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Ordering Phenomena in Undercooled Alloys

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1. Overview

Most of the work performed under this grant was on the topic of kinetics of ordering in undercooled alloys. Especially in the last year, however, the direction changed to a study of vibrational entropy of alloy phases.

2. Kinetics of Ordering

2.A. Theory and Monte Carlo Simulations

Much of the work performed under this grant was devoted to using modern ideas in kinetics to understand atom movements in metallic alloys far from thermodynamic equilibrium. Kinetics arguments were based explicitly on the vacancy mechanism for atom movements. The emphasis was on how individual atom movements are influenced by the local chemical environment of the moving atom, and how atom movements cause changes in the local chemical environments. I formulated a kinetic master equation method to treat atom movements on a crystal lattice with a vacancy mechanism. Some of these analyses [3,10,16] are as detailed as any treatment of the statistical kinetics of atom movements in crystalline alloys. Three results came from this work. Chronologically they were:

- 1) A recognition that tracking time dependencies is not necessarily the best way to study kinetic phenomena. If multiple order parameters can be measured in a material, the "kinetic path" through the space spanned by these order parameters may be just as informative about the chemical factors that affect atom movements [2,3,5-7,9-11,14-16,18,19,21,23,24,26,36,37].
- 2) Kinetic paths need not follow the steepest gradient of the free energy function (this should be well-known), and for alloys far from equilibrium the free energy function can be almost useless in describing kinetic behavior. This is why the third result surprised me.
- 3) In cluster approximations with multiple order parameters, saddle points are common features of free energy functions. Interestingly, kinetic processes stall or change time scale when the kinetic path approaches a state at a saddle point in the free energy function, even though these states exist far from thermodynamic equilibrium. I term such a state a "pseudostable" (falsely stable) state [6,21,26]. I have also studied these phenomena by more "exact" Monte Carlo simulations. The kinetic paths showed features similar to those found in analytical theories. I found that a microstructure with interfaces arranged in space as a periodic minimal surface is a probably an alloy at a saddle point in its free energy function [21,26,37].

2.B. Experiments

It is optimistic to expect direct connections between theoretical work in statistical kinetics and experimental measurements with real alloys. Nevertheless, our experimental studies of kinetic paths of ordering in Fe_3Al and other alloys identified phenomena that are purely kinetic, and contrary to expectations from thermodynamics. This work used primarily Mössbauer spectrometry [2,7,8,11,14,17,20,22,23,24,26], to provide a detailed counting of ^{57}Fe atoms having different numbers of first neighbor Al atoms. Some work was also performed with EXELFS spectrometry [1,29]. We have measured a wide variety of kinetic paths [5,7,11,14,17,23,24,26], but my favorite is our observation of B32 ordering in Fe_3Al , which seems thermodynamically impossible [7,11,26]. We can explain well the transient appearance of the B32 state by the kinetic master equation method or by Monte Carlo simulations, and we can show that these states of B32 order are at saddle points of a free energy in the point approximation.

In other work performed under this grant, we measured the critical temperatures for the D0_3 -B2 order transformation in Fe_3Al alloys with dilute ternary substitutions for Fe [30]. We made measurements on alloys of $\text{Fe}_{3-x}\text{AlX}_x$, where the ternary solutes were $X = \{\text{Ti}, \text{V}, \text{Cr}, \text{Zr}, \text{Nb}, \text{Mo}, \text{Hf}, \text{Ta}, \text{W}\}$. The effects of the solutes on the critical temperatures were well-explained by an argument based on the differences in metallic radii of the X atom and the Al atom.

3. Vibrational Entropy

3.A. Magnitude of Vibrational Entropy

The free energy, F , of an alloy phase is:

$$F = E - T(S_{\text{config}} + S_{\text{vibr}})$$

The configurational entropy, S_{config} , is well-known. Until now, however, the importance of the vibrational entropy contribution, S_{vibr} , has been essentially unknown or neglected improperly. Towards the end of this grant we made efforts to measure the size of the difference in S_{vibr} between different alloy phases. This work used precision cryogenic calorimetry, and for a variety of alloys found values of ΔS_{vibr} in the range of 0.1 - 0.3 kJ/atom [13,27,28,31,33,34].

We began a parallel effort to understand vibrational entropy by measuring differences in phonon densities of states in different alloy phases [31,33,35]. This work involved inelastic neutron scattering, and was performed at the High Flux Isotope Reactor at the Oak Ridge National Laboratory. Our Oak Ridge collaborators have been Drs. R. M. Nicklow, S. Spooner, and J. L. Robertson. The changes vibrational entropy predicted from the changes in the phonon DOS have generally been in good agreement with the measurements by calorimetry, and both methods have complemented each other nicely. It seems that determining changes in vibrational entropy by measuring inelastic neutron scattering will be a rich new direction for further work.

1992 – June, 1995

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Hao Ouyang, "Grain Boundaries of Nanophase Materials", Ph.D. in Materials Science, California Institute of Technology, October 12, 1992.
Presently: Assistant Professor of Materials Science, National Chung Hsing University, Taiwan

James Okamoto, "Temperature-Dependent Extended Electron Energy Loss Fine Structure Measurements from K, L₂₃, and M₄₅ Edges in Metals, Intermetallic Alloys, and Nanocrystalline Materials", Ph.D. in Applied Physics, California Institute of Technology, May 6, 1993.

Presently: Law student, Stanford University

Lawrence Anthony, "Kinetics of Disorder→Order Transformations in Highly Nonequilibrium Materials", Ph.D. in Materials Science, California Institute of Technology, May 24, 1993.

Presently: Assistant Professor of Physics, University of Toledo. Toledo, Ohio

Zheng-Qiang Gao, "The Kinetics of Ordering, Grain Growth, and Chemical Segregation in Nonequilibrium Fe₃X Alloys (X = Al, Si, and Ge)", Ph.D. in Materials Science, California Institute of Technology, May 18, 1994.

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Laura Jeanne Nagel, "Vibrational Entropy Differences in Materials", Ph.D. in Materials Science, California Institute of Technology, June 18, 1996.

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