The X-ray Photoabsorption Spectrum of Potassium near the K-edge

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Abstract

We have used a heat-pipe target in an X-ray beam to obtain photoabsorption spectra of potassium at the K- and KM-edges, in the photon energy range 3600 to 3650 eV. Preliminary identifications of most of the peaks observed are made using Dirac Hartree-Fock calculations. We compare these results with those obtained previously in closed-shell rare gas absorption spectra.

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Notes for Yoshiro (and others): (1) Is there any Japanese work which should be referenced? (2)
All sections have been rewritten (3) still needs a few references

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Introduction

The photoabsorption spectra of rare gases were amongst the first measurements made when x-ray light sources became available¹. Studies of the K-shell absorption edges in most of the rare gases have become possible with the advent of synchrotron radiation sources equipped with efficient X-ray crystal spectrometers². Understanding of the absorption features at the K-edges has been a more gradual process. Many-electron processes are especially important in these studies where much of the time, more than one electron is removed from the inner shells, with strong shakeup and shakeoff processes occurring. The many-electron processes provide strong tests of calculations of the electron correlation processes in many-electron atoms. Much work has been performed on the Argon K-edge region. The landmark paper of Breinig et al³ gave a careful interpretation of the complex features of several of these edges; most notably, for relevance to this work, their interpretation of the K-edge absorption features of argon. Deslattes et al⁴ have analyzed the multivacancy spectrum above the K-edge. Subsequently, higher-energy edges corresponding to double and triple inner-shell vacancy production, have also been observed in x-ray absorption spectra of some of the rare gases⁵. The identification of the many configurations giving rise to these complex features is recent and still requires further theoretical and experimental investigation. Many fewer studies of atomic open shell elements have been made, principally because of the generally lower densities possible in vapor targets, since most open shell atoms are solids or liquids at room temperatures. Some edges have been studied in the lighter alkalis; for example, lithium, by Ederer et al.⁶, and the sodium K edge, by Tuillier et al.⁷

The K edge spectra of heavier atomic systems are strongly affected by relativistic effects: Crasemann et al ⁸ have made several studies of the heavy rare gases to compare with calculations of edge energies, the Breit and other fine structure splittings. These measurements can lead to very precise tests of the many-body relativistic Hamiltonians used in the calculations. The coupling of open outer-shell electrons to the inner shell vacancies produced in these edges has not been studied in detail, either theoretically or experimentally.

In this paper we describe a first attempt to measure absorption spectra in potassium vapor. The heat-pipe technique described below should be applicable to measure X-ray absorption of most alkali vapors and other atoms and molecules which are not gaseous at normal room temperatures. We make identifications of resonances observed near the K threshold and the KM threshold in potassium, and compare these results with measurements made in argon by ourselves and other groups, and with calculations using a multi-configurational Dirac Hartree-Fock code.

Experiment

The experiment took place at the X-24A beamline of the X-ray ring at the Brookhaven National Synchrotron Light Source. The X-24A beamline, jointly operated by NIST(National Institute of Science and Technology) and Argonne, provides a well-focussed and monochromatized beam of photons in the energy range of 2 to 5 keV, and is equipped with a double-crystal monochromator with pre- and post-focussing mirrors⁹. The double crystal arrangement of the monochromator

strongly reduces any second order harmonics. Silicon 111 crystals were used for these experiments. The photon beam was focussed to less than 1 mm² spot size at the center of the absorption path.

Figure 1 is a schematic of the experimental arrangement. The apparatus consists of a half-meter long heat-pipe¹⁰, 5 cm in diameter, with three-mil kapton windows. The tube is connected to a gas-handling system, and was filled with argon, typically to a pressure of 20 torr. The whole system was pumped out initially to better than 2×10^{-7} torr. Pressure and vacuum gauges with electronic readouts allowed us to monitor the pressure to less than 1% at all pressures, and record the pressure simultaneous to each photoabsorption measurement. The center of the heat-pipe is surrounded by a heating coil encased in thermal insulating material. Water-cooling coils at each end of the pipe help to define a sharp cut off of the hot vapor section and prevent the vapor from reaching the end-windows. The output of a temperature sensor inside the heat-pipe acts as a thermostat to control the heater current and maintain a fixed temperature inside the heat-pipe. Heat-pipe conditions are obtained when the temperature at the center of the heat-pipe is such that the vapor pressure of the potassium (or other vaporizing material) is close to that of the buffer gas within the entire system. A "wick", typically a cylindrical metal gauze fitting close to the inside wall of pipe provides a returning path for vapor condensing at the ends of a stabilized center column of potassium vapor. When heat-pipe action¹¹ is attained the vapor pressure of this central column of potassium is balanced by the buffer gas pressure on the outside, preventing any potassium vapor reaching the windows. In practice, the condition is realized from the increase in the reading of the gas pressure in the tube, which matches the expected potassium vapor pressure at the observed temperature setting in the middle of the tube.

The approach to equilibrium of the heat-pipe is made slowly to allow heat-pipe action to commence; otherwise, the return flow is not sufficient to prevent potassium, or other hot vapor, from escaping the tube and depositing on or disintegrating the heat-pipe end-windows. A second problem for absorption measurements is the maintenance of a constant absorption path. A thermostatic system based on an on/off current device can cause cycling in temperature, and thus in the absorption path-length. A preferable technique found was to maintain the equilibrium temperature with a constant heater current system.

A neon-filled ionization chamber was used to record the photon flux before the absorption tube, and a PIN diode monitored the flux following the absorption tube. The ratio of these two signals provide the raw absorption data. The attenuation observed includes that of the two kapton windows. Kapton has a monotonically decreasing absorption index with increasing energy in the regions near the K edges of potassium and argon. Four-jaw slits just in front of the ionization chamber defined the photon beam size to be less than 2 mm diameter within the chamber. The effective area of the PIN diode is about 1 cm², and its position could be adjusted vertically and horizontally so that the 2 mm x 2 mm beam was in the center of the PIN signal plateau region.

A second experiment to study the K-edge, and the doubly excited KL and KM edges in argon was undertaken to allow an accurate comparison between argon and potassium. In this experiment the heat-pipe was replaced by a 10 cm long tube filled with argon to a pressure of

about 10 torr. In other respects the two experiments were basically the same.

The potassium experiment consisted of making wavelength scans over the energy range of 3590 to 3670 eV in steps of about 0.2 eV. Each scan took approximately 30 minutes, and groups of up to 20 scans were made sequentially. Instabilities in the heat pipe occasionally produced small vapor pressure bursts (typically an abrupt few percent effective path length change). Such scans which were eliminated during final data analysis.

Results

Figure 2 shows overview spectra of the argon and potassium K edge regions. The most noticeable feature is the 4p resonance which is much stronger in the potassium spectrum. The ratio of 4p resonance intensity to background is approximately 2.4 for potassium and only 1.2 for argon. The potassium ratio should be affected more by the instrumental width which was close to 0.8 eV, compared with about 0.6 eV in the argon case. The relative strength of this resonance appears to be close to that observed by Esteva et al¹² for the 3p resonance in neon. This resonance is also enhanced in the corresponding alkali (sodium) spectrum of Tuillier et al⁷. The observed sodium and neon ratios are approximately 3.5 and 2.5 respectively. The enhancement could be due to mixing between the [1s]4s4p and the 4p singly-excited state: however, our relativistic Hartree Fock calculations indicate almost no mixing in potassium. We suggest that the enhancement of the 4p resonance arises due to the larger number of states in the final configuration. The coupling of the outer shell electron leads to more states in the final configuration, and to six instead of two transitions. An averaging over the individual oscillator strengths then suggests an increase of close to 2, as observed. A more systematic investigation of heavier alkalis and alkaline earth K edge spectra will help to verify this suggestion.

We have used the GRASP-II multi-configurational relativistic Hartree-Fock code¹³ to calculate the transition energies and transition rates to fit the measured potassium K edge resonances. The calculation, which includes the ground state configuration and 15 excited state configurations, leads to the 46 transitions shown in the "stick spectrum" in the upper part of Fig. 3. The heights of the lines represent our calculated transition rates. By folding in an experimental halfwidth of 0.8 eV, and adding an arctangent function to represent the ionization limit, we are able to reproduce well the shape of the edge observed by experiment. The results of the calculation are shown in the center section of Fig. 3, and should be compared with the fitted potassium experimental curve shown in the bottom section of Fig. 3. The ionization limit and upper rydberg states are represented in the fit by an arctangent as shown in the lowest section of Fig. 3. The strengths but not the positions of the resonance peaks are sensitive to the midpoint energy and the width of the arctangent function. The fit shown in Fig. 3 is obtained with a midpoint energy of $E_M = 3914.5$ eV, and a "halfwidth" D of 1.2 eV, where E_M and D are defined in the function as $\text{arctangent}\{(E-E_M)/D\}$.

The absolute energy calibration of the potassium 4p resonance was performed by measuring the energies of the known L_1 , L_2 and L_3 edges of silver¹⁴, and the K edge of argon. These measurements result in a value for the [1s]4s4p resonance at 3609.94 ± 1.0 eV. The [1s]4s4p

resonance is a sum of six transitions. The result is in reasonable agreement with previously tabulated values¹⁴. The weighted average over these transitions of our calculations leads to a theoretical value for the observed resonance of 3610.33 eV. The calculation includes a correction of about 4.0 eV from Breit and quantum electrodynamic (QED) effects. Thus the measurement is quite sensitive to these relativistic terms. The agreement between theory and experiment (the difference is 0.39 ± 1.0 eV) is approximately the expected precision of the calculations.

The other strong peaks arise mainly from transitions to the $[1s]4s5p$, the $[1s]4p5s$ and the $[1s]5s5p$ upper states. The $[1s]3d4p$ and $[1s]3d5p$ states lie close to the $[1s]4s5p$ states but the transition rates to these states are much lower. We list some identifications and energies of the relevant transitions in Table I. We include only those transitions with the transition probabilities greater than $1.2 \cdot 10^{10} \text{ s}^{-1}$. Most of the final states are strongly mixed in the LS configurations given, and we have included the principal wavefunctions for each upper state. The experimental energies given in the fifth column of Table I show good agreement with the centroids of the theoretical values after adjustment for the difference in experiment and theory for the 4p resonance energy.

The KM doubly-excited satellites are also seen in both the argon and potassium spectra. Figure 4 shows an enhancement of these spectral regions. The identifications in the argon spectrum are taken from the work of Dyall et al¹⁵ who used a multi-configurational Hartree-Fock code, plus an extended shake theory to estimate the relative intensities. We have verified many of their identifications in this spectrum, and further detailed analysis is underway, benefitting from our improved signal to noise ratio and resolution for this argon spectrum. Our preliminary identifications in the potassium KM spectrum are indicated in the lower half of Fig. 4. The multi-configurational Dirac-Fock calculations included up to 250 excited configurations. We derived energies and transition rates for all the $[1s3p]4snp^2$ levels, where $n=4,5,6$, and 7. This resulted in 188 allowed transitions from the ground state by single photon excitation. The strongest 20 transitions, taken within each np^2 manifold are listed in table II.

The first three peaks of the strong quartet of peaks in the KM potassium spectrum of Fig. 4 arise from transitions to the $[1s3p]4s4p^2$ configuration. None of the calculated transitions can account for the fourth peak of this quartet. The next doublet can be assigned to transitions to the $[1s3p]4s5p^2$ configuration. It is likely that transitions to the $[1s3p]4s4p5p$ overlap with this latter transition. Unfortunately, this involves a configuration with five open shells which are too many to be accommodated with the present code. The calculations cannot be considered complete until several of these configurations with five-open shells have been included. We suggest that the fourth peak of the strong quartet of lines belongs to these $[1s3p]4s4p5p$ and $[1s3p]4s4p3d$ configurations. Its experimental energy is at 3642.4 eV. The $[1s3p]4s$ limit is indicated above the $[1s3p]4s7p^2$ position at an energy of 3655 eV. This limit includes four $[1s3p]^3P \ 4s \ ^4P$ and 2P levels, the lowest of which is calculated to be the $^4P_{5/2}$ level at 3654.41 eV, and two $[1s3p]^1P \ 4s^2P_{1/2,3/2}$ levels calculated to lie at 3655.9 eV. The energy range between 3645 eV and 3655 eV must contain many transitions to other levels with 5 open shells of the general type $[1s3p]4s4pn/$, $[1s3p]4s5pn/$, etc. Above this limit, we observe, but have not classified, peaks due to transitions

to configurations of the type $[1s3s]n/n'l'$.

Conclusion

We have measured the K and KM edge absorption spectra in potassium for the first time. We note that the 4p resonance is stronger relative to the ionization continuum in potassium than in the neighboring argon noble gas spectrum. The same enhancement appears in the respective sodium and neon spectra. We have made identifications of many of the observed structures using GRASP-II, a multi-configurational relativistic Hartree Fock code. The measured energies are accurate to ± 0.1 eV. The theoretical values of the singly excited core-hole states appear accurate to ± 0.5 eV, and the double-hole states have an accuracy of ± 2 eV. Our work shows the need to incorporate five open shell systems in this code in order to treat double inner-shell vacancy promotions in already open shell atoms. Within these limitations the calculations show reasonable agreement with the measurements for the K and the KM edges giving confidence in these preliminary identifications.

Acknowledgements

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Table II. Transition energies and transition rates for KM-double excitation in potassium to the $[1s3p]4s4p^2$, $5p^2$, $6p^2$, $7p^2$ levels - for the strongest transitions only. The experimental errors are given relative to the $4p$ resonance energy.

Excited state configuration	Transition energy eV	Transition rate 10^4 s^{-1}	Experimental Energy eV
$[1s3p]^3P \ 4s \ ^2P \ 4p^2 \ ^3P_2 \ ^2P_{1/2}$	3638.336	99278	3636.05 ± 0.20
$[1s3p]^3P \ 4s \ ^2P \ 4p^2 \ ^3P_2 \ ^2S_{1/2}$	3638.436	69474	
$[1s3p]^3P \ 4s \ ^2P \ 4p^2 \ ^3P_2 \ ^2P_{3/2}$	3638.473	147808	
$[1s3p]^3P \ 4s \ ^4P \ 4p^2 \ ^1S_0 \ ^4P_{1/2}$	3640.683	23228	3638.48 ± 0.20
$[1s3p]^1P \ 4s \ ^2P \ 4p^2 \ ^1D_2 \ ^2P_{1/2}$	3640.761	76844	
$[1s3p]^1P \ 4s \ ^2P \ 4p^2 \ ^1D_2 \ ^2P_{3/2}$	3640.792	90111	
$[1s3p]^3P \ 4s \ ^4P \ 4p^2 \ ^3P_2 \ ^2P_{3/2}$	3641.748	59334	3640.21 ± 0.20
$[1s3p]^1P \ 4s \ ^2P \ 4p^2 \ ^1S_0 \ ^2P_{3/2}$	3641.899	59519	
$[1s3p]^3P \ 4s \ ^4P \ 4p^2 \ ^3P_2 \ ^2P_{1/2}$	3641.986	108316	
$[1s3p]^3P \ 4s \ ^2P \ 4p^2 \ ^3P_2 \ ^2P_{1/2}$	3642.696	33867	3642.32 ± 0.20
$[1s3p]^3P \ 4s \ ^2P \ 4p^2 \ ^3P_2 \ ^2P_{3/2}$	3642.782	37803	
$[1s3p]4s4p5p$	-	-	
$[1s3p]4s4p3d$			
$[1s3p]^3P \ 4s \ ^4P \ 5p^2 \ ^3P_2 \ ^2P_{3/2}$	3646.826	66.4	3646.25 ± 0.20
$[1s3p]^3P \ 4s \ ^4P \ 5p^2 \ ^3P_2 \ ^2P_{1/2}$	3646.878	115.6	
$[1s3p]^3P \ 4s \ ^2P \ 5p^2 \ ^3P_2 \ ^2P_{1/2}$	3646.894	69.50	
$[1s3p]^1P \ 4s \ ^2P \ 5p^2 \ ^1D_2 \ ^2P_{3/2}$	3648.239	192.3	3649.0 ± 0.2
$[1s3p]^1P \ 4s \ ^2P \ 5p^2 \ ^1D_2 \ ^2P_{1/2}$	3648.267	205.1	
$[1s3p]^3P \ 4s \ ^4P \ 6p^2 \ ^3P_2 \ ^2P_{3/2}$	3649.259	7.24	3655.5 ± 0.3
$[1s3p]^3P \ 4s \ ^4P \ 6p^2 \ ^3P_2 \ ^6D_{3/2}$	3649.373	5.27	
$[1s3p]^3P \ 4s \ ^4P \ 6p^2 \ ^3P_2 \ ^4D_{1/2}$	3649.414	5.31	
$[1s3p]^3P \ 4s \ ^4P \ 6p^2 \ ^1D_2 \ ^4D_{1/2}$	3649.574	8.90	
$[1s3p]^3P \ 4s \ ^4P \ 7p^2 \ ^3P_2 \ ^2P_{3/2}$	3650.576	2.49	3655.5 ± 0.3
$[1s3p]^3P \ 4s \ ^4P \ 7p^2 \ ^3P_2 \ ^2P_{1/2}$	3650.738	2.01	
$[1s3p]^1P \ 4s \ ^2P \ 6p^2 \ ^1D_2 \ ^2P_{3/2}$	3650.823	13.1	
$[1s3p]^1P \ 4s \ ^2P \ 6p^2 \ ^1D_2 \ ^2P_{1/2}$	3650.853	18.03	
$[1s3p]^3P \ 4s \ ^4P \ 7p^2 \ ^3P_2 \ ^2P_{1/2}$	3650.884	2.26	
$[1s3p]^1P \ 4s \ ^2P \ 7p^2 \ ^1D_2 \ ^2P_{3/2}$	3652.107	2.63	
$[1s3p]^1P \ 4s \ ^2P \ 7p^2 \ ^1D_2 \ ^2P_{1/2}$	3651.140	4.17	
$[1s3p]^1P \ 4s \ ^2P \ 7p^2 \ ^1D_2 \ ^2D_{3/2}$	3652.147	1.52	
$[1s3p]4s$ Ion limit			

Figure Captions

- Figure 1. The experimental arrangement for absorption measurements at the NSLS X-24A beamline. The heat pipe was replaced by a short cell of argon gas for the argon absorption measurements.
- Figure 2. Spectra of the K-shell absorption edges in argon (upper) and potassium (lower). The spectra show the strong 4p resonances at the edge plus multivacancy excitation at higher energies.
- Figure 3. Calculated transitions in potassium. The (upper) stick spectra have heights proportional to the Einstein A-values for the 46 transitions. Note scale change of the dominating 4p transition. The middle spectrum shows the result after folding in an experimental halfwidth of 0.8 eV, and an arctangent function for the ionization edge at 3615.5 eV. The lowest spectrum shows a fit to the experimental spectrum.
- Figure 4. Comparisons of the KM satellite structures in argon and potassium. Theoretical identifications are given.

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