EFFECT OF COMPOSITION ON THE PROCESSING AND PROPERTIES OF SINTERED REACTION-BONDED SILICON NITRIDE

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INTRODUCTION

Silicon nitride materials are the leading ceramics for structural applications at elevated temperatures because of their excellent combination of properties. However, cost is the major impediment for the use of silicon nitride in commercial applications. Sintered reaction-bonded silicon nitride (SRBSN) uses silicon powder as the raw material source and is an attractive alternative to the more conventional approaches that begin with Si,N, powders. Because of the low silicon powder cost, the process is cost-effective compared to the expensive high-purity powders.

A significant amount of work on SRBSN has been done since the 1970's and a large database of information is available. However, within the last 10 years the level of research has been minor. In that same time period, a considerable amount of understanding on silicon nitride processing, microstructure development and property relationships has been established. The purpose of the present study was to examine the current state of SRBSN materials and determine the effect of different silicon precursor powders and different sintering additives on the processing and final mechanical properties.

The silicon powders are important because they determine to a large extent the overall purity of the system and the α-Si,N, phase content developed during nitridation of the silicon. Much of the previous work reported in the literature used rather coarse Si powders (>10 μm) and relatively low purity (>0.5 wt. % Fe). Considerably finer particle sizes and higher purity are presently available. The sintering additives also influence the α-Si,N, phase content and, in addition, affect the microstructural development during densification. Most previous work on SRBSN employed Y2O3 and Al2O3 as the sintering additives. It is also well known that alternative sintering aids have been found to result in very different final material properties. Consequently, several alternative additives were...
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studied to see if they provided any advantage over the Y\textsubscript{2}O\textsubscript{3}-Al\textsubscript{2}O\textsubscript{3} system in terms of densification, microstructural development or final properties.

**EXPERIMENTAL PROCEDURES**

The starting materials consisted of appropriate amounts of Si, \(\alpha\)-Si\textsubscript{3}N\textsubscript{4}, Al\textsubscript{2}O\textsubscript{3}, MgO, MgAl\textsubscript{2}O\textsubscript{4} and Y\textsubscript{2}O\textsubscript{3} to give final compositions after nitriding as shown in Table 1. The silicon powders were chosen so that they represented a large variation in impurity content. The Y\textsubscript{2}O\textsubscript{3}-Al\textsubscript{2}O\textsubscript{3} additive system was used as a base material because of the large database already established and it was also used to test the different silicon powders. SRBSN without the \(\alpha\)-Si\textsubscript{3}N\textsubscript{4} seed additive was tested because elimination of the seed is desirable since it represents a significant portion of the raw materials cost.\textsuperscript{7} The MgO-Y\textsubscript{2}O\textsubscript{3} additives were studied because they were reported to produce materials with very high fracture toughneses.\textsuperscript{11} High toughness has also been reported for the MgAl\textsubscript{2}O\textsubscript{4}-Y\textsubscript{2}O\textsubscript{3} additive system.\textsuperscript{12} Previous work on the Y\textsubscript{2}O\textsubscript{3}-SiO\textsubscript{2} system showed that it resulted in materials that had good high temperature strength retention.\textsuperscript{13} All of the samples testing the alternative sintering aids used the same silicon powder (Elkem Metallurgical Grade).

The powders were turbomilled in isopropanol with 0.7 wt.% PVP K-15\textsuperscript{b} and 1 wt.% Darvan 821A\textsuperscript{c} added as dispersants. The mixtures were dried, screened to 200 mesh and isopressed at 207 MPa into discs approximately 7 cm in diameter and 2 cm thick. Binder burnout consisted of a heat-treatment to 550\degree C in air prior to sintering. The nitridation was carried out in either a tungsten-mesh or a graphite-resistance furnace with a N\textsubscript{2}-4 \% H\textsubscript{2} gas flow and controlled heating to 1450\degree C. The samples were then sintered in a graphite-element furnace with Si\textsubscript{3}N\textsubscript{4} packing powder under one atmosphere N\textsubscript{2} to 1780-1800\degree C for 2 hours.

Densities were determined by the Archimedes method. Samples of high density were machined into bend bar specimens with nominal dimensions of 3 mm x 4 mm x 50 mm. Flexural strength testing was done in four point bending with inner and outer spans of 20 mm and 40 mm, respectively. Fracture toughness was determined by an indentation and fracture method. Three batches of samples were fabricated and because the values were similar between batches, the results were averaged together to provide a more reliable set of data.

**RESULTS AND DISCUSSION**

After nitridation the \(\alpha\)-Si\textsubscript{3}N\textsubscript{4} phase contents were determined by x-ray diffraction and the results are summarized in Fig. 1. As shown, for the different Si powders, the \(\alpha\)-content can be well correlated with the starting powder Fe content. This is similar to previous studies.\textsuperscript{8,9} The \(\alpha\)-content developed with alternate sintering aids shows significant differences with the higher \(\alpha\)-contents associated with the

\textsuperscript{b} GAF Chemicals, Wayne, NJ: Polyvinylpyrrolidone K-15
\textsuperscript{c} R. T. Vanderbilt, Norwalk, CT
MgO-Y$_2$O$_3$ and Y$_2$O$_3$-SiO$_2$ systems. The presence of the α-Si$_3$N$_4$ seed was found to enhance the formation of α-phase during nitridation of the silicon.

As shown in Fig. 2, high densities were achieved for nearly all of the samples with the exception of the Y$_2$O$_3$-SiO$_2$ system. The sintering conditions had been selected to obtain high densities, however, the one set that did not sinter well was also the one that contained the most refractory of the grain boundary phases and therefore densification was marginal. All of the other materials had high enough densities to obtain mechanical property measurements.

The results on the room temperature flexural strength and fracture toughness for the various Si powder types are shown in Fig. 3. For the strength, a decrease was observed as the Fe content was also decreased. X-ray diffraction indicated that some unreacted Si was present in the samples from the highest purity Si powder. This unreacted Si may explain the lower strength and indeed, optical examination showed the fracture origins to be black spots that were most likely regions of unreacted Si. The fracture toughness did not show any significant differences for the samples from the various Si types. Microstructurally, the higher Fe content silicon powders resulted in a slightly finer overall grain size as shown in a comparison of Figs. 4 a and b. The finer grain size is probably due to the higher starting α-phase content after nitridation.

The results on the room temperature flexural strength and fracture toughness for the alternate sintering additives are shown in Fig. 5. High strengths and toughness were observed for the Y$_2$O$_3$-Al$_2$O$_3$ and MgAl$_2$O$_4$-Y$_2$O$_3$ containing materials. A relatively fine grain microstructure was observed with the MgAl$_2$O$_4$-Y$_2$O$_3$ containing materials (Fig. 4d) and may help explain the good mechanical properties. The low strength and high toughness exhibited by the MgO-Y$_2$O$_3$ samples is related to its lower sintered density. However, very large grains were observed with this system (Fig. 4e) which would also contribute to the high toughness and low strength. The samples without the α-Si$_3$N$_4$ seed addition had a slightly coarser grain microstructure as indicated by a comparison of Figs. 4 and c. The mechanical properties for the samples with alternative sintering aids correlated well with the starting α-Si$_3$N$_4$ contents as shown in Fig. 6.

The high temperature strength and strength retention are shown in Figs. 7 and 8. As shown previously, the room temperature strength for the different Si powders increased with increasing Fe content. However, the high temperature strength decreased more rapidly with the higher Fe content so that the samples made from higher purity Si had better strength retention.

Results on the weight gain during oxidation at 1000°C for 100 hours is shown in Fig. 9. Only the samples with MgAl$_2$O$_4$-Y$_2$O$_3$ additives showed any significant oxidation weight gain and that is most likely related to the low sintered densities of those materials. Flexural strength (measured at room temperature) after oxidation showed some decrease from the initial values for all samples except the one containing no added α-Si$_3$N$_4$ seed (Fig. 10). The α-Si$_3$N$_4$ seed in the present study was nitrided material that was generally of higher purity than the silicon
powders used so impurity effects are negligible. It is a common practice to give these types of powders a hydrofluoric acid treatment to reduce the oxygen content and this may have contributed to the strength reduction due to oxidation. However, the results need to be confirmed to support such an explanation.

CONCLUSIONS

The type of silicon powder and sintering additive were found to influence the processing and final mechanical properties of sintered reaction bonded silicon nitride. High purity silicon powders produced low α-Si₃N₄ content during nitridation. The Si powder type had no apparent effect on densification. More complete nitridation and higher room temperature mechanical properties were observed for the Si powders with higher Fe contents. However, the higher Fe contents resulted in greater high temperature strength degradation and so there was better high temperature strength retention with the higher purity Si.

High α-Si₃N₄ contents were found after nitridation with α-Si₃N₄ seeded materials and with MgO-Y₂O₃ as the sintering additive. Densification was inhibited by refractory additives, such as Y₂O₃-SiO₂. The highest room temperature strength and fracture toughness values correlated to high nitrided α-Si₃N₄ contents. The high temperature strength behavior was similar for all additive types.

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REFERENCES


Table 1. Summary of compositions used for SRBSN study.

<table>
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<tr>
<th>Sample ID</th>
<th>Final Compositionab</th>
<th>α-Si₃N₄ Contentc</th>
<th>Si Powder Fe Contentd</th>
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<tr>
<td>Elkem or A-Y</td>
<td>Si₃N₄-9% Y₂O₃-3% Al₂O₃</td>
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<td>0.29f</td>
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<tr>
<td>K.N. 4C</td>
<td>Si₃N₄-9% Y₂O₃-3% Al₂O₃</td>
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<td>Albemarle</td>
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<td>0.001g</td>
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<tr>
<td>A-Y</td>
<td>Si₃N₄-9% Y₂O₃-3% Al₂O₃</td>
<td>10</td>
<td>0.29d</td>
</tr>
<tr>
<td>A-Y*</td>
<td>Si₃N₄-9% Y₂O₃-3% Al₂O₃</td>
<td>0</td>
<td>0.29d</td>
</tr>
<tr>
<td>M-Y</td>
<td>Si₃N₄-6.4% Y₂O₃-3.2% MgO</td>
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<tr>
<td>M-A-Y</td>
<td>Si₃N₄-5% Y₂O₃-5% MgAl₂O₄</td>
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<td>Y-S</td>
<td>Si₃N₄-7.6% Y₂O₃-6.5% SiO₂</td>
<td>10</td>
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a All compositions contained approximately 1.7 wt. % SiO₂ to simulate typical Si₃N₄ powder.
b Y₂O₃—MolyCorp, White Plains, NY; Grade 5600, >99.99%; Al₂O₃—Reynolds, Malakoff, TX; Grade RC-HP DBM; MgAl₂O₄—Ceralox, Tucson, AZ; MgO, Mallinckrodt, St. Louis, MO, Reagent Grade; SiO₂, U. S. Silica, 5 μm.
c Starck, Berlin, Germany; Grade LC-10N. The 10% is based on Si₃N₄ content after nitridation.
d Elkem Metals Co., Buffalo, NY; Grade Metallurgical Si; 3.4 μm mean particle size.
e KemaNord, Stockholm, Sweden, Grade 4C; 6.4 μm mean particle size. KemaNord, Stockholm, Sweden, Grade 5C; 6.2 μm mean particle size.
g Albemarle Corp., Baton Rouge, LA; 0.35 μm mean particle size.
h Sample also contained 0.2% CaO added as CaCO₃.
Fig. 1. $\alpha$-Si$_3$N$_4$ content after nitridation for (a) different Si powder types correlated with starting Fe content, and (b) alternate sintering additives. Sample designations listed in Table 1.

Fig. 2. Summary of densification results on (a) different Si powder types, and (b) alternate sintering additives. Sample designations listed in Table 1.

Fig. 3. Room temperature flexural strength (a) and fracture toughness (b) of SRBSN from different Si powder types. Additive composition was Si$_3$N$_4$-9% Y$_2$O$_3$-3% Al$_2$O$_3$. Sample designations listed in Table 1.
Fig. 4. Microstructures of SRBSN, (a) composition Si$_3$N$_4$-9% Y$_2$O$_3$-3% Al$_2$O$_3$ with Elkem Si, (b) composition Si$_3$N$_4$-9% Y$_2$O$_3$-3% Al$_2$O$_3$ with KemaNord 4C Si, (c) composition Si$_3$N$_4$-9% Y$_2$O$_3$-3% Al$_2$O$_3$ with Elkem Si and no α-seed, (d) composition Si$_3$N$_4$-5% Y$_2$O$_3$-5% MgAl$_2$O$_4$ with Elkem Si, (e) composition Si$_3$N$_4$-6.4% Y$_2$O$_3$-3.2% MgO with Elkem Si.
Fig. 5. Room temperature flexural strength (a) and fracture toughness (b) of SRBSN with alternate sintering additives. Sample designations listed in Table 1.

Fig. 6. Correlation of flexural strength and fracture toughness with starting $\alpha$-Si$_3$N$_4$ content after nitridation for SRBSN with alternate sintering additives. Sample designations listed in Table 1.

Fig. 7. High temperature flexural strength (a) and strength retention (b) of SRBSN from different Si powder types. Additive composition was Si$_3$N$_4$-9% Y$_2$O$_3$-3% Al$_2$O$_3$. Sample designations listed in Table 1.
Fig. 8. High temperature flexural strength (a) and strength retention (b) of SRBSN with alternate sintering additives. Sample designations listed in Table 1.

Fig. 9. Oxidation weight gains for (a) different Si powder types, and (b) alternate sintering additives. Oxidation at 1000°C for 100 hours in air. Sample designations listed in Table 1.

Fig. 10. Strength retention after oxidation for (a) different Si powder types correlated with starting Fe content, and (b) alternate sintering additives. Oxidation at 1000°C for 100 hours in air. Sample designations listed in Table 1.

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