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SURFACE ACOUSTIC WAVE SENSING OF VOCs IN HARSH CHEMICAL ENVIRONMENTS

by

K. B. Pfeifer, S. J. Martin, and A. J. Ricco Microsensor Research Department, 1315 Sandia National Laboratories Albuquerque, NM 87185

ABSTRACT

The measurement of VOC concentrations in harsh chemical and physical environments is a formidable task. A surface acoustic wave (SAW) sensor has been designed for this purpose and its construction and testing are described in this paper. Included is a detailed description of the design elements specific to operation in 300° C steam and HCl environments including temperature control, gas handling, and signal processing component descriptions. In addition, laboratory temperature stability was studied and a minimum detection limit was defined for operation in industrial environments. Finally, a description of field tests performed on steam reforming equipment at Synthetica Technologies Inc. of Richmond, CA is given including a report on destruction efficiency of CCl, in the Synthetica moving bed evaporator. Design improvements based on the field tests are proposed.



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SURFACE ACOUSTIC WAVE SENSING OF VOCs IN HARSH CHEMICAL ENVIRONMENTS

1.0 INTRODUCTION

Much attention has been paid recently to the collateral damage that has been done to the environment as a result of the Cold War. In addition to the well known environmental disaster sites at Hanford (WA), Savannah River (SC), and Rocky Flats (CO), there are literally hundreds of other sites around the United States where soil and ground-water contamination has occurred. In fact, any site where high-technology materials and devices are produced has the potential for soil and ground-water contamination due to the volume of volatile organic compounds (VOCs) associated with those industries.

It has become clear that in order to restore the environment and protect the public health, it will be necessary to cease the practices that have resulted in this contamination. In addition to the development of environmentally conscious manufacturing techniques, the old damage must be repaired. A whole range of environmental sensors must be developed to help assess the level and extent of the damage and evaluate the remediation technologies. This report is a description of a sensor developed for such a purpose.

One current method of cleaning up VOC contamination of ground water is called **in-situ** air stripping and involves the drilling of wells into the water table that are used to pump air down to

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the contamination, and then to pull the air and the volatilized contaminant out of the ground. The exhaust stream is then scrubbed of the contaminant by passing it through beds of granular activated charcoal (GAC) before it is released into the atmosphere. Eventually, the charcoal becomes saturated with the contaminant and the efficiency of the process diminishes. At this point, the charcoal is replaced and the process continued.

Replacement of the charcoal is undesirable because of the cost of new GAC (\$2/1b) and because the old GAC (loaded with the contaminant) is classified as hazardous waste and must be disposed of as such. In the past, reactivation of the charcoal has not been a viable option due to the inefficiency and high cost of the processes; however, a relatively efficient, low cost process has been developed by Synthetica Technologies, Inc. to remove the organic waste from the charcoal and reactivate the charcoal absorber. This process is called the Synthetica Technologies Detoxifier (STD) and uses steam to oxidize the organic chemicals to CO, and H,O, which can then be vented into the atmosphere.¹ The process works effectively on organics without heteroatoms (i.e. organics with only carbon, hydrogen, and oxygen); organics with heteroatoms, such as halogenated solvents (e.g. contain chlorine or fluorine), are reformed into CO_2 , H_2O and mineral acids such as HCl or HF. The acids eventually destroy the stainless steel vessel and heating elements used in the construction of the STD. This problem has been addressed by the development of a moving-bed evaporator

(MBEv), that neutralizes the acid from the waste stream by passing it through a bed of spheres coated with a molten base such as NaOH.

In order to evaluate the efficiency of the MBEv at destroying VOCs, it was necessary to develop a sensor to measure the concentration of VOC vapors in the exhaust stream of the MBEv. The exhaust stream from the MBEv is a high-temperature, potentially corrosive, steam environment that requires special engineering effort to measure the concentration of VOCs without damaging the sensor.

The sensor is based on surface acoustic wave (SAW) technology.² A polyisobutylene (PIB) film is spin cast onto the SAW device (2000 Å) to absorb species from the gas phase. The SAW device is used as the feedback element in an oscillator circuit whose frequency is then monitored as a function of time (F.gure 1.1). The amount of VOCs absorbed by the polyisobutylene film is proportional to their gas-phase concentration; absorbed VOCs change the film's mass density and viscoelastic properties, resulting in changes in the acoustic wave velocity and hence the oscillator frequency.³

A prototype sensor unit was assembled that included the SAW sensor, oscillator electronics, on-board SAW test case, and device temperature controller. The SAW case was designed to allow diffusion of the gas from the waste stream to the SAW sensor; sufficient distance was allowed for the hot waste stream sample to cool sufficiently before reaching the sensor,





preventing damage to the SAW device and its chemically and temperature sensitive film.

Preliminary tests of the device on CCl_4 in the Synthetica quartz reactor (a small scale version of the MBEv) indicated that the device responds linearly to CCl_4 in a background atmosphere of steam and CO_2 with a proportionality constant of 1.6 Hz/ppm. It appears that the sensor can monitor the destruction of VOCs (CCl_4) with little interference from the other reaction products (steam and HCl), and has a minimum sensitivity (detection limit) on the order of 200 ppm CCl₄.

The remainder of this report outlines the details of the design and testing of the SAW sensor. Section 2 gives a detailed overview of the physical description of SAW devices and their application to chemical sensing. Sections 3 and 4 deal with the design of the prototype SAW sensor system and the laboratory testing of this system at Sandia National Laboratories (SNL). Section 5 describes the tests of the sensor system mounted on the Synthetica steam reforming equipment, while Section 6 presents some important concluding points and provides some suggestions for future studies.

2.0 SURFACE ACOUSTIC WAVE SENSORS

The technology used to implement a CCl₄ sensor into the Synthetica process involves the application of a thin, chemically-sensitive polymer film to the surface of a surface acoustic wave (SAW) device. The SAW device (Figure 1.1) consists

of a guartz (SiO,) substrate with two sets of interdigital transducers (IDTs) on either end of the substrate; interdigital transducers consist of a series of conductors (fingers) alternately placed on the quartz with a center-to-center spacing If an electric field is placed across one of the sets of d. IDTs, a strain is induced in the surface of the quartz as a result of its piezoelectric nature. In addition, if the electric field is oscillating at frequency ν , then an oscillating strain field is generated at the surface, giving rise to a surface acoustic wave that propagates across the surface of the quartz. The acoustic wave induces a potential on the output transducer which is amplified and fed back into the first transducer, resulting in an oscillation whose frequency is determined by the physical properties of the transducer and substrate. The wavelength (λ) most efficiently excited by the transducer is determined by the periodicity of the transducer finger spacing d: $\lambda = d$. The synchronous frequency (ν) at which the transducers operate most efficiently is related to the wavelength and velocity (v) by:

$$\nu = \frac{v}{d} \qquad (2.1)$$

From Equation 2.1, it is clear that a change in the velocity of the wave will result in a corresponding change in the frequency of oscillation of the circuit:

$$\frac{\Delta \nu}{\nu_0} = \frac{\Delta \mathbf{v}}{\mathbf{v}_0} , \qquad (2.2)$$

where ν_0 and v_0 are the initial SAW frequency and wave velocity, respectively.

Application to the quartz surface of a thin, sensitive film capable of binding species (from the gas phase) and changing the wave velocity of the quartz surface of the SAW device results in a system capable of sensing. In the system described herein, a thin polyisobutylene (PIB) film with a thickness of 1800 Å is spin cast onto a SAW device. The system detects the presence of an organic vapor when the vapor is physisorbed by the thin film, because the resulting mass and viscoelasticity changes modify the velocity of wave propagation on the quartz surface and result in a measurable and reproducible frequency shift. The following sections contain a detailed explanation of the fundamentals of SAW behavior and can be skipped.

2.1 PERTURBATIONS IN ACOUSTIC WAVE PROPAGATION

In acoustic wave sensor applications, the measurand is typically a change in wave velocity v and/or attenuation α induced by the interaction of the device with its environment. Consequently, it is important to understand how these interactions cause a sensor response. Changes in wave velocity and attenuation are fundamentally related to changes in wave energy density and power dissipation. With regard to velocity changes, the power density P (power/area) carried by a wave can be related to the wave energy density U (energy/volume) stored in a lossless medium. Considering a unit cube through which a wave is passing, the transit time for the wave across the cube is $\tau =$ 1/v. When the wave passes through the cube, the energy density in the cube increases by the incident power times the transit time: U = $P\tau = P/v$. Thus,

$$\mathbf{P} = \mathbf{U}\mathbf{V} \quad . \tag{2.3}$$

This relation can be used to relate changes in wave energy density to changes in wave velocity in a lossless medium, i.e., one in which P is constant. Differentiating Equation 2.3 yields

$$\frac{\Delta \mathbf{v}}{\mathbf{v}_0} = -\frac{\Delta \mathbf{U}}{\mathbf{U}_0} \tag{2.4}$$

where v_0 and U_0 denote unperturbed propagation velocity and energy density, respectively. This represents a fundamental relation between wave velocity and energy density: the fractional change in wave velocity is proportional to minus the fractional change in wave energy density. Thus, changes to the medium that affect the wave energy density change the wave velocity.

The peak kinetic energy density U_K for a solid is given by:

$$U_{\rm K} = \frac{1}{2} \rho \sum_{i=1}^{3} \dot{u}^{2}_{i}$$
 (2.5)

where ho is the mass density of the medium and u_i is the

displacement of the medium along the ith dimension. Implicit differentiation of the varying quantities leads to

$$\frac{\Delta U_{\rm K}}{U_0} = \frac{\Delta \rho}{\rho_0} \qquad (2.6)$$

Thus, from Equation 2.4, the fractional change in wave velocity is minus the fractional change in mass density of the medium:

$$\frac{\Delta \nu}{\nu_0} = -\frac{\Delta \rho}{\rho_0} \qquad (2.7)$$

Thus, if a plane wave propagates in a medium in which the mass density changes, the wave velocity is shifted as indicated in Equation 2.7.

Next, the effect on wave propagation of power dissipation in a lossy medium is considered. From conservation of energy, the power dissipated by the wave P_d (power/volume) must be balanced by a reduction in power transmitted by the wave P. If the wave is propagating in the x direction, then

$$P_{d} = -\frac{\partial P}{\partial x} \qquad (2.8)$$

Since energy density and power flow are proportional to the square of wave amplitude, in a lossy medium

$$P(x) = P_0 e^{-2\alpha x}$$
, (2.9)

where P_0 is the input power density in the medium. Therefore,

$$\frac{\partial P}{\partial x} = -2\alpha P \quad . \tag{2.10}$$

Combining Equations 2.8 and 2.10 yields

$$\alpha = \frac{P_d}{2P} \qquad (2.11)$$

Equation 2.11 gives the relationship between wave attenuation and power dissipation in the medium: specifically, attenuation is half the ratio of power dissipated to power transmitted by the wave. Note that in the derivations of this section, velocity and attenuation changes depend on <u>ratios</u> of energy and power, not on absolute levels. Consequently, in the small-signal limit, velocity and attenuation changes are independent of wave amplitude.

In general, perturbations change both energy storage and power dissipation and thus result in a combination of velocity and attenuation changes. The manner in which the propagation of the wave is described is therefore important and will be discussed here briefly.

The continuous propagation in the x direction of a wave with angular frequency ω is described generally as

$$u(x,y,z,t) = u(x,y,z) e^{(j\omega t - \gamma)}$$
, (2.12)

where γ is a complex propagation factor representing both attenuation and wavenumber (k):

$$\gamma = \alpha + jk = \alpha + j - \frac{\omega}{v} \qquad (2.13)$$

Changes in wave propagation can thus be represented by

$$\Delta \gamma = \Delta \alpha + j \left[\frac{\Delta \omega}{v_0} - k_0 \frac{\Delta v}{v_0} \right] . \qquad (2.14)$$

When a SAW device is included as the feedback element in an oscillator circuit, there are two conditions that must be met for oscillation to occur. The first condition requires that the round-trip phase shift be an integral multiple of 2π radians (boundary conditions are met). Thus, the bracketed terms in Equation 2.14 must be zero. Setting these terms to zero results in Equation 2.2 as expected. The second condition requires that the sum of all transduction and propagation losses be less than the gain in the circuit (i.e., the circuit gain is above threshold and constructive interference occurs).

20 log (
$$\Delta \alpha L$$
) + 2T < Gain (dB) (2.15)

where L is the center-to-center spacing between the IDTs, and T is the loss in dB associated with each transducer.

2.2 SURFACE ACOUSTIC WAVE (SAW) DEVICES

The stress-free boundary imposed by the surface of a crystal gives rise to a unique acoustic mode whose propagation is confined to the surface and is therefore known as a surface acoustic wave (SAW). In 1887, Lord Rayleigh discovered this mode of propagation in which acoustic energy is confined very near the surface of an isotropic solid.⁴ This mode, now known as the Rayleigh wave, is of interest to seismologists because it is excited by earthquakes. The utility of Rayleigh waves in sensor applications is also due to the surface confinement of energy, allowing them to be excited by surface electrodes in piezoelectric materials and also making the wave extremely sensitive to surface perturbations.

In order to satisfy the stress-free boundary condition, coupled compressional and shear waves propagate together in a SAW such that surface traction forces are zero (i.e., $\mathbf{T} \cdot \mathbf{y} = 0$, where \mathbf{y} is a unit vector normal to the device surface and \mathbf{T} is the stress tensor for the medium). The generalized surface acoustic wave, propagating in the z-direction, generates a displacement profile $\mathbf{u}(\mathbf{y})$ that varies with depth \mathbf{y} into the crystal as

$$\mathbf{u}(\mathbf{x},\mathbf{y},\mathbf{z},\mathbf{t}) = \begin{bmatrix} \mathbf{u}_{\mathbf{x}}(\mathbf{y})e^{\mathbf{j}\boldsymbol{\phi}_{1}}\mathbf{\mathbf{x}} + \mathbf{u}_{\mathbf{y}}(\mathbf{y})e^{\mathbf{j}\boldsymbol{\phi}_{2}}\mathbf{\mathbf{y}} + \mathbf{u}_{\mathbf{z}}(\mathbf{y})e^{\mathbf{j}\boldsymbol{\phi}_{3}}\mathbf{\mathbf{z}} \end{bmatrix} e^{(\mathbf{j}\omega\mathbf{t}-\gamma z)} ,$$

$$(2.16)$$

where u_x , u_y , and u_z represent displacement components and ϕ_i the phases of the components with respect to u_z . The terms x, y, and z represent unit vectors in the x, y, and z directions,

respectively (see Figure 2.1). The component u_y is perpendicular to the surface, u_z is in the direction of propagation, and u_x is transverse to the yz-plane (i.e., the sagittal plane).

The displacement components $u_i(y)$ vary approximately as $e^{-2\pi y/\lambda}$, where λ is the SAW wavelength along the surface and y is the distance into the substrate; amplitude thus decays rapidly with distance into the bulk interior of the crystal (see Figure 2.1). A cross-sectional view of the strain field generated by a surface wave propagating along the surface of a crystal is shown in Figure 2.1. The strain energy density, also shown in the figure, indicates that the majority of wave energy is contained well within one wavelength of the surface. At higher frequencies (i.e., shorter wavelengths), acoustic energy is confined more closely to the surface and wave sensitivity to surface perturbations increases.

The sensitivity of SAW devices to surface perturbations is dependent upon the wave amplitude at the surface, which can be represented by the surface particle velocities v_{XO} , v_{YO} , and v_{ZO} in the x-, y-, and z-directions, respectively. For ST-cut⁵ quartz, these values are $v_{XO}^2=0.13\omega P$, $v_{YO}^2=1.34\omega P$, and $v_{ZO}^2=0.88\omega P$ where ωP is the product of the angular frequency and the power density.

For propagation in an isotropic medium or along a pure-mode direction of a crystal (e.g., a plane of symmetry), Equation 2.16 reduces to a Rayleigh wave, characterized by having no transverse component: $u_x = 0$. Since u_y and u_z are $\pi/2$ out of phase, the





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molecules of the film move in an elliptical orbit in the sagittal plane; the surface motion resembles that of the ocean under the influence of a passing wave.

The presence of the surface-normal displacement component makes the SAW poorly suited for liquid sensing applications. When the SAW medium is contacted by a liquid, this component generates compressional waves in the liquid; the power thus dissipated leads to excessive attenuation of the SAW.⁶

2.2.1 SAW EXCITATION AND DETECTION

The discovery, by R. M. White⁴ of the University of California at Berkeley, that surface acoustic waves could be excited and detected by lithographically patterned interdigital electrodes on the surface of piezoelectric crystals has led to widespread use of SAW devices in a number of signal-processing applications. These include frequency filters, resonators, delay lines, convolvers, and correlators.^{7,8,9}

A surface acoustic wave (SAW) is most conveniently excited on a piezoelectric crystal using an interdigitated electrode pattern, or interdigital transducer (IDT), as shown in Figure 1.1. Application of a voltage between alternately connected electrodes causes a periodic electric field to be imposed on the crystal. When an alternating voltage is applied, an oscillating spatially periodic strain field is generated in the piezoelectric crystal that generates a standing surface acoustic wave. This standing wave gives rise to propagating waves that are launched

in both directions away from the transducer; the wavefronts are parallel to the transducer fingers.

As discussed above, the transducer operates most efficiently when the SAW wavelength matches the transducer periodicity d (Figure 2.2). This occurs when the transducer is excited at the synchronous frequency, defined by $\nu_0 = v_0/d$, where v_0 is the SAW propagation velocity.

The propagation of the mechanical wave in a piezoelectric medium is accompanied by an associated wave potential Φ . When the wave is incident on a receiving transducer, this potential induces a current flow in each transducer electrode; these currents combine to produce a current flow in the external detection circuit. The addition of current contributions in the receiving transducer is optimized when the transducer periodicity matches the acoustic wavelength. Thus, a reciprocity relation holds, as it must for a passive linear device, between the wave and external signals.

2.2.2 INTERDIGITAL TRANSDUCER FREQUENCY RESPONSE

Each transducer finger may be considered to be a discrete source for the generation of surface waves in a piezoelectric medium. A simple transfer function relates the continuous wave (CW) voltage V_1 applied to a finger and the electrical potential (Φ) associated with the waves radiated in each direction:

$$\Phi^{\pm} = \mu_{\rm S} \, \mathbf{V}_{\rm I} \quad , \tag{2.17}$$



igure 2.2. Relationship between IDT periodicity and coherently excited waves. Cross sectional view of alternately polarized transducer fingers are shown with their corresponding n=1 and n=3 excitation modes. Figure 2.2. coherently

where $\mu_{\rm S}$ is a substrate-dependent parameter, Φ^+ is associated with the rightward propagating SAW and Φ^- with a leftward propagating SAW. The parameter $\mu_{\rm S}$ may be considered frequency independent. The frequency response of the transducer arises mainly from interference between finger contributions, and is relatively insensitive to the frequency response of the individual fingers. This approximation is typically made in analyzing wave scattering from an array of elements: the "element factor" (individual fingers) is typically considered frequency-independent when compared with the "array factor" (entire set of fingers).

When an array of fingers is excited, as occurs with an interdigital transducer (IDT), the wave potential for a rightward propagating wave Φ^+ evaluated at position z is a vector sum of the contributions from each finger:

$$\Phi^{+}(z) = \mu_{s} \sum_{n=0}^{N_{f}-1} V_{n} e^{jk(z-z_{n})} , \qquad (2.18)$$

where z_n is the position of each finger excited with voltage V_n , and N_f is the total number of fingers. Equation 2.18 has the form of a discrete Fourier transform of the sequence V_n . Consequently, the frequency response of the device is proportional to the Fourier transform of the sequence of transducer finger contributions. Schemes have been devised to

vary the individual finger contributions in order to achieve a desired frequency response (see for example the excellent books on SAW filter design by Datta⁷, Morgan⁸, and Ristic⁹). If N_f identical fingers are spaced periodically a distance d/2 apart and excited with alternating voltages $V_n = (-1)^n V_0$, Equation 2.18 becomes

$$\Phi^{+}(0) = \mu_{s} V_{0} \sum_{n=0}^{N_{f}-1} (-1)^{n} e^{-(jnkd/2)} . \quad (2.19)$$

The sum in Equation 2.19 is a geometric series whose elements become unity and add constructively when $kd/2 = m\pi$ (m is an odd integer). This condition defines the relationship between SAW wavelength λ and transducer periodicity d for coherent addition, as shown in Figure 2.2. The IDT excites harmonics at odd multiples of the synchronous frequency: $\nu_m = m\nu_1$.

Moving away from the synchronous frequency, the addition of components from individual fingers becomes incoherent, giving rise to the frequency response

$$\left|\Phi^{+}(f)\right| = \left|\frac{\sin(X)}{X}\right| \qquad (2.20a)$$

in which

$$X = \frac{N_{\rm P}\pi (\nu - \nu_0)}{\nu_0} , \qquad (2.20b)$$

where ν_0 is the transducer's synchronous frequency and N_p is the number of IDT periods: $N_p = N_f/2$. The wave potential as a function of the detuning parameter X, described by Equation 2.20b, is shown in Figure 2.3. Note that when X is a multiple of π , Φ^+ is zero (a result of complete cancellation between finger contributions). Consequently, the frequency interval B between the first nulls on either side of the synchronous frequency is

$$B = \frac{2}{N_{\rm P}} .$$
 (2.21)

Thus, the transducer bandwidth is inversely proportional to the number of IDT fingers. Since a narrow bandwidth is desirable for oscillator applications in order to avoid spurious oscillations and to improve stability, a large number of fingers and a small finger spacing are desirable.

The frequency response measured between a pair of transducers having $\lambda_0 = 32 \ \mu m$ and $N_p = 50$ finger pairs is shown in Figure 2.4. The amplitude, shown on a log (decibel) scale, shows the characteristic sin(X)/X behavior.

2.2.3 SAW PERTURBATION MECHANISMS

When SAW devices are used for sensors or thin-film characterization, the measured responses arise from perturbations in wave propagation characteristics, specifically wave velocity and attenuation, resulting from interactions between the SAW and a surface film. Because a SAW propagating in a piezoelectric



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Figure 2.3. Sin(X)/X vs. X.



medium generates both mechanical deformation and an electrical potential, both mechanical and electrical coupling between the SAW and surface film are possible Consequently, a number of interactions between surface waves and surface films have been found that give rise to velocity and attenuation responses. SAW-film interactions that arise from mechanical coupling between the wave and film include mass loading caused by the translation of surface mass by the SAW surface displacement, and elastic and viscoelastic effects caused by SAW-induced deformation of a surface film. The next section examines the velocity and attenuation changes caused by mass loading interactions between SAWs and surface films.

2.2.4 SAW MASS LOADING

The simplest interaction, and the one most utilized for SAW sensor applications, is the response due to changes in the mass density (mass/area) on the device surface. The harmonic motion of the crystal surface caused by the passing surface wave causes molecules bound to the surface to move in an elliptical orbit synchronously with the SAW surface displacement. The effect on wave velocity and attenuation of this interaction may be derived from energy considerations.

Synchronous movement with the surface acoustic wave of a surface film causes an increase in the kinetic energy density U_K of the wave without dissipating any wave energy. From the discussion in Section 2.1, this is expected to change the wave

propagation velocity without affecting attenuation. The change in average kinetic energy per area of surface is

$$\Delta U_{\rm K} = - \frac{\rho_{\rm S}}{4} (v_{\rm XO}^2 + v_{\rm YO}^2 + v_{\rm ZO}^2) , \quad (2.22)$$

where v_{XO} , v_{YO} , and v_{ZO} are the molecule velocities at the surface and ρ_S is the surface mass density. Particle velocities are related to particle displacement u by $v_i = j\omega u_i$. This increase in kinetic energy density results in a decrease in wave velocity, according to Equation 2.4. Combining Equations 2.3, 2.4, and 2.22 yields an expression for the change in wave velocity arising from surface mass loading:

$$\frac{\Delta \mathbf{v}}{\mathbf{v}_{0}} = -\frac{\mathbf{v}_{0}\omega\rho_{s}}{4} \left[\frac{\mathbf{v}^{2}\mathbf{x}_{0}}{\omega\mathbf{P}} + \frac{\mathbf{v}^{2}\mathbf{y}_{0}}{\omega\mathbf{P}} + \frac{\mathbf{v}^{2}\mathbf{z}_{0}}{\omega\mathbf{P}} \right] . \quad (2.23)$$

Due to the greater confinement of wave energy near the surface that occurs as operating frequency increases, surface molecule velocities increase in proportion to $(P\omega)^{1/2}$. Thus, the quantities in parentheses $(v_{10}^2/\omega P)$, being independent of wave amplitude and dopending only on the substrate material, remain constant. Slobodnik et al. have tabulated these normalized surface particle velocities for a large number of substrates.¹⁰ Propagation in the x-direction in the ST-cut of quartz (a rotated cut chosen for its desirable temperature characteristics) results in three components of particle velocity because of the lack of

crystal symmetry. Grouping all the substrate-dependent constants together results in the expression for the mass-induced change in SAW propagation velocity:

$$\frac{\Delta v}{v_0} = -c_m v_0 \rho_s , \qquad (2.24)$$

where the mass sensitivity factor c_m is

$$c_{\rm m} = \frac{2\pi v_{\rm o}}{4} \left[\frac{v^2 x_{\rm o}}{\omega P} + \frac{v^2 y_{\rm o}}{\omega P} + \frac{v^2 z_{\rm o}}{\omega P} \right] \qquad (2.25)$$

Note that the frequency dependence of the SAW mass sensitivity (fractional velocity change $\Delta v/v_0$) varies with operating frequency ν_0 . Because the mass layer is assumed (in this case) to be lossless, Equation 2.24 implies that attenuation is unchanged by mass loading. The assumption that the mass layer is lossless is not valid for many practical applications; however, the effect on the mass sensitivity is typically small and can generally be neglected. For lossy materials, additional interaction mechanisms such as viscoelasticity arise making the SAW response no longer due simply to mass loading.

The velocity and attenuation changes resulting from depositing a gold layer on a 97-MHz SAW device using an ST-cut quartz substrate are shown in Figure 2.5. Velocity decreases linearly in this thickness regime, yielding $c_m = 1.32 \times 10^{-6}$ cm^2 -s/g, in good agreement with the mass sensitivity factor calculated above for a 100 MHz SAW. The figure demonstrates



that, as predicted from the model, the relative attenuation change $(\Delta \alpha/k)$ due to mass loading is negligible in comparison with $\Delta v/v_{0}$ (shown on the same scale).

2.2.5. SUMMARY OF SAW PROPERTIES

In the above sections, the basic SAW sensor was described including the relationship between the change in velocity of the wave and the change in frequency. Next, a more detailed mathematical model of SAW propagation was given to demonstrate how an increase in mass density will result in a decrease in wave velocity. These arguments were expanded to a lossy medium and it was shown that the same relationship between change in frequency and change in wave velocity (Equation 2.2) holds for a lossy medium. The nature of the interdigital transducers were discussed including the frequency response and bandwidth of the devices.

Finally, it was shown how mass loading of a film on a surface acoustic wave device changes the velocity of the wave, and that the change in velocity is independent of the amplitude of the wave. A relationship between the mass sensitivity of the SAW device and its frequency and material properties was derived. The next section describes a system design that implements these SAW properties in a sensor.

3.0 SYSTEM DESIGN

In this section, the design of the sensor system is

discussed, including the construction of the SAW oscillator board and the SAW test cases. In addition, the designs of the mixer and signal processing circuitry and the heater controller circuitry are presented.

The system consists of two identical SAW oscillator circuits (Figure 3.1). The SAW device of one of the oscillator circuits is open to the atmosphere via a diffusion tube and is referred to as the sensor oscillator. The SAW device of the other oscillator circuit (reference) is identical to the first except that it is isolated from the atmosphere (sealed in air). The reference circuit allows for the correction of common-mode noise signals such as temperature fluctuations.

The output of the reference and signal oscillators are mixed and the difference frequency is converted to a TTL-compatible signal by the mixer/signal processing section of the system. The TTL output frequency of the circuit is measured with the frequency counter which in turn is interfaced via the RS-232 bus with a computer.

In addition, the temperature of each individual SAW device is controlled by an operational amplifier (op amp) based temperature controller. The temperature of the sensor SAW device is more strongly coupled to the environment than that of the reference SAW device. Therefore, the temperature set point of the reference SAW device is controlled by the temperature of the sensor SAW device rather than being set to a fixed temperature.

Finally, the entire system's temperature is controlled via



gure 3.1. Block diagram of SAW VOC sensor system showing signal and reference SAW devices, signal processing, chassis and sensor temperature controls, and data controls, temperature chassis and sensor analysis sections. Figure 3.1.

the chassis temperature controller. The chassis temperature controller consists of an Omega Model CN9000 miniature controller coupled to a 52 W, 120 V heating element mounted to the chassis.¹¹ The system temperature controller can maintain the chassis temperature to within approximately 1°C; the chassis controller can compensate for gross changes in the room temperature, while the individual SAW temperature controllers compensate for fine (<1°C) temperature fluctuations in the SAW devices.

3.1 SAW OSCILLATOR CIRCUIT

The SAW oscillator circuits consist of a SAW device followed by a Mini-Circuits Laboratories PSC-2-1 3-dB splitter (Figure 3.2) that was placed on the circuit board to provide a measurement of the output power from the SAW device.¹² It, along with another splitter on the input to the SAW device, allows for measurement of the attenuation change in a film due to exposure to the chemical environment. Chemical discrimination techniques employing measurement of both frequency shift and attenuation change have been demonstrated³; however, this feature was not used in this prototype.

The other output of the splitter (Pin-6) is input to a current-controlled attenuator (PAS-3) that is included in the circuit to allow adjustment of the oscillation threshold of the circuit.¹² By increasing the attenuation, the circuit can be made to oscillate in only the most efficient mode of the SAW

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device, resulting in a much lower noise level.

The signal from the attenuator is amplified by two radio-frequency (RF) amplifiers in series (UTO-104 and UTO-250) that provide 24 dB of gain and 30 dB of gain, respectively, at saturation.¹³ The output from the second amplifier is fed into a low-pass filter (PLP-150) that has a 3-dB roll-off at 150 MHz and is included in the circuit to eliminate harmonics generated by the saturated amplifier.¹² The signal is next coupled to a 10-dB splitter (TDC-10-1) that is used to couple a portion of the RF signal out of the circuit to the mixer processor system. Finally, the remaining signal is coupled back to the SAW device, completing the oscillator loop. This circuit was constructed on a 2.75" by 3.75" printed circuit board.

3.2 SAW CASES

The SAW cases designed for this project had to meet several distinct requirements. First, they had to fit on a 1.25" x 0.75" area of the circuit board; this restriction arose because circuit boards that had been designed for a separate program were being used. At the time this project was initiated, these boards were designed to have SMA connections to connect the circuit to a separate SAW device test case. Second, they had to allow easy device replacement in the field. The second objective was achieved using spring-loaded electrical contacts built into the SAW test case. In previous SAW case designs, SAW devices were wire bonded to the contacts in the case, making replacement
in the field impractical. Finally, since the SAW cases are exposed to a chemically harsh environment, they were constructed from corrosion resistant 300-series stainless steel.

The SAW device cases were constructed (as depicted in Figure 3.3) from a stainless steel block with a rectangular depression milled in the center to contain the SAW device. Appendix A, Figures A.1, A.2, A.3, and A.4 show mechanical drawings of the SAW case. There are four holes in the bottom of the depression through which spring-loaded electrical connectors protrude. Two of the holes are 0.0625" in diameter and two are 0.045" in diameter. The smaller holes are designed for a press fit of the spring-loaded "pogo" contacts (P/N C-S-R).¹⁴ Augat, Inc. The larger holes are designed for a press fit of the pogo contacts with a 0.009"-thick cylindrical Teflon bushing that electrically insulates the pins from the case. The larger holes provide the RF signal input and output to the SAW devices while the smaller holes provide circuit ground connections.

The top of the case has a 1/16" O-ring groove that forms a gas-tight seal when the lid is attached. In addition, the pogo contacts are placed at a depth that allows the SAW device to be positively located by the walls of the depression when the lid is removed from the case. This requires that the 0.025" thick SAW device extend 0.010" below the surface of the case when the lid is off of the case and the pogo contacts are not compressed. When the lid is placed on the case and the back of the device is flush with the top of the case, the SAW device is forced down



electrical contacts. Figure 3.3.

0.015". The pogo contacts are compressed to approximately half of their range of motion, making excellent contact with the 0.071" square bond pad of the SAW device.

Temperature control of the SAW case is accomplished by glueing a heater resistor in the $(0.590" \times 0.150" \times 0.150")$ depression milled into the side of the case. In addition, a cut $(0.125" \times 0.125" \times 0.063")$ is milled just below the heater resistor depression to provide room for a thermistor to measure the case temperature.

There are two variations of the SAW case used in the sensor system. The first variation is hermetically sealed to contain the <u>reference</u> SAW device. The second variation has a 1/8" x 1" stainless steel inlet tube welded to the bottom of the depression to provide a path for the gas sample to diffuse into the <u>sensor</u> SAW device.

3.3 MIXER SIGNAL PROCESSOR CIRCUIT

The output of each SAW oscillator (Pin 4 TDC-10-1) is a sinusoidal signal with an amplitude of approximately 100 mV (peak-to-peak) and a frequency of approximately 97 MHz. The outputs are mixed by an Anzac MD-123 mixer (see Figure 3.4) and then "squared" (for TTL counter compatibility) with a comparator circuit.¹⁵ The MD-123 is a double-balanced mixer with a 10 MHz to 3 GHz RF bandwidth. The mixer IF (intermediate frequency) bandwidth is DC to 3 GHz; it provides 20 dB of isolation between the RF inputs and the IF output in the frequency range of



gure 3.4. Schematic diagram of signal processing subsystem. In this system, two sinusoidal RF signals are mixed and their difference frequency is converted into a TTL compatible signal. Figure 3.4.

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interest. The output of the mixer is capacitively coupled through a 0.1 μ F capacitor to a signal processing circuit consisting of an Analog Devices AD-790 high speed comparator and associated hardware.¹⁶ The input signal is biased at 2.5 V by the divider circuit (150 k Ω resistors) and coupled to a 390 Ω resistor. The 390 Ω resistor, along with a 65 pF capacitor, form a low-pass filter with a 3-dB cut off at approximately 6.0 MHz; this effectively passes the difference frequency while removing the sum-frequency signal at 194 MHz. This signal is then coupled into the inverting input of an AD-790 high-speed comparator that converts the difference signal to a square-wave signal with the same frequency as the input sine wave. The AD-790 is wired to have an adjustable trigger threshold (20 k Ω variable resistor) and adjustable hysteresis (10 k Ω variable resistor). The output of the comparator is input to a Signetics 7413 four-input Schmitt trigger to buffer the output of the comparator and provide a signal that switches between 0 and 5 V (TTL).17

The TTL signal is then fed into a Newport Electronics Model P6430 TTL frequency counter (Figure 3.1) whose bandwidth is 7 MHz and has 6 digits of output resolution.¹⁸ Thus, for a difference signal that is less that 1 MHz, the output resolution is on the order of 1 Hz. In addition, the P6430 is interfaced via RS-232C to a Dell 320N notebook (PC-compatible) computer that reads the difference frequency and plots it as a function of time.¹⁹

3.4 HEATER CIRCUITS

In order to keep liquid water from condensing on the sensing SAW device, it is necessary to elevate its temperature to a temperature above that of the input atmosphere - approximately 38°C at the sensor end of the diffusion tube. This was accomplished with the circuit shown in Figure 3.5. A heater resistor is imbedded in the SAW case and the temperature of the case is controlled by a LM255 op amp circuit.²⁰ The temperature of the reference SAW case is also elevated to the same temperature as the sensing SAW case to reduce temperature-induced noise in the system.

The sensing SAW case is strongly coupled to the outside environment; thus, it is more susceptible to variations in the temperature of the outside environment. The temperature control circuit was constructed to behave as a master temperature controller for the signal SAW case and a slave temperature controller for the reference SAW case. The master/slave arrangement allows the reference temperature control circuit to have a variable set point that is electronically controlled by the temperature of the sensing SAW case.

To understand the operation of the temperature control system, it is necessary to explain the master temperature controller. As indicated in Figure 3.5, the master (sensor) temperature controller consists of a resistive bridge circuit connected to an LM224 op amp.²⁰ The output of the op amp drives an npn power transistor that in turn controls current through the heater resistor.





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The LM224 op amp is designed to operate from a single-sided power supply (+5 V) and can only output positive voltage values. The op amp will output a non-zero voltage when the non-inverting input voltage (Pin 3) is larger than the inverting input voltage (Pin 2).

The resistance bridge consists of two metal film 10 k Ω resistors, one 20 k Ω variable resistor, and a negative temperature-coefficient thermistor that is 10 k Ω at 25°C. When the bridge is balanced, the voltages at the inverting and non-inverting inputs of the op amp are equal and the output of the op amp is zero. However, when the temperature of the SAW case drops below the set point (i.e., the thermistor resistance rises above the variable resistance value), the voltage at Pin 13 becomes larger than the voltage at Pin 12, turning on the op amp and causing current flow through the heater resistor. Dissipated heat flows into the SAW case and raises the temperature of the SAW case back to the set point. The 12 k Ω feedback resistor on the op amp decreases the slope of the response of the op amp, reducing thermal oscillation of the SAW case.

The temperature controller for the reference SAW case works similarly to the temperature controller for the signal SAW case. However, instead of the temperature being set independently by the variable resistor, the temperature of the reference SAW case is set by the temperature of the signal SAW case. This is done by replacing the variable resistor arm of the resistance bridge with the thermistor arm of the signal

SAW case control circuit as is illustrated in Figure 3.5. If the input gas to the signal SAW case is hot enough to heat the SAW case, the temperature of the reference SAW case will also rise automatically. Since the temperature coefficient of each SAW device is similar, each SAW oscillator changes in frequency by a similar amount. Thus, the difference frequency will remain relatively unchanged.

In the previous sections, the SAW sensing system was described in detail. This included discussion of the oscillator and signal processing section of the circuit, mechanical design of the SAW cases, and the master/slave temperature control circuits. In the next sections, the results of the laboratory and field demonstrations of the system will be discussed.

4.0 LABORATORY TESTS

In this section, the laboratory tests of the sensor are discussed. First, the coating of the devices are described, and studies of the short-term and long-term noise in the system, including the temperature response, are reported. In addition, the response as a function of CCl_4 concentration is described and the minimum detectable limit is determined.

4.1 POLYISOBUTYLENE FILMS

The coating of the SAW devices was accomplished as follows. First, 0.90 g of polyisobutylene (PIB) was dissolved in 44.1 g of chloroform (CHCl₃) to obtain a 2*-by-weight solution of PIB; the mixture of chloroform and PIB was placed in an ultrasonic cleaner to dissolve the PIB. The bare SAW devices were cleaned in boiling acetone, methanol, and isopropanol to remove any residual photoresist and other surface contaminants. Finally, the bare devices were baked at 120°C for approximately 18 min to remove any residual organic solvent from the surfaces of the devices.

The 2% PIB solution was then spin cast at 2000 rpm for 30 s onto the bare SAW device. The freshly coated devices were baked at 100°C for 1 hr in a vacuum oven (pressure was approximately 1 Torr) to remove any residual chloroform. In addition to the SAW devices, a silicon coupon was also coated using the same procedure and the thickness was measured using a Sloan Dektak profilometer. The thickness of the PIB films prepared as described was found to be approximately 1800 Å.

4.2 TEMPERATURE STABILITY

The freshly coated SAW devices were placed in the reference and signal SAW cases. In addition, two thermistors (5 k Ω at 25°C) were placed in close thermal contact to the SAW cases to provide an independent measurement of the SAW case temperature. The unit was then closed, placed in a styrofoam insulating box, and allowed to warm to operating temperature.

Next, the room temperature was cycled between 14°C and 26°C with the sensing SAW device exposed to room air while the difference frequency was monitored via computer. Figure 4.1



changes sensor oscillator temperatures nort dashed line represents room of SAW output solid and long dashed lines represent in stability change line showing long-term depict circles short and and Graph Filled in reference respectively, temperature. frequency, 4.1. system. Figure

shows the difference frequency along with room, sensor SAW case, and reference SAW case temperatures plotted versus time. The room temperature was allowed to fluctuate rapidly over a span of 12°C, resulting in a difference frequency shift of about 450 Hz over 30 hrs. Rapid shifts in room temperature (>5°C/hr) result a larger change than do slower temperature shifts. in Examination of the interval from 5 to 10 hrs indicates that as the temperature of the room rose from 16°C to 26°C, the frequency first dropped by 300 Hz and then rose by 300 Hz. The sign change is a result of a thermal phase delay between the temperature of the signal (master) SAW case and the reference (slave) SAW case. Since the signal SAW case was better coupled to the environment, its temperature initially rose faster than the reference SAW case; however, as the reference SAW temperature rose to compensate for the change, the sign of the effect changed and the net temperature effect was eliminated. When the room temperature shift is slower, the phase delay becomes less significant, resulting in a smaller temperature difference between the two devices and a smaller difference-frequency shift. Ideally, to minimize the temperature drift further, both SAW devices should be placed in a single case so that equal temperature changes are experienced. This approach is planned for future generations of this sensor.

During the interval in which the room temperature is most stable (11 to 21 hrs), the frequency fluctuates on the order of 150 Hz. Thus, the minimum detectable signal over a 10 hour run

with room temperature stability of ± 2 °C is on the order of 200 Hz.

A gas flow system (Figure 4.2) was constructed to calibrate the response of the SAW sensor to CCl₄ exposure. Ultra high purity (99.999%) N_2 is used as a carrier gas and is passed through a gas washing bottle containing CCl₄. The CCl₄-saturated flow is mixed with another known flow of N_2 to adjust the concentration and the resultant CCl₄-laden stream is then passed by the SAW sensor and out the vent. The CCl₄-laden stream diffuses from the main tube to the sensor and the difference frequency from the SAW device is monitored as a function of time. The steady-state frequency shift was recorded vs. CCl₄ a concentration.

Figure 4.3 is a plot of the change in frequency vs. CCl_4 concentration in dry N_2 . The measured data (points) indicate that the sensor responds linearly to concentration with a slope of 1.1 ppm/Hz. Sensor response is linear in the infinite dilution limit (Raoult's Law) but saturation is expected at much higher concentrations. In the application to be described, the concentrations are low and the system is assumed to be linear. All concentration vs. frequency change data are presented with frequency as the independent variable in order to simplify conversion of measured frequency shifts into concentrations during later experiments.

The above discussion of the frequency noise limit leads to an estimate for the minimum CCl_4 detection limit of 200 ppm. The









minimum detection limit can be lowered substantially by modifying the gas sampling technique so that the SAW sensor can be exposed quickly to a reference atmosphere and then exposed to the sample gas of interest at a rate much faster then the temperature drift. This could be accomplished by evacuation of the signal SAW sensor volume, monitoring the response of the SAW to the vacuum, back-filling the volume with gas from the atmosphere of interest, and then observing the frequency change between the reference atmosphere and the sample atmosphere.

In the previous sections, a discussion of the laboratory tests of the sensor was presented. This included a description of the method for coating PIB films onto the SAW devices. In addition, thermal stability was discussed. It was found that the system has a larger temperature dependence for rapidly changing environmental temperatures. In the next generation of system development, the signal and reference device temperatures will be more closely thermally coupled by placing both the sensor and reference SAW devices in the same metal SAW case. Finally, the concentration dependence of the response was measured and found to be linear with a slope of 1.1 ppm/Hz, indicating a minimum detection limit of about 200 ppm CCl₄ in N₂. In the next section, the field test results of the device are presented.

5.0 FIELD TESTS

This sensor was designed specifically to monitor the level of CCl, that is emitted from the Synthetica moving bed evaporator

(MBEv). To accomplish this, several field trials of the system were performed at Synthetica's Richmond, CA facility. In the following section, the results of the these trials are presented. First, the response as a function of CCl_4 concentration in CO_2 and steam was measured to determine the effects of a moist, corrosive environment on the function of the SAW sensor. Next, small-scale steam reforming of CCl_4 in the Synthetica quartz reactor was monitored and the sensor response was measured. Finally, large-scale steam reforming of CCl_4 in the MBEv was demonstrated.

5.1 QUARTZ REACTOR

During the period of March 23-25, 1992, the sensor was tested on the Synthetica quartz reactor depicted in Figure 5.1. The reactor consists of a steam source with a controllable feed rate and a CO_2 source connected through a CCl_4 gas washing bottle. The CCl_4 -saturated stream was mixed with steam and piped into a furnace where the CCl_4 was reformed into CO, CO_2 , H_2O , and HCl. The exhaust stream was then passed by the inlet to the SAW sensor and the CCl_4 level measured with the output of the sensor recorded as a function of time by the notebook computer.

The first experiment on the quartz reactor was designed to determine the CCl_4 concentration dependence of the SAW sensor response in CO_2 and steam and what effect a corrosive steam atmosphere would have on the sensor. The data as a function of time are presented in Figure 5.2 as follows: Interval 1 is the





response for 12900 ppm CCl₄ in dry CO₂. Intervals 2 through 5 are various concentrations of CCl₄ in steam with the various flow rates and CCl₄ vapor pressures for each peak given in Table 5.1. The variation in the CCl₄ vapor pressure is a result of temperature changes in the room and radiative heating of the CCl₄ reservoir by the boiler heaters.

Table 5.1. Flow data used to calculate CCl₄ concentrations in Synthetica quartz reactor.

INTER. (#)	CCl. FLOW (mL/min)	CO2 FLOW (mL/min)	H ₂ O(g) FLOW (mL/min)	CCl, PRÉSS. (Torr)	CCl. CONC. (ppm)	Δν (kHz)
1	523	3161	0	69.8	12900	15.9
2	522	3161	1060	80.3	11500	16.2
3	265	3161	1350	80.3	5830	12.3
4	265	6322	690	80.3	3830	7.3
5	532	1611	1270	80.3	16200	25.4

a. Atmospheric pressure was 760 Torr. Quartz reactor temperature was 330°C.

b. Concentration in ppm by volume.

The concentrations in Table 5.1 are calculated based on an atmospheric pressure of 760 Torr (Richmond, CA has an elevation of approximately 20' above sea level). Initially, the quartz reactor was kept at a temperature of 330°C to avoid destroying the CCl, and the steam boiler was kept at an approximately constant temperature of 550°C.

The arrows in Figure 5.2 indicate when the sensor test case volume was evacuated using a small diaphragm vacuum pump and a three way valve to switch between vacuum and sample stream. As indicated, the response dropped to baseline rapidly and then returned to a signal level when the volume was backfilled with a





CCl₄-laden stream. The responses in Intervals 4 and 5 on the plot are not significantly faster for the backfilled sampling technique over the diffusion sampling technique of Interval 1. This is due to a leak in the system that brought the sample chamber to atmospheric pressure before the valve could be switched. However, further testing in the laboratory with the backfill sampling technique indicates that an order of magnitude increase in response time can be achieved by the backfill technique over the diffusion technique.

Figure 5.3 is a plot of the change in frequency versus CCl_4 concentration for the concentrations listed in Table 5.1. The responses measured in the presence of steam are plotted as filled circles and the data point measured in dry CO_2 is plotted as an upside-down triangle. Error bars are calculated assuming an uncertainty in the CCl_4 flow rate of 10 mL/min, an uncertainty in the steam and CO_2 flow rates of 100 mL/min, and an uncertainty in the temperature of the CCl_4 of 1°C. Figure 5.3 shows that, within these uncertainties, the sensor responded linearly and has a slope of 0.6 ppm/Hz.

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Next, with a fixed CCl₄ concentration of 16200 ppm in steam (mixture 5 in Table 5.1), the temperature of the quartz reactor was slowly ramped from 330°C to 560°C. The change in SAW frequency and corresponding change in CCl₄ concentration are plotted as a function of temperature in Figure 5.4 with circles representing the data from the SAW sensor. The solid line is the CCl₄ concentration as a function of temperature for pyrolysis as







concentration g corresponds to $\Delta CC1,=0$ of CC1. ACC1 16200 ppm of CC1

reported in the literature, and the dashed line is the literature data translated to the left by 80°C.²¹ The literature data is reported as the weight percent remaining after the pyrolysis process at each temperature; this data was converted to a destruction efficiency and multiplied by our starting concentration to obtain a comparison.

It is clear that the steam reforming process is more efficient than pyrolysis since equal destruction occurs at a temperature that is 80°C lower in the steam reformer than in the pyrolysis chamber. This is due to the addition of steam and the greater resultant thermodynamic driving force associated with the formation of the oxidized products. Examination of the chemical equation that describes the CCl₄ destruction process indicates that addition of H_2O to the system results in more destruction of CCl₄ as LeChatelier's principle says it must.

 $2 H_2O(g) + CCl_4(g) = CO_2(g) + 4 HCl(g)$ 5.1 Additional destruction efficiency may in part be due to the longer residence time of the molecules in the reactor. The pyrolysis literature reports a residence time of at most 6 sec while the Synthetica quartz reactor residence time is on the order of 18 sec with the flows given.

The next large scale test of the sensor occurred during Oct. 13-15, 1992 and was designed to determine the destruction efficiency of the Synthetica moving-bed evaporator (MBEv). A schematic diagram of the system is given in Figure 5.5 and shows



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the MBEv, STD (Synthetica Technologies Detoxifier), SAW sensor, CCl, delivery system, NaOH delivery system, and gas sampling mechanism. The hazardous waste (in this case CCl_4) is introduced directly into the MBEv where it is steam reformed into CO, CO_2 , hydrocarbon fragments, and mineral acids such as HCl.

As depicted in Figure 5.5, the MBEV consists of a bucket elevator system to lift vitreous Al_2O_3 spheres (3/8" diameter) to the top of a hopper where they are coated with NaOH and allowed to settle to the bottom of the hopper. In the entry arm of the hopper, the moving spheres roll through a stream of concentrated NaOH that coats their surfaces. Neutralization of any molecules of mineral acid produced by steam reforming in the MBEV occurs by contact with the NaOH coated spheres. The Al_2O_3 bed consists of about 600 lbs of 3/8" spheres and represents about 50 m² of surface area.

The output gas from the MBEv is pumped by the turbine into the higher temperature (1100°C) reaction chamber of the STD where any remaining organic fragments are converted to CO, CO_2 and H_2O and then vented to the atmosphere. It is absolutely necessary that the total Cl concentration be kept below two weight percent or serious damage will occur to the STD vessel and heating elements. Thus, the SAW sensor was placed on the outlet of the MBEv to monitor the output CCl₄ level from the MBEv and hence determine its efficiency at destroying the CCl₄ and protecting the STD. With the addition of a pH monitoring apparatus, it is possible to assess the efficiency of the MBEv at

removing the CCl₄ from the system and determine the level of acid neutralization provided by the NaOH-coated alumina spheres.

The experimental procedure employed was as follows: first, with the MBEv effluent steam stream at 230°C (a temperature too low to cause CCl, to be destroyed by the steam reforming), CCl_4 was introduced into the system via the metering apparatus while the SAW sensor monitored a sample of gas from the MBEv outlet. The flow rate was kept below 10 ml/min to keep the total Cl concentration below 2%. The gas was sampled by pumping a fresh volume of sample gas from the MBEv outlet with Valves 2 and 3 open and Valve 1 closed. Next, Valve 3 was closed so that the sensor could be evacuated, and the sensor was backfilled with the sample gas by first closing Valve 2 and then opening Valves 3 and 1 in order. A calibration point was measured that indicated a shift of 800 Hz for a calculated concentration of 1228 ppm. From the quartz reactor experiments, it was shown that the frequency response is a linear function of concentration; thus, a scale factor of 1.5 ppm/Hz was found from this calibration measurement. The calibration found for the quartz reactor measurement (0.6 ppm/Hz) was not in agreement with this value and was ignored. This inconsistency was probably due to uncertainties in the steam flow rate in the Synthetica equipment, uncertainties in the CCl₄ flow rate, and errors introduced by condensation of CCl, on the walls of the diffusion tube.

Next, the concentration was changed and the above process was repeated as the temperature of the MBEv steam stream was

increased causing steam destruction of CCl, by steam reforming. These data are tabulated in Table 5.2.

NaOH flow (ml/min)	CCl, flow (ml/min)	CCl Measured (ppm) ^a	CCl. Calculated (ppm) ^b	MBEV Temp. inlet/outlet (°C)	ξ (%) [℃]
0 0 0 50 60 60 120	6.2 4.6 3.0 7.0 8.0 10 20 20	1200 975 300 750 460 300 280 300	1230 924 600 1390 1590 1990 3980 3980	497/236 516/243 517/272 517/272 522/297 514/298 501/283 503/281	98 106 50 54 29 15 7 8

Data collected from MBEv and SAW sensor as Table 5.2. a function of MBEv outlet temperature.

a. At outlet from MBEv.

b. At inlet to MBEv.

c. Percent of inlet CCl_4 remaining at outlet of MBEv.

In Table 5.2, the measured CCl, level at the MBEv outlet is the product of the measured frequency change in Hz and the scale factor 1.5 discussed earlier. The calculated CCl, concentration at the MBEv inlet is derived as follows:

$$ppm CCl_{\bullet} = \frac{R \rho}{MW \beta \rho} \times 10^{6}$$
(5.1)

is the density of CCl_4 (1.6 g/ml), ρ where MW is the molecular weight of CCl, (154 g/mole), β is the measured steam flow rate in the machine (45 scfm), R is the CCl, flow rate into the machine, and p is the conversion ratio from scfm to moles steam (1.16 moles/scfm).

The parameter ξ in Table 5.2 is the ratio of the measured CCl, outlet level to the calculated CCl, inlet level converted to percent. This parameter is plotted as a function of MBEv outlet temperature in Figure 5.6 and shows a large decrease with increasing temperature. This decrease indicates that CCl, is being steam reformed into its reaction products, and that the MBEv is capable of destroying at least 90% of the input CCl, at MBEv inlet and outlet temperatures of 500°C and 300°C respectively. As indicated in the table, the inlet temperature of the MBEv is much higher than the outlet temperature, indicating that the CCl, is actually exposed to a bed whose temperature is somewhere between the temperature of the inlet and the outlet. Measurement of the actual moving-bed temperature is difficult due to the constant motion of the spheres; thus, the destruction temperature of CCl, in the MBEv can only be inferred from the inlet and outlet temperatures.

A significant limitation of the destruction capacity of the MBEv is the heating capacity of the steam from the STD. The MBEv has no steam generation or heating capacity of its own and therefore is heated by steam passing from the STD through the turbines to the MBEv. As indicated in Table 5.2, the temperature of both the inlet and outlet of the MBEv began to drop with increasing NaOH and CCl_4 flow. This was due to the cooling caused by the water in the NaOH solution evaporating as it came in contact with the hot spheres. While NaOH (MP=318°C) is molten

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Figure 5.6. Measured CCl, outlet concentration normalized to calculated CCl, inlet concentration as a function of MBEv outlet temperature.

in the MBEv, too much cooling of the bed will cause the NaOH to freeze and convert the moving bed into a mass of solid NaOH and Al_2O_3 spheres that will jam the machine. To avoid this problem, the experiment was terminated before the MBEv cooled below the melting point of NaOH. This premature termination of the experiment prevented testing with higher flow rates of CCl_4 , that in turn precluded demonstration of higher than 90% destruction.

As indicated in Figure 5.5, the gas sampling tubing had a water trap to prevent moisture from reaching the vacuum pump and destroying it. By measuring the pH of the water in the trap it was possible to qualitatively verify than the MBEv was producing acid (i.e., reforming CCl,) and that the NaOH was neutralizing the acid. Since the trap was clear glass, it was possible to place wide range pH paper in the water and visually monitor the color change, and thus the pH change. At several intervals during the course of the experiment, the H₂O in the trap and the pH paper were changed and fresh samples of steam were pulled through the trap. It was verified that by running either an excess of CCl, or NaOH it was possible to change the pH of the trap water from about 3 with excess CCl, to 8 with excess NaOH in the MBEv. This verifies qualitatively that the MBEv was destroying CCl, and neutralizing the HCl released by the destruction, and that the SAW measurements at high temperatures are correct.

In summary, through various experiments with the SAW sensor,

it was possible to determine the following information about the operation of the SAW sensor and the MBEV. First, from the quartz reactor experiment, the SAW sensor was shown to respond linearly with CCl₄ concentration in moist, corrosive background atmospheres with no discernable damage to the sensor (the SAW devices used in this system were constructed from 20 Å of Cr and 2000 Å of Au lithographically patterned on a quartz substrate; conventional SAW devices are patterned in Al and would corrode rapidly in this environment). It was also shown that the sensor can measure CCl₄ concentrations in the harsh environment of the full-scale Synthetica system. Finally, it was demonstrated that the Synthetica MBEV can destroy CCl₄ with at lest 90% efficiency at MBEV inlet and outlet temperatures of 500°C and 300°C respectively.

6.0 CONCLUSIONS

In the above discussions, we have described a sensor based on surface acoustic wave technology for the measurement of VOCs in very harsh chemical and physical environments. A mathematical description of the physical phenomena occurring in the SAW device was given to describe the sensing mechanism. Next, a detailed description of the mechanical and electrical construction of the device was presented. This included descriptions of the efforts made to minimize temperature effects on the response of the sensor and calibration of the device as a function of concentration of CCl₄. It was found that the sensor is stable to on the order of 200 Hz over 10 hr periods and that the sensor responds linearly to CCl₄ concentration. A minimum detection limit of about 200 ppm CCl₄ was given for industrial environments even though the sensor can be operated with better stability ir the laboratory.

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Sensor operation was then demonstrated in the harsh physical and chemical environment of the Synthetica MBEv and it was verified that the MBEv can destroy a finimum of 90% CCl₄ at MBEv inlet and outlet temperatures of 500°C and 300°C respectively. From the measurements in the quartz reactor, the sensor was shown to function in a corrosive environment of HCl and steam and that using a vacuum pump to evacuate the SAW case and then backfilling the system with the VOC laden gas was an efficient method of increasing the response time in the sensor.

Finally, several recommendations for improvements in the sensor were suggested. For example, embedding both SAW devices in the same metal case would provide better thermal coupling and reduce the temperature induced fluctuations in the sensor. In addition, adding a voltage follower between the output of the op amp and the current switching transistor in the sensor heater control circuit will reduce the temperature induced feedback that is translated back to the resistance bridge and result in better long term stability of the system. Another recommended modification to the system would be the addition of a electrically actuated three-way valve and a vacuum pump to allow

automatic switching between the sample gas and baseline.

Future additions to this type of system include the development of SAW devices with built-in resistance temperature devices (RTDs) for on chip measurement of the SAW surface temperature and additional interface hardware and software to allow compensation for temperature induced fluctuations in SAW frequency. Finally, integration of the system with a built-in microprocessor would allow stand alone operation and remote sensing via modem and telephone lines.

Future development of high temperature films would allow the direct sensing of gases within hot environments and reduce the problems associated with sensing via diffusion tubes. Currently there are no films that show the required sensitivity at temperatures much higher than the 40°C used here. It is clear that optimization of this type of system will require much further research and development; however, as we have shown, this type of system displays great potential for becoming the standard technology for measurements in harsh environments.

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Mechanical drawing of reference SAW test case. Figure A.1


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Figure A.2 Mechanical drawing of signal SAW test case.



Figure A.3 Mechanical drawing of test case lid.

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Figure A.4 Mechanical drawing of teflon bushing.

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