some time, and then the exponential decline equation can be used thereafter. This exercise makes it clear why the curves which become horizontal in Fig. 1 look so much like each other.

Clearly, systems at other radii will behave in this same way. Thus we could derive closed form solutions for the times to switch from infinite acting to exponential behavior, and define the slopes and interrupts of these equations. I've not done that here, for the constant pressure case is the one most commonly used in water influx calculations.

Closed Outer Boundary

The lines that rise above the semi-log straight line in Fig. 1 are for the closed outer boundary. They curve on this graph, but if they are plotted on arithmetic paper, we find that they are straight lines. The reason for this is simple. At late times, the entire system approaches pseudo-steady state flow. We'll address this concept next.

A commonly used equation relates the difference between the initial pressure and the inner boundary pressure to the reservoir parameters. This equation is derived by Brigham (1998) and repeated from Eq. 30b of those notes.

$$\frac{2\pi k h (p_i - p_w)}{q\mu} = \frac{(r_e/r_w)^4 \ln(r_e/r_w)}{[(r_e/r_w)^2 - 1]^2} - \frac{3(r_e/r_w)^2 - 1}{4[(r_e/r_w)^2 - 1]} + \frac{2t_D}{(r_e/r_w)^2 - 1} \quad (14)$$

What we would like to do is to compare Chatas' pressures in Table 4, for the closed outer boundary, with the results one would calculate using various assumptions about the flow. At early times, one would expect that the outer boundary would have no effect, while at later times Eq. 14 should be valid. To test this idea, I've listed pressure data from Chatas' Table 1, $(p_D)_\infty$, and from his Table 4, $(p_D)_c$, and from calculations using Eq. 14, $(p_D)_{pss}$, at values of $r_D = 2, 5$ and 10 in Table 1. There are also data available from Katz *et al.* (1968) for larger radii. I list their $r_D = 100$ data in the same way in Table 1.

Figure 1. Values of $p_i (= \Delta p_D l_{well})$ for infinite reservoirs, for finite circular reservoirs with no flow at the external boundary and for finite circular reservoirs with constant pressure at the external boundary. From Aziz and Flock (1963).

Figure 2. Exponential decline for constant rate with a constant Pressure outer boundary.

Table 1. Comparisons of Calculated p_D 's at Various Values of r_D and t_D for Closed Systems

$r_D = 2$				$r_D = 5$			
t_D	$(p_D)_\infty$	$(p_D)_c$	$(p_D)_{pss}$	t_D	$(p_D)_\infty$	$(p_D)_c$	$(p_D)_{pss}$
0.20	0.4241	0.427	0.4489	3.0	1.1665	1.167	1.2255
0.30	0.5024	0.507	0.5156	4.0	1.2750	1.281	1.3088
0.40	0.5645	0.579	0.5823	5.0	1.3625	1.378	1.3922
0.50	0.6167	0.648	0.6489	6.0	1.4362	1.469	1.4755
				7.0	1.4997	1.556	1.5588
				10.0	1.6509	1.808	1.8088
$r_D = 10$							
t_D	$(p_D)_\infty$	$(p_D)_c$	$(p_D)_{pss}$				
15	1.8294	1.832	1.8973				
20	1.9601	1.968	1.9983				
30	2.1470	2.194	2.2003				
40	2.2824	2.401	2.4024				
50	2.3884	2.604	2.6044				

Comparisons of p_D 's at $r_D = 100$,

Closed Outer Boundary (Katz *et al.*, 1968)

t_D	$(p_D)_\infty$	$(p_D)_c$	$(p_D)_{pss}$
100	2.7233	2.723	3.8760
250	3.1726	3.173	3.9060
500	3.5164	3.516	3.9560
1,000	3.8584	3.861	4.0560
2,500	4.3166	4.335	4.3561
5,000	4.6631	4.856	4.8561
10,000	5.0097	5.856	5.8562
25,000	5.4679	8.857	8.8569

If we look at these results in detail, certain trends and comparisons become obvious. First, the infinite system tables show the smallest pressure drops, but at early times the pressure behavior for the finite systems closely follows that of the infinite system. The pseudo-steady state equations predict the greatest pressure drops. But again, note that the later time behavior of all the real systems closely follow the pseudo-steady state equations.

A most important conclusion can be reached by evaluating the tabulated data in detail. We see that one or the other of these simpler equations will predict the values in the tables with an error of only about 1% over the entire range of data. In brief, the tables for finite systems are not needed at all! We can use the infinite system equations at early times and then switch to the pseudo-steady state equation to calculate for later times.

To carry the idea out in detail, I have performed these same calculations for a host of r_D 's ranging from 1.5 to 100, and have evaluated the t_D 's at the crossover times. These results are shown in Fig. 3 as circles. These data were straightened by graphing against $r_D - 1$ rather than r_D .

The resulting data are shown as diamonds on Fig. 3. They clearly fall on a straight line, whose slope is almost equal to 1.00. The equation is ,

$$t_D = 0.328 (r_D - 1)^{1.945} \quad (15)$$

In summary, to make calculations for a closed system at constant rate; at early times the equations for an infinite system can be used, and at late times, Eq. 14 can be used. Equation 15 defines the time, t_D , to switch from early to late time calculations.

Constant Pressure Inner Boundary

For the constant rate cases, Chatas looked at three outer boundaries: infinite, closed and constant pressure. For the constant pressure case, in Table 2, he lists the infinite system, and in Table 3 he lists closed outer boundaries. He did not look at the constant pressure case. It is not too important to consider this case, so we'll ignore it, and begin by looking at the infinite system in Chatas' Table 2, and also at Ehlig-Economides more recent work, (Appendix B) which is more accurate.

Infinite Aquifer

Note the headings in Chatas' Table 2 for the infinite aquifer. Dimensionless time is labeled, t , while we commonly use t_D now. The fluid influx term is labeled $q(t)$. This is cumulative influx, and the present nomenclature is $Q_D(t_D)$. Dimensionless influx rate is not listed in this table, but we will discuss this later, and its symbol is now $q_D(t_D)$.

For the constant rate inner boundary, we noted that the very early time data closely followed the $(t_D)^{1/2}$ equation. This is also true for the constant pressure case. It seemed likely that this idea could be extended empirically by adding a term using t_D to some other power. The following equation was found to fit the tabulated data for $0.01 \leq t_D \leq 10.0$,

$$Q_D(t_D) = 1.058 t_D^{1/2} + 0.510 t_D^{0.90} \quad (16)$$

The fit is quite accurate over this range. The values at $t_D = 0.01$ and 0.05 show rather large errors of up to 2.5 %, but these are usually not too important in practical use. Further, there is likely some inherent error in Chatas' table for these low values of t_D for they do not quite behave logically. The rest of the data show errors of 1% or less.

At late time, it seemed likely that a semi-logarithmic approach would work,

$$\frac{t_D}{Q_D(t_D)} = a + b \ln(t_D) \quad (17)$$

It does! However, we would like to extend this equation to a shorter time range. I added an empirical constant to the t_D on the left-hand side of Eq. 17, as follows,

$$Q_D(t_D) = \frac{t_D - 1.4}{0.0407 + 0.4887 \ln(t_D)} \quad (18)$$

This equation fit Ehlig-Economides' data from $t_D = 10.0$ to $t_D = 100,000$, with a maximum error of 0.40%; very good indeed.

Closed Outer Boundary

In thinking about a closed outer boundary, with a constant pressure inner boundary, we should realize that, after a period of time, water influx will stop. This will occur when the entire

Figure 3. Time for crossover from infinite acting to pseudo-steady state.
Constant rate-closed outer boundary.

aquifer has been depleted to the inner boundary pressure. We can calculate the maximum cumulative influx using simple material balance principles, as follows,

$$Q_D(\infty) = \frac{(r_e / r_w)^2 - 1}{2} = \frac{r_D^2 - 1}{2} \quad (19)$$

The long time results in Chatas' Table 3 verify this equation.

The early time data in these tables also behave logically. We would expect that, at early time, the effect of the outer boundary would not be felt. So the finite systems should act the same way as an infinite system. They do! This is why Chatas started his listings in Table 3 after a period of time.

Ehlig-Economides (1979), in her Ph.D. dissertation, looked at the flow behavior for a constant inner boundary pressure. One important conclusion she reached was that all the finite systems exhibit exponential decline once the outer boundary is felt. It is interesting that neither van Everdingen and Hurst (1949) nor Chatas recognized this fact. I'll show how this idea can be used to transform Chatas' results into simple equation forms.

For any finite system, after a time, the flow rate at the inner boundary is proportional to the difference between the average pressure and the inner boundary pressure, as follows.

$$q(t) = \frac{C_1 [\bar{p}(t) - p_w]}{(p_i - p_w)} \equiv \frac{dQ(t)}{dt} \quad (20)$$

Also, we can define cumulative influx, using the material balance as follows,

$$Q(t) = \frac{C_2 [p_i - \bar{p}(t)]}{(p_i - p_w)} \quad (21)$$

We can now combine Eqs. 20 and 21, rearrange and integrate, to get,

$$t = \frac{C_2}{C_1} \ln \left[\frac{C_2}{C_2 - Q(t)} \right] \quad (22)$$

Next we need to evaluate the constants, C_1 and C_2 in Eq. 22. At time, $t = 0$, $Q(t) = 0$, and from Eq. 20 and Eq. 21, we can define C_1 ,

$$q(t) = q(0) = C_1 \quad (23)$$

When $t = \infty$, the log term in Eq. 22 must be infinite, so we can conclude,

$$Q(t) = Q(\infty) = C_2 \quad (24)$$

As a result, Eq. 22 becomes,

$$t = \frac{Q(\infty)}{q(0)} \ln \left[\frac{Q(\infty)}{Q(\infty) - Q(t)} \right] \quad (25)$$

Equation 25 is only valid from the time when exponential decline begins. However we can extrapolate the equation back to $t = 0$, and it can be changed to the following dimensionless form, which is not quite correct, but very nearly so.

$$t_D = \frac{Q_D(\infty) - Q_D(0)}{q_D(0)} \ln \left[\frac{Q_D(\infty) - Q_D(0)}{Q_D(\infty) - Q_D(t_D)} \right] \quad (26)$$

It only remains to evaluate these terms from the data in Chatas' tables.

The evaluation problem is more difficult than it first seems. It requires a number of graphical procedures and calculations, plus some correlation work to correct for the slight errors involved in getting from Eq. 25 to Eq. 26. The procedure I used was as follows. First I graphed $[Q_D(\infty) - Q_D(t_D)]$ from Chatas' Table 3 against t_D on semilog paper, as suggested by Eq. 26. The straight line portions of these graphs were extrapolated to $t_D = 0$. These are the correct values of $[Q_D(\infty) - Q_D(0)]$ for Eq. 26. After all of Chatas' tables had been evaluated, an empirical equation was derived to calculate $Q_D(0)$. It was a function of r_e / r_w , as expected.

Finally, I evaluated the slopes of the semilog straight lines, and compared them to the slopes one would calculate using pseudo-steady state for the terms in front of the logarithm in Eq. 26. There was a slight error since the flow is not quite pseudo-steady state, which I correlated against r_e / r_w .

Since all this procedure may be a bit hard to follow, I'll show a detailed set of example calculations, for $r_D = 4.0$, to show the procedure. For $r_D = 4.0$, from Eq. 19, we can calculate $Q_D(\infty)$, as follows.

$$Q_D(\infty) = [(r_e / r_w)^2 - 1] / 2 = [(4.0)^2 - 1] / 2 = 7.500 \quad (27)$$

The data from Chatas' Table 3 are graphed in Fig. 4. It is clear they fit a semilog straight line. To evaluate the slope, I used values from his table at $t_D = 2.0$ and $t_D = 26.0$.

$$t_D = 2.00 \quad , \quad Q_D(t_D) = 2.442 \quad ; \quad Q_D(\infty) - Q_D(t_D) = 5.058$$

$$t_D = 26.0 \quad , \quad Q_D(t_D) = 7.377 \quad ; \quad Q_D(\infty) - Q_D(t_D) = 0.123$$

The slope is,

$$\frac{\Delta t_D}{\ln[\Delta Q_D(2.00)] - \ln[\Delta Q_D(26.0)]} = \frac{26.0 - 2.00}{\ln(5.058) - \ln(0.123)} = 6.4576 \quad (28)$$

Using the slope from Eq. 28, and the value of $Q_D(\infty) - Q_D(2.00)$ equal to 5.058, the value of $Q_D(\infty) - Q_D(0)$ can be easily calculated,

$$\begin{aligned} \ln[Q_D(\infty) - Q_D(0)] &= \frac{2.00}{6.4576} + \ln(5.058) \\ Q_D(\infty) - Q_D(0) &= 6.8942 \end{aligned} \quad (29)$$

Thus the value of $Q_D(0)$ for this r_D ($r_D = 4.0$) is,

$$Q_D(0) = 7.500 - 6.8942 = 0.6058 \quad (30)$$

The values of $Q_D(0)$ for all the radii were correlated into the following equation.

$$Q_D(0) = 0.013 + 0.1756 (r_D - 1)^{1.111} \quad (31)$$

Next I calculated the approximate value for $q_D(0)$ assuming the pseudo-steady state equation was valid. This equation (Brigham, 1998) is,

$$\frac{1}{q_D(0)} = \frac{r_D^2 \ln r_D}{r_D^2 - 1} - \frac{1}{2} = \frac{(4.0)^2 \ln(4)}{(4.0)^2 - 1} - \frac{1}{2} = 0.9787 \quad (32)$$

Thus the approximate value for the slope is,

$$\frac{Q_D(\infty) - Q_D(0)}{q_D(0)} = 6.8942 (0.9787) = 6.7474 \quad (33)$$

The actual slope is Eq. 28, while the approximate slope is Eq. 33. The error is thus,

$$\text{Slope Error} = \frac{6.7474}{6.4576} = 1.045 \quad (34)$$

The values of these errors were correlated into an equation for all radii, as follows,

$$\text{Error} = 1.070 - \frac{0.057}{(r_D^2 - 1)^{0.297}} \quad (35)$$

In brief then, we now know we can calculate the early time constant pressure aquifer data using Eq. 16 or Eq. 18, depending on the time range, and we can calculate the later time (depletion) history using Eqs. 26, 33 and 35. It only remains to define the time to switch from infinite-acting to finite-acting (depletion) behavior. Again this required correlating the data in Chatas' Tables 2 and 3 as a function of r_D . The equation I came up with was,

$$(t_D)_{\text{switch}} = 0.1600 (r_D - 1)^{2.21} \quad (36)$$

Equation 36 is not very accurate. The reason, is that for all r_D 's, the infinite-acting data and the finite-acting data were quite close to each other over a rather broad time range. Thus the precise times could not be defined accurately. This, of course, is good news, for it means that all the resulting calculations would be reasonably accurate.

To evaluate the accuracy of these equations at the times when we switch from infinite acting to finite acting behavior, I have listed all the influx values from Chatas' tables and from my equations in Table 2. In this table, the first column shows all the r_D 's (r_e / r_w values) listed in Chatas' Table 3. The second lists the switchover times calculated by Eq. 36; while the third column shows the actual times used. These are near the calculated times, and also compatible with Chatas' tables.

Table 2. Influxes From Chatas' Tables 2 and 3
Compared to Approximate Equations

r_D	Calc. (t_D) _{switch} Eq. 36	(t _D) _{switch} Used	∞ Acting Q_D		Finite Acting Q_D		Max. Diff. (%)
			Eqs. 16 and 18	Chatas' Table 2	Eq. 26	Chatas' Table 3	
1.5	0.035	0.050	0.2710	0.278	0.2753	0.276	2.58
2.0	0.160	0.15	0.5022	0.520	0.5064	0.507	3.54
2.5	0.392	0.40	0.8927	0.898	0.8940	0.897	0.59
3.0	0.740	0.70	1.255	1.251	1.257	1.256	0.46
3.5	1.212	1.00	1.568	1.569	1.574	1.571	0.56
4.0	1.814	2.00	2.448	2.447	2.443	2.442	0.24
4.5	2.55	2.50	2.836	*	2.836	2.835	0.04
5.0	3.43	3.50	3.554	*	3.552	2.542	0.34
6.0	5.61	6.0	5.150	5.153	5.144	5.148	0.17
7.0	8.39	9.0	6.859	6.869	6.853	6.861	0.23
8.0	11.80	12.0	8.446	8.457	8.436	8.431	0.31
9.0	15.85	15.0	9.970	9.949	9.932	9.945	0.38
10.0	20.6	20.0	12.36	12.32	12.29	12.30	0.57

Figure 4. Exponential decline graph for constant pressure depletion, $r_D = 4.0$.

The fourth and fifth columns compare the Q_D 's for the infinite acting system. The fourth column is from my Eq. 16 for r_D 's up to 7.0, and from Eq. 18 for the three larger r_D 's; while the fifth column lists the results from Chatas' Table 2 for these same times.

The sixth and seventh columns show the same kind of information for the finite systems. The sixth column shows the predicted values of Q_D from Eq. 26, while the seventh column shows values listed in Chatas' Table 3. It is important that, at any r_D , all four values of Q_D are very close to each other. To compare them in detail, I've listed the maximum differences in the Q_D listings in the eighth column. Note that the first two show differences of 2.58 and 3.54%, while all the others are less than 1% in maximum difference. This is a remarkably accurate result!

The final evaluation is to compare the calculated exponential decline slopes (using all the material discussed here) with the slopes found from Chatas' Table 3. The equation

for the decline using my method is as follows,

$$\text{Calculated Slope} = \frac{Q_D(\infty) - Q_D(0)}{q_D(0)(\text{Error})} \quad (37)$$

The decline rates calculated for these systems are quite accurate. The maximum error is 0.71%.

Linear Geometry

The behavior of a linear aquifer is far simpler than a radial aquifer. The mathematics of the problem were first published by Miller in 1962, but shortly after that a quite elegant piece of work by Nabor and Barham (1964) presented the entire linear aquifer equations and curves in a three-page paper. The remarkable result of Nabor and Barham's work was to show that all six possible boundary conditions (Interior Boundary, constant pressure or constant rate; Outer Boundary, closed, constant pressure or infinite), could be shown with only three equations, or alternatively, three lines on a single graph. Their paper is attached as Appendix C.

The reason for this behavior becomes obvious when one looks at their equations. Their Eq. 1 shows the pressure drop for the infinite system with a constant rate inner boundary, while their Eq. 4 shows the cumulative water influx for the infinite system with a constant pressure inner boundary. Notice that the time relationship is the same for both of them. In dimensionless form, it is defined in their Eqs. 9 and 12, as,

$$F_{1/2}(t_D) = 2\sqrt{t_D/\pi} \quad (38)$$

$$\text{where } t_D = kt/\phi\mu c_t L^2$$

and L = any arbitrary distance, for the infinite system

Other comparisons are also interesting. Note in Nabor and Barham's Eq. 11, for the finite aquifer with the constant pressure at the outer boundary at constant rate, the pressure drop behavior fits the $F_0(t_D)$ function. In their Eq. 13, for the finite aquifer with a closed outer boundary and constant pressure inner boundary, the cumulative water influx solution uses the same $F_0(t_D)$ function. $F_0(t_D)$ is defined in Eq. 15 by Nabor and Barham.

It is not too surprising that these two cases give the same result. For the pressure drop case, their Eq. 11, at late time the pressure drop follows Darcy's Law. For the water influx case, Eq. 13, the total water influx is finite due to the sealed outer boundary. The dimensionless equations are defined so that these constants are both equal to 1.00.

The $F_0(t_D)$ function can be expressed almost exactly using simple analytic solutions. With a little thought, we should realize that we should expect exponential behavior, as found in radial flow. Using ideas similar to those in Eqs. 25 and 26, we would expect the tabulated data would be a straight line on semi-log paper. Their $F_0(t_D)$ data are graphed in Fig. 5 for $t_D \geq 0.18$. These data veer away from the infinite acting data; but note that the first data point in this graph fits the infinite aquifer solution, Eq. 38. So these data act the same as the radial system. At early times they fit the infinite acting equation. Then they switch immediately to exponential decline at time, $t_D = 0.18$, as Fig. 5 shows.

The data in Fig. 5 are a perfect straight line. So a simple equation could be used to predict the cumulative influx behavior with time. Since the pressure behavior for constant rate with a constant pressure outer boundary fit the same curve, we could also predict the pressure decline history for this case.

The last two cases which fit together are the pressure drop prediction for a constant rate with a sealed outer boundary, and the water influx prediction for constant pressure inner and outer boundaries. These both use the $F_1(t_D)$ equation, Nabor and Barham's Eq. 17. This behavior is also logical. With the closed outer boundary, the system reaches pseudo-steady state, with pressure a linear function of time. While, with the constant pressure boundaries, the system reaches steady state and the cumulative water influx will rise linearly with time. Notice in their Eq. 17, that the long time result for the $F_1(t_D)$ function is,

$$F_1(t_D) = t_D + 1/3 \quad (39)$$

This is the pseudo-steady state equation for linear systems.

Since a simplification of the $F_0(t_D)$ curve worked well, it seems logical that a similar approach would work for the $F_1(t_D)$ curve. To test this idea, I compared $F_1(t_D)$ for various times against the values of Eq. 38 and Eq. 39, as shown on the following table.

Table 3. Comparisons of $F_1(t_D)$ with Eq. 38 and Eq. 39
to Approximate $F_1(t_D)$

t_D	$F_1(t_D)$	Eq. 39 $t_D + 1/3$	Eq. 39 Error	Eq. 38 $F_{1/2}(t_D)$	Eq. 38 Error
0.225	0.536	0.558	0.022	0.535	-0.001
0.25	0.566	0.583	0.017	0.564	-0.002
0.28	0.601	0.613	0.012	0.597	-0.004
0.31	0.634	0.643	0.009	0.628	-0.006
0.35	0.677	0.683	0.006	0.668	-0.009
0.40	0.729	0.733	0.004	0.714	-0.015
0.45	0.781	0.783	0.002	0.757	-0.024
0.50	0.832	0.833	0.001	0.798	-0.034
0.56	0.893	0.893	0.000	0.844	-0.049

Figure 5. Exponential decline behavior for linear aquifers.

A comparison of the results in this table is similar to the behavior we saw for radial systems. At late time the pseudo-steady state assumption, is valid, while at early time the infinite-acting equation is valid. The cross-over occurs at a time, t_D , between 0.31 and 0.35 with an error of about 1.4%. Clearly this procedure would simplify calculations of influx for linear aquifers.

Spherical Geometry

The spherical geometry is not too common in water influx, but it can arise whenever there is a small oil "bubble" surrounded at the sides and bottom by a large aquifer. The paper by Chatas (1966) discusses this geometry. It is far longer than it needs to be, for it could have been simplified in much the same way that Nabor and Barham simplified the linear systems.

One important point in spherical flow is that vertical permeability, k_v , is often far less than horizontal permeability, k_h . Chatas discusses this fact and his Eq. 10 is supposed to give us the correct value of average permeability, $\bar{k}$, to use when these permeabilities differ. This equation is wrong! The correct average permeability is,

$$\bar{k} = (k_h^2 k_v)^{1/3} \quad (40)$$

as discussed in my notes on injectivity (Brigham, 1985).

We also need to look at the transformed inner boundary that results from scaling. The z direction coordinate will be elongated, while the x and y directions will be shrunk. As a result, the inner sphere will be changed into "rugby ball" shape, with the ball standing on its end, an ellipsoid. To the best of my knowledge, the equivalent radius equations have not been worked out for this geometry.

Chatas' solutions are listed in voluminous tables. In his table for the infinite system, he lists e_D , which is $q_D(t_D)$, F_D , which is $Q_D(t_D)$, and p_D . He shows that there are simple equations for these terms. They are,

$$q_D(t_D) = 1 + (\pi t_D)^{-1/2} \quad (41)$$

$$Q_D(t_D) = t_D + 2(t_D / \pi)^{1/2} \quad (42)$$

$$\text{and } p_D(t_D) = 1 - \exp(-t_D) \operatorname{erfc}(t_D^{1/2}) \quad (43)$$

So it really wasn't necessary to list these values. Equations 41 and 42 are very easy to evaluate. Even Eq. 43, the most complex one, can be evaluated easily.

Chatas also lists results for finite systems with closed exterior boundaries, and constant pressure exterior boundaries. Based on the work we have done for the linear and radial systems, we would expect that these tables could also be handled with simple analytic solutions. His table results indicate this can be done, for some follow exponential decline, while others vary linearly with time.

I am positive that one could develop simple appropriate equations to handle these systems. However, I have not done it, for I've seldom seen field cases where the spherical geometry is required.

Conclusions

We've seen that the results of all three geometries (linear, radial and spherical) can be put into simple equation forms. These equations are based on the flow physics. Thus for all the possible

boundary conditions (Inner Boundary; constant rate or constant pressure: Outer Boundary; infinite, constant pressure or closed) the solutions behave in a logical manner.

The linear and spherical systems behave in similar ways, because the spherical equation can be transformed into a form identical to linear flow. Thus both geometries show a square root of time relationship for the infinite system for predicting encroachment. For the linear system, the pressure prediction is also proportional to the square root of time. While for the spherical system, the equation is slightly more complex, but still simple.

For the infinite radial systems, the very early time data also follow square root of time behavior. For a limited time, simple empirical extensions of this idea are valid for either the constant pressure or constant flow rate inner boundary.

The very long time behavior of the infinite radial systems are also logical, being functions of the logarithm of time. Simple empirical adjustments to these late time results are shown for both the constant pressure and constant flow rate inner boundary.

For all the finite systems, either a constant pressure or a closed outer boundary can be assumed. The early time data for these systems all follow the infinite curves. Simple equations define the times when this short time behavior is no longer valid. As expected, these equations are functions of the sizes of the systems.

Once the outer boundary begins to be felt, the equations, for all practical purposes, jump immediately to the long time form expected for that geometry and boundary condition. For example, for a constant pressure inner and outer boundary, the cumulative influx varies linearly with time, following the steady state Darcy equation. Similarly, for the constant rate inner boundary and a closed outer boundary, the pseudo-steady state equations define the linear pressure decline behavior. These statements are true for all three geometries.

By comparison, for a constant rate inner boundary and constant pressure outer boundary, exponential decline is seen. The pressure history is a logarithmic function of dimensionless time. On the logarithmic coordinate we graph, $p_D(\infty) - p_D(t_D)$, to depict this exponential behavior.

Similarly, for the constant pressure inner boundary and the closed outer boundary, we also see exponential behavior. This, too, has a limit at infinite time, $Q_D(\infty)$, which is defined by the geometry. The variable graphed on the arithmetic coordinate is again the dimensionless time, while the logarithmic coordinator is $Q_D(\infty) - Q_D(t_D)$.

Thus we've seen that the exact solutions can be transformed into very accurate simple approximations which make calculations much easier, and which also give great insight into the behavior of the various solutions. These approximate equations can be used to greatly simplify calculations.

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Appendix A

A Practical Treatment of Nonsteady-State Flow Problems in Reservoir Systems

Part 3 (Appendix) Practical information is presented in the form of tables, definitions, and a complete resume of the Hurst-van Everdingen equations

ANGELOS T. CHATAS*

Definitions

1. **Annulus Problem.** The determination of the flowing bottom-hole pressure history of a flowing well, undergoing a flow test, that produces through tubing at a constant surface-rate of production. During the test some of the oil comes from the pay formation, but some is also unloaded from the annulus between the tubing and casing.

2. **Annulus Volume Adjustment.** In the "annulus problem" the volume of fluid unloaded from the annulus per unit pressure drop per unit sand thickness. It may be expressed approximately by

$$\pi (r_c^2 - r_w^2) \rho_o g h$$

which in dimensionless form becomes

$$C = \frac{r_c^2 - r_w^2}{2g\phi h \rho_o c_o r_w^2}$$

3. **Boundary Conditions.** The location of the interior and exterior boundaries and the specification of pressure and/or flow at these boundaries at a given instant of time.

4. **Boundary Variations.** The changes in the boundary conditions of a reservoir system undergoing exploitation.

5. **Continuous Succession of Steady States.** A method of solving flow problems in a reservoir system that suffers boundary variations but fulfills instantaneous steady-state conditions. The history of such a system is divided into an appropriate number of stages, each of which is treated by steady-state analysis.

6. **Dimensionless Time Ratio.** The dimensionless ratio defined by the following relations:

Radial System

$$t = \frac{k\theta}{\phi \mu c_r r_w^2}$$

Linear System

$$t = \frac{k\theta}{\phi \mu c_r x_c^2} = \frac{K\theta}{\phi M C_t}$$

7. **Fluid-Influx Terms.** Terms that appear in the Hurst-van Everdingen equation which treats the "pressure case." Such a term, denoted by $q(t)$, is a dimensionless, numerical quantity representing the total volume of fluid per unit thickness that passes the interior boundary of a reservoir system over the time

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TABLE 1. Infinite radial system — Rate case.

Dimensionless time t	Pressure change $p(t)$	Dimensionless time t	Pressure change $p(t)$	Dimensionless time t	Pressure change $p(t)$	Dimensionless time t	Pressure change $p(t)$
0	0	.06	.2500	3.0	1.1685	150.0	2.6212
.005	.0250	.07	.2680	4.0	1.2750	200.0	3.0636
.001	.0352	.08	.2845	5.0	1.3625	250.0	3.1728
.002	.0495	.09	.2999	6.0	1.4382	300.0	3.2630
.003	.0663	.1	.3144	7.0	1.4997	350.0	3.3394
.004	.0824	.15	.3750	8.0	1.5557	400.0	3.4057
.005	.0974	.2	.4241	9.0	1.6057	450.0	3.4641
.006	.0845	.3	.5024	10.0	1.6509	500.0	3.5194
.007	.0911	.4	.5645	15.0	1.8294	550.0	3.5643
.008	.0971	.5	.6167	20.0	1.9601	600.0	3.6076
.009	.1028	.6	.6622	30.0	2.1470	650.0	3.6476
.01	.1081	.7	.7024	40.0	2.2824	700.0	3.6842
.015	.1312	.8	.7387	50.0	2.3884	750.0	3.7184
.02	.1503	.9	.7716	60.0	2.4753	800.0	3.7505
.025	.1669	1.0	.8019	70.0	2.5501	850.0	3.7805
.03	.1818	1.2	.8672	80.0	2.6147	900.0	3.8088
.04	.2077	1.4	.9160	90.0	2.6718	950.0	3.8355
.05	.2301	2.0	1.0193	100.0	2.7233	1000.0	3.8584

span $t = t_M - t_i$, which is caused by a unit pressure drop at this boundary at time t_i .

8. **Infinite Reservoir System.** A reservoir system analyzed over a period of time during which the presence of an exterior boundary is not felt.

9. **Linear Reservoir System.** A system defined by two parallel planes, which serve as boundaries, over which pressure and flow are specified according to prescribed conditions, and whose physical properties of interest vary only with the perpendicular distance x between the two planes.

10. **Nonsteady-State Flow.** In reservoir systems undergoing exploitation, fluid flow that is characterized predominantly by time variations and necessitates the formal introduction of time as an explicit variable in the basic flow equations.

11. **Oil Field Equivalent.** An oil field lumped together such that its outer limits serve as the interior boundary of a reservoir system, with a pressure equal to the average oil field pressure, and with a flux equal to the sum of the fluxes of the individual wells.

12. **Pressure Case.** In the analysis of a reservoir system the situation that presumes knowledge of the pressure history and predicts the cumulative fluid influx.

13. **Pressure Change Terms.** Terms that appear in the Hurst-van Everdingen equation which treats the "rate case." Such a term, denoted by $p(t)$, is a dimensionless, numerical quantity representing the change in pressure at the interior boundary of a reservoir system over the time span $t = t_M - t_i$, which is caused by a unit rate of production

per unit thickness at time t_i .

14. **Pressure Correction Terms.** Terms that appear in the Hurst-van Everdingen equation which treats the "annulus problem." Such a term, denoted by $p'(t)$, is a dimensionless numerical quantity representing the total pressure drop in the well bore, which is caused by a unit rate of production at the surface and is corrected for the unloading of fluid from the annulus.

15. **Radial Reservoir System.** A system defined by two concentric circular cylinders, which serve as boundaries, over which pressure and flow are specified according to prescribed conditions, and whose physical properties of interest vary only with the distance r from the axis of symmetry.

16. **Rate Case.** In the analysis of a reservoir system the situation that presumes knowledge of the production or fluid-influx rate history and predicts the cumulative pressure drop for a reservoir at its interior boundary.

17. **Steady-State Flow.** In reservoir systems undergoing exploitation, fluid flow whose time variations are insignificant, and, which permit the formal neglect of time as an explicit variable in the basic flow equations.

18. **Time of Readjustment.** The approximate time required for the readjustment of the internal pressure distribution in a reservoir system to a steady-state distribution when pressure variations occur at the boundaries. For radial systems this time is given by

$$\theta_R = \frac{\phi c_r r_o^2}{4k/\mu}$$

while for linear systems it is

EXPLANATION

OF SYMBOLS

- A** = proven area of oil field
- A'** = estimated area of aquifer
- B** = bulk modulus of reservoir fluid
- C** = dimensionless "annulus volume adjustment term"
- C₁** = arbitrary constant
- F_c** = cumulative influx of fluid
- C** = mass rate of flow
- I₀** = modified Bessel function of the first kind and order zero
- J_n** = Bessel function of the first kind and order n
- K_n** = modified Bessel function of the second kind and order zero
- P** = pressure
- P_T** = transform of pressure
- P_B** = pressure at a boundary
- P_{BP}** = bubble point pressure
- P_e** = pressure at exterior boundary; effective reservoir pressure
- P_r** = flowing bottom hole pressure
- P_i** = original reservoir pressure
- P_w** = pressure at interior boundary;
- Q** = volume rate of flow; average rate of production
- Q_T** = transform of production
- Q_s** = volume rate of flow measured at the stock tank
- W_e** = cumulative water influx into oil field
- X** = arbitrary function
- Y_n** = Bessel function of the second kind and order n
- a** = cross-sectional area perpendicular to the direction of flow in linear system
- c** = speed of sound
- c_f** = compressibility of fluid
- c_o** = compressibility of oil
- c_w** = compressibility of water
- g** = gravitational acceleration
- h** = net effective formation thickness

TABLE 3. Finite radial system with closed exterior boundary—Pressure case.

re/rw=1.5		re/rw=2.0		re/rw=2.5		re/rw=3.0		re/rw=3.5		re/rw=4.0		re/rw=4.5	
Dimensionless time	Fluid influx	Dimensionless time	Fluid influx	Dimensionless time	Fluid influx	Dimensionless time	Fluid influx	Dimensionless time	Fluid influx	Dimensionless time	Fluid influx	Dimensionless time	Fluid influx
t	q(t)	t	q(t)	t	q(t)	t	q(t)	t	q(t)	t	q(t)	t	q(t)
5.0(10) ⁻¹	0.278	5.0(10) ⁻¹	0.278	1.0(10) ⁻¹	0.408	3.0(10) ⁻¹	0.755	1.00	1.571	2.00	2.442	2.5	2.835
6.0	0.304	7.5	0.345	1.5	0.509	4.0	0.895	1.20	1.761	2.20	2.598	3.0	3.196
7.0	0.330	1.0(10) ⁻¹	0.404	2.0	0.599	5.0	1.023	1.40	1.940	2.40	2.748	3.5	3.537
8.0	0.354	1.25	0.458	2.5	0.681	6.0	1.143	1.60	2.111	2.60	2.893	4.0	3.859
9.0	0.375	1.50	0.507	3.0	0.758	7.0	1.256	1.80	2.273	2.80	3.034	4.5	4.165
1.0(10) ⁻¹	0.396	1.75	0.553	3.5	0.829	8.0	1.363	2.00	2.427	3.00	3.170	5.0	4.454
1.1	0.414	2.00	0.597	4.0	0.897	9.0	1.465	2.20	2.574	3.20	3.334	5.5	4.727
1.2	0.431	2.25	0.638	4.5	0.962	1.00	1.563	2.40	2.715	3.50	3.493	6.0	4.986
1.3	0.446	2.50	0.678	5.0	1.024	1.25	1.791	2.60	2.849	3.75	3.645	6.5	5.231
1.4	0.461	2.75	0.715	5.5	1.083	1.50	1.997	2.80	2.976	4.00	3.792	7.0	5.464
1.5	0.474	3.00	0.751	6.0	1.140	1.75	2.184	3.00	3.098	4.25	3.932	7.5	5.684
1.6	0.486	3.25	0.785	6.5	1.195	2.00	2.353	3.25	3.242	4.50	4.068	8.0	5.892
1.7	0.497	3.50	0.817	7.0	1.248	2.25	2.507	3.50	3.379	4.75	4.198	8.5	6.089
1.8	0.507	3.75	0.848	7.5	1.299	2.50	2.648	3.75	3.507	5.00	4.323	9.0	6.276
1.9	0.517	4.00	0.877	8.0	1.348	2.75	2.772	4.00	3.628	5.50	4.560	9.5	6.453
2.0	0.525	4.25	0.905	8.5	1.395	3.00	2.886	4.25	3.742	6.00	4.779	10	6.621
2.1	0.533	4.50	0.932	9.0	1.440	3.25	2.990	4.50	3.850	6.50	4.982	11	6.930
2.2	0.541	4.75	0.958	9.5	1.484	3.50	3.084	4.75	3.951	7.00	5.169	12	7.208
2.3	0.548	5.00	0.983	1.0	1.528	3.75	3.170	5.00	4.047	7.50	5.343	13	7.457
2.4	0.554	5.50	1.028	1.1	1.605	4.00	3.247	5.50	4.122	8.00	5.504	14	7.680
2.5	0.559	6.00	1.070	1.2	1.679	4.25	3.317	6.00	4.178	8.50	5.653	15	7.880
2.6	0.565	6.50	1.108	1.3	1.747	4.50	3.381	6.50	4.518	9.00	5.790	16	8.060
2.8	0.574	7.00	1.143	1.4	1.811	4.75	3.439	7.00	4.639	9.50	5.917	18	8.365
3.0	0.582	7.50	1.174	1.5	1.870	5.00	3.491	7.50	4.749	10	6.035	20	8.611
3.2	0.588	8.00	1.203	1.6	1.924	5.50	3.581	8.00	4.846	11	6.246	22	8.809
3.4	0.594	9.00	1.253	1.7	1.975	6.00	3.656	8.50	4.932	12	6.425	24	8.968
3.6	0.599	1.00	1.295	1.8	2.022	6.50	3.717	9.00	5.009	13	6.580	26	9.097
3.8	0.603	1.1	1.335	1.9	2.066	7.00	3.778	9.50	5.078	14	6.715	28	9.200
4.0	0.606	1.2	1.358	2.2	2.173	7.50	3.809	10.00	5.138	15	6.825	30	9.283
4.5	0.613	1.3	1.382	2.4	2.241	8.00	3.843	11	5.241	16	6.922	34	9.404
5.0	0.617	1.4	1.402	2.6	2.294	9.00	3.894	12	5.321	17	7.004	38	9.481
6.0	0.621	1.6	1.432	2.8	2.340	10.00	3.928	13	5.385	18	7.076	42	9.532
7.0	0.623	1.7	1.444	3.0	2.380	11.00	3.951	14	5.435	20	7.189	46	9.565
8.0	0.624	2.0	1.453	3.4	2.442	12.00	3.978	15	5.476	22	7.272	50	9.586
		2.5	1.487	4.2	2.525	18.00	3.993	17	5.531	26	7.377	70	9.621
		3.0	1.495	4.6	2.551	18.00	3.997	18	5.551	30	7.434	80	9.623
		4.0	1.499	6.0	2.570	20.00	3.999	20	5.579	34	7.464	90	9.624
		5.0	1.500	6.0	2.599	22.00	3.999	25	5.611	38	7.481	100	9.625
				7.0	2.613	24.00	4.000	30	5.621	42	7.490		
				8.0	2.619			35	5.624	46	7.494		
				9.0	2.622			40	5.625	50	7.497		
				10.0	2.624								

re/rw=5.0		re/rw=6.0		re/rw=7.0		re/rw=8.0		re/rw=9.0		re/rw=10.0	
Dimensionless time	Fluid influx	Dimensionless time	Fluid influx	Dimensionless time	Fluid influx	Dimensionless time	Fluid influx	Dimensionless time	Fluid influx	Dimensionless time	Fluid influx
t	q(t)	t	q(t)	t	q(t)	t	q(t)	t	q(t)	t	q(t)
3.0	3.195	6.0	5.148	9.00	6.861	9	6.861	10	7.417	15	9.965
3.5	3.542	6.5	5.440	9.50	7.127	10	7.398	15	9.945	20	12.32
4.0	3.875	7.0	5.724	10	7.389	11	7.920	20	12.26	22	13.22
4.5	4.193	7.5	6.002	11	7.902	12	8.431	22	13.13	24	14.09
5.0	4.499	8.0	6.273	12	8.397	13	8.930	24	13.98	26	14.95
5.5	4.792	8.5	6.537	13	8.876	14	9.418	26	14.79	28	15.75
6.0	5.074	9.0	6.795	14	9.341	15	9.895	28	15.59	30	16.59
6.5	5.345	9.5	7.047	15	9.791	16	10.361	30	16.35	32	17.38
7.0	5.605	10.0	7.293	16	10.23	17	10.82	32	17.10	34	18.16
7.5	5.854	10.5	7.533	17	10.65	18	11.26	34	17.82	36	18.91
8.0	6.094	11	7.767	18	11.06	19	11.70	36	18.52	38	19.65
8.5	6.325	12	8.220	19	11.46	20	12.13	38	19.19	40	20.37
9.0	6.547	13	8.651	20	11.85	22	12.95	40	19.85	42	21.07
9.5	6.760	14	9.063	22	12.58	24	13.74	42	20.48	44	21.76
10	6.965	15	9.456	24	13.27	26	14.50	44	21.09	46	22.42
11	7.350	16	9.820	26	13.92	28	15.23	46	21.89	48	23.07
12	7.708	17	10.19	28	14.53	30	15.92	48	22.26	50	23.71
13	8.035	18	10.53	30	15.11	34	17.22	50	22.82	52	24.33
14	8.339	19	10.85	35	16.30	38	18.41	52	23.36	54	24.94
15	8.620	20	11.16	40	17.49	40	18.97	54	23.89	56	25.53
16	8.879	22	11.74	45	18.43	45	20.26	56	24.39	58	26.11
18	9.338	24	12.26	50	19.24	50	21.42	58	24.88	60	26.67
20	9.731	25	12.50	60	20.51	60	22.46	60	25.36	65	28.02
22	10.07	31	13.74	70	21.45	65	23.40	65	25.48	70	29.39
24	10.35	35	14.40	80	22.13	70	24.08	70	26.02	75	30.49
26	10.59	39	14.93	90	22.63	80	26.26	75	26.48	80	31.61
28	10.80	51	18.05	100	23.00	90	27.22	80	26.98	85	32.67
30	10.98	60	18.56	120	23.47	100	28.11	85	30.13	90	33.66
34	11.28	70	19.91	140	23.71	120	29.31	90	30.93	95	34.60
38	11.46	80	17.14	160	23.85	140	30.09	95	31.63	100	35.48
42	11.61	90	17.27	180	23.92	160	30.51	100	32.27	120	38.51
46	11.71	100	17.38	200	23.96	180	30.91	120	34.39	140	40.89
50	11.79	110	17.41	500	24.00	200	31.12	140	35.92	160	42.75
60	11.91	120	17.45			240	31.34	160	37.04	180	44.21
70	11.96	130	17.46			280	31.43	180	37.85	200	45.36
80	11.98	140	17.48			320	31.47	200	38.44	240	46.95
90	11.99	150	17.49			360	31.49	240	39.17	280	47.94
100	12.00	160	17.49			400	31.50	280	39.59	320	48.54
		180	17.50			500	31.50	320	39.77	360	48.91
		200	17.50					360	39.88	400	49.14
		220	17.50					400	39.94	440	49.28
								440	39.97	480	49.30
								480	39.98		

3. Rate of Fluid Influx. The rate of fluid influx has been presented as a function of the fluid-influx terms and, therefore, is a modification of the "pres-

sure case." Thus the rate of fluid-influx at the interior boundary of a radial reservoir system is given at time θ_M by the algebraic equation

$$Q(\theta_M) = 2\pi\phi c_f r_w^2 h \{ [P_o - P(\theta_1)] \frac{[q(t_M) - q(t_M - t_1)]}{\theta_1} + [P(\theta_1) - P(\theta_2)] \}$$

no/rwp=1.5	no/rwp=2.0	no/rwp=2.5	no/rwp=3.0	no/rwp=3.5	no/rwp=4.0
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re/rw=1.5		re/rw=2.0		re/rw=2.5		re/rw=3.0		re/rw=3.5		re/rw=4.0			
Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change		
t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)		
6.0	(10) ⁻² 0.251	2.2	(10) ⁻¹ 0.443	4.0	(10) ⁻¹ 0.565	5.2	(10) ⁻¹ 0.827	1.0	0.802	1.5	0.927		
8.0	0.288	2.4	0.459	4.2	0.576	5.4	0.836	1.1	0.830	1.6	0.948		
1.0	(10) ⁻¹ 0.322	2.6	0.476	4.4	0.587	5.6	0.845	1.2	0.857	1.7	0.968		
1.2	0.335	2.8	0.492	4.6	0.598	5.8	0.863	1.3	0.882	1.8	0.985		
1.4	0.357	3.0	0.507	4.8	0.608	6.0	0.883	1.4	0.906	1.9	1.004		
1.6	0.379	3.2	0.522	5.0	0.618	7.0	0.703	1.5	0.929	2.0	1.025		
1.8	0.452	3.4	0.536	5.2	0.628	7.5	0.721	1.6	0.951	2.2	1.059		
2.0	0.484	3.6	0.551	5.4	0.638	8.0	0.740	1.7	0.973	2.4	1.092		
2.2	0.516	3.8	0.565	5.6	0.647	8.5	0.758	1.8	0.994	2.6	1.123		
2.4	0.548	4.0	0.579	5.8	0.657	9.0	0.776	1.9	1.016	2.8	1.154		
2.6	0.580	4.2	0.593	6.0	0.666	9.5	0.791	2.0	1.034	3.0	1.184		
2.8	0.612	4.4	0.607	6.5	0.688	1.0	0.806	2.25	1.083	3.5	1.255		
3.0	0.844	4.6	0.621	7.0	0.710	1.2	0.865	2.50	1.130	4.0	1.324		
3.5	0.724	4.8	0.634	7.5	0.731	1.4	0.920	2.75	1.176	4.5	1.392		
4.0	0.804	5.0	0.648	8.0	0.752	1.6	0.973	3.0	1.220	5.0	1.460		
4.5	0.886	5.0	0.715	8.5	0.772	2.0	1.076	4.0	1.401	5.5	1.527		
5.0	0.964	7.0	0.782	9.0	0.792	3.0	1.328	5.0	1.579	6.0	1.594		
5.5	1.044	8.0	0.849	9.5	0.812	4.0	1.575	6.0	1.757	6.5	1.660		
6.0	1.124	9.0	0.915	1.0	0.832	5.0	1.828			7.0	1.727		
		1.0	0.982	2.0	1.215					8.0	1.861		
		2.0	1.649	3.0	1.977					9.0	1.994		
		3.0	2.316	4.0	2.735					10.0	2.127		
		5.0	3.849	5.0	2.358								
re/rw=4.5	re/rw=5.0	re/rw=6.0	re/rw=7.0	re/rw=8.0	re/rw=9.0	re/rw=10.0							
Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change						
t	p(t)	t	p(t)	t	p(t)	t	p(t)						
2.0	1.023	3.0	1.187	4.0	1.275	6.0	1.436	8.0	1.559	10.0	1.651	12.0	1.732
2.1	1.040	3.1	1.200	4.1	1.292	6.1	1.470	8.1	1.623	10.1	1.733	12.1	1.793
2.2	1.058	3.2	1.192	5.0	1.384	7.0	1.501	9.0	1.607	11.0	1.693	13.0	1.788
2.3	1.072	3.3	1.204	5.5	1.404	7.5	1.531	9.5	1.631	11.5	1.734	13.5	1.844
2.4	1.087	3.4	1.215	6.0	1.441	8.0	1.559	10.0	1.653	12.0	1.732	14.0	1.810
2.5	1.102	3.5	1.227	6.5	1.477	8.5	1.586	10.5	1.675	12.5	1.750	14.5	1.817
2.6	1.117	3.6	1.238	7.0	1.511	9.0	1.610	11.0	1.697	13.0	1.768	15.0	1.830
2.7	1.130	3.7	1.249	7.5	1.544	9.5	1.638	11.5	1.717	13.5	1.786	15.5	1.847
2.8	1.144	3.8	1.259	8.0	1.576	10.0	1.663	12.0	1.737	14.0	1.803	16.0	1.862
2.9	1.158	3.9	1.270	8.5	1.607	11.0	1.711	12.5	1.757	14.5	1.819	17.0	1.876
3.0	1.171	4.0	1.281	9.0	1.638	12.0	1.737	13.0	1.776	15.0	1.835	18.0	1.891
3.2	1.197	4.2	1.301	9.5	1.668	13.0	1.810	13.5	1.795	15.5	1.851	19.0	1.905
3.4	1.222	4.4	1.321	10.0	1.698	14.0	1.845	14.0	1.813	16.0	1.867	20.0	1.968
3.6	1.246	4.6	1.340	11.0	1.757	15.0	1.888	14.5	1.831	17.0	1.897	22.0	2.017
3.8	1.269	4.8	1.360	12.0	1.815	16.0	1.931	15.0	1.849	18.0	1.926	24.0	2.065
4.0	1.292	5.0	1.378	13.0	1.873	17.0	1.974	17.0	1.919	19.0	1.955	26.0	2.108
4.5	1.340	5.5	1.424	14.0	1.931	18.0	2.016	18.0	1.936	20.0	1.983	28.0	2.151
5.0	1.403	6.0	1.469	15.0	1.988	19.0	2.058	21.0	2.051	22.0	2.037	30.0	2.194
5.5	1.457	5.5	1.513	16.0	2.045	20.0	2.100	23.0	2.116	24.0	2.090	32.0	2.236
6.0	1.510	7.0	1.556	17.0	2.103	22.0	2.184	25.0	2.180	26.0	2.143	34.0	2.278
7.0	1.615	7.5	1.598	18.0	2.160	24.0	2.267	30.0	2.340	28.0	2.192	36.0	2.319
8.0	1.719	8.0	1.641	19.0	2.217	25.0	2.335	35.0	2.444	32.0	2.244	38.0	2.360
9.0	1.823	9.0	1.684	20.0	2.274	28.0	2.421	40.0	2.658	34.0	2.345	40.0	2.400
10.0	1.927	10.0	1.805	25.0	2.560	30.0	2.517	45.0	2.817	38.0	2.446	50.0	2.604
11.0	2.031	11.0	1.892	30.0	2.846					40.0	2.498	60.0	2.806
12.0	2.135	12.0	1.975							45.0	2.621	70.0	3.000
13.0	2.239	13.0	2.059							50.0	2.746		
14.0	2.343	14.0	2.142										
15.0	2.447	15.0	2.225										

 ϕ = average effective porosi

$$\times \left[\frac{q(t_M - t_1) - q(t_M - t_2)}{\theta_2 - \theta_1} \right] + \dots + [P(\theta_{n-1}) - P(\theta_n)] \left[\frac{q(t_M - t_{n-1}) - q(t_M - t_n)}{\theta_n - \theta_{n-1}} \right] \Bigg\}$$

B. Linear Systems. The corresponding expressions for the rate case, pressure case, and rate of fluid influx for infinite linear systems are as follows:

$$P_0 - P(0, \theta_M) = \frac{2}{a} \sqrt{\frac{\mu}{\pi \phi k c_T}} \{ Q(\theta_0) \sqrt{\theta_M} + [Q(\theta_1) - Q(\theta_0)] \sqrt{\theta_M - \theta_1} + \\ + [Q(\theta_2) - Q(\theta_1)] \sqrt{\theta_M - \theta_2} + \dots + [Q(\theta_n) - Q(\theta_{n-1})] \sqrt{\theta_M - \theta_n} \}$$
$$F_c(\theta_M) = 2a \sqrt{\frac{\phi c_1 k}{\pi \mu}} \{ [P_0 - P(\theta_1)] \sqrt{\theta_M} + [P(\theta_1) - P(\theta_2)] \sqrt{\theta_M - \theta_1} + \dots + [P(\theta_{n-1}) - P(\theta_n)] \sqrt{\theta_M - \theta_{n-1}} \}$$
$$Q(\theta_M) = a \sqrt{\frac{\phi c_1 k}{\pi \mu}} \left\{ \frac{[P_0 - P(\theta_1)]}{\sqrt{\theta_M}} + \frac{[P(\theta_1) - P(\theta_2)]}{\sqrt{\theta_M - \theta_1}} + \frac{[P(\theta_2) - P(\theta_3)]}{\sqrt{\theta_M - \theta_2}} + \dots + \frac{[P(\theta_{n-1}) - P(\theta_n)]}{\sqrt{\theta_M - \theta_{n-1}}} \right\}$$

TABLE 5. Finite radial system with fixed constant pressure at exterior boundary—Rate case.

re/rw=1.5		re/rw=2.0		re/rw=2.5		re/rw=3.0		re/rw=3.5		re/rw=4.0		re/rw=6.0		re/rw=8.0	
Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change
t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)
5.0(10) ⁻¹	0.230	2.0(10) ⁻¹	0.424	3.0(10) ⁻¹	0.502	5.0(10) ⁻¹	0.617	5.0(10) ⁻¹	0.620	1.0	0.802	4.0	1.275	7.0	1.499
5.5 "	0.240	2.2 "	0.441	3.5 "	0.535	5.5 "	0.640	5.5 "	0.665	1.2	0.857	4.5	1.320	7.5	1.527
6.0 "	0.249	2.4 "	0.457	4.0 "	0.564	6.0 "	0.662	6.0 "	0.705	1.4	0.905	5.0	1.361	8.0	1.554
7.0 "	0.266	2.5 "	0.472	4.5 "	0.591	7.0 "	0.702	8.0 "	0.741	1.6	0.947	5.5	1.398	8.5	1.580
8.0 "	0.282	2.6 "	0.485	5.0 "	0.616	8.0 "	0.738	9.0 "	0.774	1.8	0.986	6.0	1.432	9.0	1.604
9.0 "	0.292	3.0 "	0.498	5.5 "	0.638	9.0 "	0.770	1.0	0.804	2.0	1.020	6.5	1.462	9.5	1.627
1.0(10) ⁻¹	0.307	3.5 "	0.527	6.0 "	0.659	1.0	0.799	1.2	0.858	2.2	1.052	7.0	1.490	10.0	1.648
1.2 "	0.328	4.0 "	0.552	7.0 "	0.696	1.2	0.850	1.4	0.904	2.4	1.080	7.5	1.516	12.0	1.724
1.4 "	0.344	4.5 "	0.573	8.0 "	0.728	1.4	0.892	1.6	0.945	2.6	1.106	8.0	1.539	14.0	1.786
1.6 "	0.356	5.0 "	0.591	9.0 "	0.755	1.6	0.927	1.8	0.981	2.8	1.130	8.5	1.561	16.0	1.837
1.8 "	0.367	5.5 "	0.606	1.0	0.778	1.8	0.955	2.0	1.013	3.0	1.152	9.0	1.580	18.0	1.879
2.0 "	0.375	6.0 "	0.619	1.2	0.815	2.0	0.980	2.2	1.041	3.4	1.190	10.0	1.615	20.0	1.914
2.2 "	0.381	6.5 "	0.630	1.4	0.842	2.2	1.000	2.4	1.065	3.8	1.222	12.0	1.667	22.0	1.943
2.4 "	0.386	7.0 "	0.639	1.6	0.861	2.4	1.016	2.6	1.087	4.5	1.266	14.0	1.704	24.0	1.967
2.6 "	0.390	7.5 "	0.647	1.8	0.876	2.6	1.030	2.8	1.106	5.0	1.290	16.0	1.730	26.0	1.986
2.8 "	0.393	8.0 "	0.654	2.0	0.887	2.8	1.042	3.0	1.123	5.5	1.309	18.0	1.749	28.0	2.002
3.0 "	0.396	8.5 "	0.660	2.2	0.895	3.0	1.051	3.5	1.158	6.0	1.325	20.0	1.762	30.0	2.018
3.5 "	0.400	9.0 "	0.665	2.4	0.900	3.5	1.069	4.0	1.183	7.0	1.347	22.0	1.771	35.0	2.040
4.0 "	0.402	9.5 "	0.669	2.6	0.905	4.0	1.080	5.0	1.215	8.0	1.381	24.0	1.777	40.0	2.055
4.5 "	0.404	1.0	0.673	2.8	0.908	4.5	1.087	6.0	1.232	9.0	1.370	26.0	1.781	45.0	2.064
5.0 "	0.405	1.2	0.682	3.0	0.910	5.0	1.091	7.0	1.242	10.0	1.376	28.0	1.784	50.0	2.070
6.0 "	0.405	1.4	0.688	3.5	0.913	5.5	1.094	8.0	1.247	12.0	1.382	30.0	1.787	60.0	2.076
7.0 "	0.405	1.6	0.690	4.0	0.915	6.0	1.096	9.0	1.250	14.0	1.385	35.0	1.789	70.0	2.078
8.0 "	0.405	1.8	0.692	4.5	0.916	6.5	1.097	10.0	1.251	16.0	1.386	40.0	1.791	80.0	2.079
		2.0	0.692	5.0	0.916	7.0	1.097	12.0	1.252	18.0	1.386	50.0	1.792		
		2.5	0.693	5.5	0.916	8.0	1.098	14.0	1.253						
		3.0	0.693	6.0	0.916	10.0	1.099	16.0	1.253						

re/rw=10.0		re/rw=15.0		re/rw=20.0		re/rw=25.0		re/rw=30.0		re/rw=40.0		re/rw=50.0		re/rw=60.0	
Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change
t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)
10.0	1.651	20.0	1.960	30.0	2.148	50.0	2.389	70.0	2.551	12.0(10) ⁻¹	2.813	20.0(10) ⁻¹	3.064	3.0(10) ⁻¹	3.267
12.0	1.730	22.0	2.003	35.0	2.219	55.00	2.434	80.0	2.615	14.0 "	2.888	22.0 "	3.111	4.0 "	3.401
14.0	1.798	24.0	2.043	40.0	2.282	60.0	2.476	90.0	2.672	16.0 "	2.933	24.0 "	3.154	5.0 "	3.512
16.0	1.856	26.0	2.080	45.0	2.338	65.0	2.514	10.0(10) ⁻¹	2.733	18.0 "	3.011	26.0 "	3.193	6.0 "	3.602
18.0	1.907	28.0	2.114	50.0	2.388	70.0	2.550	12.0 "	2.812	20.0 "	3.063	28.0 "	3.229	7.0 "	3.676
20.0	1.952	30.0	2.146	60.0	2.475	75.0	2.583	14.0 "	2.886	22.0 "	3.109	30.0 "	3.263	8.0 "	3.739
25.0	2.043	35.0	2.218	70.0	2.547	80.0	2.614	16.0 "	2.950	24.0 "	3.152	35.0 "	3.339	9.0 "	3.792
30.0	2.111	40.0	2.279	80.0	2.609	85.0	2.643	18.5 "	2.965	26.0 "	3.191	40.0 "	3.405	10.0 "	3.832
35.0	2.160	45.0	2.332	90.0	2.658	90.0	2.671	17.0 "	2.979	28.0 "	3.226	45.0 "	3.461	12.0 "	3.908
40.0	2.197	50.0	2.379	10.0(10) ⁻¹	2.707	95.0	2.697	17.5 "	2.992	30.0 "	3.259	50.0 "	3.512	14.0 "	3.959
45.0	2.224	60.0	2.455	10.5 "	2.723	10.0(10) ⁻¹	2.721	18.0 "	3.006	35.0 "	3.321	55.0 "	3.556	16.0 "	3.996
50.0	2.245	70.0	2.513	11.0 "	2.747	12.0 "	2.807	20.0 "	3.054	40.0 "	3.391	60.0 "	3.595	18.0 "	4.023
55.0	2.260	80.0	2.558	11.5 "	2.764	14.0 "	2.878	25.0 "	3.150	45.0 "	3.440	65.0 "	3.630	20.0 "	4.043
60.0	2.271	90.0	2.592	12.0 "	2.781	16.0 "	2.936	30.0 "	3.219	50.0 "	3.482	70.0 "	3.661	25.0 "	4.071
65.0	2.278	10.0(10) ⁻¹	2.619	12.5 "	2.796	18.0 "	2.984	35.0 "	3.269	55.0 "	3.515	75.0 "	3.688	30.0 "	4.084
70.0	2.285	12.0 "	2.655	13.0 "	2.810	20.0 "	3.024	40.0 "	3.306	60.0 "	3.545	80.0 "	3.713	35.0 "	4.090
75.0	2.290	14.0 "	2.677	13.5 "	2.823	22.0 "	3.057	45.0 "	3.332	65.0 "	3.568	85.0 "	3.735	40.0 "	4.092
80.0	2.293	16.0 "	2.689	14.0 "	2.835	24.0 "	3.085	50.0 "	3.351	70.0 "	3.588	90.0 "	3.754	45.0 "	4.093
90.0	2.297	18.0 "	2.697	14.5 "	2.846	26.0 "	3.107	60.0 "	3.375	80.0 "	3.619	95.0 "	3.771	50.0 "	4.094
10.0(10) ⁻¹	2.300	20.0 "	2.701	15.0 "	2.857	28.0 "	3.126	70.0 "	3.387	90.0 "	3.640	10.0(10) ⁻¹	3.787	55.0 "	4.094
11.0 "	2.301	22.0 "	2.704	16.0 "	2.876	30.0 "	3.142	80.0 "	3.394	10.0(10) ⁻¹	3.655	12.0 "	3.833		
12.0 "	2.302	24.0 "	2.706	18.0 "	2.906	35.0 "	3.171	90.0 "	3.397	12.0 "	3.672	14.0 "	3.862		
13.0 "	2.302	26.0 "	2.707	20.0 "	2.929	40.0 "	3.189	10.0(10) ⁻¹	3.399	14.0 "	3.681	16.0 "	3.881		
14.0 "	2.302	28.0 "	2.707	24.0 "	2.958	45.0 "	3.200	12.0 "	3.401	16.0 "	3.685	18.0 "	3.892		
16.0 "	2.303	30.0 "	2.708	28.0 "	2.975	50.0 "	3.207	14.0 "	3.401	18.0 "	3.687	20.0 "	3.900		
				30.0 "	2.980	60.0 "	3.214			20.0 "	3.688	22.0 "	3.904		
				40.0 "	2.992	70.0 "	3.217			25.0 "	3.689	24.0 "	3.907		
				50.0 "	2.995	80.0 "	3.218					26.0 "	3.909		
						90.0 "	3.219					28.0 "	3.910		

re/rw=70.0		re/rw=80.0		re/rw=90.0		re/rw=100.0		re/rw=200.0		re/rw=300.0		re/rw=400.0		re/rw=500.0	
Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change	Dimension- less time	Pressure change
t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)
5.0(10) ⁻¹	3.512	6.0(10) ⁻¹	3.603	8.0(10) ⁻¹	3.747	1.0(10) ⁻¹	3.859	1.5(10) ⁻¹	4.061	6.0(10) ⁻¹	4.754	1.5(10) ⁻¹	5.212	2.0(10) ⁻¹	5.356
6.0 "	3.603	7.0 "	3.680	9.0 "	3.806	1.2 "	3.949	2.0 "	4.205	8.0 "	4.898	2.0 "	5.356	2.5 "	5.468
7.0 "	3.680	8.0 "	3.747	1.0(10) ⁻¹	3.858	1.4 "	4.026	2.5 "	4.317	10.0 "	5.010	3.0 "	5.556	3.0 "	5.559
8.0 "	3.746	9.0 "	3.805	1.2 "	3.949	1.6 "	4.092	3.0 "	4.408	12.0 "	5.101	4.0 "	5.689	3.5 "	5.636
9.0 "	3.803	10.0 "	3.867	1.3 "	3.988	1.8 "	4.150	3.5 "	4.485	14.0 "	5.177	5.0 "	5.781	4.0 "	5.702
10.0 "	3.854	12.0 "	3.946	1.4 "	4.025	2.0 "	4.200	4.0 "	4.552	16.0 "	5.242	6.0 "	5.845	4.5 "	5.759
12.0 "	3.937	14.0 "	4.019	1.5 "	4.058	2.5 "	4.303	5.0 "	4.663	18.0 "	5.299	7.0 "	5.889	5.0 "	5.810
14.0 "	4.003	15.0 "	4.051	1.8 "	4.144	3.0 "	4.379	6.0 "	4.754	20.0 "	5.348	8.0 "			

TABLE 5. Finite radial system with fixed constant pressure at Exterior Boundary — Rate case. — (Continued)

re/rw=600.0		re/rw=700.0		re/rw=800.0		re/rw=900.0		re/rw=1000.0		re/rw=1200.0		re/rw=1400.0		re/rw=1600.0	
Dimensionless time	Pressure change	Dimensionless time	Pressure change	Dimensionless time	Pressure change	Dimensionless time	Pressure change	Dimensionless time	Pressure change	Dimensionless time	Pressure change	Dimensionless time	Pressure change	Dimensionless time	Pressure change
t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)
4.0(10) ⁴	5.703	5.0(10) ⁴	5.814	7.0(10) ⁴	5.983	8.0(10) ⁴	6.049	1.0(10) ⁵	6.181	2.0(10) ⁵	6.807	2.0(10) ⁵	6.807	2.5(10) ⁵	6.819
4.5 "	5.752	6.0 "	5.905	8.0 "	6.108	9.0 "	6.161	1.2 "	6.252	3.0 "	6.704	2.5 "	6.819	3.0 "	6.710
5.0 "	5.814	7.0 "	5.982	9.0 "	6.161	10.0 "	6.214	1.4 "	6.329	4.0 "	6.833	3.0 "	6.709	3.5 "	6.787
6.0 "	5.904	8.0 "	6.048	10.0 "	6.180	12.0 "	6.251	1.6 "	6.395	5.0 "	6.918	3.5 "	6.785	4.0 "	6.853
7.0 "	5.979	9.0 "	6.105	12.0 "	6.249	14.0 "	6.327	1.8 "	6.452	6.0 "	6.975	4.0 "	6.849	5.0 "	6.962
8.0 "	6.041	10.0 "	6.156	14.0 "	6.322	16.0 "	6.392	2.0 "	6.503	7.0 "	7.013	5.0 "	6.950	6.0 "	7.046
9.0 "	6.094	12.0 "	6.229	16.0 "	6.382	18.0 "	6.447	2.5 "	6.605	8.0 "	7.038	6.0 "	7.028	7.0 "	7.114
10.0 "	6.139	14.0 "	6.305	18.0 "	6.432	20.0 "	6.494	3.0 "	6.681	9.0 "	7.056	7.0 "	7.082	8.0 "	7.167
12.0 "	6.210	16.0 "	6.357	20.0 "	6.474	25.0 "	6.587	3.5 "	6.738	10.0 "	7.067	8.0 "	7.123	9.0 "	7.210
14.0 "	6.282	18.0 "	6.398	25.0 "	6.551	30.0 "	6.652	4.0 "	6.781	12.0 "	7.080	9.0 "	7.154	10.0 "	7.244
16.0 "	6.299	20.0 "	6.430	30.0 "	6.599	40.0 "	6.729	4.5 "	6.813	14.0 "	7.085	10.0 "	7.177	15.0 "	7.334
18.0 "	6.326	25.0 "	6.494	35.0 "	6.630	45.0 "	6.781	5.0 "	6.837	16.0 "	7.088	15.0 "	7.229	20.0 "	7.364
20.0 "	6.345	30.0 "	6.514	40.0 "	6.650	50.0 "	6.786	5.5 "	6.854	18.0 "	7.089	20.0 "	7.241	25.0 "	7.373
25.0 "	6.374	35.0 "	6.530	45.0 "	6.663	55.0 "	6.777	6.0 "	6.868	19.0 "	7.089	25.0 "	7.243	30.0 "	7.376
30.0 "	6.387	40.0 "	6.540	50.0 "	6.671	60.0 "	6.785	7.0 "	6.885	20.0 "	7.090	30.0 "	7.244	35.0 "	7.377
35.0 "	6.392	45.0 "	6.545	55.0 "	6.675	70.0 "	6.794	8.0 "	6.895	21.0 "	7.090	31.0 "	7.244	40.0 "	7.378
40.0 "	6.395	50.0 "	6.548	60.0 "	6.679	80.0 "	6.798	9.0 "	6.901	22.0 "	7.090	32.0 "	7.244	42.0 "	7.378
50.0 "	6.397	60.0 "	6.550	70.0 "	6.682	90.0 "	6.800	10.0 "	6.904	23.0 "	7.090	33.0 "	7.244	44.0 "	7.378
60.0 "	6.397	70.0 "	6.551	80.0 "	6.684	10.0(10) ⁵	6.801	12.0 "	6.907	24.0 "	7.090				
		80.0 "	6.551	100.0 "	6.684			14.0 "	6.907						
								16.0 "	6.908						

re/rw=1800.0		re/rw=2000.0		re/rw=2200.0		re/rw=2400.0		re/rw=2600.0		re/rw=2800.0		re/rw=3000.0	
Dimensionless time	Pressure change	Dimensionless time	Pressure change	Dimensionless time	Pressure change	Dimensionless time	Pressure change	Dimensionless time	Pressure change	Dimensionless time	Pressure change	Dimensionless time	Pressure change
t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)	t	p(t)
3.0(10) ⁵	6.710	4.0(10) ⁵	6.854	5.0(10) ⁵	6.966	6.0(10) ⁵	7.057	7.0(10) ⁵	7.134	8.0(10) ⁵	7.201	1.0(10) ⁶	7.312
4.0 "	6.854	5.0 "	6.986	6.0 "	7.057	7.0 "	7.134	8.0 "	7.201	9.0 "	7.260	1.2 "	7.403
5.0 "	6.965	6.0 "	7.056	7.0 "	7.097	8.0 "	7.200	9.0 "	7.259	10.0 "	7.312	1.4 "	7.480
6.0 "	7.054	7.0 "	7.132	8.0 "	7.196	9.0 "	7.259	10.0 "	7.312	11.0 "	7.365	1.6 "	7.545
7.0 "	7.120	8.0 "	7.196	9.0 "	7.251	10.0 "	7.310	11.0 "	7.365	12.0 "	7.418	1.8 "	7.602
8.0 "	7.188	9.0 "	7.251	10.0 "	7.298	11.0 "	7.365	12.0 "	7.418	13.0 "	7.471	2.0 "	7.651
9.0 "	7.238	10.0 "	7.298	11.0 "	7.329	12.0 "	7.398	13.0 "	7.458	14.0 "	7.508	2.2 "	7.732
10.0 "	7.280	12.0 "	7.374	13.0 "	7.429	14.0 "	7.481	15.0 "	7.533	15.0 "	7.558	2.4 "	7.794
15.0 "	7.407	14.0 "	7.431	16.0 "	7.474	17.0 "	7.526	18.0 "	7.578	16.0 "	7.631	2.6 "	7.820
20.0 "	7.459	16.0 "	7.474	18.0 "	7.506	19.0 "	7.558	20.0 "	7.611	17.0 "	7.664	2.8 "	7.871
30.0 "	7.489	18.0 "	7.506	20.0 "	7.530	21.0 "	7.582	22.0 "	7.635	18.0 "	7.687	3.0 "	7.908
40.0 "	7.495	20.0 "	7.530	22.0 "	7.562	23.0 "	7.611	24.0 "	7.664	19.0 "	7.739	3.2 "	7.935
50.0 "	7.495	25.0 "	7.566	25.0 "	7.594	25.0 "	7.631	25.0 "	7.664	20.0 "	7.799	3.4 "	7.955
51.0 "	7.495	30.0 "	7.593	30.0 "	7.621	30.0 "	7.658	30.0 "	7.695	21.0 "	7.845	3.6 "	7.979
52.0 "	7.495	35.0 "	7.621	35.0 "	7.658	35.0 "	7.695	35.0 "	7.732	22.0 "	7.890	3.8 "	7.992
53.0 "	7.495	40.0 "	7.650	40.0 "	7.686	40.0 "	7.723	40.0 "	7.760	23.0 "	7.935	4.0 "	8.002
54.0 "	7.495	50.0 "	7.680	50.0 "	7.716	50.0 "	7.753	50.0 "	7.790	24.0 "	7.980	4.2 "	8.004
56.0 "	7.495	60.0 "	7.701	60.0 "	7.737	60.0 "	7.774	60.0 "	7.811	25.0 "	8.025	4.4 "	8.006
		64.0 "	7.801	64.0 "	7.801	64.0 "	7.838	64.0 "	7.875	26.0 "	8.070	4.6 "	8.006
										27.0 "	8.115	4.8 "	8.006
										28.0 "	8.160	5.0 "	8.006
										29.0 "	8.205	5.2 "	8.006
										30.0 "	8.250	5.4 "	8.006
										31.0 "	8.295	5.6 "	8.006
										32.0 "	8.340	5.8 "	8.006
										33.0 "	8.385	6.0 "	8.006
										34.0 "	8.430	6.2 "	8.006
										35.0 "	8.475	6.4 "	8.006
										36.0 "	8.520	6.6 "	8.006
										37.0 "	8.565	6.8 "	8.006
										38.0 "	8.610	7.0 "	8.006
										39.0 "	8.655	7.2 "	8.006
										40.0 "	8.700	7.4 "	8.006
										41.0 "	8.745	7.6 "	8.006
										42.0 "	8.790	7.8 "	8.006
										43.0 "	8.835	8.0 "	8.006
										44.0 "	8.880	8.2 "	8.006
										45.0 "	8.925	8.4 "	8.006
										46.0 "	8.970	8.6 "	8.006
										47.0 "	9.015	8.8 "	8.006
										48.0 "	9.060	9.0 "	8.006
										49.0 "	9.105	9.2 "	8.006
										50.0 "	9.150	9.4 "	8.006
										51.0 "	9.195	9.6 "	8.006
										52.0 "	9.240	9.8 "	8.006
										53.0 "	9.285	10.0 "	8.006
										54.0 "	9.330	10.2 "	8.006
										55.0 "	9.375	10.4 "	8.006
										56.0 "	9.420	10.6 "	8.006
										57.0 "	9.465	10.8 "	8.006
										58.0 "	9.510	11.0 "	8.006
										59.0 "	9.555	11.2 "	8.006
										60.0 "	9.600	11.4 "	8.006
										61.0 "	9.645	11.6 "	8.006
										62.0 "	9.690	11.8 "	8.006
										63.0 "	9.735	12.0 "	8.006
										64.0 "	9.780	12.2 "	8.006
										65.0 "	9.825	12.4 "	8.006
										66.0 "	9.870	12.6 "	8.006
										67.0 "	9.915	12.8 "	8.006
										68.0 "	9.960	13.0 "	8.006
										69.0 "	10.005	13.2 "	8.006
										70.0 "	10.050	13.4 "	8.006
										71.0 "	10.095	13.6 "	8.006
										72.0 "	10.140	13.8 "	8.006
										73.0 "	10.185	14.0 "	8.006
										74.0 "	10.230	14.2 "	8.006
										75.0 "	10.275	14.4 "	8.006
										76.0 "	10.320	14.6 "	8.006
										77.0 "	10.365	14.8 "	8.006
										78.0 "	10.410	15.0 "	8.006
										79.0 "	10.455	15.2 "	8.006
										80.0 "	10.500	15.4 "	8.006
										81.0 "	10.545	15.6 "	8.006
										82.0 "	10.590	15.8 "	8.006
										83.0 "	10.635	16.0 "	8.006
										84.0 "	10.680	16.2 "	8.006
										85.0 "	10.725	16.4 "	8.006
										86.0 "	10.770	16.6 "	8.006
										87.0 "	10.815	16.8 "	8.006
										88.0 "	10.860	17.0 "	8.006
										89			

Appendix B

Flow Behavior for Constant Pressure Inner Boundary and Infinite Outer
Boundary, Skin = 0; Eltig-Economides (1979)

t_D	Q_D	q_D	t_D	Q_D	q_D
1.00D-01	4.0433D-01	2.2487D 00	1.00D 04	2.1486D 03	1.4594D-01
2.00D-01	5.9803D-01	1.7153D 00	2.00D 04	4.0887D 03	1.8371D-01
3.00D-01	7.5642D-01	1.4764D 00	3.00D 04	5.8410D 03	1.7727D-01
4.00D-01	8.9674D-01	1.3376D 00	4.00D 04	7.6405D 03	1.7209D-01
5.00D-01	1.0244D 00	1.2336D 00	5.00D 04	9.3527D 03	1.6766D-01
6.00D-01	1.1439D 00	1.1601D 00	6.00D 04	1.1036D 04	1.6712D-01
7.00D-01	1.2569D 00	1.1025D 00	7.00D 04	1.2697D 04	1.6502D-01
8.00D-01	1.3648D 00	1.0550D 00	8.00D 04	1.4338D 04	1.6329D-01
9.00D-01	1.4641D 00	1.0167D 00	9.00D 04	1.5765D 04	1.6172D-01
1.00D 00	1.5684D 00	9.8384D-01	1.00D 05	1.7573D 04	1.6037D-01
2.00D 00	2.4452D 00	8.0043D-01	2.00D 05	3.3140D 04	1.5203D-01
3.00D 00	3.1797D 00	7.1674D-01	3.00D 05	4.8102D 04	1.4753D-01
4.00D 00	3.6882D 00	6.6443D-01	4.00D 05	6.2697D 04	1.4450D-01
5.00D 00	4.5339D 00	6.2021D-01	5.00D 05	7.7030D 04	1.4223D-01
6.00D 00	5.1480D 00	6.0071D-01	6.00D 05	9.1161D 04	1.4043D-01
7.00D 00	5.7377D 00	5.7940D-01	7.00D 05	1.0513D 05	1.3844D-01
8.00D 00	6.3077D 00	5.6160D-01	8.00D 05	1.1896D 05	1.3767D-01
9.00D 00	6.8615D 00	5.4671D-01	9.00D 05	1.3267D 05	1.3650D-01
1.00D 01	7.4021D 00	5.3394D-01	1.00D 06	1.4628D 05	1.3561D-01
2.00D 01	1.2321D 01	4.6116D-01	2.00D 06	2.7050D 05	1.2957D-01
3.00D 01	1.6742D 01	4.2612D-01	3.00D 06	4.0637D 05	1.2678D-01
4.00D 01	2.0806D 01	4.0399D-01	4.00D 06	5.3144D 05	1.2400D-01
5.00D 01	2.4043D 01	3.8020D-01	5.00D 06	6.5462D 05	1.2237D-01
6.00D 01	2.6642D 01	3.7609D-01	6.00D 06	7.7631D 05	1.2103D-01
7.00D 01	3.2372D 01	3.6330D-01	7.00D 06	8.9678D 05	1.1992D-01
8.00D 01	3.5796D 01	3.5233D-01	8.00D 06	1.0162D 06	1.1897D-01
9.00D 01	3.9544D 01	3.5149D-01	9.00D 06	1.1348D 06	1.1815D-01
1.00D 02	4.3027D 01	3.4257D-01	1.00D 07	1.2526D 06	1.1742D-01
2.00D 02	7.5575D 01	3.1081D-01	2.00D 07	2.4012D 06	1.1286D-01
3.00D 02	1.0573D 02	2.9330D-01	3.00D 07	3.5164D 06	1.1035D-01
4.00D 02	1.3447D 02	2.8204D-01	4.00D 07	4.6110D 06	1.0863D-01
5.00D 02	1.6224D 02	2.7382D-01	5.00D 07	5.6908D 06	1.0734D-01
6.00D 02	1.8930D 02	2.6744D-01	6.00D 07	6.7590D 06	1.0621D-01
7.00D 02	2.1578D 02	2.6226D-01	7.00D 07	7.8179D 06	1.0530D-01
8.00D 02	2.4178D 02	2.5792D-01	8.00D 07	8.8686D 06	1.0471D-01
9.00D 02	2.6739D 02	2.5421D-01	9.00D 07	9.9126D 06	1.0408D-01
1.00D 03	2.9264D 02	2.5097D-01	1.00D 08	1.0931D 07	1.0351D-01
2.00D 03	5.3254D 02	2.3151D-01	2.00D 08	2.1102D 07	9.9943D-02
3.00D 03	7.5862D 02	2.2142D-01	3.00D 08	3.0992D 07	9.7967D-02
4.00D 03	9.7651D 02	2.1477D-01	4.00D 08	4.0718D 07	9.6412D-02
5.00D 03	1.1085D 03	2.0906D-01	5.00D 08	5.0327D 07	9.5366D-02
6.00D 03	1.3967D 03	2.0402D-01	6.00D 08	5.9845D 07	9.4764D-02
7.00D 03	1.6011D 03	2.0207D-01	7.00D 08	6.9287D 07	9.4080D-02
8.00D 03	1.8026D 03	2.0002D-01	8.00D 08	7.8666D 07	9.3492D-02
9.00D 03	2.0017D 03	1.9794D-01	9.00D 08	8.7991D 07	9.2901D-02

Linear Aquifer Behavior

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INTRODUCTION

Linear aquifers, either limited or essentially infinite, may be encountered in reservoir engineering practice. In areas where faulting fixes reservoir boundaries, the fault block reservoir may have an aquifer of limited extent whose geometry is best approximated as linear. An infinite linear aquifer can occur as a regional feature whenever water movement through the aquifer member is constrained to one direction. Such constraints can arise from major faults, facies changes or pinchout of the member.

Miller* pointed out that linear aquifers have received only meager attention in the past. He analyzed the performance of finite and infinite aquifers, developed working equations and curves, and presented examples. While Miller's curves may be used fairly easily, a separate one is required for each size of aquifer. In this paper, Miller's equations have been used as a starting point. By modifying them slightly, they can be reduced to a form which yields a single working curve, applicable to any size of aquifer. Thus, interpolation between curves is eliminated and accuracy is improved.

Miller's results for finite aquifers covered only the boundary condition of no flow across the outer aquifer boundary. This paper also includes the case of constant pressure at the outer aquifer boundary.

DEVELOPMENT OF EQUATIONS FOR LINEAR AQUIFERS

Miller's equations give pressure drop or cumulative influx at the linear aquifer-reservoir boundary as a function of time for the boundary conditions of an infinite aquifer and a finite aquifer with sealed outer boundary. In addition to these equations, those appropriate for the boundary condition of a finite aquifer with constant pressure at the outer boundary have been developed. The approach used in developing these equations was the same as that used by Miller.

BOUNDARY CONDITION 1: CONSTANT RATE OF INFLUX ACROSS AQUIFER-RESERVOIR BOUNDARY

Infinite Linear Aquifer

$$\Delta p = \frac{q\mu}{kbh} \left[2 \sqrt{\frac{kt}{\pi\phi\mu c_i}} \right] \quad (1)$$

Finite Linear Aquifer, Sealed Outer Boundary

$$\Delta p = \frac{q\mu}{kbh} \left[\left(\frac{L}{3} + \frac{kt}{\phi\mu c_i L} \right) - \frac{2L}{\pi^2} \sum_{n=1}^{\infty} \left(\frac{1}{n^2} \right) \exp \left(-\frac{n^2 \pi^2 kt}{\phi\mu c_i L^2} \right) \right] \quad (2)$$

Finite Linear Aquifer, Constant Pressure at Outer Boundary

$$\Delta p = \frac{q\mu}{kbh} \left[L - \frac{8L}{\pi^2} \sum_{n \text{ odd}} \left(\frac{1}{n^2} \right) \exp \left(-\frac{n^2 \pi^2 kt}{4\phi\mu c_i L^2} \right) \right] \quad (3)$$

BOUNDARY CONDITION 2: CONSTANT PRESSURE AT AQUIFER-RESERVOIR BOUNDARY

Infinite Linear Aquifer

$$W_e = \phi bh c_i (\Delta p) \left[2 \sqrt{\frac{kt}{\pi\phi\mu c_i}} \right] \quad (4)$$

Finite Linear Aquifer, Sealed Outer Boundary

$$W_e = \phi bh c_i (\Delta p) \left[L - \frac{8L}{\pi^2} \sum_{n \text{ odd}} \left(\frac{1}{n^2} \right) \exp \left(-\frac{n^2 \pi^2 kt}{4\phi\mu c_i L^2} \right) \right] \quad (5)$$

Finite Linear Aquifer, Constant Pressure at Outer Boundary

$$W_e = \phi bh c_i (\Delta p) \left[\left(\frac{L}{3} + \frac{kt}{\phi\mu c_i L} \right) - \frac{2L}{\pi^2} \sum_{n=1}^{\infty} \left(\frac{1}{n^2} \right) \exp \left(-\frac{n^2 \pi^2 kt}{\phi\mu c_i L^2} \right) \right] \quad (6)$$

These equations are usually put in a form where dimensionless time is defined by

$$t_D = \frac{kt}{\phi\mu c_i x_e^2} \quad (7)$$

Here, x_e is a reference distance and is usually taken to be a unit distance. However, the choice is really arbitrary, as long as consistency is maintained. We choose $x_e = L$; then

$$t_D = \frac{kt}{\phi\mu c_i L^2} \quad (8)$$

For finite aquifers, L is the length of aquifer; for infinite cases, it may be considered as an arbitrarily chosen length. The reason for this choice will be clear later when the performances of finite and infinite aquifers are compared.

Substituting t_D from Eq. 8, the first six equations become:

BOUNDARY CONDITION 1:

Infinite Linear Aquifer

$$\Delta p = \frac{q\mu L}{kbh} F_1(t_D) \quad (9)$$

Finite Linear Aquifer, Sealed Outer Boundary

$$\Delta p = \frac{q\mu L}{kbh} F_1(t_D) \quad (10)$$

Finite Linear Aquifer, Constant Pressure at Outer Boundary

$$\Delta p = \frac{q\mu L}{kbh} F_0(t_D) \quad (11)$$

BOUNDARY CONDITION 2:

Infinite Linear Aquifer

$$W_e = \phi bh L c_i (\Delta p) F_1(t_D) \quad (12)$$

Original manuscript received in Society of Petroleum Engineers office Dec. 13, 1963.

*Miller, F. G.: "Theory of Unsteady-State Influx of Water in Linear Reservoirs", *Journal Institute of Petroleum* (Nov., 1962) 48, 365.

Finite Linear Aquifer, Sealed Outer Boundary

$$W_s = \phi b h L c_i (\Delta p) F_0(t_D) \quad (13)$$

Finite Linear Aquifer, Constant Pressure at Outer Boundary

$$W_s = \phi b h L c_i (\Delta p) F_1(t_D) \quad (14)$$

In these equations, the functions of dimensionless time are:

$$F_0(t_D) = 1 - \frac{8}{\pi^2} \sum_{n \text{ odd}} \left(\frac{1}{n^2} \right) \exp \left(-\frac{n^2 \pi^2 t_D}{4} \right) \quad (15)$$

$$F_1(t_D) = 2 \sqrt{t_D/\pi} \quad (16)$$

$$F_1(t_D) = \left(t_D + \frac{1}{3} \right) - \frac{2}{\pi^2} \sum_{n=1}^{\infty} \left(\frac{1}{n^2} \right) \exp(-n^2 \pi^2 t_D) \quad (17)$$

These functions are shown in Fig. 1. The subscripts 0, 1/2 and 1 correspond to the slopes of the curves as t_D approaches large values. Table 1 gives numerical values.

THE F-FUNCTIONS

The F-functions, in contrast to those used in Eqs. 1 through 6, do not depend explicitly on L , the aquifer length. This dependence has been removed by the particular definition used for dimensionless time.

The advantages are obvious. Only a single curve or table of values need be examined for a given boundary condition. Even in cases where L is not known precisely, it is easy to assume several trial values, calculate the corresponding t_D values, and read the required F-values from a single curve. All interpolation between curves is eliminated.

The behavior of the F-functions at small and large times is of interest. At $t_D=0$, the summation appearing in $F_0(t_D)$ reduces to a summation of $(1/n^2)$, for odd values of n . The summation of $F_1(t_D)$ reduces to a summation of $(1/n^2)$, for all values of n . These sum to $(\pi^2/8)$ and $(\pi^2/6)$, so F_0 and F_1 both approach zero as t_D approaches zero. As t_D approaches infinity, $F_0(t_D)$ approaches 1, and $F_1(t_D)$ approaches $(t_D + 1/3)$; i.e., becomes linear with t_D . If the boundary conditions applying to Eqs. 9 through 14 are examined, it is apparent this must be the case. For example, in the constant rate case, with sealed outer boundary, we expect the total pressure drop to become linear with time as the quasi-steady-state condition is reached at large time.

It would be expected that finite linear aquifers would behave, at small times, just as infinite aquifers do. In other words, at low values of t_D , $F_0(t_D)$ and $F_1(t_D)$ should both approach $F_{1/2}(t_D)$. This is indeed the case, as a referral to Table 1 or Fig. 1 will show.

Table 1 was computed digitally and Fig. 1 was prepared from the results. Fig. 1 clearly indicates that the functions $F_0(t_D)$ and $F_1(t_D)$, for finite aquifers, begin to deviate from the infinite linear aquifer curve, $F_{1/2}(t_D)$, when t_D exceeds 0.25. When t_D exceeds 2.5, the finite aquifer functions may be very well approximated by their limiting forms. Hence,

$$\text{For } t_D \leq 0.25, F_0 = F_1 = F_{1/2} = 2 \sqrt{t_D/\pi} \quad (18)$$

$$\text{For } 0.25 < t_D < 2.5, \text{ use Table 1 or Fig. 1 for } F_0 \text{ and } F_1$$

$$\text{For } t_D \geq 2.5, F_0 = 1 \quad (19)$$

$$F_1 = t_D + \frac{1}{3} \quad (20)$$

Eqs. 19 and 20 show, of course, that the finite systems have closely approached the steady or quasi-steady state at longer times.

APPLICATIONS

UNITS

The equations and graphs presented in this paper require that a dimensionally consistent set of units be used. This is perhaps an obvious point, but may easily be overlooked when attempting to work in terms of practical units.

The metric or cm-sec-cp-atm-darcy system is dimensionally consistent and applications can be carried out without difficulty.

The so-called practical system, which uses one basic unit, ft, for measuring length and area and another, bbl, for measuring volume, is subject to error in application. The most rational approach appears to be that of defining a special permeability unit (spu) such that

$$k_{\text{spu}} = 6.3283 k_{\text{darcies}} \quad (21)$$

Then, a consistent ft-day-cp-psi-spu system of units may be used, remembering that influx rates and volumes must be expressed in terms of cubic feet rather than barrels.

SUPERPOSITION

To calculate pressure drop when rate of influx has varied:

$$\Delta p = \frac{\mu L}{k b h} \left[q_0 F_k(t_D) + (q_1 - q_0) F_k(t_D - t_{D1}) + \dots \right] \quad (22)$$

To calculate influx for a series of pressure drops:

$$W_s = \phi b h L c_i [\Delta p_0 F_k(t_D) + \Delta p_1 F_k(t_D - t_{D1}) + \dots] \quad (23)$$

where

$$(\Delta p)_j = (p_{j-1} - p_{j+1})/2 \quad (24)$$

In these equations, $F_k(t_D)$ refers to the particular F-function for boundary conditions appropriate to the case of interest.

Example No. 1

Over the months of April through June, the pressure of a reservoir dropped from 2,810 to 2,780 psi during initial production. The aquifer associated with this reservoir is estimated to have the properties in Table 2.

Estimate the water influx from the aquifer over this period of time, assuming (1) an infinite aquifer, (2) a finite, sealed aquifer 2 miles long, and (3) a finite aquifer 2 miles long with constant pressure at the outer boundary.

Since the data are given in practical units,

$$k_{\text{spu}} = 6.3283 k_{\text{darcies}} = 6.3283 (0.3) = 1.8985$$

FIG. 1—PRESSURE DROP AND CUMULATIVE INFLUX FORMATIONS, LINEAR AQUIFERS.

TABLE 1—PRESSURE DROP AND CUMULATIVE INFLUX FUNCTIONS FOR LINEAR AQUIFERS

t_D	$F_0(t_D)$	$F_{1/2}(t_D)$	$F_1(t_D)$	t_D	$F_0(t_D)$	$F_{1/2}(t_D)$	$F_1(t_D)$
1.00(10 ⁻²)	1.128379(10 ⁻¹)	1.128379(10 ⁻¹)	1.128379(10 ⁻¹)	3.10(10 ⁻¹)	6.226824(10 ⁻¹)	6.282549(10 ⁻¹)	6.338276(10 ⁻¹)
1.10(10 ⁻²)	1.183454(10 ⁻¹)	1.183454(10 ⁻¹)	1.183454(10 ⁻¹)	3.50(10 ⁻¹)	6.581891(10 ⁻¹)	6.675581(10 ⁻¹)	6.769283(10 ⁻¹)
1.25(10 ⁻²)	1.261566(10 ⁻¹)	1.261566(10 ⁻¹)	1.261566(10 ⁻¹)	4.00(10 ⁻¹)	6.978819(10 ⁻¹)	7.136496(10 ⁻¹)	7.294231(10 ⁻¹)
1.40(10 ⁻²)	1.335116(10 ⁻¹)	1.335116(10 ⁻¹)	1.335116(10 ⁻¹)	4.50(10 ⁻¹)	7.329537(10 ⁻¹)	7.569398(10 ⁻¹)	7.809461(10 ⁻¹)
1.60(10 ⁻²)	1.427299(10 ⁻¹)	1.427299(10 ⁻¹)	1.427299(10 ⁻¹)	5.00(10 ⁻¹)	7.639503(10 ⁻¹)	7.978846(10 ⁻¹)	8.318760(10 ⁻¹)
1.80(10 ⁻²)	1.513880(10 ⁻¹)	1.513880(10 ⁻¹)	1.513880(10 ⁻¹)	5.60(10 ⁻¹)	7.964332(10 ⁻¹)	8.444016(10 ⁻¹)	8.925272(10 ⁻¹)
2.00(10 ⁻²)	1.595769(10 ⁻¹)	1.595769(10 ⁻¹)	1.595769(10 ⁻¹)	6.20(10 ⁻¹)	8.244456(10 ⁻¹)	8.884866(10 ⁻¹)	9.528875(10 ⁻¹)
2.25(10 ⁻²)	1.692569(10 ⁻¹)	1.692569(10 ⁻¹)	1.692569(10 ⁻¹)	7.00(10 ⁻¹)	8.558930(10 ⁻¹)	9.440697(10 ⁻¹)	10.331309(10 ⁻¹)
2.50(10 ⁻²)	1.784124(10 ⁻¹)	1.784124(10 ⁻¹)	1.784124(10 ⁻¹)	8.00(10 ⁻¹)	8.874029(10 ⁻¹)	10.092530(10 ⁻¹)	11.332578(10 ⁻¹)
2.80(10 ⁻²)	1.888139(10 ⁻¹)	1.888139(10 ⁻¹)	1.888139(10 ⁻¹)	9.00(10 ⁻¹)	9.120229(10 ⁻¹)	10.704744(10 ⁻¹)	12.333052(10 ⁻¹)
3.10(10 ⁻²)	1.986717(10 ⁻¹)	1.986717(10 ⁻¹)	1.986717(10 ⁻¹)	1.00	9.312597(10 ⁻¹)	1.128379	1.333323
3.50(10 ⁻²)	2.111004(10 ⁻¹)	2.111004(10 ⁻¹)	2.111004(10 ⁻¹)	1.10	9.462902(10 ⁻¹)	1.183454	1.433329
4.00(10 ⁻²)	2.256758(10 ⁻¹)	2.256758(10 ⁻¹)	2.256758(10 ⁻¹)	1.25	9.629049(10 ⁻¹)	1.261566	1.583332
4.50(10 ⁻²)	2.393654(10 ⁻¹)	2.393654(10 ⁻¹)	2.393654(10 ⁻¹)	1.40	9.743799(10 ⁻¹)	1.335116	1.733333
5.00(10 ⁻²)	2.523133(10 ⁻¹)	2.523133(10 ⁻¹)	2.523133(10 ⁻¹)	1.60	9.843590(10 ⁻¹)	1.427299	1.933333
5.60(10 ⁻²)	2.670232(10 ⁻¹)	2.670232(10 ⁻¹)	2.670232(10 ⁻¹)	1.80	9.904512(10 ⁻¹)	1.513880	2.133333
6.20(10 ⁻²)	2.809641(10 ⁻¹)	2.809641(10 ⁻¹)	2.809641(10 ⁻¹)	2.00	9.941705(10 ⁻¹)	1.595769	2.333333
7.00(10 ⁻²)	2.985411(10 ⁻¹)	2.985411(10 ⁻¹)	2.985411(10 ⁻¹)	2.25	9.968541(10 ⁻¹)	1.692569	2.583333
8.00(10 ⁻²)	3.191537(10 ⁻¹)	3.191537(10 ⁻¹)	3.191537(10 ⁻¹)	2.50	9.983024(10 ⁻¹)	1.784124	2.833333
9.00(10 ⁻²)	3.385133(10 ⁻¹)	3.385133(10 ⁻¹)	3.385133(10 ⁻¹)	2.80	9.991902(10 ⁻¹)	1.888139	3.133333
1.00(10 ⁻¹)	3.568234(10 ⁻¹)	3.568234(10 ⁻¹)	3.568234(10 ⁻¹)	3.10	9.996137(10 ⁻¹)	1.986717	3.433333
1.10(10 ⁻¹)	3.742370(10 ⁻¹)	3.742370(10 ⁻¹)	3.742370(10 ⁻¹)	3.50	9.998560(10 ⁻¹)	2.111004	3.833333
1.25(10 ⁻¹)	3.989280(10 ⁻¹)	3.989280(10 ⁻¹)	3.989280(10 ⁻¹)	4.00	9.999581(10 ⁻¹)	2.256758	4.333333
1.40(10 ⁻¹)	4.221615(10 ⁻¹)	4.222008(10 ⁻¹)	4.222401(10 ⁻¹)	4.50	9.999878(10 ⁻¹)	2.393654	4.833333
1.60(10 ⁻¹)	4.512368(10 ⁻¹)	4.513517(10 ⁻¹)	4.514665(10 ⁻¹)	5.00	9.999964(10 ⁻¹)	2.523133	5.333333
1.80(10 ⁻¹)	4.784617(10 ⁻¹)	4.787307(10 ⁻¹)	4.789997(10 ⁻¹)	5.60	9.999992(10 ⁻¹)	2.670232	5.933333
2.00(10 ⁻¹)	5.040878(10 ⁻¹)	5.046265(10 ⁻¹)	5.051652(10 ⁻¹)	6.20	9.999998(10 ⁻¹)	2.809641	6.533333
2.25(10 ⁻¹)	5.341424(10 ⁻¹)	5.352372(10 ⁻¹)	5.363320(10 ⁻¹)	7.00	10.000000(10 ⁻¹)	2.985411	7.333333
2.50(10 ⁻¹)	5.622333(10 ⁻¹)	5.641896(10 ⁻¹)	5.661456(10 ⁻¹)	8.00	10.000000(10 ⁻¹)	3.191538	8.333333
2.80(10 ⁻¹)	5.936127(10 ⁻¹)	5.970821(10 ⁻¹)	6.005516(10 ⁻¹)	9.00	10.000000(10 ⁻¹)	3.385138	9.333333

For all cases, $t = 91$ days and L is 2 miles = 10,560 ft.

$$t_D = \frac{kt}{\phi \mu c_i L^2} = \frac{(1.8985)(91)}{(0.25)(1)(6.2)(10^{-6})(10,560)^2} = 1.0.$$

Since the pressure drop occurs over a period of time, superposition should be used. Assumption of a linear drop with time is shown in Table 3.

The water influx is given by Eq. 23:

$$W_e = \phi b h L c_i \sum \Delta p_j F_k(t_D - t_{Dj}) = (0.25)(2,000)(41)(10,560)(6.2)(10^{-6})(\Sigma) = 1342.2(\Sigma).$$

Example No. 2

Assume an aquifer of the same properties as used in Example 1, Case (2). Estimate the pressure drop at the aquifer-reservoir boundary for a constant influx rate of 53.1 B/D over a 91 day time period.

$$q, \text{ cu ft/D} = 5.6146(53.1) = 298.1 \text{ cu ft/D}$$

Since the same t_D equation applies in this case as previously, $t_D = 1.0$. Also, $k_{\text{spu}} = 1.8985$. The pressure drop is given by Eq. 22:

$$\Delta p = \frac{q \mu L}{k b h} F_1(t_D) = \frac{(298.1)(1)(10,560)}{(1.8985)(2,000)(41)}(1.333) = 27 \text{ psi.}$$

TABLE 2—AQUIFER PROPERTIES

$b = 2,000$ ft
$h = 41$ ft
$k = 300$ md
$\phi = 0.25$
$\mu = 1$ cp
$c_i = 6.2(10^{-6}) \text{ psi}^{-1}$ (rock & water)

TABLE 3—INFLUX FUNCTION CALCULATIONS

t_{Dj}	p_j	$t_D - t_{Dj}$	Δp_j	$F_0(t_D - t_{Dj})$	$F_{1/2}(t_D - t_{Dj})$	$F_1(t_D - t_{Dj})$
0.0	2810	1.0	3	.931	1.128	1.333
0.2	2804	0.8	6	.887	1.009	1.133
0.4	2798	0.6	6	.815	.870	.930
0.6	2792	0.4	6	.698	.714	.729
0.8	2786	0.2	6	.504	.505	.506
1.0	2780	0.0				
$\Sigma \Delta p_j F_k(t_D - t_{Dj})$				20.217	21.972	23.788

TABLE 4—WATER INFLUX RESULTS

Case	F_k	W_e (cu ft)	W_e (bbbl)	Avg. q (B/D)
1	$F_{1/2}$	29,491	5,253	57.7
2	F_0	27,135	4,833	53.1
3	F_1	31,928	5,687	62.5

CONCLUSIONS

The equations describing linear aquifer behavior may be reduced, by a particular definition of dimensionless time, to three working curves. These curves are appropriate for both infinite and finite aquifers of any size, for all common boundary conditions.

A single curve applies for any given set of boundary conditions. This eliminates interpolation within a family of curves, and thereby improves the speed and accuracy of calculations.

Very simple limiting forms of the equations may be used in place of the working curves at short times ($t_D \leq 0.25$) and at long times ($t_D \geq 2.5$).

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NOMENCLATURE*

- b = Width of aquifer, cm or ft
- c_i = Aquifer compressibility (total), atm⁻¹ or psi⁻¹
- h = Thickness of aquifer, cm or ft
- k = Aquifer permeability, darcies or spu**
- Δp = Pressure drop, atm or psi
- q = Flow rate, cc/sec or cu ft/D
- t = Time, sec or days
- x_e = Unit distance, cm or ft
- F = Function of t_D , dimensionless
- L = Length of aquifer, cm or ft
- W_e = Water influx volume, cc or cu ft
- μ = Viscosity, cp
- ϕ = Porosity, dimensionless

SUBSCRIPTS

- j = index
- k = index
- D = dimensionless.

*Where two units are given, the metric system appears first followed by the practical system unit.

**spu = special permeability units
 $k_{\text{spu}} = 6.8288 \text{ k.darcies}$

PROJECT 5: FIELD SUPPORT SERVICES

To provide technical support for design and monitoring of DOE sponsored or industry initiated field projects.

5.1 A COMPARISON OF MASS RATE AND STEAM QUALITY REDUCTIONS TO OPTIMIZE STEAMFLOOD PERFORMANCE

(Gregory Messner and William Brigham)

Abstract

Many operators of steamdrive projects will reduce the heat injection rate as the project matures. The major benefit of this practice is to reduce the fuel costs and thus extend the economic life of the project. However, there is little industry consensus on whether the heat cuts should take the form of; (1) mass rate reductions while maintaining the same high steam quality, or (2) steam quality decreases while keeping the same mass rate. Through the use of a commercial three-phase, three-dimensional simulator, the oil recovery schedules obtained when reducing the injected steam mass rate or quality with time were compared under a variety of reservoir and operating conditions. The simulator input was validated for Kern River Field conditions by using the guidelines developed by Johnson, *et al.* (1989) for four steamflood projects in Kern River.

The results indicate that for equivalent heat injection rates, decreasing the steam injection mass rate at a constant high quality will yield more economic oil than reducing the steam quality at a constant mass rate. This conclusion is confirmed by a sensitivity analysis which demonstrates the importance of the gravity drainage/steam zone expansion mechanism in a low-pressure, heavy oil steamflood with gravity segregation. Furthermore, the impact of discontinuous silts and nonuniform initial temperatures within the steamflood zone was studied, indicating again that a decreasing mass rate injection strategy is a superior operating practice.

1. Introduction

It is routine to reduce the heat injection rate in a steamflood project as it approaches the economic limit. The major benefit of this practice is to reduce the fuel costs and thus extend the economic life of the project. Furthermore, there is evidence that frequent reductions in the heat injection rate are both theoretically sound and economically advantageous in steamflooding operations (Vogel, 1984). There are two general approaches to accomplishing these reductions: (1) cutting the mass rate of the injected steam while maintaining the same high quality; and (2) reducing the steam quality while keeping the same rate. Interestingly, there appears to be no industry consensus on which heat reduction process should be applied under a given set of reservoir conditions, and operators will often follow different steam injection rate or quality strategies within the same field.

The purpose of this study is to examine the recovery consequences of reducing mass rates or steam qualities using a thermal reservoir simulation program. In addition to the insights gained, it was hoped that general guidelines could be developed which would help determine which heat reduction scheme should be applied under a given set of steamflood conditions.

2. Literature Review

Several investigators have studied the effect of varying the heat injection rate with time in thermal recovery projects. Chu and Trimble (1975) conducted a reservoir simulation study using a three-dimensional, three-phase numerical model (Coats, *et al.*, 1974). After history matching the oil recovery performance for a single pattern in the Kern River Field, they investigated the influence of varying the steam injection schedule. By using the concept of cumulative discounted net oil (CDNO) as the optimizing criterion, they found that the economic performance of a steamflood could be improved over the constant rate injection case by increasing the initial steam rate and then decreasing the steam rate with time. In all these cases, the steam quality was held constant at 70%. While they mentioned that further work was needed to determine the optimal variations of steam rates, a hyperbolic decline was superior to a linear variation. They also concluded that the improvements increased with sand thickness.

The work of Vogel (1984) provides a more theoretical foundation for placing the Chu and Trimble results in perspective. By assuming that steam override is instantaneous, he derived an analytical descending steam chest model using the equations for linear heat flow from an infinite plane. Inspection of his equations showed that the heat rate requirements in a steamflood should start high and then decline with time. Figure 1, reproduced from Vogel's paper, illustrates typical steam injection rate requirements using his method. These steam requirements decline in a somewhat hyperbolic manner. Although not explicitly expressed, Vogel appeared to assume that mass rates rather than steam qualities should be cut, because the key to his method involved estimating the rate of downward growth of the steam zone.

In a similar study, Neuman (1985) developed equations to predict steamflood performance with steam override. His approach enabled calculation of steam zone growth, oil displaced, and consequences of reduced heat injection. With these equations, a reasonable match of oil production was made for Chevron's 10-pattern steamflood in the Kern River Field, California (Oglesby, *et al.*, 1982 and Blevins and Billingsley, 1975). Like Vogel, Neuman also showed the benefits of decreased heat injection with time, but he stated that a reduction in steam quality is the preferred method. However, he did not present any field or theoretical evidence to support this assertion.

Other studies have supported reduced injection rates with time in steamfloods. Using scaled physical laboratory models and field data, Myhill and Stegemeier (1978) found that high initial steam injection rates were desirable to promote faster heating around the producers. After breakthrough, however, large amounts of heat were produced, indicating that injection rates should be reduced. Also, from a mathematical heat balance model, they demonstrated that oil-steam ratios were improved with increased steam quality. Farouq Ali and Meldau (1979) expanded upon these findings.

Spivak and Muscatello (1987) performed a reservoir simulation study of steamflooding in the South Belridge Field, Kern County, California. After peak oil production, they found that tapering the mass injection rate at constant quality, increased the oil-steam ratio over a constant injection rate case. The optimum taper rate was 10% per year. Furthermore, they showed that, for equivalent heat injection, tapering the steam quality was much less effective than tapering the mass rate. However, these comparisons were made for a layered reservoir model with no vertical communication, so steam override was not a factor.

In the field, reduced injection rates have been reported to be successful in several steamflood projects. In Kern River, Bursell and Pittman (1975) saw improved steam-oil ratios resulting from reduced steam injection rates in three steamflood pilots, presumably with steam

at a constant high quality. Ault, *et al.* (1985) also reported economic success in reducing heat injection rates in two Kern River projects, but they accomplished these reductions by cutting the steam quality to approximately 10%.

There seems to be common agreement that the heat injection rate should be reduced with time. However, contradictory evidence exists over whether heat cuts should take the form of reduced rate at constant quality or reduced quality at constant rate. Since fuel consumption is the highest operating cost item in a steamflood, deciding how best to use that fuel is of great importance. The purpose of this work is to answer that question.

3. Methodology

This study relies heavily on a commercial three-phase, three-dimensional thermal simulation program (THERM) developed by Scientific Software-Intercomp (SSI). This program is capable of simulating steamflood, hot waterflood, cyclic steam and in-situ combustion processes and is fully implicit in the pressure, temperature, saturation and concentration terms. All simulation runs used a 9-point finite difference scheme which accurately handles reservoir heterogeneities and unequal grid spacings, and considerably reduces any grid orientation effect (SSI, 1988). All runs were made on a CRAY supercomputer.

For model input, the general guidelines developed for the Kern River Field by Johnson, *et al.* (1989) were incorporated. These include:

1. Kern River oil can be represented by a single-component, non-distillable heavy oil.
2. To induce the rapid steam overlay observed in the field, the vertical transmissibility at the injector is multiplied by 100 until 0.4 pore volumes (PV) of steam are injected.
3. A uniform set of relative permeability curve shapes and endpoints can be used.

Reservoir input data used in all of the following models are listed in Table 1. All cases comparing either mass rate or steam quality reductions have identical heat injection schedules.

One further point should be stressed here. The Johnson, *et al.* (1989) study successfully history matched the performance of high and low quality steamfloods using a uniform set of input guidelines. Not only were oil and water production rates and cumulatives matched; but also satisfactory predictions were achieved for steam breakthrough times, oil saturation profiles and temperature profiles. Successful history matching of several types of field data reduces the problem of non-uniqueness and indicates that the model is properly simulating the recovery mechanisms in low-pressure, heavy oil steamfloods. In a similar vein, this study will use several different operating scenarios and initial reservoir conditions to arrive at general conclusions regarding the preference of rate or quality reductions.

4. Discussion of Results

Several different types of reservoir settings and production schemes were studied to assess their effects on the economics of heat reduction operations. Each of these will be discussed in some detail.

4.1 One-Eighth of a Five-Spot Model

The first segment of this study involved using a one-eighth pattern element of symmetry to model a five-spot steamflood configuration. A $7 \times 4 \times 5$ parallel grid (Fig. 2) was set up for two homogeneous sand thicknesses, 40 and 80 feet. Table 2 lists input assumptions for these cases. The first set of runs assumed a 24% heat injection rate cut for two timing scenarios: (1) after two years of high quality, constant rate injection; and (2) at the high quality, constant rate economic limit, which was set to occur at a steam-oil ratio of 10.5. "High quality" in these cases means 52.5% of the steam mass is vapor at the sandface, a good average for Kern River operations. Key results from these simulation runs are shown in Figs. 3 and 4.

In brief, the recovery results for the 40 ft. reservoirs show that rate cuts at constant quality are superior to quality cuts at constant rate, no matter when the reduction in heat rate begins. Clearly comparing Figs. 3 and 4, it is better to start heat reduction sooner. The resulting total recoveries are about the same, but the operating costs would be significantly less.

For the 80 ft. reservoir case, the recoveries are also better for rate reduction compared to quality reduction. But clearly these differences are minor. In terms of practicality, we should recognize that we seldom see steam floods in sands that have as much as 80 ft. of homogeneous section. Later results will include cases where the systems are not homogeneous, but first we will discuss the effects of assumed relative permeability relationships and residual oil saturations.

4.2 Recovery Mechanisms and Sensitivity to Residual Oil Saturation and Relative Permeability Assumptions

From examining these simulation runs, it is clear that tapering mass rates promotes more rapid steam zone expansion. For example, in the 40 ft. example, the average steam saturation in the entire reservoir was 9.7% at the end of four years in the rate cut case. This is 34% higher than the quality cut case (7.1%). Furthermore, the rate case had 10% greater volumetric sweep of the steam zone.

These differences in steam zone expansion rates should be expected, since more steam vapor is injected in the rate cut case. However, the oil recovery differences suggest that steam zone expansion/gravity drainage is a more efficient recovery mechanism than hot water displacement. Looking at our input data, this idea is reasonable, given that at 225°F the residual oil saturation to liquid water (S_{orw}) is assumed to be 0.233, while the residual oil saturation to steam vapor (S_{org}) is only 0.05. These endpoint residual oil saturation (ROS) assumptions are extremely important in determining whether to reduce steam mass rates or qualities.

Since economic limits are built into the recovery comparisons, the shapes of the relative permeability curves should also have some influence on the recovery efficiencies. Changing the shapes and endpoints of these curves did have some effect on the results. The endpoint effect was the stronger of the two.

4.3 Validity of ROS and Relative Permeability Input Assumptions

A key factor influencing the recovery comparisons is the residual oil saturation to water and steam vapor. The ROS endpoint values used here are the same as those used by Johnson *et al.* (1989). The S_{orw} values were based on laboratory coreflood experiments, and the S_{org} endpoints were determined from both post-steamflood coring and laboratory corefloods. The low ROS to steam (0.05 PV) agrees with field results presented by other authors (Bursell and Pittman, 1975; and Blevins and Billingsley, 1975).

There are two major mechanisms by which steam vapor reduces oil saturation to very low values: (1) distillation, and (2) three-phase film flow with gravity (Hirasaki, 1989). These two phenomena were visually observed by Bruining, *et al.* (1984) during gravity-stable coreflood experiments. For longer flood times their residual oil saturations were below 0.10. These studies explain the low ROS's observed in steam-swept field cores, and tend to validate the S_{org} endpoint values used herein.

As previously mentioned, the S_{orw} endpoints were determined from laboratory measurements (Johnson, *et al.*, 1989). For 225°F, the S_{orw} averaged 0.234, but the laboratory measurement scatter was significant. Other experimental work discussed by Prats (1982, Chapter 6) indicates that the S_{orw} endpoint at higher temperatures tends to remain above 0.20. Thus the ROS endpoints to hot water used in this study appear to be realistic.

The relative permeability curves used in any reservoir simulation study are always subject to question, as recently explained by Saleri and Toronyi (1988). However, the chosen curves have been successfully used in history matching several steamflood projects (Johnson, *et al.*, 1989) and thus are within a reasonable uncertainty range.

In summary, the input assumptions which have the greatest impact on the comparison of rate and quality cuts, i.e., the ROS endpoints and the oil relative permeability curve to water, are considered to be quite reasonable. Thus, the conclusions regarding rate and quality reductions, appear to be valid. The behavior is consistent with observations by Vogel (1984) and Myhill and Stegemeier (1978). The primary recovery mechanism after steam breakthrough, is downward steam zone expansion with gravity drainage of heated oil. Cutting the rate but keeping the steam quality high will promote this effect. Cutting steam quality, and maintaining rate, depends on less efficient hot water displacement.

4.4 Multiple Pattern Model

In the previous model, it was implicitly assumed that all fluid flow was confined within the five-spot pattern. In an actual steamflood project, there will usually be some inter-pattern communication. For example, in the Kern River 10-pattern steamflood pilot, approximately 10% of the increased production came from hot production wells outside of the original project area (Blevins and Billingsley, 1975). Thus a multiple pattern model was constructed to incorporate asymmetric conditions. With less pattern confinement, it was thought that the recovery mechanisms identified in the confined model, might change in their relative importance.

The multiple pattern model consisted of a $16 \times 16 \times 3$ triangular grid representing one-eighth of 50 patterns covering 125 acres in a 60 ft. sand, and is similar to the model set-up described by Chu (1987). The reservoir conditions were identical to those of the single pattern model, except that the grid thicknesses were now 9, 18 and 33 ft, from top to bottom, with the injection wells completed only in the bottom layer. To initialize asymmetrical conditions the

exterior 34 patterns were steamflooded for six years while the interior 16 patterns remained idle (Fig. 5). After steam stimulating the interior producers (Wells 1, 2 and 7), a steamflood was started in the interior 16 patterns, with steam injection rates of 331 BPD/well.

First, a test case of constant rate injection was run, and immediately apparent was the low oil recovery in the multiple pattern flood. Recovery was 48,000 bbl/pattern compared with 88,000 bbl from the single pattern after five years. Nevertheless, several years of economic oil production were attained in this model set-up. Figure 6 illustrates the oil recovery versus time for tapering the mass rates and steam qualities at two different decline rates. It is apparent that tapering the rate is significantly better than tapering the quality.

Interestingly, the recovery differences between the rate and quality cases are more pronounced than those in the confined model. At least part of this effect is due to the lower pressures in the multipattern model, which allow more steam zone expansion.

4.5 Comparison of Rate and Quality Reductions in Silty Sands

Both the single and multiple pattern models discussed so far have assumed homogeneous rock properties for the entire flood area. However, most steamflood projects are conducted in reservoirs with some degree of heterogeneity. For example, several Kern River steamfloods have been shown to have discontinuous shale layers within the displacement interval (Bursell and Pittman, 1975; and Blevins and Billingsley, 1975).

For this segment of the study, a $7 \times 4 \times 7$ grid was used to model one-eighth of a 2.5 acre, 60 ft. thick pattern. Input data are listed in Table 3. The presence of silts was simulated by altering the vertical fluid transmissibilities to correspond to a 2 ft. thick, 5 millidarcy shale layer. The first set of runs assumed one silt interbedded between grid layers 4 and 5, corresponding to a position 24 ft. below the top of the formation. Also, no shale was placed in the vicinity of the injection well, as illustrated in Fig. 7. This was done to create the maximum opportunity for steam override above the silt layer, as might result from a poor quality cement bond in the injector, and thus counteract the rate cut's main advantage seen in the clean sand cases: the downward steam zone expansion/gravity drainage recovery mechanism.

Figure 8 shows oil recovery vs. time for 10% and 30% per year taper rates beginning at 548 days into the project. Instead of the modest differences in recovery seen in Fig. 4, the mass rate reductions recover far more economic oil. Furthermore both the 10% and 30% per year taper cases are still producing economically after eight years of steamflooding, with a recovery over 60% of the OOIP indicated for the 10% case. The oil recoveries are well above those in the clean sand model. The reasons for these results will be discussed later.

For the second discontinuous silt model, three silt interbeds were placed in positions corresponding to 9, 24, and 48 ft. from the top of the 60 ft. thick sand. The areal extent of the silts was assigned using a random number generator (Figs. 9-11). Like the one silt case, Fig. 12 shows that mass rate cuts are far more efficient in this model. Again, high economic oil recoveries are indicated.

These high recoveries need further discussion. Comparing Figs. 8 and 12 with Fig. 4, it is apparent that the sand/silt models recover significantly more oil than the homogeneous sand case. The saturation and temperature distributions show that the silt layers improve the vertical sweep of the steamflood by trapping steam below them. Figure 13 illustrates this phenomenon for the one silt model. On this cross-section, an additional steam zone lies under

the silt layer. Thus, the silt layers accelerate the downward heating of the formation and extend the economic production time. Furthermore, high steam quality promotes this effect.

4.6 Preheated Reservoir Sands

Restine (1983) has pointed out that the productive section of the Kern River Field consists of a sequence of sands averaging 60 ft. in thickness and separated by continuous silt or clay layers. While these shaly layers may be impermeable to steam, the sands immediately above a steamflooded interval are heated by conduction. Restine discussed field results and simulation work, and he concluded that this preheating effect of upper sands is an important consideration in a steamflood operation.

To see the impact of reservoir preheating on the comparison of rate and quality cuts, a one-eighth pattern model with a $7 \times 4 \times 7$ grid was again used to simulate a steamflood in a 60 ft. thick sand. The reservoir conditions were identical to the discontinuous silts model (Table 3) except that the initial reservoir temperature varied with depth. This temperature relationship is illustrated in Fig. 14 and is similar to the field data presented by Restine (1983).

Table 4 summarizes the production results for clean and silty sand cases, and we see again that for the equivalent heat injection rate, one should cut the mass rate at constant quality. Note, however, that while the 30% per year tapers show large differences between rate and quality reductions, the 10% rate decreases yielded only modest improvement (3-6% of OOIP). It is likely that the 10% rate cases were relatively overinjected with heat. Evidence of overinjection can be seen in the almost identical final oil recoveries for the 10% and 30% cases.

Another interesting result is the high ultimate oil recoveries seen in the clean sand cases with tapered injection rates. In the last section it was noted that the presence of discontinuous silts could increase recovery, in sands of uniform temperature, by hindering the gravity override of the steam and thus accelerating the downward heating of the formation. However, when a zone has already been heated by underlying conduction, this effect is minor; in fact, discontinuous silts in preheated sand steamfloods could actually reduce oil recovery by hindering gravity drainage due to reduced vertical permeabilities. This concept is suggested in the Table 4 data, where the clean sands produced slightly more oil than the silty sand cases when the mass rates were tapered.

5. Conclusions

Under a variety of reservoir and operating conditions, the oil recovery consequences of reducing the injected steam mass rate or quality with time have been compared with the help of a thermal simulator. Examination of the simulation results combined with the literature review leads us to the following conclusions:

1. Under the conditions assumed in this study, declining the steam injection mass rate at a constant high quality will recover more economic oil than reducing the steam quality at a constant mass rate. This result assumes equivalent heat injection for the two cases.
2. This constant steam quality strategy agrees with some researchers' observations that in heavy oil, low-pressure steamfloods with gravity segregation, the dominant recovery mechanism is gravity drainage of heated oil accompanied by downward steam zone expansion. High steam quality will promote this effect more than reducing steam quality, which depends on the less effective hot water displacement.

3. Based on cumulative discounted net oil (CDNO), steam injection should start at a high rate than taper with time for the most economic operating strategy. This observation is consistent with the results of Chu and Trimble (1975) and Vogel (1984).
4. Discontinuous silts within an otherwise uniform interval can enhance oil recovery by accelerating the downward heating of the formation. Furthermore, a reduced mass rate approach will be far superior to reducing the quality under these reservoir conditions.
5. If a reservoir sand has been preheated by an underlying steamflood, steam injection rate cuts with time appear to be even more effective than the same strategy in non-preheated reservoirs.

Recommendations

These general conclusions do not answer the question of the best strategy for a given reservoir. The question is: "When should rate reductions occur, and at what decline rate?" Clearly this question can only be answered by making runs for the specific field case of interest. Such calculations would be quite important economically.

Nomenclature

CDNO	=	Cumulative discounted net oil, bbl
CWE	=	Cold water equivalent volume of steam
k_{rg}	=	Gas relative permeability, dimensionless
$(k_{rg})_{ro}$	=	Gas (steam) relative permeability at residual oil saturation, dimensionless
$(k_{ro})_{iw}$	=	Oil relative permeability at irreducible water saturation, dimensionless
k_{rog}	=	Oil relative permeability in a gas (steam/oil system, dimensionless
k_{row}	=	Oil relative permeability in a gas (steam/oil system, dimensionless
k_{rw}	=	Water relative permeability, dimensionless
$(k_{rw})_{ro}$	=	Water relative permeability at residual oil saturation, dimensionless
n_g	=	Exponent for gas relative permeability calculation, dimensionless
n_{og}	=	Exponent for oil relative permeability calculation in a gas (steam)/oil system, dimensionless
n_{ow}	=	Exponent for oil relative permeability calculation in a water/oil system, dimensionless
n_w	=	Exponent for water relative permeability calculation, dimensionless
OOIP	=	Original oil in place, bbl
ROS	=	Residual oil saturation
S_{iw}	=	Irreducible water saturation, fraction PV
S_{org}	=	Residual oil saturation to steam fraction
S_{orw}	=	Residual oil saturation to steam, fraction pv
T	=	Temperature, degrees Fahrenheit

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TABLE 1

RESERVOIR DATA USED IN ALL SIMULATION RUNS

20.

Reservoir Description

Rock heat capacity, Btu/ft ³ -°F	35.0
Rock thermal conductivity, Btu/ft-D-°F	25.6
Overburden heat capacity, Btu/ft ³ -°F	39.0
Overburden thermal conductivity, Btu/ft-D-°F	27.6
Rock compressibility, psi ⁻¹	0.000735
Initial pressure at top of reservoir, psia	40

Oil Properties

Heat capacity, Btu/lbm-°F	.50
Thermal Expansion, °F ⁻¹	.00039
Compressibility, psi ⁻¹	.5 X 10 ⁻⁵

Relative Permeability Curve Shape and End Point Data (See Appendix A)

n_w	=	1.94	
n_{ow}	=	2.24	
n_{og}	=	1.20	
n_g	=	2.58	
			$(k_{rg})_{ro} = 0.46$ (all temperatures)

TEMP, °F	S_{iw}	S_{orw}	S_{org}	$(k_{rw})_{ro}$	$(k_{ro})_{iw}$
50	0.515	0.400	0.400	0.0486	0.751
225	0.590	0.234	0.050	0.0558	0.575
400	0.665	0.229	0.050	0.0630	0.400

TABLE 2
RESERVOIR DATA USED IN ONE-EIGHTH PATTERN MODEL

Model and grid	1/8 of 5-spot X x Y x Z: 7 x 4 x 5	
Vertical Grid Thickness	<u>40' Sand</u>	<u>80' Sand</u>
Z = 1	5 ft	5 ft
Z = 2	5 ft	10 ft
Z = 3	10 ft	15 ft
Z = 4	10 ft	20 ft
Z = 5	10 ft	30 ft
Distance between injector and producer, ft	233.35 (2.5 Acre Pattern)	
Initial reservoir temperature °F	85	
Porosity, %	31	
Initial water saturation, %	55	
Initial oil saturation, %	45	
Initial gas saturation, %	-0-	
Horizontal permeability, md	3,000	
Vertical permeability, md	2,700	
Oil viscosity @ 80°F, cp	7,500	
Oil viscosity @ 300°F, cp	8.38	
Injectors completed in the bottom 30' of zone		
Producers completed throughout thickness		
P _{wf} = 25 psia		
P _{inj} varies according to injectivity		
Injected fluid enthalpies set at 366°F		

TABLE 3

RESERVOIR DATA USED IN MODEL WITH DISCONTINUOUS SILTS

Model: grid	1/8 of 5-spot pattern 2.5 acres X x Y x Z: 7 x 4 x 7
Distance between injection and producer, ft	233.35
Initial Reservoir Temperature, °F	85
Porosity, %	31.0
Initial Water Saturation, %	55
Initial Oil Saturation, %	45
Initial Gas Saturation, %	0
Horizontal Permeability, md	3500
Vertical Permeability, md	3150
Oil Viscosity, cp	80°F 7175 300°F 7.40
Vertical Grid Thickness	
Z = 1	3'
Z = 2	6'
Z = 3	6'
Z = 4	9'
Z = 5	12'
Z = 6	12'
Z = 7	<u>12'</u>
60' Total	
Injector completed in bottom 3 layers	
Producer completed in all layers	

TABLE 4
COMPARISON OF OIL RECOVERIES
FOR TAPERING INJECTION RATES/QUALITIES
IN PREHEATED 60' SANDS

<u>CASE</u>	<u>OIL RECOVERY AT THE E. L., % OOIP</u>	<u>LIFE, YEARS</u>
<u>Clean Sand</u>		
1. 10%/Yr Taper		
Rate	63.4	4.25
Quality	57.4	3.50
2. 30%/Yr Taper		
Rate	63.3	5.75
Quality	46.2	3.00
<u>One Shale</u>		
1. 10%/Yr Taper		
Rate	62.5	3.75
Quality	59.7	3.75
2. 30%/Yr Taper		
Rate	61.8	5.0
Quality	46.7	3.4
<u>3 Random Shales</u>		
1. 10%/Yr Taper		
Rate	61.9	4.0
Quality	58.3	3.75
2. 30%/Yr Taper		
Rate	61.1	5.5
Quality	45.0	3.15

TAPERING INJECTION SCHEDULE
FOR PREHEATED SAND CASES

<u>TIME, DAYS</u>	<u>10%/YR RATE TAPER BPD</u>	<u>10%/YR QUALITY TAPER %</u>	<u>30%/YR RATE TAPER BPD</u>	<u>30%/YR QUALITY TAPER %</u>
0-183	331	52.5	331	52.5
183-548	300	43.885	245	28.59
548-913	271	35.82	182	11.07
913-1278	245	28.59	142.17	0
1278-1643	222	22.20	142.17	0
1643-2008	201	16.36	142.17	0
2008-2373	182	11.08	142.17	0
2373-2738	164	6.07	142.17	0

Figure 1. Theoretical steam requirements using planar heat flow model (Vogel, 1984).

Figure 2. One-eighth of a five-spot pattern, areal view of parallel grid.

Figure 3. Oil recovery vs. time. One-eighth pattern model. Heat injection reduction at the E.L.

Figure 4. Oil recovery vs. time. One-eighth pattern model. Heat injection in Year 3.

Figure 5. Multipattern model configuration sixteen patterns surrounded by previously steamflooded reservoir.

Figure 6. Oil recovery vs. time numerical model -- field case (sixteen) patterns). Heat injection taper starting in Year 2.

Figure 7. Location of silt between Layers 4 and 5, one silt model
(24 ft. from top, 36 ft. from bottom).

Figure 8. Oil recovery vs. time to the E.L. One-eighth pattern model. Heat injection taper starting at 548 days. One discontinuous silt in the 60 ft. sand.

**Figure 9. Location of silts between Layers 2 and 3, three-silt model
(9 ft. from top).**

Figure 10. Location of silt between Layers 4 and 5, three-silt model
(24 ft. from top).

Figure 11. Location of silt between Layers 6 and 7, three-silt model
(48 ft. from top).

Figure 12. Oil recovery vs. time to the E.L. One-eighth pattern model. Heat injection taper starting at 548 days, 3 random silt layers, 60 ft. sand.

STEAM SATURATIONS @ 2190 DAYS
 ONE DISCONTINUOUS SILT CASE
 INJECTION RATE=331 BPD CWE
 STEAM QUALITY=52.5%

STEAM SATURATIONS @ 2190 DAYS
 CLEAN SAND CASE
 INJECTION RATE=331 BPD CWE
 STEAM QUALITY=52.5%

$$I=1-7, J=1$$

Figure 13. Comparison of steam saturation profiles for clean and silty sands.

Figure 14. Initial temperature profile for 60 ft. preheated sand.