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Ultrahard Carbon Nanocomposite Films

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Abstract

Modest thermal annealing to 600 °C of diamondlike amorphous-carbon (a-C) films grown at room temperature results in the formation of carbon nanocomposites with hardness similar to diamond. These nanocomposite films consist of nm-sized regions of high density a-C embedded in an a-C matrix with a reduced density of 5 - 10 %. We report on the evolution of density and bonding topologies as a function of annealing temperature. Despite a decrease in density, film hardness actually increases ~ 15% due to the development of the nanocomposite structure.

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Portions of this document may be illegible in electronic image products. Images are produced from the best available original document. Hydrogen-free diamondlike amorphous-carbon (a-C) films consist of a mixture of 3-fold (sp²-like) and 4-fold (sp³-like) coordinated atoms. Interest in a-C films is due to its room-temperature growth (1-3), high-temperature thermal stability (4) and diamondlike tribological properties (5-7). A significant drawback is the as-grown high residual compressive stresses of these films (8). Stresses on the order of several GPa limit the usefulness of a-C, causing delamination of films with thickness 100 - 200 nm. However, it has been shown that modest thermal annealing to 600 °C dramatically reduces the as-grown residual stress, enabling the ability to grow a-C films several microns thick (9). The tribological properties of such thick, stress-free a-C films are similar to those of diamond, with hardness near 88 GPa (10). Such stress-reduction is leading to the development of new a-C applications, such as stiff membranes for flexural plate wave sensors and even as structural materials for microelectromechanical systems fabrication (11).

We report on the evolution of bonding structures within a-C films as a result of thermal annealing. We find that the annealing process not only reduces stress, but also rearranges carbon-carbon bonding resulting in a nanocomposite material composed purely of a-C. We grow a-C films onto Si(100) substrates at room-temperature using pulsed-laser deposition (PLD), as described elsewhere (8). Other energetic growth methods, such as filtered cathodic vacuum arc and mass-selected ion beam deposition, may also be used (1, 3). Briefly, highly-focussed 248 nm excimer pulsed-laser radiation (KrF) ablates a rotating pyrolytic graphite target with energy density = 45 J/cm², generating a highly energetic (tens of eV) carbon species which deposit onto a rotating substrate. The base pressure of the PLD vacuum chamber is less than 10⁻⁷ torr and rises to nearly 10⁻⁶ torr during deposition. We use a commercial rapid thermal processing system for post-deposition annealing to temperatures ranging from 200 to 600 °C in a pure argon atmosphere in a fused silica chamber with external heating lamps.

Figure 1(a) is a cross-sectional, high-resolution transmission electron microscopy (HRTEM) image of a 32.5 nm thick a-C film grown onto Si. Film thickness was kept low due to the typical 5 – 7 GPa residual stresses that causes thick films to literally crumble during foil formation for HRTEM analysis. The contrast variation within the film image is due solely to changes in carbon density. In general, every a-C film has a distinguishable layer of low density near both the substrate and surface regions, and a region of higher density in the middle, or bulk portion, of a film. Depending upon growth energy, this bulk region can have a density gradient that decreases, remains constant, or increases with distance from the substrate interface (12). The decreasing density gradient seen here is typical of films grown using PLD (13). Nevertheless, note the homogeneity of density parallel to the substrate, and the smooth evolution of density in the growth direction.

We processed by RTA a second sample cut from the same as-grown film shown in fig. 1(a) at 600 °C in Ar for 30 minutes. This results in an elimination of measurable residual film stress and enables the growth of thicker films via multiple deposition and anneal cycles (10). Figure 1(b) is an HRTEM image of this residual stress-free film. Annealed films show dramatic changes in density variation compared with the as-grown films. The dark spots, on the order of 3-5 nm in diameter, are 5-10 % more dense than the lighter contrast surrounding regions. These comparative measurements are made parallel to the substrate surface to exclude the effects of the density gradient perpendicular to the surface found in the as-grown films.

Upon observing the development of this nanocomposite structure, we became curious about the nature of the nearly diamondlike hardness reported for such annealed films. Amorphous carbon films contain considerable fractions of 3-fold coordinated carbon. Furthermore, the stress-reduction annealing process further increases the 3-fold coordinated carbon content at the expense of the 4-fold coordinated fraction. Perhaps the annealing process used for stress reduction is also

responsible for the reported 88 GPa hardness values. To test this hypothesis, we grew a 110 nm thick a-C film on a two-inch Si(100) wafer, cleaved it in half, and annealed one half by RTA at 600 °C in Ar for 30 minutes. Hardness on both the as-grown and annealed samples was measured using nanoindentation accomplished by pushing a small diamond indenter, with an effective tip radius < 50 nm, into a sample and then withdrawing it, recording the force required as a function of position. Since these relatively thin a-C films are on a significantly softer Si substrate, determination of the true hardness values requires finite-element modeling (14). However, since our two samples are of identical thickness on the same substrate, differences in the measurements will approximate a direct ratio of the actual hardness values. We find that the hardness of these a-C films increases nearly 15 % with an anneal at 600 °C. We believe that the development of an a-C nanocomposite, as shown in figure 1B, is responsible for these increased values. The remainder of this letter studies the nature of this unique, pure carbon, totally amorphous nanocomposite structure.

X-ray reflectivity (XRR) infers film density from the critical angle of total internal reflection. Using Cu $K_{\alpha 1}$ radiation from a 12 kW rotating anode source with a resolution of 0.001 Å⁻¹, we collected XRR spectra from nominally 95 nm thick a-C films cut from the same growth substrate and annealed for two hours at temperatures ranging from 200 – 600 °C. Figure 2 shows the XRR inferred densities of the films as a function of annealing temperature. Density values ~ 3 g/cm³ are typical for a-C films of this thickness grown using similar PLD conditions (15). Film density decreases appreciably with annealing above 200 °C, dropping to ~ 2.8 g/cm³ by 600 °C. This ~ 7 % drop in overall density is in good agreement with the ratio determined from HRTEM image contrast. Decreases in a-C density may result in either increases in the ratio of 3-fold to 4-fold coordinated C atoms or a rearrangement of bonding configurations, and is consistent with the general decrease in residual film stress and electrical resistivity reported elsewhere (16).

We investigated the annealing-induced changes in bonding topologies of a-C using Raman spectroscopy. Exciting Raman scatter with the 514.5 nm wavelength (green light) of an argon ion laser preferentially samples 3-fold coordinated carbon atoms due to resonant enhancement through low-lying electronic orbitals. Figure 3(a) shows the as-grown and 600 °C annealed Raman spectra for the same films measured by XRR. We normalize the spectra to the carbon peak height. A change in surface roughness with annealing temperature, detected by XRR (17), contributes to differences in Raman intensities. The spectra for the films annealed from 200 – 500 °C fit monotonically between these two spectra and are not shown for clarity. The feature occurring between 925 and 1025 cm⁻¹ is a second-order phonon band from the silicon substrate. Note that the intensity of the Si band decreases with increasing temperature, indicating that a-C becomes less transparent upon annealing. In fact, the decrease in the ratio of the Si:C Raman band intensities shown in fig. 3(b) represents a monotonic decrease in optical transparency that is consistent with an increase in the 3-fold coordinated carbon fraction of the films; i.e. a-C becomes less diamondlike with thermal annealing. This is not surprising given that graphitic bonding is thermodynamically more stable than diamondlike bonding.

The broad Raman band between 1050 and 1850 cm⁻¹ and centered near 1570 cm⁻¹ is due to resonantly-enhanced vibrational modes of 3-fold coordinated carbon. We further analyze the subtle changes in these carbon spectra by considering the physical characteristics of thermodynamically stable solid-state π -bonded structures. The topographical distribution of 3-fold coordinated carbon atoms in highly diamondlike a-C should include small, relatively isolated graphene clusters. Conjugated chain-like structures may also exist, perhaps even dominate the real structure, but they cannot account for the observed Raman spectra, most likely due to resonant-enhancement issues (15). While 6-membered ring topologies are the most stable, rings of other sizes are possible in solid-state materials. For example, fullerenes include 5- and 7-membered carbon rings to impart curvature. We deconvolute the C bands using vibrational frequencies

calculated from idealized 5-, 6-, and 7-membered π -bonded carbon ring clusters (18) in a manner described elsewhere (15). While idealized ring structures are unlikely to occur in a-C films with high local strains, vibrational shifts due to distortions tend to broaden the range of frequencies rather than to shift the average vibrational frequency for a given ring structure. Since 6-membered carbon rings are the most thermodynamically stable π -bonded structure, we assume that small clusters of such rings are present and exhibit Raman-active breathing (A-type symmetry class) and stretching (E-type symmetry class) vibrational modes. The highly strained environment within these a-C films will force curvature in these clusters, resulting in the evolution of both 5- and 7-membered rings. However, since 5- and 7-membered rings lack inversion symmetry, the A modes dominate their Raman scatter. Hence, four vibrational bands, three with A-type symmetry (from 5-, 6-, and 7-membered rings) and one with E-type symmetry (from 6-membered rings), were used to simulate the Raman spectra.

Figure 4 shows the fractional contribution of each vibrational frequency to the entire 3-fold coordinated carbon portion of the Raman spectra as a function of annealing temperature. Since asgrown films have the smallest population of π -bonded carbon, we expect to find the smallest and most isolated clusters, leading to a preponderance of 6-membered rings. This distribution remains relatively constant for annealing temperatures up to 300 °C. However, recall that both the density and transparency measurements imply a steady in bonding topologies. In particular, the densities plotted in figure 2 show measurable changes by 300 °C. Above this temperature, the fraction of 5-membered rings begins to increase dramatically. This is consistent with an increase in the π -bonded fraction which leads to the development of larger ring clusters. Despite the reduction occurring in residual film stress, the local bonding environments continue to exist within considerable strain fields. Therefore, for the π -bonded ring clusters to grow in size without forming layered graphite-like structures, they must develop curvature to maintain a high degree of atomic packing. This transformation of σ -bonded atoms into π -bonded atoms is driven by

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thermodynamic considerations, hence, 5-membered ring structures, which are more stable than 7-

membered rings, increase in population (16). This does not mean that the 6-membered ring

population is decreasing, only that newly formed π -bonded atoms are grouping into 5-membered

rings.

In summary, we find that annealing smooth and homogeneous diamondlike a-C films to

temperatures near 600 °C transforms them into nanocomposites of pure amorphous carbon. This

new structure consists of highly dense a-C clusters with diameters of 3 - 5 nm in a matrix of 5 -

10 % less dense a-C. Average film density reduces monotonically with annealing temperature,

consistent with the decrease in optical transparency. These reductions result from the conversion

of 4-fold coordinated carbon atoms into thermodynamically more stable, 3-fold coordinated atoms.

Nanoindentation measurements of the final nanocomposite structure, known to have hardness near

88 GPa, find it to be ~ 15 % harder than as-grown films.

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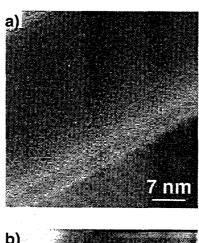
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Figure Captions

- Cross-sectional HRTEM images of (a) as-grown and, (b) 600 °C annealed a-C films from the same growth sample. The second image was enhanced by Fourier filtering for clarity. All of the features can also be observed in the unfiltered image.
- 2. Film density inferred by XRR vs. annealing temperature.
- 3. (a) Raman spectra for the as-grown and 600 °C annealed a-C samples. (b) Si/C Raman band intensity ratio (an indication of optical transparency) vs. annealing temperature.
- 4. Contribution to the carbon Raman band from the vibrations of 5-, 6-, and 7-membered rings vs. annealing temperature.



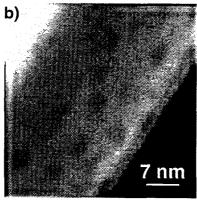


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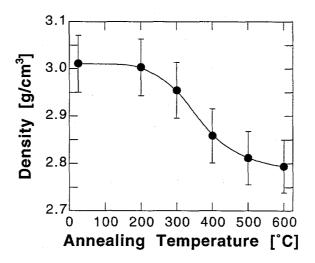


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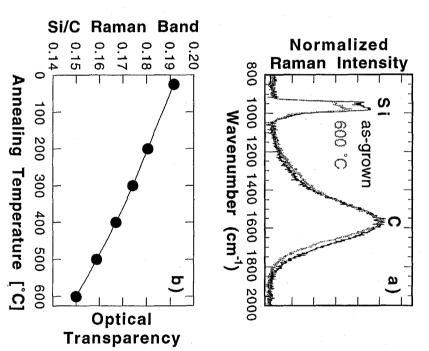


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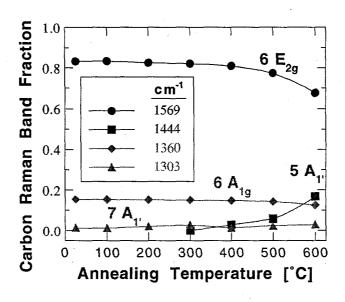


Fig 4 (0f4) M.P. Siegal Appl. Phys. Lett.