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RECUPLEX PROTOTYPE ANION EXCHANGE COLUMN

By:
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FINISHED PRODUCTS TECHNOLOGY RESEARCH AND ENGINEERING CHEMICAL PROCESSING DEPARTMENT

March 27, 1959
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RECUPLEX PROTOTYPE ANION EXCHANGE COLUMN

Future Z-Plant needs for reducing the metallic impurities in plutonium nitrate feeds, reducing the amounts of plutonium sent to cribs in wastes, and the development of new technology for plutonium recycle recovery place emphasis on expanding the application of anion exchange systems. It is proposed to install a fixed bed ion exchange prototype in Recuplex to make feasibility studies on feed and waste streams. Depending on the results from these studies, and the status of development at Savannah River, the decision may be made toward the end of CY 1959 to construct a second prototype to develop design data on the fluid bed principle of operation.

The following discussion treats the need and scope of this prototype development study.

DISCUSSION

Anion exchange resins are used in the Purex Plant to purify plutonium nitrate solutions following the final solvent extraction cycle. The plutonium nitrate solution is then concentrated and shipped to Z-Plant for conversion to metal. The purity of Purex plutonium nitrate solutions has been routinely higher than nitrate solutions received from either Redox(1) or Recuplex. With the probable requirement for production of large quantities of alpha plutonium, high purity plutonium nitrate will be routinely required. Recuplex can produce such high purity nitrate solution from the solvent extraction system, but with occasional upsets in the system, lower purity nitrate solution occurs. The need for high purity nitrate solution in the future may require that the Recuplex solvent extraction system be supplemented by or replaced with an anion exchange system.

Also, in Recuplex, a measurable amount of plutonium is contained in the solvent extraction waste. This waste is now routinely discharged to underground cribs. The long range goal for Chemical Processing Department is a reduction in the quantity of plutonium discharged to ground. Laboratory studies(2) show that anion exchange will remove the plutonium almost quantitatively, thereby permitting a decrease in the plutonium being cribbed.

To test the principal of removing plutonium from wastes, a prototype column is required. The prototype column would provide data for (1) number of column throughputs required to load the column with Pu, (2) allowable volume velocity
in column to permit removal of Pu, (3) required volume of solution to remove plutonium from a loaded column, and (4) number and size of columns to provide sufficient capacity to treat all Recuplex waste. The same prototype column could be used to test Recuplex feed by repiping the equipment, and the data obtained may provide the required information for design of an anion exchange system for a new plutonium recovery facility.

The data available at HAPO indicate that the solids present in the Recuplex feeds or wastes would soon plug a fixed bed ion exchange column. At the present time, a prototype fixed bed ion exchange column is being operated at the Purex Plant to remove plutonium from concentrated wastes. This bed plugs frequently but has been freed by a short back-flush to momentarily fluidize the bed, allowing some solids to pass. This system would probably be satisfactory for an exploratory prototype column in the Recuplex facility. Installation of the prototype column in Recuplex rather than in a laboratory facility is recommended because of the presence of the volume of feed solution required and the difficulty of transferring these solutions to a laboratory.

The duPont Company at Savannah River has under development an agitated bed anion exchange column which permits passage through the column of solids of smaller diameter than the resin beads. Such a column would be taller than a fixed bed to accomplish the same amount of plutonium absorption, but the operating feature which allows the solids to pass through the column on out to waste would eliminate the need for extensive feed clarification steps prior to the columns. The agitated bed system may make anion exchange more economical to operate than solvent extraction with the elimination of solvent treatment and solvent reprocessing problems.

Mr. F. Tober of duPont Company indicated during his recent visit to HAPO that they will have a full scale agitated bed anion exchanger in operation in about two months. Data available from this unit might be used as the design basis for such columns in a new recovery facility following prototype studies in Recuplex.

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Manager, Finished Products Technology Research and Engineering

REFERENCES

(1) HW-59460, "Revised Metallic Impurity Specification for Z-Plant Feed from Purex and Redox", R. E. Smith, March 6, 1959.