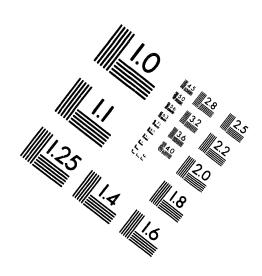
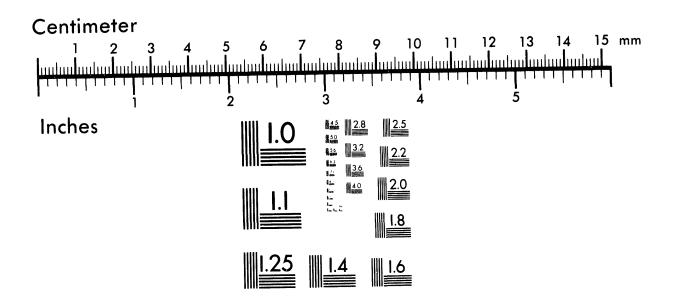


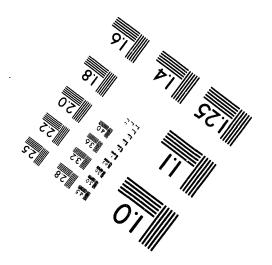


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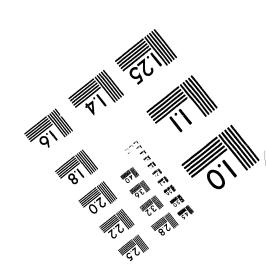
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C. P. EXPUER (2)

Jan. 9, 1944

TURBIDITY FLOCCULATION IN COLUMBIA RIVER WATER

Object:

To study the character of floc obtained by use of commercial aluminum sulfate alone and together with other chemicals designed to improve flocculation.

Equipments

The equipment consisted of a motor driver variable speed laboratory mixer with six agitators. The flocculation tests were performed in one liter beakers.

Procedure:

To series of one liter beakers containing raw Columbia River water were added known amounts of alum, alum and lime, alum and sulfuric acid, alum, sulfuric acid, and lime, and alum and artificial turbidity. The water and chemicals were mixed for fifteen minutes and the character of floc noted.

Results:

1. Enough sulfuric acid was added to the water to give a pH as tabulated before the alum additions:

P.P.H.	Resul	te						
10								
10	No £3	Loc						
	Floc	small	and	light,	settled	very	slowly	
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	Alum	10 Result 10 No find 1	Alum Results 10 No floc 10 No floc 10 Floc small 10 " " " 10 10 " " " 20 20 " "	Alum Results 10 Ho floc 10 Ho floc 10 Floc small and 10 H H 10 H H 20 H 20	Alum Results 10 No floc 10 No floc 10 Floc small and light, 10 " " " " 10 10 " " " " 20 " " " "	Alum Results 10 No floc 10 No floc 10 Floc small and light, settled 10 " " " " " " " " " " " " " " " " " " "	Alum Results 10 No floc 10 No floc 10 Floc small and light, settled very 10 " " " " " " " " " " " " " " " " " " "	Alum Results 10 No floc 10 No floc 10 Floc small and light, settled very slowly 10 " " " " " " " " " " " " " " " " " " "

2. Sandy leas from River bluffs was added in amounts tabulated as follows:

P.P.H. Turba	P.P.M.	Results
2 5 10 20	20 20 20	Small light floo, settleddelowly Slight improvement Fairly large floo, settled faster Very good floe, settled rapidly
20	ĨÓ	N N N N Naturi
40	10	A section of the sect

C. P. Kidder

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3. A filtered solution of Ga(OH), was added to the water in amount tabulated and pH adjusted to 7 with sulfurio acid before the alum additions:

P.P.M. Ga(OH) ₂	P.P.N.	Result	18			
0	10	Small	light	floc.	settled	slowly
20	10	#	Ħ	#	*	W
40	10	41	#		18	¥
40 60 80	10	9	*	*	*	Ħ
80	10	Slight	tly be	tter		•
40	20	Batter	but	still	light an	d fine
40 60	20	Ħ		=	# #	*

4. Calcium Hydroxide, suspended in water, was added to the raw water slong with the Calculated amount of acid necessary to bring the solution to a pH of 7 before the alum additions:

P.P.N. Lime	P.P.W.	Results
0	10	Small light floc, settled slowly
15	10	Slightly better
25	10	Slightly better
50	10	Very marked improvement, settled rapidly
75	10	H H H H

Conclusions:

The following conclusions apply only to the laboratory beaker tests as noted above.

- 1. Commercial alum (0-20 ppm) in raw Columbia River water containing less than 5 ppm of turbidity does not give a satisfactory floc. The floc which is formed settles slowly and probably would not settle out in the conventional type settling basin on a filter plant. It would probably carry over to the sand filters and some of it might pass through the filter into the clean well.
- 2. Then 10 or more p.p.m. of turbidity is added to Columbia river water along with 10 to 20 p.p.m. of alum, a heavy floc is formed which settles rapidly. To refore, it is indicated that when the turbidity in the river is greater than 10 p.p.m., flocculation should be satisfactory.
- 3. The addition of 40 or more p.p.m. of a filtered calcium hydroxide solution to Columbia River water with 20 property. Of almost resulted in a satisfactory flow. The pH of this water was adjusted to 7 with H-SO... An addition of this type does not seem practical on a large scale plant.
- 4. The addition of 50 or sorregions of line in a suspension to Columbia River water along with 10 property of columbia gave a satisfactory floc. The pH of the water was adjusted to 7 with Hosol. It is thought



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that the particles of line act as a moleus around which the floc can form, similar to turbidity.

Recommendations:

Of the four different types of treatment which were tested, the addition of turbidity gave the best results and seemed to be the more practical from an economy and simplicity viewpoint. It is indicated that the plant will have difficulty in the coagulation of turbidities below 5 p.p.m. with alum. If coagulation of such low turbidity is necessary, it is suggested that consideration be given to equipment for feeding at least 20 p.p.m. of turbidity to the water.

R. D. FEARK

By W. R. Conley h.

WRC/jhm

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