Abstract. The principal scientific thrust of this project was to demonstrate a novel method for precision matrix-assisted laser desorption-ionization (MALDI) mass spectrometry (MS) of model tank-waste materials using, using the sodium nitrate component of the tank waste both as the matrix and as an internal calibration standard. Conventional nanosecond and femtosecond, single-frequency lasers and a tunable, mid-infrared free-electron laser were used in the development of the MS protocols and in measurements of the MALDI dynamics. In addition to developing a model of the processes which lead to efficient desorption and ionization of organic molecules (e.g., toluene, benzene, chelators, various organic acids, crown ethers) from sodium nitrate, we developed protocols for quantitative analysis based on the use of the sodium nitrate in tank waste as an internal standard. Comparisons of MALDI-MS using nanosecond and picosecond lasers, and of infrared and ultraviolet lasers, have been especially instructive, and demonstrate the superior potential of IR-MALDI for this purpose, as well as for a number of related analytical and thin-film applications.

Key experimental results. The main thrust of our effort has been to elucidate the mechanisms of MALDI in tank-waste simulants using NaNO₃ itself as the matrix, and to demonstrate the potential for quantitative analysis. Most of the experiments involve using a tunable infrared free-electron laser to create ions by ablating the samples at resonant vibrational bands of the matrix. This approach makes it possible to carry out high-precision analysis of the organic content of the simulants without having to add additional matrix material, thus avoiding the possibility of generating unwanted photochemistry during the MS procedure and of creating a potential secondary waste stream. During the project, we have:

• Delineated the roles of laser wavelength, pulse duration, sample pH and mixture complexity in the mass spectra; in particular, we have shown that it is possible to ablate and ionize all the components of a complex tank-waste simulant and to quantitate the results using the characteristic ion peaks of NaNO₃ as internal standards.

• Demonstrated that vibrationally excited matrix materials used in analyzing tank waste — such as NaNO₃ microcrystallites and 2,5-dihydroxybenzoic acid — are directly implicated in the creation of analyte ions from neutral organic molecules embedded in the matrix, when the laser is tuned to a strong vibrational absorption band.

• Developed a physical model of the processes of desorption, ablation and ionization which can be used both to underpin quantitative analysis and to give insight into the MALDI mechanism in the infrared.

The success of IR-MALDI using NaNO₃ as a matrix is surprising (UV-MALDI works best when salts are removed from the matrix-analyte sample), but also points the way to mass spectrometry of organics in other poorly characterized hosts, such as those found associated with the silicate complexes in ordinary dirt. This would make possible, for example, tracking contaminants in the
vadose zone. We have already initiated collaborative activities along these lines with our colleagues in the Vanderbilt Department of Civil and Environmental Engineering, and expect to submit relevant proposals in the future.

Our experiments and studies of mechanistic models appropriate to IR-MALDI in tank-waste simulants have confirmed the following results:

First, we can indeed detect molecular ions desorbed from NaNO₃ with high sensitivity, and there is a linear regime within which one could hope to obtain relatively accurate quantitation. While we are presently only in the low ppm range in terms of sensitivity, we are aware of a number of things we can and will do to enhance the sensitivity by several orders of magnitude. It may be of particular interest that we find the crown ethers, upon desorption, also can abstract and carry away an attached Na atom which shows up very clearly in the time-of-flight spectrum. [See references 13, 14 below.]

Second, particularly efficient ion production in UV-MALDI is obtained if one irradiates doped NaNO₃ at 248 nm after adding an appropriate matrix material (e.g., dihydroxybenzoic acid), possibly with some penalty in mixtures due to UV-induced photochemistry. On the contrary, however, the addition of infrared MALDI matrices (e.g., succinic acid or 4-hydroxy-[β]-cyano-cinnamic acid) to the NaNO₃ does not materially affect the ion yield, even though the added matrix materials have significant absorption at the 7.1 µm wavelength used to excite the NO₃ stretching vibration. The difference between UV-MALDI and IR-MALDI seems to result from the comparatively greater efficiency for generating photoelectrons in the ultraviolet. [Ref. 17] However, comparison of results across a broad range of wavelengths and pulse durations shows that mass spectra by themselves are not particularly useful in accounting for these mechanistic effects. [Ref. 7]

Third, the “phase explosion” model proposed by a number of researchers seems to describe our results quite nicely — the first time there has been any experimental evidence for this in the infrared. A phase explosion can occur when the optical absorption depth of the material is short compared to the characteristic thermal diffusion length. It is then possible to create very high volumetric density of vibrational excitation at high laser intensity, because of the strong thermal confinement. (In CaCO₃, the isoelectronic cousin to NaNO₃, the picosecond free-electron-laser pulses deposit an average of two vibrational quanta in every CO₃ group in the irradiated volume.) The irradiated material reaches the critical temperature Tc for explosive vaporization before the deposited thermal energy has time to equilibrate, leading to the so-called “phase explosion.” This should be especially productive of highly energetic ions and neutrals, and gives extremely efficient material removal from surfaces. [Ref. 3]

The major quantitative results for IR-MALDI mixtures in tank simulants will be summarized in a pair of longer papers already prepared for submission in August 2002 [Ref. 13, 14].

Collateral results. The key idea underlying our approach to MALDI-MS in tank-waste simulants is the idea of selective excitation, ablation and ionization based on resonant vibrational, as contrasted with electronic, excitation. This is a novel concept whose experimental realization was dependent upon the availability of a tunable, ultrashort-pulse mid-infrared laser capable of exciting specific vibrational modes in many different materials. Our early success in demonstrating the efficacy of this approach to MALDI-MS led us to apply this concept to three important “spin-off” projects that were also investigated during the course of the project.

Desorption of proteins from separation media. Both genomics and proteomics are heavily dependent on mass spectrometry for identification of nucleotides and proteins. In general, MALDI-MS on separation media is a complex and time-
A consuming process, at each step of which the risk of unwanted modification of the analyte is increased. We have shown that resonant IR-MALDI can induce desorption and ionization of proteins directly from a standard polyacrylamide gel. [Refs. 11, 21].

Desorption of proteins from water ice. The native environment for biological molecules is water. However, previous attempts by many investigators to apply IR- and UV-MALDI to the analysis of proteins in water ices have been disappointing. We have been able to reproducibly generate protein ions from water ices both at the O-H stretch mode (2.94 μm) and O-H bending mode (5.9 μm) wavelengths. There is reason to believe that the picosecond temporal pulse profile of the free-electron laser plays a major role in the ablation and ionization of the embedded proteins, but that remains as a subject of future investigations. [Refs. 13, 21]

Resonant infrared pulsed laser deposition of polymers. While a few polymers have been successfully deposited as thin films by ultraviolet pulsed laser deposition (PLD), the high probability of photofragmentation prevents this in most cases. Using infrared wavelengths resonant with various vibrational modes of polymers, we have been able to show deposition of intact polymers relevant to electronic, photonic, sensor and medical applications, even for polymers of high molecular weight. [Refs. 5, 8, 14]

Each of these research directions has borne fruit in areas relevant to Department of Energy missions, the first two in the area of proteomics and the last in the development of biological and chemical sensor technology. Two patents have been filed involving these discoveries (described on page 7 of this report), and further proposals and research in these areas are anticipated.

Development of laboratory apparatus. In our proposal, we stated our intent to construct a novel miniature time-of-flight mass spectrometer which would also be suitable for post-ionization studies of neutral atoms and molecules emitted during laser-induced desorption. This apparatus became fully operational during the summer of 1999, and has provided much of the data needed to test our conjecture about the mechanism of desorption and ionization in the infrared, in particular, the “phase explosion” hypothesis. It was particularly useful in studies of IR-MALDI mechanisms using single-ion counting. [Ref. 6]

Training of postdoctoral scholars and graduate students. Substantial technical leadership in our program was provided by Dr. David Ermer (Ph.D., 1997), who was exceptionally adept in developing new hardware and initiating our experiments. He left Vanderbilt for a regular tenure-track faculty position at Mississippi State University in the fall of 2000. Two graduate students are working on the project: Michael Papantonakis is two or three months away from completing his dissertation; he spent the summer of 1999 working with Drs. Michael Alexander and Wayne Hess in the Environmental Molecular Sciences Laboratory at the Pacific Northwest National Laboratory. His research focus has been the problems of quantitation associated with tank waste and the potential for the use of compact ion-trap mass spectrometers with MALDI ion sources. Michelle Baltz-Knorr is just also nearing the end of her dissertation program, and has overseen the development and fabrication of a cryogenic-target ion source for MALDI using the reflectron time-of-flight spectrometer acquired with research funding from the Office of Naval Research. This has made possible much higher-resolution measurements than we were previously able to carry out, as well as reproducible MALDI-MS using semisolid slurries. Dr. Kenneth Schriber, Research Assistant Professor of Physics, provided much of the day-to-day scientific direction for
the development of IR-MALDI in water ices during the last months of the project, working with Ms. Baltz-Knorr.

Refereed Publications. The following publications in archival journals and conference proceedings were submitted, accepted or published during this reporting period, all acknowledging Department of Energy sponsorship.


*Conference proceedings.* The following publications in conference proceedings were submitted, accepted or published during this reporting period, all acknowledging Department of Energy sponsorship. Those articles that were peer reviewed are marked with a *.


*Conference contributions.* A number of conference contributions and publications were presented as a result of this year’s work, including papers at the annual meeting of the American Society for Mass Spectrometry and a number of invited talks. The following talks acknowledged DOE sponsorship (the name of the presenting author is italicized):


**Patent applications.** We filed two applications for patents relating to this research program during the course of the past three years.

“Method and technology for pulsed laser evaporation of thin polymer films using resonant infrared laser ablation,” provisional patent application filed jointly by the Naval Research Laboratory and Vanderbilt University; inventors were Daniel M. Bubb, John H. Callahan, Richard F. Haglund, Jr., James S. Horwitz and Michael R. Papantonakis, July 2001.

“Method and Apparatus for the Mass Spectrometry of Proteins and Nucleotides Embedded in Electrophoresis Gels,” international patent application filed August 2001 at Vanderbilt University on behalf of David R. Ermer, Michelle L. Baltz-Knorr and Richard F. Haglund, Jr.