AIM

Association for Information and Image Management

1100 Wayne Avenue, Suite 1100
Silver Spring, Maryland 20910
301/587-8202

Centimeter

Inches

MANUFACTURED TO AIM STANDARDS
BY APPLIED IMAGE, INC.

1 of 1

SAND 94-8704
Unlimited Release
Printed August 1994

THE MODIFIED EMBEDDED ATOM METHOD

M. I. Baskes
Materials and Process Research Department

ABSTRACT

Recent modifications have been made to generalize the Embedded Atom Method (EAM) to describe bonding in diverse materials. By including angular dependence of the electron density in an empirical way, the Modified Embedded Atom Method (MEAM) has been able to reproduce the basic energetic and structural properties of 45 elements. This method is ideally suited for examining the interfacial behavior of dissimilar materials.

This paper will explain in detail the derivation of the method, show how the parameters of the MEAM are determined directly from experiment or first principles calculations, and examine the quality of the reproduction of the database.

Materials with fcc, bcc, hcp, and diamond cubic crystal structure will be discussed. A few simple examples of the application of the MEAM to surfaces and interfaces will be presented. Calculations of pullout of a SiC fiber in a diamond matrix as a function of applied stress show non-uniform deformation of the fiber.

THE MODIFIED EMBEDDED ATOM METHOD

INTRODUCTION

Empirical and semi-empirical computational methods for metals and covalent materials have evolved rapidly in the past few years. These methods have recently been reviewed by Carlsson (Ref. 13), in an MRS Symposium (Ref. 35), and at a special symposium of the World Materials Congress (Ref. 37). The Embedded Atom Method (EAM) has been the mainstay of the new methods used for metallic systems. Daw and Baskes (Refs. 13, 14) introduced this method over a decade ago. The EAM, which is based on density functional theory, has successfully been applied to the fcc or nearly-filled d-band transition metals and also to bcc metals (Refs. 1, 28). A number of recent papers summarize the technique and its many applications (Refs. 5, 6, 14, 15, 17, 18, 21, 22, 23). Finnis and Sinclair (Ref. 20) derived a similar method based on a second moment approximation to tight binding and originally applied it to the bcc or half-filled d-band transition metals.

A basic limitation of the EAM is that it spherically averages the electron density which precludes directional bonding. Baskes (Ref. 2) modified the EAM to include directional bonding and applied it to silicon. The silicon EAM model was extended by Baskes, et al. (Ref. 8) to the silicon/germanium system where the Modified Embedded Atom Method (MEAM) was developed. Most recently the MEAM has been extended to fcc, bcc (Ref. 3) and hcp (Ref. 7) metals. Savino et al. (Ref. 34) developed a related method based on second order invariants.

All of these methods are mathematically similar and have in common the attribute that the interaction between two atoms depends upon their local environment. It is mainly this fact that accounts for the success that these methods have had in predicting effects at metallic surfaces where the atomic environment is significantly different from the bulk.

This paper will review the derivation of the MEAM and its application to a large number of elements showing its wide range of applicability. An underlying theme in the development of the method is computational simplicity so that the interactions may be readily used for large molecular dynamics or Monte Carlo simulations. MEAM functions for the Si/C system will be developed and applied to the deformation of a SiC fiber in a diamond matrix.

THEORY

The total energy E of a system of atoms in the Embedded Atom Method (EAM) has been shown (Ref. 16) to be given by an approximation of the form:

$$E = \sum_i \left(F_i(\bar{\rho}_i) + \frac{1}{2} \sum_{j \neq i} \phi_{ij}(R_{ij}) \right) \quad (1)$$

where the sums are over the atoms i and j .^{*} In this approximation, the embedding function F_i is the energy to embed an atom of type i into the background electron density at site i , $\bar{\rho}_i$; and ϕ_{ij} is a pair interaction between atoms i and j whose separation is given by R_{ij} . In the EAM, $\bar{\rho}_i$ is given by a linear supposition of spherically averaged atomic electron densities, while in the Modified Embedded Atom Method (MEAM), $\bar{\rho}_i$ is augmented by angularly dependent terms (Refs. 2, 3, 7, 8). Let us denote the term in brackets in Eq. 1, i.e., the direct contribution to the energy from the i^{th} atom, as E_i . Of course, atom i also indirectly contributes to the energy through its interactions with its neighbors. Then E_i may be written as follows:

$$E_i = F_i(\bar{\rho}_i) + \frac{1}{2} \sum_{j \neq i} \phi_{ij}(R_{ij}). \quad (2)$$

As in Baskes et al. (Refs. 3, 8) consider the case of a homogeneous monatomic solid with interactions limited to first neighbors only. In a specific reference structure (usually the equilibrium structure) for an atom of type i we have:

$$E_i^u(R) = F_i(\bar{\rho}_i^0(R)) + \frac{1}{2} Z_i \phi_{ii}(R) \quad (3)$$

where $\bar{\rho}_i^0(R)$ is the background electron density for the reference structure of atom i , Z_i is the coordination, and R is the nearest neighbor distance. Here $E_i^u(R)$ is the energy per atom of the reference structure as a function of nearest neighbor distance, obtained, e.g., from first principles calculations or the universal equation of state of Rose et al. (Ref. 32). Here we choose the latter:

$$E_i^u(R) = -E_c \left(1 + \alpha \left(\frac{R}{r_c} - 1 \right) \right) e^{-\alpha(R/r_c - 1)} \quad (4)$$

with

$$\alpha^2 = \frac{9\Omega B}{E_c} \quad (5)$$

where E_c , r_c , Ω , and B are the cohesive energy, nearest-neighbor distance, atomic volume, and bulk modulus, respectively, all evaluated at equilibrium in the reference state. The pair potential for like atoms is then given by:

$$\phi_{ii}(R) = \frac{2}{Z_i} \{ E_i^u(R) - F_i(\bar{\rho}_i^0(R)) \}. \quad (6)$$

At equilibrium (denoted by $|_e$), with the use of this form, $E|_e = -E_c$, $dE/dr|_e = 0$, and $d^2E/dr^2|_e = 9\Omega B/r_c^2$ so that agreement is assured between the model and the input cohesive energy, atomic volume, and bulk modulus. The pair potential for unlike atom pairs will be discussed below.

^{*}Throughout the paper the subscripts (i,j,k) denote either an atom at a particular site or the type of that atom.

In the MEAM the embedding function $F(\rho)$ is taken as

$$F(\rho) = AE_c \frac{\rho}{\rho_0} \ln \frac{\rho}{\rho_0} \quad (7)$$

where A is an adjustable parameter and ρ_0 is a density scaling parameter. See Baskes et al. (Refs. 2, 8) for a justification of this functional form. The density scaling parameter was initially (Ref. 3) taken to be the coordination times the atomic density scaling factor (see Eq. 11 below) and more recently (Ref. 7) as the density in the equilibrium reference structure. The SiC calculations presented below use the latter definition. For fcc and bcc reference lattices the definitions are identical. For hcp and diamond cubic there is a small difference.

The background electron density, $\bar{\rho}$, is assumed to be a function of what we call partial electron densities. These partial electron densities contain the angular information in the model. The reader should be cautioned that even though the electron density may be thought of as qualitatively similar to a real electron density, there is no expectation that the electron densities calculated here would be in agreement with those obtained from first principles calculations. A number of functional forms have been used previously (Refs. 2, 3, 4, 7, 8).

For example, the square of the electron density at a given site has previously (Ref. 3) been defined as the sum of terms with s, p, d, and f symmetry from the neighboring atoms. By including these angular terms in the background electron density, we introduce angular forces into the model. Thus at a particular atom:

$$\bar{\rho}^2 = \sum_{h=0}^3 t^{(h)} \rho^{(h)2} \quad (8)$$

with $h=0$ to 3 corresponding to s, p, d, and f symmetry, respectively, and for convenience we take $t^{(0)} = 1$. We note that in a crystal the s, p, d, and f terms may be considered as measures of volume, polarization, shear, and lack of inversion symmetry, respectively. For example, as we vary the volume of a perfect fcc lattice, only $\rho^{(0)}$ is not equal to zero and, thus, it can simply be related to the volume. Similarly as we shear the fcc lattice, contributions from $\rho^{(2)}$ arise. An alternative exponential form with the same asymptotic behavior near the perfect lattice has also been used (Ref. 7):

$$\bar{\rho} = \rho^{(0)} e^{\left[\frac{1}{2} \sum_{h=1}^3 t^{(h)} (\rho^{(h)} / \rho^{(0)})^2 \right]}. \quad (9)$$

The contributions to the density are given by:

$$\rho^{(0)} = \sum_i \rho^{a(0)}(r^i) \quad (10a)$$

$$(\rho^{(1)})^2 = \sum_{\alpha} \left[\sum_i \rho^{a(1)}(r^i) \frac{r_{\alpha}^i}{r^i} \right]^2 \quad (10b)$$

$$(\rho^{(2)})^2 = \sum_{\alpha, \beta} \left[\sum_i \rho^{a(2)}(r^i) \frac{r_{\alpha}^i r_{\beta}^i}{r^{i2}} \right]^2 - \frac{1}{3} \left[\sum_i \rho^{a(2)}(r^i) \right]^2 \quad (10c)$$

$$(\rho^{(3)})^2 = \sum_{\alpha, \beta, \gamma} \left[\sum_i \rho^{a(3)}(r^i) \frac{r_{\alpha}^i r_{\beta}^i r_{\gamma}^i}{r^{i3}} \right]^2. \quad (10d)$$

Here, the $\rho^{a(h)}$ are radial functions which represent the decrease in the contribution with distance r^i from the site in question, the superscript i indicates neighboring atoms to the site in

question, and the α , β , and γ summations are each over the three coordinate directions with r_i being the distance from the site in question in that direction. The functional forms for the partial electron densities ($h=1,3$) were chosen to be translationally and rotationally invariant and equal to zero for crystals with cubic symmetry. Finally, the individual contributions are assumed to decrease exponentially, i.e.,

$$\rho^{a(h)}(R) = \rho_0^a e^{-\beta^{(h)}(R/r_e-1)} \quad (11)$$

where ρ_0^a and $\beta^{(h)}$ are constants. For alloys the coefficients $t^{(h)}$ were initially assumed to depend on the properties of the atom at which the average electron density was calculated (Ref. 3). More recently (Ref. 26) it has been found that better agreement with defect properties in SiC could be obtained if the properties of the atoms surrounding this atom were included. The latter averaging procedure is used below for the SiC calculations:

$$t^{(h)} = \sum_i t_i^{(h)} \rho_0^{a(0)} / \rho^{(0)}. \quad (12)$$

The determination of the parameters has been discussed previously (Refs. 3, 7) in great detail. Basically, analytic expressions are obtained for the elastic constants, vacancy formation energy, and structural energy differences. Using these expressions and experimental data each parameter (or sets of parameters) is uniquely defined. The relationship between the parameters and the experimental input is summarized in Table 1. A number of the β parameters are not well determined for the fcc and bcc elements and nominal values are chosen for convenience.

Table 1 - Parameters and their correspondence with experimental data.

Parameter	Experimental Input
E_c	sublimation energy
r_e	atomic volume
α	bulk modulus
A	structural energy difference
$\beta^{(0)}, t^{(2)}$	shear moduli
$\beta^{(1)}$	internal relaxation for hcp
$\beta^{(2)}, \beta^{(3)}$	shear moduli for hcp
$t^{(1)}$	vacancy formation energy
$t^{(3)}$	stacking fault energy or c/a for hcp

We begin this section with a diagram of the periodic table that shows the elements for which MEAM functions have been developed. These functions are now available for the 45 elements denoted by shaded squares. As you can see in Fig. 1, MEAM functions exist for most of the

Figure 1 - Periodic table of the elements showing the current status of MEAM function development. The shaded squares denote those elements for which MEAM functions currently exist.

technologically interesting elements. We are currently in the process of developing MEAM functions for the additional elements whose names are present in unshaded squares. It should be noted that once functions are available for a pair of elements it is relatively simple to produce the pair interactions so that alloy or compound calculations may be performed. We have no current interest in the other elements in the table.

To examine the ability of the MEAM to represent the properties of "real materials" a number of examples are presented. The MEAM parameters used here are those previously presented (Refs. 3, 7). As a first example the surface energy of a number of materials is shown in Fig. 2. Here we plot the calculated surface energy of a (100) face using MEAM vs. the experimental value of an "average" surface extrapolated to 0 K. Also included in the "experimental" data set are calculated estimates from de Boer, et al. (Ref. 19). As the figure shows the agreement with experiment is quite good for most of the elements. In contrast, EAM calculations of surface energy (Ref. 24) are frequently in disagreement with experiment by up to 40%.

As a second example calculated structural energies are shown in Fig. 3. The energies shown in the figure are those obtained after minimization with respect to lattice constant. The calculations are compared to an estimate of structural energy differences derived from analyses of experimental phase diagrams (Ref. 33), stacking fault energies (Refs. 25, 29, 31) and first principles calculations (Ref. 30).

Figure 2 - Comparison of experimental (Refs. 19, 27, 36) and calculated surface energies for various elements. The elements shown are those denoted in Fig. 1.

Figure 3 - Comparison of "experimental" and calculated structural energy differences for various elements. The elements shown are those denoted in Fig. 1. See text for how the "experimental" values are determined. The inset shows the detail for very small energy differences.

The final example is that of a small SiC fiber embedded in a matrix of diamond. The MEAM potentials for silicon and carbon have been modified slightly from those previously developed (Ref. 3) to include use of the exponential form for the average electron density and more recent elastic constants of carbon (Ref. 38). The parameters for the new potentials are given in Table 2.

Table 2 - Parameters for the MEAM

Values listed are the cohesive energy E_c (eV), the equilibrium nearest neighbor distance r_e (Å), the exponential decay factor for the universal energy function α , the scaling factor for the embedding energy A , the exponential decay factors for the atomic densities $\beta^{(i)}$, the weighting factors for the atomic densities $t^{(i)}$, and the density scaling factor ρ_0^a .

	E_c	r_e	α	A	$\beta^{(0)}$	$\beta^{(1)}$	$\beta^{(2)}$	$\beta^{(3)}$	$t^{(1)}$	$t^{(2)}$	$t^{(3)}$	ρ_0^a
C	7.3	1.5	4.3	1.0	4.1	4.2	5.0	3.0	5.0	9.34	-1.0	2.5
	7	4	8									
Si	4.6	2.3	4.8	1.0	4.8	4.8	4.8	4.8	3.3	5.11	-0.8	1.0
	3	5	7									
SiC	6.4	1.8	4.3	-	-	-	-	-	-	-	-	-
	3	9	7									

In Table 3 the elastic constants and vacancy formation energy calculated from these functions are compared to experiment. Agreement is excellent.

Table 3 - Elastic constants and vacancy formation energies for C and Si.

The experimental elastic constants are from Refs. 10 and 38 and the first principles calculations of vacancy formation energies are from Refs. 9, 11, and 12.

	c_{11} (eV/Å ³)		c_{12} (eV/Å ³)		c_{44} (eV/Å ³)		E_{lv}^f (eV)	
	Calc.	Expt.	Calc.	Expt.	Calc.	Expt.	Calc.	Theory
C	6.77	6.74	0.77	0.78	3.66	3.61	6.5	7.2
Si	1.01	1.03	0.41	0.40	0.46	0.48	3.0	5 - 6

The reference state for the alloy system is taken to be the diamond cubic structure (β) of SiC. The parameters for the equation of state are given in Table 2. Using a method similar to that described in Baskes (Ref. 3), the following expression is obtained for the Si-C pair potential:

$$\phi_{SiC} = \frac{1}{4} \{ 2E_{SiC}'' - F_C(\bar{\rho}_C^0) - F_{Si}(\bar{\rho}_{Si}^0) \}, \quad (13)$$

where

$$\bar{\rho}_C^0 = 4\rho_{Si}^{a(0)} e^{\frac{1}{4}t_{Si}^{a(1)}(\rho_{Si}^{a(1)}/\rho_{Si}^{a(0)})^2} \quad (14)$$

and

$$\bar{\rho}_{\text{Si}}^0 = 4\rho_{\text{C}}^{a(0)} e^{\frac{1}{2}t_{\text{C}}^{a(3)}(\rho_{\text{C}}^{a(3)}/\rho_{\text{C}}^{a(0)})^2} \quad (15)$$

The elastic constants of SiC using the above potentials are calculated to be: bulk modulus 1.32 eV/Å³; shear modulus (c44) 1.28 eV/Å³; and second shear modulus 0.89 eV/Å³. By construction the cohesive energy, lattice constant, and bulk modulus agree exactly with experiment.

Calculations were performed to simulate the deformation of a SiC fiber in a diamond carbon (C) matrix. The geometry is shown in Fig. 4. Both the SiC and C are single crystals with a (100) orientation. The computational cell is about 40 Å long in the axial direction which is 11 unit cells of C and 9 unit cells of SiC. Choosing this ratio of cells allows almost perfect matching of the periodicity of the C and SiC. The SiC fiber is 10 Å in diameter. The surrounding C matrix is effectively infinite in radial extent with C atoms outside of a 20 Å diameter cylinder held fixed in space. The fiber/matrix cell is allowed to relax to its minimum energy configuration. To restrict interactions to first neighbors, screening is included as in Ref. 7. Axial forces are then applied to each atom of the SiC fiber and the atoms are allowed to relax to their minimum energy configuration.

Figure 4 - Geometry of the SiC fiber in a C matrix.

In Fig. 5 we see the resulting displacement of the fiber and matrix atoms. The displacement is relatively constant across the fiber and decreases rapidly in the matrix. A significant amount of shear occurs in the fiber near the fiber/matrix interface. If we assume that the applied force, f , acts directly on the matrix, in a continuum model the predicted displacement in the matrix, d , at a radial position, r , is given by:

$$d = \frac{f(R-r)}{A_0 C_{44}} \quad (16)$$

where R is the position of the fixed boundary ($10\text{\AA}$) and A_0 is the area ($6.32\text{\AA}^2/\text{atom}$) that the force acts on. The predicted displacement, as shown in Fig. 5, is in excellent agreement with the atomistic calculation. At the higher forces the displacement near the interface is somewhat under predicted by the continuum calculation.

Figure 5 - Axial displacement as a function of radial position. The fiber boundary is at 5 Å. Curves are labeled by the applied axial force (eV/Å/atom). The solid lines represent the displacements predicted by a rigid continuum model (Eq. 16).

We have investigated the uniformity of displacement in the fiber and matrix. In a continuum model one would expect a uniform displacement field throughout the length of the computational cell. In the atomistic calculation, however, the SiC/C boundary structure varies as we move along the axial direction potentially leading to regions of different adhesion. In essence we have introduced dislocations into the interface to account for the mismatch in the SiC and diamond lattice constants. We clearly see this effect in Fig. 6 (a). Here we show the axial displacement as a function of axial position for the case of the applied axial force of $0.3\text{ eV/\AA/atom}$. Away from the interface the displacements are relatively uniform as expected from continuum theory. However, the displacements in the SiC fiber near the interface ($r=2.75$ and 4.25 Å) vary by almost a factor of two. A region of large displacements occur near the center of the periodic fiber. The fact that the displacements show a minimum at the boundary of the periodic cell is related to the specific boundary structure. At this axial position the C and SiC share a common plane most likely leading to higher adhesion and hence smaller displacements.

Similarly the SiC/C boundary structure varies circumferentially. In Fig. 6 (b) we see that the axial displacements do not vary much as a function of circumferential position away from the boundary, but at $r=4.25$ Å the displacements are significantly smaller at 90° and larger at about 300° .

(a)

(b)

Figure 6 - Axial displacement as a function of (a) axial position and (b) angle for an applied axial force of $0.3 \text{ eV}/\text{\AA}/\text{atom}$. Curves are labeled by radial position (Å). The fiber radius is $5 \text{ \AA}$.

SUMMARY

The derivation of the Modified Embedded Atom Method (MEAM) has been presented including a short description of how its parameters are determined. Surface energies and structural energies of a large number of elements are shown to be in good agreement with experiment. Parameters for new MEAM functions for the Si/C system have been determined that fit the properties of Si, C, and SiC quite well. Calculations of a SiC fiber in a diamond matrix show that application of uniform axial forces to the fiber produces non-uniform axial deformation which has been attributed to variation in adhesion at the SiC/C interface.

REFERENCES

1. Adams, J. B. and Foiles, S. M., "Development of an Embedded Atom Method Potential for a bcc Metal: Vanadium," *Phys. Rev. B*, Vol. 41, No. 6, 1990, pp. 3316-3328.
2. Baskes, M. I., "Application of the Embedded-Atom Method to Covalent Materials: A Semiempirical Potential for Silicon," *Phys. Rev. Lett.*, Vol. 59, No. 23, 1987, pp. 2666-2669.
3. Baskes, M. I., "Modified Embedded-Atom Potentials for Cubic Materials and Impurities," *Phys. Rev. B*, Vol. 46, No. 5, 1992, pp. 2727-2742.
4. Baskes, M. I., Angelo, J. E., and Bisson, C. L., "Atomistic Calculations of Composite Interfaces," *Modelling Simul. Mater. Sci. Eng.*, Vol. 2, No. 3A, 1994, pp. 505-518.
5. Baskes, M. I., Daw, M. S., and Foiles, S. M., "The Embedded Atom Method: Theory and Application," in *Atomic Scale Calculations in Materials Science*, Tersoff, J., Vanderbilt, D., and Vitek, V. (eds.), MRS, Boston, MA, 1988, pp. 31.
6. Baskes, M. I., Foiles, S. M., and Daw, M. S., "Application of the Embedded Atom Method to the Fracture of Interfaces," *J. de Physique*, Vol. C5, 1988, pp. 483-503.
7. Baskes, M. I. and Johnson, R. A., "Modified Embedded Atom Potentials for hcp Metals," *Modelling Simul. Mater. Sci. Eng.*, Vol. 2, No. 1, 1994, pp. 147-163.
8. Baskes, M. I., Nelson, J. S., and Wright, A. F., "Semiempirical Modified Embedded-Atom Potentials for Silicon and Germanium," *Phys. Rev. B*, Vol. 40, No. 9, 1989, pp. 6085-6100.
9. Bernholc, J., *et al.*, "Mechanism of Self-Diffusion in Diamond," *Phys. Rev. Lett.*, Vol. 61, No. 23, 1988, pp. 2689-2692.
10. Brandes, E. A. (ed.), *Smithells Metals Reference Book*, 6th ed., Butterworths, London, 1983.
11. Car, R., *et al.*, "Microscopic Theory of Atomic Diffusion Mechanism in Silicon," *Phys. Rev. Lett.*, Vol. 52, No. 20, 1984, pp. 1814-1817.
12. Car, R., *et al.*, "Microscopic Theory of Impurity - Defect Reactions and Impurity Diffusion in Silicon," *Phys. Rev. Lett.*, Vol. 54, No. 4, 1985, pp. 360-363.
13. Carlsson, A. E., *Beyond Pair Potentials in Elemental Transition Metals and Semiconductors*, Academic Press, Boston, 1990.
14. Daw, M. S., "Embedded Atom Method: Many-Atom Description of Metallic Cohesion," in *Atomistic Simulation of Materials*, Vitek, V. and Srolovitz, D. (eds.), Plenum Publishing Co., Chicago, 1988, pp. 181.
15. Daw, M. S., "Model of Metallic Cohesion: The Embedded-Atom Method," *Phys. Rev. B*, Vol. 39, No. 11, 1989, pp. 7441-7452.
16. Daw, M. S. and Baskes, M. I., "Semiempirical, Quantum Mechanical Calculation of Hydrogen Embrittlement in Metals," *Phys. Rev. Lett.*, Vol. 50, No. 17, 1983, pp. 1285-1288.
17. Daw, M. S. and Foiles, S. M., "Theory of Hydrogen on Metal Surfaces," in *First International Conference on the Structure of Surfaces*, Van Hove, M. A. and Tong, S. Y. (eds.), Berkeley, CA, 1984, Springer-Verlag, Berlin.
18. Daw, M. S., Foiles, S. M., and Baskes, M. I., "The Embedded Atom Method: A Review of Theory and Applications," *Mater. Sci. Rep.*, Vol. 9, No. 7-8, 1993, pp. 251-310.
19. de Boer, F. R., *et al.*, *Cohesion in Metals Transition Metal Alloys*, North-Holland, Amsterdam, 1988.
20. Finnis, M. W. and Sinclair, J. E., "A Simple Empirical N-Body Potential for Transition Metals," *Phil Mag A*, Vol. 50, No. 1, 1984, pp. 45-55.
21. Foiles, S. M., "Calculation of the Surface Segregation of Alloys Using the Embedded Atom Method," in *Surface Segregation and Related Phenomena*, Dowben, P. A. and Miller, A. (eds.), CRC Press, Boca Roton, FL, 1990, pp. 79.

22. Foiles, S. M., Baskes, M. I., and Daw, M. S., "Atomistic Studies of Interfacial Structure and Properties," in *Interfacial Structure, Properties and Design*, Yoo, M. H., Clark, W. A. T., and Briant, C. L. (eds.), Materials Research Society, Pittsburgh, PA, 1988, pp. 343-365.
23. Foiles, S. M., Baskes, M. I., and Daw, M. S., "The Calculation of the Equilibrium Interface Structure of Alloys," *Mat. Sci. Forum*, Vol. 37, 1989, pp. 223-234.
24. Foiles, S. M., Baskes, M. I., and Daw, M. S., "Embedded-Atom-Method Functions for the fcc Metals Cu, Ag, Au, Ni, Pd, Pt, and Their Alloys," *Phys. Rev. B*, Vol. 33, No. 12, 1986, pp. 7983-7991.
25. Hirth, J. P. and Lothe, J., *Theory of Dislocations*, McGraw-Hill, New York, 1968.
26. Huang, H., *et al.*, "Molecular Dynamics Determination β -SiC Using Three Representative Empirical Potentials," *in preparation*.
27. Hyland, B., private communication, 1991.
28. Johnson, R. A. and Oh, D. J., "Analytic Embedded Atom Method Model for bcc Metals," *J. Mater. Res.*, Vol. 4, No. 5, 1989, pp. 1195-1201.
29. Korner, A. and Karnthaler, H. P., "Weak-Beam Study of Glide Dislocations in hcp Cobalt," *Phil. Mag. A*, Vol. 48, No. 3, 1983, pp. 469-477.
30. Legrand, P. B., "Relations Entre La Structure Electronique et la Facilite de Glissement Dans Les Metaux Hexagonaux Compacts," *Phil. Mag. B*, Vol. 49, No. 2, 1984, pp. 171-184.
31. Partridge, P. G., "The Crystallography and Deformation Modes of Hexagonal Close-Packed Metals," *Met. Rev.*, Vol. 12, 1967, pp. 169.
32. Rose, J. H., *et al.*, "Universal Features of the Equation of State of Metals," *Phys. Rev. B*, Vol. 29, No. 6, 1984, pp. 2963-2969.
33. Saunders, N., Miodownik, A. P., and Dinsdale, A. T., "Metastable Lattice Stabilities for the Elements," *CALPHAD*, Vol. 12, No. 4, 1988, pp. 351-374.
34. Savino, E. J., Rao, S., and Pasianot, R., "Beyond the Embedded Atom Interatomic Potential," in *Atomic Scale Calculations in Materials Science*, Tersoff, J., Vanderbilt, D., and Vitek, V. (eds.), Materials Research Society, Boston, MA, 1988, pp. 43.
35. Tersoff, J., Vanderbilt, D., and Vitek, V. (ed.), *Atomic Scale Calculations in Materials Science*, 1988, Materials Research Society, Boston, 1988, pp. 471.
36. Tyson, W. R. and Miller, W. A., "Surface Free Energies of Solid Metals: Estimation From Liquid Surface Tension Measurements," *Surf. Sci.*, Vol. 62, 1977, pp. 267-276.
37. Vitek, V. and Srolovitz, D. (ed.), *Atomistic Simulation of Materials*, Plenum Publishing Co., New York, 1989.
38. Wilks, J. and Wilks, E., *Properties and Applications of Diamond*, Butterworth-Heinemann Ltd., Stoneham, MA, 1991.

UNLIMITED RELEASE

INITIAL DISTRIBUTION

U. S. Department of Energy
Office of Basic Energy Sciences
Attn: A. L. Dragoo
Washington, D.C. 20545

U. S. Department of Energy
Office of Basic Energy Sciences
Attn: Iran L. Thomas
Washington, D.C. 20545

U. S. Department of Energy
Office of Basic Energy Sciences
Attn: R. Gottschall
Washington, D.C. 20545

Alcoa Technical Center
Alloy Technology Division
Alcoa Laboratories
Attn: Barbara O. Hall
Alcoa Center, PA 15069

AT&T Bell Laboratories
Attn: M. J. Cardillo
Murray Hill, NJ 07974

AT&T Bell Laboratories
Attn: Y. C. Chabal
Murray Hill, NJ 07974

AT&T Bell Laboratories
Attn: M. Grabow
Murray Hill, NJ 07974

AT&T Bell Laboratories
Attn: D. R. Hamann
Murray Hill, NJ 07974

AT&T Bell Laboratories
Attn: J. C. Tully
Murray Hill, NJ 07974

Abteilung für Oberflächenchemie
und Katalys
Universität Ulm
Attn: R. J. Behm
Albert-Einstein-Allee 11
D-89081 Ulm/Donau
Federal Republic of Germany

Ames Laboratory
Iowa State University
221 Metals Development
Attn: J. H. Rose
Ames, IA 50011

Argonne National Laboratory
9700 S. Cass Avenue
Attn: D. M. Gruen
Argonne, IL 60439

Argonne National Laboratory
Materials Science Division
Bldg. 212
Attn: Dr. Dieter Wolf
9700 S. Cass Avenue
Argonne, IL 60439

Battelle Pacific Northwest
Attn: R. H. Jones
P. O. Box 999
Richland, WA 99352

Brookhaven National Laboratory
Attn: J. W. Davenport
Upton, Long Island, NY 11973

Brookhaven National Laboratory
Attn: K. Lynn
Upton, Long Island, NY 11973

Brookhaven National Laboratory
Attn: D. Welch
Upton, Long Island, NY 11973

Brown University
Department of Physics
Attn: P. J. Estrup
Providence, RI 02912

Dr. R. Bullough
Materials Development Div.
B552 Harwell Laboratory
Oxfordshire OX11 0RA
England

California Inst. of Technology
Attn: T. C. McGill
Pasadena, CA 91125

California Inst. of Technology
Attn: T. Tombrello
Pasadena, CA 91125

California Inst. of Technology
Attn: W. H. Weinberg
Pasadena, CA 91125

Carnegie-Mellon University
Attn: A. W. Thompson
Dept of Met. Engr. and
Matls. Science
3325 Science Hall
Pittsburgh, PA 15213

Carnegie-Mellon University
Attn: P. Wynblatt
Dept of Met. Engr. and
Matls. Science
3325 Science Hall
Pittsburgh, PA 15213

Cornell University
Laboratory of Atomic and
Solid State Physics
Attn: N. W. Ashcroft
Ithaca, NY 14853-2508

Cornell University
Bard Hall
Attn: J. M. Blakely
Ithaca, NY 14850

Cornell University
School of Chemical Engineering
Attn: Prof. Paulette Clancy
Olin Hall
Ithaca, NY 14853-1501

Cornell University
Dept. of Materials Science
and Engineering
Attn: Prof. S. L. Sass
Ithaca, NY 14853-1501

Eastman Kodak Company
Research Laboratories, Bldg. 81
Attn: Dr. J. R. Lavine
Rochester, NY 14650-2008

EG&G Idaho, Inc.
Attn: Clinton Van Sien
Physics Group, MS 2211
P. O. Box 1625
Idaho Falls, ID 83415

Eindhoven University
Attn: H. H. Brongersma
Dept. of Physics
P. O. Box 513
5600 M. B. Eindhoven
The Netherlands

Fritz-Haber Institut
Faradayweg 4-6
Attn: G. Ertl
1000 Berlin 33
Federal Republic of Germany

Fysik Institut
Odense Universitat
Attn: P. Sigmund
DK-5230 Odense M
Denmark

George Washington University
School of Engineering and Applied Science
Attn: Charles Gilmore
Associate Dean for Research
Tompkins Hall
Washington, D. C. 20052

Georgia Institute of Technology
Attn: Prof. Uzi Landman
225 North Avenue N.W.
Atlanta, GA 30332

Hahn-Meitner-Institut
Attn: H. Wollenberger
Glienicke Str. 100
1000 Berlin 39, West Germany

Harvard University
Attn: Prof. H. Ehrenreich
Cambridge, MA 02183

Harvard University
Attn: Prof. J. R. Rice
Cambridge, MA 02183

Haverford College
Department of Physics
Attn: L. Roelofs
Haverford, PA 19041

IBM
Almaden Research
K33.801
Attn: Farid F. Abraham
650 Harry Road
San Jose, CA 95120-6099

IBM
Fishkill Laboratory
Dept. 62G
Attn: Dr. Cheruvu S. Murthy
Hopewell Junction, NY 12533

IBM.
T. J. Watson Research Center
Attn: P. M. Marcus
P. O. Box 218
Yorktown Heights, NY 10598

IBM.
T. J. Watson Research Center
Attn: J. Tersoff
P. O. Box 218
Yorktown Heights, NY 10598

International School for
Advanced Studies
Attn: E. Tosatti
Strada Costiera 11,
Miramare, Trieste 34014
Italy

Institut für Festkörperforschung
der Kernforschungsanlage
Attn: P. H. Dederichs
Postfach 1913
D-5170 Jülich 1, West Germany

Institut für Festkörperforschung
der Kernforschungsanlage
Attn: H. Ibach
Postfach 1913
D-5170 Jülich 1, West Germany

Institut für Festkörperforschung
der Kernforschungsanlage
Attn: W. Schilling
Postfach 1913
D-5170 Jülich 1, West Germany

Institut für Festkörperforschung
der Kernforschungsanlage
Attn: H. Wenzl
Postfach 1913
D-52405 Jülich, Germany

Institut für Physikalische Chemie
Freie Universität Berlin
Attn: K. Christman
Takush. 3
1000 Berlin 33
Federal Republic of Germany

Institute of Corrosion and Protection of
Metals Academia Sinica
No. 62 Wencui Road
Shenyang 110015
China

Interfaculty Reactor Institute
Attn: Dr. A. van Veen
Mekelweg 15
NL-2629JB, Delft
The Netherlands

International School for
Advanced Studies
Attn: V. Bortolani
I-34014 Trieste
Italy

International School for
Advanced Studies
Attn: Prof. F. Ercolessi
I-34014 Trieste
Italy

International School for
Advanced Studies
Attn: E. Tosatti
I-34014 Trieste
Italy

Iowa State University
Ames Laboratory
Attn: A. E. DePristo
Ames, IA 50011

Iowa State University
Ames Laboratory
Attn: K.-M. Ho
Ames, IA 50011

Iowa State University
Ames Laboratory
Attn: P. A. Thiel
Ames, IA 50011

Lab. Voor Fysische Metaalkunde
Materials Science Centre
Attn: J. Th. M. De Hosson
Groningen, Nijenborgh 18
The Netherlands

Lawrence Berkeley Laboratory
Materials and Molecular Res. Div.
Attn: M. L. Cohen
Berkeley, CA 94720

Lawrence Berkeley Laboratory
Materials and Molecular Res. Div.
Attn: L. M. Falicov
Berkeley, CA 94720

Lawrence Berkeley Laboratory
Materials and Molecular Res. Div.
Attn: G. A. Somorjai
Berkeley, CA 94720

Lawrence Berkeley Laboratory
Materials and Molecular Res. Div.
Attn: S. G. Louie
Berkeley, CA 94720

Lawrence Berkeley Laboratory
Materials and Molecular Res. Div.
Attn: M. A. van Hove
Berkeley, CA 94720

Los Alamos Natl. Scientific Lab.
P. O. Box 1663
Attn: S. P. Chen
Los Alamos, NM 87545

Los Alamos Natl. Scientific Lab.
Attn: Dr. Terry C. Lowe
MST-5, MS G730
P.O. Box 1663
Los Alamos, NM 87545

Los Alamos Natl. Scientific Lab.
P. O. Box 1663
Attn: D. L. Smith
Los Alamos, NM 87545

Los Alamos Natl. Scientific Lab.
P. O. Box 1663
Attn: A. Voter
Los Alamos, NM 87545

Loyola College
Department of Physics
Attn: Prof. R. S. Jones
4501 N. Charles St.
Baltimore, MD 21210

Massachusetts Inst. of Technology
Attn: Prof. A. S. Argon
Cambridge, MA 02139

Massachusetts Inst. of Technology
Attn: Prof. R. M. Latanision
Cambridge, MA 02139

Massachusetts Inst. of Technology
Attn: S. Yip
Cambridge, MA 02139

Max-Planck-Institut für Plasmaphysik
Attn: R. Behrisch
D8046 Garching, West Germany

Max-Planck-Institut für Plasmaphysik
Attn: Professor V. Dose
D-8046 Garching, West Germany

Max-Planck Inst. für Metallforschung
Inst. für Werkstoffwissenschaften
Attn: M. W. Finnis
Seestrass 92
7000 Stuttgart 1
Germany

Max-Planck-Institut für Metallforschung
Institut für Werkstoffwissenschaften
Attn: Prof. Dr. Phil. H. Fischmeister
Seestrass 92
7000 Stuttgart 1
Federal Republic of Germany

Max-Planck-Institut für Metallforschung
Institut für Werkstoffwissenschaften
Attn: Dr. W. Mader
Seestrasse 92
7000 Stuttgart 1
Federal Republic of Germany

Max-Planck-Institut für Metallforschung
Institut für Werkstoffwissenschaft
Attn: Dr. S. Schmauder
Seestrasse 92
70174 Stuttgart
Federal Republic of Germany

Max-Planck-Institut für
Stromungsforschung
Attn: Prof. J. P. Toennies
Bunsenstrasse 10
D-3400 Göttingen
Federal Republic of Germany

Michigan Technological University
Dept. of Metallurgical Engineering
Attn: J. K. Lee
Houghton, MI 49931

Michigan Technological University
Materials Science Department
Attn: J. R. Waber
Houghton, MI 49931

National Institute of Standards
and Technology
Attn: J. M. Rowe, 235/A106
Washington, DC 20234

National Institute of Standards
and Technology
Materials Bldg., Rm. A 113
Attn: R. Thomson
Washington, DC 20234

Naval Research Laboratory
Attn: T. Andreadis, Code 4651
Washington, DC 20375

Naval Research Laboratory
Attn: Michael J. Mehl, Code 4693
4555 Overlook Ave. SW
Washington, DC 20375-5000

Nippon Steel Corporation
Attn: Dr. Tooru Matsumiya
Advanced Materials & Technology Research
Laboratories
1618 Ida. Nakahara-ku
Kawasaki 211, Japan

North Carolina State University
Attn: J. R. Beeler
Raleigh, NC 27607

Northwestern University
Dept. of Mat. Science & Engr.
Attn: Dr. David Seidman
2145 Sheridan Road
Evanston, IL 60201-9990

Oak Ridge National Laboratory
Bldg. 4500S, MS 6114
Box 2008
Attn: W. H. Butler
Oak Ridge, TN 37831-6114

Oak Ridge National Laboratory
P. O. Box X
Attn: H. L. Davis
Oak Ridge, TN 37830

Oak Ridge National Laboratory
P. O. Box X
Attn: J. Noonan
Oak Ridge, TN 37830

Oak Ridge National Laboratory
P.O. Box 2008
Attn: D. Pedraza
Oak Ridge, TN 37831-6069

Oak Ridge National Laboratory
P. O. Box X
Attn: J. O. Steigler
Oak Ridge, TN 37830

Oak Ridge National Laboratory
P. O. Box X
Attn: M. Stocks
Oak Ridge, TN 37830

Oak Ridge National Laboratory
Attn: M. H. Yoo
Metals and Ceramics Division
P. O. Box 2008, Bldg. 5500, MS-6376
Oak Ridge, TN 37830

The Ohio State University
Division of Materials Research
Attn: S. A. Dregia
Columbus, OH 43210-1106

The Ohio State University
Attn: Prof. P. Shewmon
Dept. Matl. Sci. & Engr.
116 W. 19th Ave.
Columbus, OH 43210-1106

The Ohio State University
Physics Department
Attn: J. W. Wilkins
174 West 18th Avenue
Columbus, OH 43210-1106

Oxford University
Attn: Prof. D. G. Pettifor
Parks Road
Oxford OX1 3PH
England

Oxford University
Dept. of Metallurgy and
Science of Materials
Attn: Adrian Sutton
Parks Road
Oxford OX1 3PH
England

Pennsylvania State University
Attn: Barbara J. Garrison
Department of Chemistry
State College, PA 16801

Queens University
Dept. of Physics
Attn: Prof. Malcolm Stott
Stirling Hall
Kingston, Canada K7L 3N6

Rice University
Attn: Prof. R. B. McLellan
Houston, TX 77251

Rutgers University
Dept. of Ceramics
Attn: Dr. S. H. Garofalini
P. O. Box 909
Piscataway, NJ 08854

Stanford University
Department of Chemistry
Attn: Hans C. Anderson
Stanford, CA 94305

Stanford University
Department of Materials Science & Engineering
Attn: Timur Halicioglu
Stanford, CA 94305

Stanford University
Dept. of Chemical Engineering
Attn: Prof. R. Madix
Stanford, CA 94305

Stanford University
Department of Chemistry
Attn: Prof. R. Zare
Stanford, CA 94305

Technical University of Denmark
Attn: J. K. Nørskov
Lab. f. Technical Physics
DK-2800 Lyngby, Denmark

Texas A&M University
Dept. of Nuclear Engineering
Attn: Dr. R. Vijay
College Station, TX 77843

Ukrainian Academy of Sciences
Attn: V. Pokropivny
V. Ogorodnikov
Institute for Problems of Matls. Science
Computer Simulation Laboratory
Ukrainian 252680 KIEV GSP-142

University of Aarhus
Institute of Physics
Attn: F. Besenbacher
DK-8000 Aarhus C, Denmark

University of Aarhus
Institute of Physics
Attn: J. Bottiger
Ivan Stensgaard
DK-8000 Aarhus C, Denmark

University of California
MANE Department
Attn: N. M. Ghoniem
6275 Boelter Hall
Los Angeles, CA 90024

University of California, Irvine
Department of Physics
Attn: Prof. A. Maradudin
Irvine, CA 92717

University of California, Irvine
Department of Physics
Attn: Prof. M. Mills
Irvine, CA 92717

University of California, Los Angeles
Department of Materials Science and
Engineering
Attn: Royoichi Kikuchi
405 Hilgard Avenue
Los Angeles, CA 90024-1595

University of Connecticut
Center for Materials Simulation
Attn: Professor Phil Clapp
97 North Eagleville Rd.
Storrs, CT 06269-3136

University of Connecticut
Department of Metallurgy
Attn: Jonathan A. Rifkin
97 North Eagleville Rd.
U-136
Storrs, CT 06269-3136

University of Copenhagen
H. C. Ørsted Institute
Attn: H. H. Andersen
Universitetsperken 5
DK-2100 Copenhagen, Denmark

University of Illinois
Dept. of Materials Science
Attn: J. B. Adams
Urbana, IL 61801

University of Illinois
Dept. of Mining and Metallurgy
Attn: H. K. Birnbaum
Urbana, IL 61801

University of Illinois
Nuclear Eng. Lab.
Attn: D. Ruzic
103 S. Goodwin Ave., Rm. 214
Urbana, IL 61801

University of Jyväskylä
Department of Physics
Attn: Prof. R. M. Nieminen
Seminaarinkatu 15
SF - 40100 Jyväskylä
Finland

The University of Liverpool
Dept. of Materials Science
and Engineering
Attn: Prof. D. J. Bacon
P. O. Box 147
Liverpool L69 3BX
England

University of Maryland
Department of Physics
Attn: T. Einstein
College Park, MD 20742-4111

University of Maryland
Department of Physics
and Astronomy
Attn: Dr. K. E. Khor
College Park, MD 20742-4111

University of Michigan
Dept. of Mat. Sci. and Engr.
Attn: D. Srolovitz
Ann Arbor, MI 48109-2136

University of Michigan
Dept. of Nuclear Engineering
Attn: G. Was
Ann Arbor, MI 48109

University of Minnesota
Dept. of Chemical Engr. and
Materials Science
Attn: J. R. Chelikowsky
151 Amundson Hall
421 Washington Avenue S. E.
Minneapolis, MN 55455

University of Minnesota
Corrosion Research Center
Attn: W. W. Gerberich
Minneapolis, MN 55455

University of Minnesota
Corrosion Research Center
Attn: R. A. Oriani
Minneapolis, MN 55455

University of New Mexico
Chemistry & Nuclear Eng. Dept.
Attn: Dr. Anil Prinja
Albuquerque, NM 87131

University of Notre Dame
Dept. Physics
Attn: J. Dow
Box E
Notre Dame, IN 46556

University of Pennsylvania
Department of Materials Science
Attn: V. Vitek
Philadelphia, PA 19104

University of Texas
Depts. of Astronomy & Physics
RLM 15-212
Attn: R. Smoluchowski
Austin, TX 78712

University of Trieste
Dipartimento di Fisica Teorica
Attn: Prof. N. Parrinello
I-34014 Trieste
Italy

University of Virginia
Department of Materials Science
Attn: R. A. Johnson
School of Engineering and
Applied Science
Charlottesville, VA 22901

University of Wisconsin
ECE Department
Attn: Prof. W. N. G. Hitchon
Madison, WI 53706

Virginia Commonwealth University
Dept. of Physics
Attn: Prof. P. Jena
Richmond, VA 23284-2000

Virginia Commonwealth University
Dept. of Physics
Attn: Prof. T. McMullen
Richmond, VA 23284-2000

Washington University
Department of Physics
Campus Box 1105
Attn: Anders Carlsson
St. Louis, MO 63130

Washington State University
Dept. of Mechanical and
Materials Engineering
Attn: Richard Hoagland
Pullman, WA 99164-2920

Westinghouse Savannah River Co.
Attn: Khalid Mansour
Scientific Computation Section
Bldg. 773-42A
Aiken, SC 29802

Westinghouse Savannah River Co.
Attn: Ralph Wolf
Scientific Computation Section
Bldg. 773-42A
Aiken, SC 29802

West Virginia University
Dept. of Physics
Attn: Bernard R. Cooper
Morgantown, WV 26506

A. Gonis, L-268
W. E. King, L-356
E. C. Sowa, L-356
M. J. Weber, L-326

MS 0300 R. L. Schwoebel, 12300
MS 0335 R. J. Egan, 1100
MS 0337 A. D. Romig, 1800
MS 0343 J. A. Borders, 1823
MS 0344 D. R. Jennison, 1114
MS 0344 G. L. Kellogg, 1114
MS 0345 B. W. Dodson, 1153
MS 0345 D. Emin, 1153
MS 0346 J. E. Schirber, 1002
MS 0349 S. T. Picraux, 1102

MS 0350	S. M. Myers, 1112	MS 0899	Technical Library Processes
MS 0350	J. S. Nelson, 1112		Dept., 7141 (4)
MS 0350	G. C. Osbourn, 1155	MS 9018	Central Technical Files (3)
MS 0513	H. W. Schmitt, 2000	MS 9022	Mail Distribution, 8533-1/ Technical Library Processes, 7141
MS 0724	D. L. Hartley, 6000	MS 9022	Mail Distribution, 8533-1/ for OSTI (2)
MS 1056	B. L. Doyle, 1111		
MS 1056	W. R. Wampler, 1111		
MS 1326	W. G. Perkins, 6312		
MS 9001	J. C. Crawford, 8000		
	Attn: D. L. Crawford, 1900		
	J. B. Wright, 5300		
	M. E. John, 8100		
	R. J. Detry, 8200		
	W. J. McLean, 8300		
	L. A. Hiles, 8400		
	P. N. Smith, 8500		
	L. A. West, 8600		
	M. T. Dyer, 8800		
MS 9006	E. E. Ives, 5200		
MS 9011	R. Cline, 1952		
MS 9036	W. D. Wilson, 8703		
MS 9052	M. Allendorf, 8361		
MS 9055	C. M. Rohlfing, 8353		
MS 9161	D. C. Chrzan, 8341		
MS 9161	S. M. Foiles, 8341		
MS 9161	R. B. James, 8341		
MS 9161	D. D. Johnson, 8341		
MS 9161	G. D. Kubiak, 8342		
MS 9161	R. H. Stulen, 8342		
MS 9161	W. G. Wolfer, 8341		
MS 9163	W. Bauer, 8302		
MS 9163	K. L. Wilson, 8304		
MS 9214	C. L. Bisson, 8117		
MS 9214	C. F. Melius, 8117		
MS 9401	C. W. Robinson, 8702		
MS 9401	R. C. Wayne, 8700		
	Attn: J. C. F. Wang, 8713		
	G. J. Thomas, 8715		
	J. M. Hruby, 8716		
	G. A. Benedetti, 8741		
	P. E. Nielan, 8742		
	M. L. Callabresi, 8743		
	R. J. Kee, 8745		
	W. A. Kawahara, 8746		
MS 9402	M. W. Perra, 8714		
MS 9402	T. E. Felter, 8715		
MS 9402	S. E. Guthrie, 8715		
MS 9403	J. E. Costa, 8711		
MS 9403	J. E. Angelo, 8712		
MS 9403	M. I. Baskes, 8712 (25)		
MS 9403	N. R. Moody, 8712		
MS 9406	S. L. Robinson, 8412		

DATE

FILMED

10/18/94

END