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NUMERICAL SIMULATION OF TRANSIENT, INCONGRUENT
VAPORIZATION INDUCED BY HIGH POWER LASER

C.H. Tsai

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MASTER

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INDUCED BY HIGH POWER LASER

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DISCLAIMER

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CONTENTS

	<u>page</u>
1. Introduction	1
2. Mathematical Model	4
2.1 Mass Balance Equation	4
2.2 Energy Balance Equation	5
2.3 Diffusive Flux and Heat Flux	7
2.4 Initial and Boundary Conditions	7
2.5 Diffusion and Heat Conduction Equations	10
2.6 Approximate Conservation Equations	11
2.7 Non-dimensionalization of Conservation Equations	13
Appendix A: Numerical Solution	17
Appendix B: Computer Programs and Sample Analyses	36
Appendix C: Thermodynamic Data and Material Properties of UO_2	84
References	89

Numerical Simulation of Transient, Incongruent Vaporization Induced by High Power Laser

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1. INTRODUCTION

When laser radiation is incident on an absorbing material, all the effects such as phase changes (melting or vaporization), thermal stress, thermal radiation and shock waves from the surface and mass transport in the bulk are associated with the surface temperatures, temperature gradients and composition gradients during the irradiation transient. The laser energy absorption and conduction is considered as a heat transfer process because the laser pulsing time and the characteristic heating time are far longer than the time for electronic relaxation and transfer of energy to the lattice phonos ($\sim 10^{-13}$ sec). In addition to the heat conduction problem, the composition redistribution during the transient due to the incongruent evaporation has to be considered when the target material is a compound.

Although laser surface heating (LSH) technique has been developed for years⁽¹⁻⁴⁾ for thermal physical property investigation of liquid phase urania, there has been no any attempt to calculate the $\sim$ msec thermal evolution of a solid urania subject to intensive laser impingement. Onse et al⁽²⁾ relied mainly on the pyrometric measurement of surface temperature and no surface oxygen depletion has been considered. The only effort they did⁽⁵⁾ for temperature profile calculation was the

adoption of Dabby and Paek⁽⁶⁾ model assuming a prior known steady state surface temperature, called "evaporation temperature" for the purpose of investigating the influence of the sub-surface temperature profile on the pyrometric measurement of surface temperatures (i.e. the difference in emitted thermal radiation intensity between a uniformly heated sample and the one with non-uniform temperature profile). Bober et al relied on firstly the gasdynamic model interpretation⁽³⁾, then the pyrometric measurement⁽⁷⁾. The surface oxygen depletion and oxygen-to-uranium ratio profile was calculated based on the "forced congruent evaporation" model by Breitung⁽⁸⁾ assuming a prior known steady state, uniform temperature and eventually an asymptotic steady state oxygen-to-uranium ratio profile. The basic requirement for the forced congruent evaporation model is the steady state temperature, and it is not applicable to $\sim$ msec transient heating condition because the characteristic time for oxygen diffusion process is in the same order as the temperature evolution and the asymptotic stationary "congruent" evaporation is not obtainable.

In this study, a mathematical model and numerical calculations developed to solve the heat and mass transfer problems specifically for uranium oxide subject to laser irradiation. It can easily be modified for other heat sources or/and other materials. In the uranium-oxygen system, oxygen is the preferentially vaporizing component, and as a result of the finite mobility of oxygen in the solid, an oxygen deficiency is set up near the surface. Because of the bivalent behavior of uranium oxide, the heat transfer problem and the oxygen diffusion problem are coupled and a numerical method of simultaneously solving the two boundary value problems is studied. The temperature dependence of the

thermal properties and oxygen diffusivity, as well as the highly ablative effect on the surface, leads to considerable non-linearities in both the governing differential equations and the boundary conditions. Based on the earlier work done in this laboratory by Olstad and Olander on Iron⁽⁹⁾ and on Zirconium hydride⁽¹⁰⁾, the generality of the problem is expanded and the efficiency of the numerical scheme is improved. The finite difference method, along with some advanced numerical techniques, is found to be an efficient way to solve this problem.

2. MATHEMATICAL MODEL

Consider a semi-infinite slab occupying the region $z \geq 0$, and the slab is irradiated by a pulse of laser beam propagating in z -direction. One-dimensional heat conduction, one-dimensional ablation (no radial liquid movement) with phase change and one-dimensional oxygen diffusion are assumed. The phase change, or fusion, is considered implicitly in the heat capacity term and heat of vaporization term (see Appendix C). Although the laser focal spot size is usually very small, it is still orders of magnitude larger than the characteristic lengths of heat conduction and diffusion in most materials, especially refractory materials like uranium oxide. Thermodynamic equilibrium during transient vaporization due to laser heating is assumed because the decrease in surface stoichiometry can only be quantified based on this assumption. Either penetration of laser radiation (volumetric heating) or the absorbing surface (surface heating) is allowed. This option is designed not only for the different optical properties of the materials subject to laser radiation, but also for the application to other volumetric heating techniques, such as exploding wires, electron beam bombardment heating and neutron pulse heating, which are also proposed for the acquisition of thermochemical data measurements. The vapor plume created by vaporization is assumed to be transparent to the incident laser radiation.

2.1 Mass Balance Equation

Due to the low diffusion coefficient of uranium ions in UO_2

compared with the diffusion coefficient of oxygen ions, the uranium ions are considered to form an immobile lattice through which oxygen can migrate.

The mass balance equation of oxygen in the solid phase is:

$$\frac{\partial C_o}{\partial t} = - \frac{\partial}{\partial z} j_o^d \quad (1)$$

where C_o = mass concentration of oxygen atom in the solid, g/cm^3
 j_o^d = oxygen diffusive flux in the solid, g/cm^2 -sec.

To account for the possibility of significant ablation from the interphase boundary, we make the following coordinate transformation:

$$\begin{cases} t' = t \\ x = z - vt \end{cases} \quad (2)$$

where z is the coordinate from the original surface, v is the ablation (surface recession) velocity.

After the coordinate transformation, the balance equation of oxygen becomes:

$$\frac{\partial C_o}{\partial t} - v \frac{\partial C_o}{\partial x} = - \frac{\partial}{\partial x} j_o^d \quad (3)$$

Neglecting the prime on t' and rearranging the equation, we have:

$$\frac{\partial C_o}{\partial t} = - \frac{\partial}{\partial x} j_o^d + v \frac{\partial C_o}{\partial x} \quad (4)$$

2.2 Energy Balance Equation

The general energy balance equation for the solid, assuming that oxygen is the only mobile component is:

$$\rho \frac{\partial \theta}{\partial t} = - \frac{\partial q}{\partial z} - \frac{\partial}{\partial z} (j_o^d h_o) + q_v \quad (5)$$

where ρ = mass density, g/cm^3

U = specific internal energy, J/g

q = heat flux, W/cm^2

$\bar{h}_o$ = partial specific enthalpy of oxygen, J/g

Q_v = volumetric heat source, W/cm^3 .

For the heat conduction through a condensed, incompressible medium with mobile oxygen and immobile uranium, the specific internal energy U is a function of temperature T and concentration C_o only. Therefore,

$$\frac{\partial U}{\partial t} = \left(\frac{\partial U}{\partial C_o}\right)_T \left(\frac{\partial C_o}{\partial t}\right) + \left(\frac{\partial U}{\partial T}\right)_{C_o} \left(\frac{\partial T}{\partial t}\right) \quad (6)$$

Assuming no mechanical work is done,

$$0 = \dot{H} = (\bar{h}_o C_o + \bar{h}_U C_U) / \rho \quad (7)$$

where $\bar{h}_U$ = partial specific enthalpy of uranium, J/g

C_U = mass concentration of uranium atom in the solid,
 g/cm^3

Since $\bar{h}_U$ and C_U are assumed independent of C_o (based on the assumption that uranium atoms are immobile),

$$\left(\frac{\partial U}{\partial C_o}\right)_T = \bar{h}_o / \rho \quad (8)$$

Similarly,

$$\left(\frac{\partial U}{\partial T}\right)_{C_o} = \left(\frac{\partial H}{\partial T}\right)_{C_o} = C_p \quad (9)$$

where C_p is the constant pressure specific heat in J/g-K . Therefore, equation (6) becomes:

$$\frac{\partial U}{\partial t} = \frac{\bar{h}_o}{\rho} \frac{\partial C_o}{\partial t} + C_p \frac{\partial T}{\partial t} \quad (10)$$

Substituting eq. (10) into eq. (5) and using the coordinate transformation of eq. (2), we have:

$$\rho C_p \frac{\partial T}{\partial t} - \rho C_p v \frac{\partial T}{\partial x} + \bar{n}_o \frac{\partial C_o}{\partial t} - \bar{n}_o v \frac{\partial C_o}{\partial x} = - \frac{\partial q}{\partial x} - \frac{\partial}{\partial x} (j_o^d \bar{n}_o) + Q_v \quad (11)$$

Multiplying eq. (4) by $\bar{n}_o$ and substituting the result into eq. (11) yields:

$$\rho C_p \frac{\partial T}{\partial t} - \rho C_p v \frac{\partial T}{\partial x} = - \frac{\partial q}{\partial x} - j_o^d \frac{\partial \bar{n}_o}{\partial x} + Q_v \quad (12)$$

2.3 Oxygen Diffusive Flux and Heat Flux

Taking into account the Soret and Dufour effects, the mass and heat fluxes are given by:

$$j_o^d = - D_o \frac{\partial C_o}{\partial x} - \frac{D_o Q^* C_o}{RT^2} \frac{\partial T}{\partial x} \quad (13)$$

$$q = - \frac{D_o Q^*}{M_o} \frac{\partial C_o}{\partial x} - k \frac{\partial T}{\partial x} \quad (14)$$

where D_o = diffusion coefficient of oxygen in uranium oxide,

cm^2/sec

Q^* = heat of transport of oxygen in uranium oxide,

J/mole

R = gas constant = 8.314 J/K-mole

M_o = atomic weight of oxygen = 16 g/g-atom

k = thermal conductivity, W/cm-K

2.4 Initial and Boundary Conditions

The two balance equations (4) and (12) are coupled through the two fluxes j_o^d and q . In order to solve those two partial differential equations, we need two initial conditions and four boundary conditions.

The initial conditions are:

$$T(x,0) = T_0 \quad \text{and} \quad C_O(x,0) = C_O^0 \quad @ \quad t=0 \quad (15)$$

where T_0 is the initial temperature before laser impingement

and C_O^0 is the initial mass concentration of oxygen.

The boundary conditions are:

(i) at the moving interphase boundary, $x = 0$:

As a result of preferential vaporization and finite supply rate from the bulk, oxygen is depleted at the surface and a concentration gradient is set up across the inside of the oxide which drives a stream of oxygen atoms towards the surface. The diffusion flux at the surface is given in eq. (13) except all the quantities are evaluated at $x=0$.

The mass balance for oxygen atoms at the interphase boundary gives:

$$j_O^d = j_O^G + vC_O^S \quad @ \quad x=0 \quad (16)$$

where C_O^S is the mass concentration of oxygen atom of the solid at the surface, j_O^G is the total mass flux of oxygen in the gas phase,

$$j_O^G = - (3\phi_{UO_3} + 2\phi_{UO_2} + \phi_{JO} + \phi_O + 2\phi_{O_2}) M_O \quad (17)$$

and v is the surface recession velocity, which is the ratio of the evaporation mass flux of uranium-bearing species j_U^G and the mass concentration of uranium atom in the solid C_U ,

$$v = - j_U^G / C_U = (\phi_{UO_3} + \phi_{JO_2} + \phi_{JO} + \phi_U) M_U / C_U \quad (18)$$

The sign convention is that a flux (mass or heat) is positive if it is in the positive x direction. (Remember that the fluxes are relative to the moving boundary.)

The Langmuir vaporization rate of species i , ϕ_i is:

$$\phi_i = \frac{P_i}{\sqrt{2\pi RT_s} \sqrt{M_i}} \quad \text{mole/cm}^2\text{-sec} \quad (19)$$

where T_s is the surface temperature and P_i is the equilibrium pressure of species i over the solid at the surface composition and temperature. The vacuum vaporization formula of eq. (19) is strictly valid only if the vapor plume in front of the solid is collisionless. However, many theoretical investigations⁽¹¹⁻¹³⁾ have shown that even in a collision-dominated vapor plume, the net vaporization rate is at least 82% of that given by eq. (19) (i.e. the fraction of the forward vaporization flux backscattered to the surface is <18%).

Combining eqs. (13) and (16), we have:

$$-D_o \frac{\partial C_o}{\partial x} \Big|_{x=0} - \frac{D_o Q^* C_o}{RT^2} \frac{\partial T}{\partial x} \Big|_{x=0} = j_o^g + v C_o^s \quad (20)$$

The heat flux in the solid at the interphase boundary is balanced by the heat loss carried by the vaporizing species (ablation), the radiation heat loss and the heat flux input from the laser (for surface absorption only), i.e.,

$$(q)_{x=0} = -\sum_i M_i \phi_i \Delta H_i^v - \epsilon_t \sigma (T_s^4 - T_b^4) + Q_s \quad (21)$$

where ΔH_i^v = heat of vaporization of species i , J/g

T_b = the ambient temperature (usually room temperature), K

ϵ_t = total hemispherical optical emissivity

σ = Stefan-Boltzmann constant

$$= 5.6686 \times 10^{-12} \text{ J/cm}^2\text{-sec-K}^4$$

and Q_s = surface heat source, W/cm².

If the laser radiation is considered to penetrate into the solid, the volumetric heat source term Q_v in the energy equation is nonzero and the surface heat source term Q_s in the boundary condition is zero. If the laser irradiation is considered as a surface source, Q_s is nonzero and Q_v is zero.

Combining eqs. (14) and (21), we have:

$$-\frac{D_o Q^*}{M_o} \frac{\partial C_o}{\partial x} \Big|_{x=0} - k \frac{\partial T}{\partial x} \Big|_{x=0} = - \sum_1 M_1 \phi_1 / H_1^v - \epsilon_t \sigma (T_s^4 - T_b^4) + Q_s \quad (22)$$

The boundary condition eqs. (20) and (22) can be separated to solve for $\partial C_o / \partial x$ and $\partial T / \partial x$ explicitly:

$$\left(\frac{\partial T}{\partial x} \right)_{x=0} = \frac{\sum_1 M_1 \phi_1 / H_1^v + \epsilon_t \sigma (T_s^4 - T_b^4) - Q_s + \frac{Q^*}{M_o} (j_o^g + v C_o^s)}{k - \frac{D_o (Q^*)^2 C_o^s}{M_o R T_s^2}} \quad (23)$$

(i = $UO_3, UO_2, UO, U, O, O_2$)

$$\left(\frac{\partial C_o}{\partial x} \right)_{x=0} = \frac{\frac{k}{D_o} (j_o^g + v C_o^s) - \frac{Q^* C_o^s}{R T_s^2} [\sum_1 M_1 \phi_1 / H_1^v + \epsilon_t \sigma (T_s^4 - T_b^4) - Q_s]}{k - \frac{D_o (Q^*)^2 C_o^s}{M_o R T_s^2}} \quad (24)$$

(ii) at the far end of the solid, $x = \infty$:

$$T(\infty, t) = T_o \quad \text{and} \quad C_o(\infty, t) = C_o^o \quad (25)$$

2.5 Oxygen Diffusion and Heat Conduction Equation

Substituting the fluxes, eqs. (13) and (14), into the partial differential equations (4) and (12), we obtain the two boundary value problems to be solved for T and C_o :

$$\frac{\partial C_o}{\partial t} = \frac{\partial}{\partial x} \left(D_o \frac{\partial C_o}{\partial x} \right) + \frac{D_o Q^* C_o \frac{\partial T}{\partial x}}{RT^2} + v \frac{\partial C_o}{\partial x} \quad (26)$$

$$\frac{\partial T}{\partial t} = \frac{1}{\rho C_p} \frac{\partial}{\partial x} \left(\frac{D_o Q^* C_o}{M_o} \frac{\partial C_o}{\partial x} + k \frac{\partial T}{\partial x} \right) + \left(\frac{D_o}{\rho C_p} \frac{\partial C_o}{\partial x} + \frac{D_o Q^* C_o \frac{\partial T}{\partial x}}{\rho C_p RT^2} \frac{\partial \bar{h}_o}{\partial x} + v \frac{\partial T}{\partial x} + \frac{Q_v}{\rho C_p} \right) \quad (27)$$

$$\text{I.C.: } T(x, 0) = T_o \text{ and } C_o(x, 0) = C_o^o @ t=0 \quad (28)$$

$$\text{B.C.: } \left(\frac{\partial T}{\partial x} \right)_{x=0} = \frac{\sum_i M_i \Phi_i \Delta H_i^v + \epsilon_t \sigma (T_s^4 - T_b^4) - Q_s + \frac{Q^*}{M_o} (j_o^s + v C_o^s)}{k - \frac{D_o (Q^*)^2 C_o^s}{M_o RT_s^2}} \quad (29)$$

$$\left(\frac{\partial C_o}{\partial x} \right)_{x=0} = \frac{\frac{k}{D_o} (j_o^s + v C_o^s) - \frac{Q^* C_o^s}{RT_s^2} [\sum_i M_i \Phi_i \Delta H_i^v + \epsilon_t \sigma (T_s^4 - T_b^4) - Q_s]}{k - \frac{D_o (Q^*)^2 C_o^s}{M_o RT_s^2}} \quad (30)$$

$$T(\infty, t) = T_o \text{ and } C_o(\infty, t) = C_o^o @ x=\infty \quad (31)$$

2.6 Approximate Conservation Equations

In this study, an approximate solution is treated by assuming that Q^* and $\partial \bar{h}_o / \partial x$ are zero; in other words, the thermal diffusion (Soret and Dufour effects) of oxygen is neglected and the oxygen diffusion does not transport any energy. Also, the ablation heat term $\sum_i M_i \Phi_i \Delta H_i^v$ is approximated by $j_{tot} \Delta H_{vap}$, where j_{tot} is the total vaporization flux ($= \sum_i M_i \Phi_i$) and ΔH_{vap} is the heat of vaporization from P_{tot} - T relation (P_{tot} is the total vapor pressure of UO_2).

The boundary value problems become:

$$\frac{\partial C_o}{\partial t} = \frac{\partial}{\partial x} \left(D_o \frac{\partial C_o}{\partial x} \right) + v \frac{\partial C_o}{\partial x} \quad (32)$$

$$\frac{\partial T}{\partial t} = \frac{1}{\rho C_p} \frac{\partial}{\partial x} \left(k \frac{\partial T}{\partial x} \right) + v \frac{\partial T}{\partial x} + \frac{Q_v}{\rho C_p} \quad (33)$$

$$\text{I.C.: } T(x, 0) = T_0 \quad \text{and} \quad C_o(x, 0) = C_o^0 \quad (34)$$

$$\text{B.C.: } \left(\frac{\partial T}{\partial x} \right)_{x=0} = \frac{1}{K} [j_{tot} \Delta H_{vap} + 4_t \sigma (T_s^4 - T_b^4) - Q_s] \quad (35)$$

$$\left(\frac{\partial C_o}{\partial x} \right)_{x=0} = \frac{1}{D_o} (-j_o^G - v C_o^s) \quad (36)$$

$$T(\infty, t) = T_0 \quad \text{and} \quad C_o(\infty, t) = C_o^0 \quad (37)$$

The validity of these approximations has been tested and found to be acceptable.

In reality, it is the oxygen-to-uranium ratio, O/U , which we are interested in, rather than the oxygen concentration C_o . Let us define the oxygen-to-uranium ratio $r = \frac{C_o/M_o}{C_U/M_U}$. Since the concentration of uranium C_U is assumed constant, we can obtain equations for r simply multiplying the equations involving C_o by $\frac{M_U}{M_o C_U}$:

$$\frac{\partial r}{\partial t} = \frac{\partial}{\partial x} \left(D_o \frac{\partial r}{\partial x} \right) + v \frac{\partial r}{\partial x} \quad (38)$$

$$\frac{\partial T}{\partial t} = \frac{1}{\rho C_p} \frac{\partial}{\partial x} \left(k \frac{\partial T}{\partial x} \right) + v \frac{\partial T}{\partial x} + \frac{Q_v}{\rho C_p} \quad (39)$$

$$\text{I.C.: } T(x, 0) = T_0 \quad \text{and} \quad r(x, 0) = r_0 \quad (40)$$

$$\text{B.C.: } \left(\frac{\partial T}{\partial x} \right)_{x=0} = \frac{1}{K} [j_{tot} \Delta H_{vap} + 4_t \sigma (T_s^4 - T_b^4) - Q_s] \quad (41)$$

$$\left(\frac{\partial r}{\partial x} \right)_{x=0} = - \frac{j_o^G}{D_o} \left(\frac{M_U}{C_U M_o} + v r_s \right) \quad (42)$$

$$T(\infty, t) = T_0 \quad \text{and} \quad r(\infty, t) = r_0 \quad (43)$$

where r_s and r_0 are the oxygen-to-uranium ratios of the solid at the surface and in the bulk, respectively.

2.7 Non-dimensionalization of Conservation Equations

Because of the non-linearity of the governing differential equations (38) and (39) and the boundary condition equations (41) and (42), the analytical solutions are not possible. Before the numerical methods are applied, eqs. (38) - (43) have to be non-dimensionalized, because the numerical values in the computations are normally assumed dimensionless and it is obvious that it is meaningless to carry on the operations among the numerical values of different units.

By introducing appropriate characteristic composition $\tilde{r}$, characteristic temperature $\tilde{T}$, characteristic time $\tilde{t}$ and characteristic length $\tilde{x}$, we can define the dimensionless quantities:

$$Y = \frac{r}{\tilde{r}}; \quad \theta = \frac{T}{\tilde{T}}; \quad \tau = \frac{t}{\tilde{t}}; \quad x = \frac{x}{\tilde{x}} \quad (44)$$

In the problem formulated in the beginning of the section, especially for UO_2 , the characteristic quantities were taken as:

$$\tilde{r} = r_o; \quad \tilde{T} = T_o; \quad \tilde{t} = t_{pul}; \quad \tilde{x} = \left(\frac{k t_{pul}}{\rho C_p} \right)_m^{\frac{1}{2}} \quad (45)$$

where r_o = initial oxygen-to-uranium ratio

T_o = initial temperature, K

t_{pul} = effective laser power pulse width

$$= \int P(t) dt / P_{max} = E_{tot} / P_{max}$$

E_{tot} = total laser energy, Joules

P_{max} = maximum laser power, Watts

and "m" means that the quantities are evaluated at the melting temperature.

After the introduction of dimensionless quantities and rearrangement, eqs. (38) - (43) become:

$$\frac{\partial^2 \theta}{\partial x^2} = A_1 \frac{\partial \theta}{\partial \tau} - A_2 \frac{\partial \theta}{\partial x} - A_3 \left(\frac{\partial \theta}{\partial x} \right)^2 - A_4 \quad (46)$$

$$\frac{\partial^2 Y}{\partial x^2} = B_1 \frac{\partial Y}{\partial \tau} - B_2 \frac{\partial Y}{\partial x} - B_3 \left(\frac{\partial Y}{\partial x} \right) \left(\frac{\partial \theta}{\partial x} \right) \quad (47)$$

$$\text{I.C.: } \theta(x, 0) = 1 \text{ and } Y(x, 0) = 1 \quad (48)$$

$$\text{B.C.: } \left(\frac{\partial \theta}{\partial x} \right)_{x=0} = \eta_T \quad (49)$$

$$\left(\frac{\partial Y}{\partial x} \right)_{x=0} = \eta_r \quad (50)$$

$$\theta(\infty, \tau) = 1 \text{ and } Y(\infty, \tau) = 1 \quad (51)$$

$$\text{where } A_1 = \left(\frac{k}{\rho C_p} \right)_m / \left(\frac{k}{\rho C_p} \right)_p$$

$$A_2 = v \sqrt{\frac{1}{t_{\text{pul}}}} \left(\frac{k}{\rho C_p} \right)_m^{\frac{1}{2}} / \left(\frac{k}{\rho C_p} \right)_p$$

$$A_3 = \frac{T_o}{k} \frac{dk}{dT}$$

$$A_4 = \left(\frac{k}{\rho C_p} \right)_m \frac{t_{\text{pul}} Q_v}{k T_o}$$

$$B_1 = \left(\frac{k}{\rho C_p} \right)_m / D_o$$

$$B_2 = v \sqrt{\frac{1}{t_{\text{pul}}}} \left(\frac{k}{\rho C_p} \right)_m^{\frac{1}{2}} / D_o$$

$$B_3 = \frac{T_o}{D_o} \frac{dD_o}{dT}$$

η_T = dimensionless surface temperature gradient

$$= \frac{\sqrt{\frac{1}{t_{\text{pul}}}} \left(\frac{k}{\rho C_p} \right)_m^{\frac{1}{2}}}{k_s} T_o [j_{\text{tot}} \Delta H_{\text{vap}} + 4 \epsilon_t \sigma (T_s^4 - T_b^4) - Q_s]$$

and η_r = dimensionless surface composition gradient

$$= \frac{\sqrt{\frac{1}{t_{\text{pul}}}} \left(\frac{k}{\rho C_p} \right)_m^{\frac{1}{2}} j_{\text{O}}^g}{D_o^s r_o} \left[\frac{C_o}{C_u} + v r_s \right]$$

The standard classical reference on the heat transfer analysis is the book by Carslaw and Jaeger⁽¹⁴⁾, in which a number of exact solutions are given for infinite and semi-infinite solids, solid bounded by two parallel planes, and the heat flow in rectangular, cylindrical and spherical geometries, that are subjected to a variety of initial and boundary conditions. In almost all of the problems for which an exact solution is possible, the thermal properties k , ρ and C_p are taken to be constants, independent of both temperature and position. And also the initial and boundary conditions are in linear form. However, in our case, the considerable non-linearities resulting from the temperature dependent thermal properties, the convectivelike term from coordinate transformation, and the nonlinear boundary conditions resulting from the strong temperature dependent ablative and radiation heat loss terms, make exact solution impossible.

The system of eqs. (46) - (51) is solved numerically by the method described in Appendix A. The most commonly used numerical method for solving this kind of problem is the finite difference method, which has been developed for decades and widely used⁽¹⁵⁻¹⁹⁾. Although it has been found that the finite difference method is not very efficient for highly non-linear problems, it turns out to work quite well with smoothly-varying heat source which drives the transient and with the help of predictor-corrector scheme for constructing a good initial guess for the iteration and the varying time and space increment scheme. A sample calculation for UO_2 is given in Appendix B. The material properties of UO_2 required in the analysis are given in Appendix C and they are permitted to vary with temperature but not with oxygen-to-uranium ratio. The overall effect of the vaporization process is to make the surface of

the urania hypostoichiometric. Although the diffusivity of oxygen in solid UO_{2-x} has not yet been measured, it is almost certainly composition-dependent and larger than that in UO_2 . However, the surface is melted very shortly at the initiation of a high energy laser pulse, and the diffusivity of oxygen in liquid UO_{2-x} is the important quantity. Its value is completely unknown but it is doubtful that concentration effects are as important as temperature effects in the liquid range.

APPENDIX A: NUMERICAL SOLUTION OF EQS. (46) - (51)

Reconstructing eqs. (46) - (51):

$$\frac{\partial^2 \theta}{\partial x^2} = A_1 \frac{\partial \theta}{\partial \tau} - A_2 \frac{\partial \theta}{\partial x} - A_3 \left(\frac{\partial \theta}{\partial x} \right)^2 - A_4 \quad (A-1)$$

$$\frac{\partial^2 Y}{\partial x^2} = B_1 \frac{\partial Y}{\partial \tau} - B_2 \frac{\partial Y}{\partial x} - B_3 \left(\frac{\partial Y}{\partial x} \right) \left(\frac{\partial \theta}{\partial x} \right) \quad (A-2)$$

$$\theta(x, 0) = 1 \quad \text{and} \quad Y(x, 0) = 1 \quad (A-3)$$

$$\left(\frac{\partial \theta}{\partial x} \right)_{x=0} = \eta_T \quad (A-4)$$

$$\left(\frac{\partial Y}{\partial x} \right)_{x=0} = \eta_r \quad (A-5)$$

$$\theta(\infty, \tau) = 1 \quad \text{and} \quad Y(\infty, \tau) = 1 \quad (A-6)$$

where $A_1 = \left(\frac{k}{\rho C_p} \right)_m / \left(\frac{k}{\rho C_p} \right)_p$

$$A_2 = v \sqrt{\frac{1}{t_{pul}}} \left(\frac{k}{\rho C_p} \right)_m^{1/2} / \left(\frac{k}{\rho C_p} \right)_p$$

$$A_3 = \frac{T_o}{k} \frac{dk}{dT}$$

$$A_4 = \left(\frac{k}{\rho C_p} \right)_m \frac{t_{pul}}{k T_o} Q_v$$

$$B_1 = \left(\frac{k}{\rho C_p} \right)_m / D_o$$

$$B_2 = v \sqrt{\frac{1}{t_{pul}}} \left(\frac{k}{\rho C_p} \right)_m^{1/2} / D_o$$

$$B_3 = \frac{T_o}{D_o} \frac{dD_o}{dT}$$

η_T = dimensionless surface temperature gradient

$$= \frac{\sqrt{\frac{1}{t_{pul}}} \left(\frac{k}{\rho C_p} \right)_m^{1/2}}{k_s} T_o \left[\phi_{to} t_{vap} + \epsilon_t \sigma (T_s^4 - T_b^4) - Q_s \right]$$

and η_r = dimensionless surface composition gradient

$$= \frac{\sqrt{\frac{1}{t_{pul}}} \left(\frac{k}{\rho C_p} \right)_m^{1/2}}{D_o^s r_o} \left[\frac{j_o^s}{C_U} + v r_s \right]$$

A.1 Crank-Nicolson Finite Difference Approximation

The Crank-Nicolson finite difference method is an implicit method in which truncated Taylor series expansions are used to approximate the derivatives in the governing differential equations and the space and time derivatives are then replaced by second order correct finite difference representations.

Let $W_{i,n}$ and $U_{i,n}$ denote the approximate solutions of the dimensionless temperature θ and the dimensionless oxygen-to-uranium ratio Y at space (dimensionless) grid x_i (called "grid" hereafter) and time (dimensionless) step τ_n (called "step" hereafter) respectively. Then the Crank-Nicolson method assumes that⁽¹⁷⁾:

$$\begin{aligned}\frac{\partial^2 W}{\partial x^2} \Big|_{i,n+\frac{1}{2}} &= \frac{1}{2} \left(\frac{\partial^2 W_{i,n+1}}{\partial x^2} + \frac{\partial^2 W_{i,n}}{\partial x^2} \right) + O((\Delta \tau)^2) \\ &= \frac{1}{2} \Delta_x^2 (W_{i,n+1} + W_{i,n}) + O((\Delta x)^2 + (\Delta \tau)^2)\end{aligned}\quad (A-7)$$

$$\begin{aligned}\frac{\partial^2 U}{\partial x^2} \Big|_{i,n+\frac{1}{2}} &= \frac{1}{2} \left(\frac{\partial^2 U_{i,n+1}}{\partial x^2} + \frac{\partial^2 U_{i,n}}{\partial x^2} \right) + O((\Delta \tau)^2) \\ &= \frac{1}{2} \Delta_x^2 (U_{i,n+1} + U_{i,n}) + O((\Delta x)^2 + (\Delta \tau)^2)\end{aligned}\quad (A-8)$$

$$\frac{\partial W}{\partial \tau} \Big|_{i,n+\frac{1}{2}} = \frac{W_{i,n+1} - W_{i,n}}{\Delta \tau} + O((\Delta \tau)^2)\quad (A-9)$$

$$\frac{\partial U}{\partial \tau} \Big|_{i,n+\frac{1}{2}} = \frac{U_{i,n+1} - U_{i,n}}{\Delta \tau} + O((\Delta \tau)^2)\quad (A-10)$$

$$\frac{\partial W}{\partial x} \Big|_{i,n+\frac{1}{2}} = \frac{1}{2} \Delta_x (W_{i,n+1} + W_{i,n}) + O((\Delta x)^2)\quad (A-11)$$

$$\frac{\partial U}{\partial x} \Big|_{i,n+\frac{1}{2}} = \frac{1}{2} \Delta_x (U_{i,n+1} + U_{i,n}) + O((\Delta x)^2)\quad (A-12)$$

where $\Delta_x^2 W_{i,n}$ = second order correct centered second

difference of $W_{i,n}$

$\Delta_x^2 U_{i,n}$ = second order correct centered second

difference of $U_{i,n}$

$\Delta_x W_{i,n}$ = second order correct centered first difference

of $W_{i,n}$

$\Delta_x U_{i,n}$ = second order correct centered first difference

of $U_{i,n}$

By utilizing these finite difference operators, the differential equations (A-1) and (A-2) can be approximated by the difference equations: (for $i \geq 2$)

$$\frac{1}{2\Delta x^2}(W_{i,n+1}+W_{i,n}) = A_1 \frac{W_{i,n+1}-W_{i,n}}{\Delta x} - A_2 \left[\frac{1}{2\Delta x}(W_{i,n+1}+W_{i,n}) \right] - A_3 \left[\frac{1}{2\Delta x}(W_{i,n+1}+W_{i,n}) \right]^2 - A_4 \quad (A-13)$$

$$\frac{1}{2\Delta x^2}(U_{i,n+1}+U_{i,n}) = B_1 \frac{U_{i,n+1}-U_{i,n}}{\Delta x} - B_2 \left[\frac{1}{2\Delta x}(U_{i,n+1}+U_{i,n}) \right] - B_3 \left[\frac{1}{2\Delta x}(W_{i,n+1}+W_{i,n}) \right] \left[\frac{1}{2\Delta x}(U_{i,n+1}+U_{i,n}) \right] \quad (A-14)$$

where the coefficients $A_1, A_2, A_3, A_4, B_1, B_2$ and B_3 are, in general, functions of $\{x_i, \tau_{n+1/2}, \frac{1}{2}(W_{i,n+1}+W_{i,n})\}$ and $\tau_{n+1/2} = \frac{1}{2}(\tau_{n+1}+\tau_n)$.

A.2 Second Order Correct Centered Finite Difference Operators

The finite difference method is an implicit method in which a truncated Taylor series expansion is used to approximate the space and time derivatives. These finite difference representations are called finite difference operators; for example, if F_i is the finite difference approximate solution of $f(x)$ at x_i , $\Delta_x F_i$ is the first difference operator of F_i , $\Delta_x^2 F_i$ is the second difference operator of F_i , etc. There are

several ways of expressing the finite difference operators of different degree of order of accuracy, depending on the truncated error from Taylor series expansion. In this section, the second order correct centered finite difference operators are discussed.

For the case of constant increment of x , denoted Δx , the Taylor series expansion of F_{i+1} around F_i is:

$$F_{i+1} = F_i + \Delta x \dot{F}_i + \frac{(\Delta x)^2}{2} \ddot{F}_i + O((\Delta x)^3) \quad (A-15)$$

The Taylor series expansion of F_{i-1} around F_i is:

$$F_{i-1} = F_i - \Delta x \dot{F}_i + \frac{(\Delta x)^2}{2} \ddot{F}_i + O((\Delta x)^3) \quad (A-16)$$

Let's approximate the first derivative $\dot{F}_i$ by the finite difference operator $\Delta_x F_i$. Then $\Delta_x F_i$ can be solved from the equations:

$$\begin{cases} F_{i+1} = F_i + \Delta x (\Delta_x F_i) + \frac{(\Delta x)^2}{2} \ddot{F}_i \end{cases} \quad (A-17)$$

$$\begin{cases} F_{i-1} = F_i - \Delta x (\Delta_x F_i) + \frac{(\Delta x)^2}{2} \ddot{F}_i \end{cases} \quad (A-18)$$

These two equations yields:

$$\Delta_x F_i = \frac{F_{i+1} - F_{i-1}}{2\Delta x} \quad (A-19)$$

which is order of $(\Delta x)^3/(\Delta x) = (\Delta x)^2$ correct.

Similarly, the second difference operator $\Delta_x^2 F_i$ can be solved from the two equations:

$$F_{i+1} = F_i + \Delta x \dot{F}_i + \frac{(\Delta x)^2}{2} \ddot{F}_i + \frac{(\Delta x)^3}{6} \dddot{F}_i \quad (A-20)$$

$$\begin{cases} F_{i-1} = F_i - \Delta x \dot{F}_i + \frac{(\Delta x)^2}{2} \ddot{F}_i - \frac{(\Delta x)^3}{6} \dddot{F}_i \end{cases} \quad (A-21)$$

These two equations yields:

$$\Delta_x^2 F_i = \frac{F_{i+1} - 2F_i + F_{i-1}}{(\Delta x)^2} \quad (A-22)$$

which is order of $(\Delta x)^4 / (\Delta x)^2 = (\Delta x)^2$ correct.

Rather than using a constant Δx as is usually the case, in this program Δx increases geometrically into the bulk of the solid according to the relation:

$$\Delta x_i = \epsilon \Delta x_{i-1} = \epsilon^{i-1} \Delta x_1 \quad (A-23)$$

where ϵ = a constant greater than 1, taken to be 1.035

Δx_i = grid increment following i th grid plane

Δx_1 = the first grid increment at the surface

Then the second order correct differences are considerably more complicated; however, the way of deriving the finite difference operators from the truncated Taylor series expansion is exactly the same.

From the following third order truncated expansions:

$$\begin{cases} F_{i+1} = F_i + \Delta x_1 \epsilon^{i-1} (\Delta_x F_i) + \frac{(\Delta x_1 \epsilon^{i-1})^2}{2} F_{i+1}'' \\ F_{i-1} = F_i - \Delta x_1 \epsilon^{i-2} (\Delta_x F_i) + \frac{(\Delta x_1 \epsilon^{i-2})^2}{2} F_{i-1}'' \end{cases} \quad (A-24)$$

$$\begin{cases} F_{i+1} = F_i + \Delta x_1 \epsilon^{i-1} (\Delta_x F_i) + \frac{(\Delta x_1 \epsilon^{i-1})^2}{2} F_{i+1}'' \\ F_{i-1} = F_i - \Delta x_1 \epsilon^{i-2} (\Delta_x F_i) + \frac{(\Delta x_1 \epsilon^{i-2})^2}{2} F_{i-1}'' \end{cases} \quad (A-25)$$

Solving for $(\Delta_x F_i)$, we have:

$$\Delta_x F_i = \frac{1}{\Delta x_1 \epsilon^{i-1}} \left[\frac{\epsilon^2}{1+\epsilon} F_{i+1} + (\epsilon-1) F_i + \frac{1}{1+\epsilon} F_{i-1} \right] \quad (A-26)$$

which is second order correct.

From the following fourth order truncated expansions:

$$F_{i+2} = F_i + (\epsilon+1) \Delta x_1 \epsilon^{i-1} (\Delta_x F_i) + \frac{(\Delta x_1 \epsilon^{i-1})^2 (1+\epsilon)^2}{2} (\Delta_x^2 F_i) + \frac{(\Delta x_1 \epsilon^{i-1})^3 (1+\epsilon)^3}{6} F_{i+2}''' \quad (A-27)$$

$$F_{i+1} = F_i + \Delta x_1 \epsilon^{i-1} (\Delta_x F_i) + \frac{(\Delta x_1 \epsilon^{i-1})^2}{2} (\Delta_x^2 F_i) + \frac{(\Delta x_1 \epsilon^{i-1})^3}{6} F_{i+1}''' \quad (A-28)$$

$$F_{i-1} = F_i \frac{\Delta x_i}{\Delta x_{i-1}} + \frac{(\Delta x_i)^2}{2\Delta x_{i-1}^2} (\Delta x_{i-1}^2 F_i) - \frac{(\Delta x_i)^3}{6\Delta x_{i-1}^3} F_i \dots \quad (A-29)$$

where $\Delta x_i = \Delta x_{i-1}^{i-1}$.

Solving for $(\Delta x_{i-1}^2 F_i)$, we have:

$$\begin{aligned} \Delta x_{i-1}^2 F_i = & \frac{2}{(\Delta x_{i-1}^{i-1})^2} \left[\frac{\Delta x_i^3 (2+\Delta x_i)}{(1+\Delta x_i)(1+\Delta x_i^2)} F_{i-1} - \frac{\Delta x_i^{2+2\Delta x_i-1} F_i}{1+\Delta x_i} \right. \\ & \left. + \frac{\Delta x_i^{2+\Delta x_i-1} F_{i+1}}{\Delta x_i (1+\Delta x_i)} - \frac{\Delta x_i-1}{\Delta x_i (1+\Delta x_i)(1+\Delta x_i^2)} F_{i+2} \right] \end{aligned} \quad (A-30)$$

Applying these formulae to the variables W and U yields:

$$\Delta x_{i,n} W_{i,n} = \frac{1}{\Delta x_{i-1}^{i-1}} \left[\frac{\Delta x_i^2}{1+\Delta x_i} W_{i-1,n} + (\Delta x_i-1) W_{i,n} + \frac{1}{1+\Delta x_i} W_{i+1,n} \right] \quad (A-31)$$

$$\begin{aligned} \Delta x_{i,n}^2 W_{i,n} = & \frac{2}{(\Delta x_{i-1}^{i-1})^2} \left[\frac{\Delta x_i^3 (2+\Delta x_i)}{(1+\Delta x_i)(1+\Delta x_i^2)} W_{i-1,n} - \frac{\Delta x_i^{2+2\Delta x_i-1} W_{i,n}}{1+\Delta x_i} \right. \\ & \left. + \frac{\Delta x_i^{2+\Delta x_i-1} W_{i+1,n}}{\Delta x_i (1+\Delta x_i)} - \frac{\Delta x_i-1}{\Delta x_i (1+\Delta x_i)(1+\Delta x_i^2)} W_{i+2,n} \right] \end{aligned} \quad (A-32)$$

$$\Delta x_{i,n} U_{i,n} = \frac{1}{\Delta x_{i-1}^{i-1}} \left[\frac{\Delta x_i^2}{1+\Delta x_i} U_{i-1,n} + (\Delta x_i-1) U_{i,n} + \frac{1}{1+\Delta x_i} U_{i+1,n} \right] \quad (A-33)$$

$$\begin{aligned} \Delta x_{i,n}^2 U_{i,n} = & \frac{2}{(\Delta x_{i-1}^{i-1})^2} \left[\frac{\Delta x_i^3 (2+\Delta x_i)}{(1+\Delta x_i)(1+\Delta x_i^2)} U_{i-1,n} - \frac{\Delta x_i^{2+2\Delta x_i-1} U_{i,n}}{1+\Delta x_i} \right. \\ & \left. + \frac{\Delta x_i^{2+\Delta x_i-1} U_{i+1,n}}{\Delta x_i (1+\Delta x_i)} - \frac{\Delta x_i-1}{\Delta x_i (1+\Delta x_i)(1+\Delta x_i^2)} U_{i+2,n} \right] \end{aligned} \quad (A-34)$$

A.3 Initial and Boundary Conditions

In the notations of the approximate solutions, the finite difference approximation to the initial conditions (A-3) are:

$$W_{i,1} = 1 \quad \text{and} \quad U_{i,1} = 1 \quad \text{for every } x_i \text{ at } \tau_1 (=0) \quad (A-35)$$

The finite difference approximation to the boundary conditions (A-4) & (A-5) are: (for $i=1$)

$$W'_{1,n+1} = \eta_T(\tau_{n+1}, W_{1,n+1}, U_{1,n+1}) \quad (A-36)$$

$$U'_{1,n+1} = \eta_r(\tau_{n+1}, W_{1,n+1}, U_{1,n+1}) \quad (A-37)$$

where $W'_{1,n+1}$ = second order correct forward first difference

of $W_{1,n+1}$

$U'_{1,n+1}$ = second order correct forward first difference

of $U_{1,n+1}$

(The boundary conditions do not have to be evaluated at $\tau_{n+\frac{1}{2}}$ as do the governing equations, because time derivatives are involved in the latter but not in the former.)

For the case of constant Δx , the second order correct forward difference can be solved from the two truncated equations:

$$\begin{cases} F_{i+2} = F_i + 2\Delta x F'_i + \frac{(2\Delta x)^2}{2} F''_i \end{cases} \quad (A-38)$$

$$\begin{cases} F_{i+1} = F_i + \Delta x F'_i + \frac{(\Delta x)^2}{2} F''_i \end{cases} \quad (A-39)$$

here F'_i denotes the second order correct forward first difference instead of exact first derivative.

Eliminating $(\Delta x)^2 F''_i$ from the above equations, we have:

$$F'_i = \frac{1}{2\Delta x} (-3F_i + 4F_{i+1} - F_{i+2}) \quad (A-40)$$

Similarly, for the case of varying Δx according to eq. (A-23), the second order forward first difference can be solved from the two truncated equations:

$$\begin{cases} F_{i+2} = F_i + (\Delta x_i + \Delta x_{i+1}) F'_i + \frac{(\Delta x_i + \Delta x_{i+1})^2}{2} F''_i \end{cases} \quad (A-41)$$

$$\begin{cases} F_{i+1} = F_i + \Delta x_i F'_i + \frac{(\Delta x_i)^2}{2} F''_i \end{cases} \quad (A-42)$$

where $\Delta x_i = \Delta x_1^{i-1}$.

The solution comes out to be:

$$F'_1 = \frac{1}{\Delta x_1} \left[-\frac{\Delta+2}{\Delta+1} F_1 + \frac{\Delta+1}{\Delta} F_{i+1} - \frac{1}{\Delta(\Delta+1)} F_{i+2} \right] \quad (A-43)$$

For $i=1$:

$$F'_1 = \frac{1}{\Delta x_1} \left[-\frac{\Delta+2}{\Delta+1} F_1 + \frac{\Delta+1}{\Delta} F_2 - \frac{1}{\Delta(\Delta+1)} F_3 \right] \quad (A-44)$$

Applying these results to the variables W and U yields:

$$W'_{1,n+1} = \frac{1}{\Delta x_1} \left[-\frac{(2+\Delta)}{1+\Delta} W_{1,n+1} + \frac{1+\Delta}{\Delta} W_{2,n+1} - \frac{1}{\Delta(1+\Delta)} W_{3,n+1} \right] \quad (A-45)$$

$$U'_{1,n+1} = \frac{1}{\Delta x_1} \left[-\frac{(2+\Delta)}{1+\Delta} U_{1,n+1} + \frac{1+\Delta}{\Delta} U_{2,n+1} - \frac{1}{\Delta(1+\Delta)} U_{3,n+1} \right] \quad (A-46)$$

A.4 Solutions of Finite Difference Equations

Step-by-step method is used to take care of the time (dimensionless) variation. At each step, eqs. (A-13), (A-14), (A-36) and (A-37) comprise a set of nonlinear equations that are to be solved for $W_{1,n+1}$ and $U_{1,n+1}$ for all i . Only a iterative technique can be used to solve a set of nonlinear equations. The Newton-Raphson's method is adopted in this study to carry on the iterations, which is found to be quite powerful and well convergent. The adoption of the predictor-corrector scheme in finding good starting values for iterations also helps to retain both efficiency and convergency of the program with reasonable size of time steps.

A.4.1 Predictor-corrector Method

The idea of the predictor-corrector method is that, instead of using the results from previous step as the first guess to start the

iteration, the approximate solution at half of the time step is solved by the simpler linear equations (called "predictor") which needs only the informations at previous step and then the "corrector" is used to obtain an approximate solution for the current step as the first guess for the following iteration of more complicated nonlinear equations.

If the difference equation has the form:

$$\begin{aligned} \frac{1}{2\Delta x}(F_{i,n+1} + F_{i,n}) - A_1 \frac{F_{i,n+1} - F_{i,n}}{\Delta t} \\ = \Psi_f[x_i, \tau_{n+\frac{1}{2}}, \frac{1}{2}(F_{i,n+1} + F_{i,n}), \frac{1}{2\Delta x}(F_{i,n+1} + F_{i,n})] \end{aligned} \quad (A-47)$$

then the predictor is: (18)

$$\begin{aligned} \frac{1}{2\Delta x}(F_{i,n+1} + F_{i,n}) - A_1 \frac{F_{i,n+1} - F_{i,n}}{\Delta t} \\ = \Psi_f(x_i, \tau_{n+\frac{1}{2}}, F_{i,n}, \Delta x F_{i,n}) \end{aligned} \quad (A-48)$$

which becomes a linear algebraic problem, because unknowns only appear in the left hand side with linear forms.

After solving $F_{i,n+\frac{1}{2}}$, it follows with the corrector: (18)

$$\begin{aligned} \frac{1}{2\Delta x}(F_{i,n+1} + F_{i,n}) - A_1 \frac{F_{i,n+1} - F_{i,n}}{\Delta t} \\ = \Psi_f(x_i, \tau_{n+\frac{1}{2}}, F_{i,n+\frac{1}{2}}, \Delta x F_{i,n+\frac{1}{2}}) \end{aligned} \quad (A-49)$$

which is also a linear problem. The solution of the corrector, denoted by $F_{i,n+1}^{(0)}$, is then used as the first guess for the iteration:

$$\begin{aligned} \frac{1}{2\Delta x}(F_{i,n+1}^{(k)} + F_{i,n}^{(k)}) - A_1 \frac{F_{i,n+1}^{(k)} - F_{i,n}^{(k)}}{\Delta t} \\ = \Psi_f[x_i, \tau_{n+\frac{1}{2}}, \frac{1}{2}(F_{i,n+1}^{(k-1)} + F_{i,n}^{(k-1)}), \frac{1}{2\Delta x}(F_{i,n+1}^{(k-1)} + F_{i,n}^{(k-1)})] \end{aligned} \quad (A-50)$$

where $k=1,2,\dots$, with repeated iteration, if necessary, to obtain final solution.

The Crank-Nicolson method with and without predictor-corrector scheme has been tested and compared and the result comes out that the predictor-corrector scheme is very powerful.

A.4.2 Iteration Procedure by Newton-Raphson's Method

Let us multiply eqs. (A-4) & (A-5) by Δx_1 and rewrite as the form:

$$f_1(\vec{W}_{n+1}) = [W'_{1,n+1} - \eta_T(\tau_{n+1}, W_{1,n+1})] \Delta x_1 = 0 \quad (A-51)$$

$$g_1(\vec{U}_{n+1}) = [U'_{1,n+1} - \eta_r(\tau_{n+1}, U_{1,n+1})] \Delta x_1 = 0 \quad (A-52)$$

Let us also multiply eqs. (A-1) & (A-2) by $2(\Delta x_1)^2$ and rewrite as the form:

$$f_i(\vec{W}_{n+1}) = \frac{(\Delta x)^2}{x} (W_{1,n+1} + W_{i,n}) - 2A_1 \frac{W_{1,n+1} - W_{i,n}}{\Delta \tau_n} + A_2 \frac{(\Delta x)^2}{x} (W_{1,n+1} + W_{i,n}) + \frac{A_3}{2} \frac{(\Delta x)^2}{x} (W_{1,n+1} + W_{i,n})^2 + 2A_4 (\Delta x_1)^2 = 0 \quad (A-53)$$

$$g_i(\vec{U}_{n+1}) = \frac{(\Delta x)^2}{x} (U_{1,n+1} + U_{i,n}) - 2B_1 \frac{U_{1,n+1} - U_{i,n}}{\Delta \tau_n} + B_2 \frac{(\Delta x)^2}{x} (U_{1,n+1} + U_{i,n}) + \frac{B_3}{2} \frac{(\Delta x)^2}{x} (W_{1,n+1} + W_{i,n}) \{ \frac{(\Delta x)^2}{x} (U_{1,n+1} + U_{i,n}) \} (\Delta x_1)^2 = 0 \quad (A-54)$$

where $\vec{W}_{n+1}$ is a vector of $[W_{1,n+1}, W_{2,n+1}, \dots, W_{m,n+1}]$,

$\vec{U}_{n+1}$ is a vector of $[U_{1,n+1}, U_{2,n+1}, \dots, U_{m,n+1}]$,

and m is the number of grid beyond which the profiles are essentially constant.

Therefore, we have a set of $2m$ nonlinear equations to be solved for

$\vec{W}_{n+1}$ and $\vec{U}_{n+1}$.

$$\left\{ \begin{array}{l} f_1(\vec{W}_{n+1}) = 0 \\ f_2(\vec{W}_{n+1}) = 0 \\ \cdot \\ \cdot \\ \cdot \\ f_m(\vec{W}_{n+1}) = 0 \\ g_1(\vec{U}_{n+1}) = 0 \\ g_2(\vec{U}_{n+1}) = 0 \\ \cdot \\ \cdot \\ \cdot \\ g_m(\vec{U}_{n+1}) = 0 \end{array} \right. \quad (A-55)$$

To do so, we have to start with an initial guess solutions $\vec{W}_{n+1}^{(0)}$ and $\vec{U}_{n+1}^{(0)}$, from predictor-corrector scheme described in the last section for example, and then find the successive new approximate solutions following the procedure:

- (i) For the first iteration of W , $W_{i,n+1}$ and $U_{i,n+1}$ in the coefficient terms of eqs. (A-51) - (A-54) are approximated by $W_{i,n+1}^{(0)}$ and $U_{i,n+1}^{(0)}$ and we can solve the linear equation: (19)

$$\vec{J}(\vec{W}_{n+1}^{(0)}) (\vec{W}_{n+1}^{(1)} - \vec{W}_{n+1}^{(0)}) + \vec{F}(\vec{W}_{n+1}^{(0)}) = 0 \quad (A-56)$$

where $\vec{J}$ is an $m \times m$ matrix with elements $\Delta_{ij} = \left(\frac{\partial f_i}{\partial W_{j,n+1}} \right)_{\vec{W}_{n+1}^{(0)}}$ if the

derivative can be obtained analytically; otherwise, the derivative is approximated by the Steffenson's method, (19)

$$\Delta_{ij} = \frac{f_i(\vec{W}_{n+1}^{(0)} + h_j \vec{e}_j) - f_i(\vec{W}_{n+1}^{(0)})}{h_j}$$

$$h_j = f_j(\vec{W}_{n+1}^{(0)})$$

$$\vec{h} = [h_1, h_2, \dots, h_m]$$

$$\text{and } \vec{F}(\vec{W}_{n+1}^{(0)}) = [f_1(\vec{W}_{n+1}^{(0)}), f_2(\vec{W}_{n+1}^{(0)}), \dots, f_m(\vec{W}_{n+1}^{(0)})]$$

The equation is then solved for $\vec{W}_{n+1}^{(1)}$ by the Gaussian elimination method.

- (ii) For the first iteration of U, every $W_{i,n+1}$ is approximated by $W_{i,n+1}^{(1)}$ and $U_{i,n+1}$ in B_1, B_2, B_3 and η_r is approximated by $U_{i,n+1}^{(0)}$ and we can solve the linear equation:

$$\vec{L}(\vec{U}_{n+1}^{(0)}, \vec{q})(\vec{U}_{n+1}^{(1)} - \vec{U}_{n+1}^{(0)}) + \vec{G}(\vec{U}_{n+1}^{(0)}) = 0 \quad (\text{A-57})$$

where $\vec{L}$ is an $m \times m$ matrix with elements $L_{ij} = \left(\frac{\partial g_i}{\partial U_{j,n+1}} \right)_{\vec{U}_{n+1}^{(0)}}$ if the

derivative can be obtained analytically; otherwise, the derivative is approximated by the Steffenson's method,

$$L_{ij} = \frac{g_i(\vec{U}_{n+1}^{(0)} + q_j \vec{e}_j) - g_i(\vec{U}_{n+1}^{(0)})}{q_j}$$

$$q_j = g_j(\vec{U}_{n+1}^{(0)})$$

$$\vec{q} = [q_1, q_2, \dots, q_m]$$

$$\text{and } \vec{G}(\vec{U}_{n+1}^{(0)}) = [g_1(\vec{U}_{n+1}^{(0)}), g_2(\vec{U}_{n+1}^{(0)}), \dots, g_m(\vec{U}_{n+1}^{(0)})]$$

The equation is again solved for $\vec{U}_{n+1}^{(1)}$ by the Gaussian elimination method.

- (iii) The approximate solutions $W_{i,n+1}^{(1)}$ and $U_{i,n+1}^{(1)}$ from the first iteration are then used to calculate solutions for the second iteration.

(iv) The process is repeated until the successive iterations are sufficiently close to each other, and then the calculations go on to the next step after determination of the next step size.

A.4.3 Gaussian Elimination Method

Either the predictor-corrector method or Newton-Raphson's method is a kind of linearization process in order to deal with a nonlinear problem. After linearization, another numerical scheme has to be chosen to solve the system of linear equations, and the Gaussian elimination back-substitution is found to be a very easy and efficient way.

In solving eqs. (A-48), (A-49) in section A.4.1 and (A-56), (A-57) in section A.4.2, we are all dealing with a set of equations with the format like:

$$\begin{aligned}
 b_1 \delta_1 + c_1 \delta_2 + d_1 \delta_3 &= e_1 \\
 a_2 \delta_1 + b_2 \delta_2 + c_2 \delta_3 + d_2 \delta_4 &= e_2 \\
 a_3 \delta_1 + b_3 \delta_2 + c_3 \delta_3 + d_3 \delta_4 &= e_3 \\
 &\dots\dots\dots \\
 a_i \delta_{i-1} + b_i \delta_i + c_i \delta_{i+1} + d_i \delta_{i+2} &= e_i \quad (A-58) \\
 &\dots\dots\dots \\
 a_{m-1} \delta_{m-2} + b_{m-1} \delta_{m-1} + c_{m-1} \delta_m &= e_{m-1} \\
 a_m \delta_{m-1} + b_m \delta_m &= e_m
 \end{aligned}$$

where δ_i denotes either the variables $W_{i, n+\frac{1}{2}}, U_{i, n+\frac{1}{2}}$ in (A-48),

$W_{i, n+1}, U_{i, n+1}$ in (A-49) or $(\vec{W}_{n+1}^{(k)} - \vec{W}_{n+1}^{(k-1)})$ in (A-56), $(\vec{U}_{n+1}^{(k)} - \vec{U}_{n+1}^{(k-1)})$ in

(A-57), a_i, b_i, c_i, d_i denote either the coefficients in (A-48), (A-49) with eqs. (A-31) - (A-34), and (A-45), (A-46), or the elements of Jacobin matrixes $\vec{J}, \vec{L}$ in (A-56), (A-57), and e_i denotes the known quantities

from either the previous step in eqs. (A-48) and (A-49) or the previous iteration in eqs. (A-56) and (A-57).

Note that the primes in b'_{m-1} , c'_{m-1} , a'_m and b'_m are because we approximated δ_{m+1} and δ_{m+2} by a linear extrapolation of δ_{m-1} and δ_m in order to eliminate the number of unknowns to m ,

$$\begin{cases} b'_{m-1} = b_{m-1} - d_{m-1}\epsilon \\ c'_{m-1} = c_{m-1} + d_{m-1}(1+\epsilon) \\ a'_m = a_m - c_m\epsilon - d_m\epsilon(1+\epsilon) \\ b'_m = b_m + c_m(1+\epsilon) + d_m(1+\epsilon\epsilon^2) \end{cases} \quad (A-59)$$

The system of equations (A-58) can be put in the matrix form:

$$\vec{M}\vec{\Delta} = \vec{E} \quad (A-60)$$

where $\vec{\Delta}$ is a column vector $\delta_1, \delta_2, \dots, \delta_m$

$\vec{E}$ is a column vector $e_1, e_2, \dots, e_m$

and $\vec{M}$ is an $m \times m$ quad-diagonal matrix since all elements are zero except those on the principal diagonal, one below and one and two above the principal diagonal.

$$\begin{bmatrix} & & & 0 \\ & & d & \\ & & c & \\ & b & & \\ a & & & \\ 0 & & & \end{bmatrix} \times \begin{bmatrix} \delta_1 \\ \delta_2 \\ \cdot \\ \cdot \\ \cdot \\ \delta_m \end{bmatrix} = \begin{bmatrix} e_1 \\ e_2 \\ \cdot \\ \cdot \\ \cdot \\ e_m \end{bmatrix} \quad (A-61)$$

The idea behind the Gaussian elimination method is to eliminate the

unknowns in a systematic way; the first equation can be used to eliminate δ_1 from the second equation, the new second equation used to eliminate δ_2 from the third equation, and so on, until finally, the new last but one equation can be used to eliminate δ_{m-1} from the last equation, giving one equation with only one unknown δ_m . The unknowns δ_i can then be found in turn by back-substitution. (19)

Generally, after $i-2$ eliminations ($i \geq 2$), we obtain the following two equations for the next elimination:

$$\begin{cases} d_{i-1}\delta_{i-1} + \beta_{i-1}\delta_i + d_{i-1}\delta_{i+1} = S_{i-1} \\ a_i\delta_{i-1} + b_i\delta_i + c_i\delta_{i+1} + d_i\delta_{i+2} = e_i \end{cases} \quad (A-62)$$

where $d_i = b_i$, $\beta_i = c_i$, $S_i = e_i$.

Eliminating δ_{i-1} leads to:

$$\begin{aligned} (b_i - \frac{a_i \beta_{i-1}}{d_{i-1}})\delta_i + (c_i - \frac{a_i d_{i-1}}{d_{i-1}})\delta_{i+1} + d_i\delta_{i+2} \\ = e_i - \frac{a_i S_{i-1}}{d_{i-1}} \end{aligned} \quad (A-63)$$

$$\text{i.e. } d_i\delta_i + \beta_i\delta_{i+1} + d_i\delta_{i+2} = S_i \quad (A-64)$$

with the recursion relations:

$$\begin{cases} d_i = b_i - \frac{a_i \beta_{i-1}}{d_{i-1}} \\ \beta_i = c_i - \frac{a_i d_{i-1}}{d_{i-1}} \\ S_i = e_i - \frac{a_i S_{i-1}}{d_{i-1}} \end{cases} \quad \text{when } 1 \leq i \leq m-2 \quad (A-65)$$

After $m-3$ eliminations, the last three equations are:

$$\begin{cases} d_{m-2}' \delta_{m-2} + \beta_{m-2}' \delta_{m-1} + d_{m-2}' \delta_m = S_{m-2} \\ a_{m-1}' \delta_{m-2} + b_{m-1}' \delta_{m-1} + c_{m-1}' \delta_m = e_{m-1} \\ a_m' \delta_{m-1} + b_m' \delta_m = e_m \end{cases} \quad (A-66)$$

Eliminating δ_{m-2} from the first two equations yields:

$$\begin{cases} d_{m-1}' \delta_{m-1} + \beta_{m-1}' \delta_m = S_{m-1} \\ a_m' \delta_{m-1} + b_m' \delta_m = e_m \end{cases} \quad (A-67)$$

$$\text{where } d_{m-1}' = b_{m-1}' - \frac{a_{m-1}' \beta_{m-2}'}{d_{m-2}'}$$

$$\beta_{m-1}' = c_{m-1}' - \frac{a_{m-1}' d_{m-2}'}{d_{m-2}'}$$

$$S_{m-1} = e_{m-1} - \frac{a_{m-1}' S_{m-2}'}{d_{m-2}'}$$

Eliminating δ_{m-1} from (A-67) yields:

$$d_m' \delta_m = S_m' \quad (A-68)$$

$$\text{where } d_m' = b_m' - \frac{a_m' \beta_{m-1}'}{d_{m-1}'}$$

$$S_m' = e_m - \frac{a_m' S_{m-1}'}{d_{m-1}'}$$

Therefore, the solutions are:

$$\begin{cases} \delta_m = \frac{S_m'}{d_m'} \\ \delta_{m-1} = \frac{S_{m-1}' - \beta_{m-1}' \delta_m}{d_{m-1}'} \\ \cdot \\ \cdot \\ \cdot \\ \delta_i = \frac{1}{d_i'} (S_i' - \beta_i' \delta_{i+1} - d_i' \delta_{i+2}) \quad 1 \leq i \leq m-1 \end{cases} \quad (A-69)$$

A.5 Determination of the Time Increment

The step-by-step methods as described above could be either one-step methods or multi-step methods, depending upon whether the information from previous steps are used in taking the next step. The multi-step methods are more efficient in the sense that generally they require fewer evaluations of the difference operators to achieve a given accuracy. The greater efficiency of the multi-step methods is obtained at the cost of requiring special provisions for changing the step size. Considering the fact that the general shape of the solutions could be closely estimated and the powerful iterational method used here, the multi-step step-by-step method is used in this study and the way of determining the step size is described as the following:

Define the ratio of the truncated second order term to the first order term as:

$$d_f = \frac{(\frac{\Delta t}{\Delta x})^2 \ddot{f}}{(\frac{\Delta t}{\Delta x}) \dot{f}} = \frac{\ddot{f}}{\dot{f}} \quad (\text{A-70})$$

where f is either T or r , $\dot{f}$ and $\ddot{f}$ are first and second order derivative of f , respectively.

In the problem we are dealing with, the composition changes much slower than the temperature, so the time increment is determined by the truncated error ratio for W :

$$d_W = \frac{(\frac{\Delta t}{\Delta x})^2 \ddot{W}}{(\frac{\Delta t}{\Delta x}) \dot{W}} = \frac{(\frac{\Delta t}{\Delta x})_{n+1}^2 \ddot{W}_n - (\frac{\Delta t}{\Delta x})_n^2 \ddot{W}_{n-1}}{(\frac{\Delta t}{\Delta x})_{n+1} \dot{W}_n - (\frac{\Delta t}{\Delta x})_n \dot{W}_{n-1}} = \frac{\left| \frac{W_{n+1} - W_n}{\Delta t_n} - \frac{W_n - W_{n-1}}{\Delta t_{n-1}} \right|}{\left| \frac{W_{n+1} - W_n}{\Delta t_n} \right|} \quad (\text{A-71})$$

And the criterion is that Δt is increased when d_W is less than 10%.

while $\frac{\Delta \psi}{\psi}$ is decreased when d_w is larger than 10%. A maximum of three times the previous step size is imposed. Both upper and lower bounds for the permissible step size are set considered the convergency and efficiency.

A flow chart of the program is shown in Fig. A.1.

Fig. A.1 The Flow Diagram of the Computer Programs "STAR" and "SURFT"

APPENDIX B: COMPUTER PROGRAMS AND SAMPLE ANALYSES

B.1 Computer Programs:

Three computer programs have been developed in this study. The program "STAR" (Surface Temperature And composition Ratio calculation) is for the materials which vaporize incongruently so that the surface composition changes with time coupled with a temperature transient. In this program, one-dimensional time dependent heat conduction and diffusion equations are solved considering melting, ablating moving boundary, ablation and radiation heat losses. The program "SURFT" (SURFace Temperature calculation) is for the materials which vaporize congruently so that no diffusion process is involved and only one-dimensional heat conduction equation with ablating moving boundary and ablation and radiation heat losses is solved. The program "SURFR" (SURFace composition Ratio calculation) is for the incongruently vaporizing materials in which the surface and bulk temperature is kept constant. Only one-dimensional diffusion equation with ablating moving boundary is solved.

All the programs have been developed based on the numerical scheme described in Appendix A and are coded in FORTRAN IV language. Variable dimensioning is used to make optimum usage of the available storage and flexible capacity controllable by the user. The total blank common block storage MTOT has to be greater than

$$NGMAX*(NPDE*7+11)+LSPUL*(3+NPDE)+NPDE*15$$

An error message will be generated and the run will be aborted if MTOT is set too small.

B.2 Program Listing:

-38-

```

C INPUT THE MATERIAL PROPERTIES AND LASER CHARACTERISTICS
$R0052 CALL FRDP
$R0057 GENERATION OF COEFFICIENT TABLE
$R0028 CALL TABLE
$R0038
$R0040 C STEP AND GRID SIZES FOR NUMERICAL STABILITY
$R0041 CALL SIZE
$R0062
$R0043 CALL SHAPE1
$R0044
$R0046 C DEFINE COMMON BLOCK ALLOCATION
$R0046 CALL ALLOC
$R0047
$R0048 C INPUT LASER NORMALIZED PULSE SHAPE
$R0049 CALL SHAPE2(CM1,CIN1)+LSPLU1
$R0070
$R0071 C LOAD CONSTANT VECTORS A'S FOR DIFFERENCE OPERATORS
$R0071 CALL LORDCM2+HNDNM1
$R0072
$R0073 C
$R0074 C CALCULATION TO BE MADE AT TSTOP (IN SEC)
$R0075 READ (5,10) TSTOP,ICFCL
$R0076 IF FOMAT(=210.0)
$R0077 IF TSTOP.GT.(CM1-11) TSTOP=CM1-11
$R0078 C DEFAULT TCYCL FOR SINGLE PULSE IS 100.
$R0078 C WHICH IS ASSUMED TO BE INFINITE.
$R0078 IF (TCYCL.EQ.0.0) TCYCL=100.
$R0080
$R0081
$R0082 C INPUT LASER ENERGY
$R0083 BB READ (5,9) EI
$R0084 IF FOMAT(=10.0)
$R0085 IF (EI.EQ.0.0) CALL EXIT
$R0086 DP=EI*(TPUL*ACFF)
$R0087 DO 1 I=1,LSPLU
$R0088 IM=I-1
$R0089 L C(IM)=DP*(CIN1+IN1*(1.-R1)
$R0090
$R0091
$R0091 C LOAD INITIAL CONDITIONS
$R0092 CALL INITIAL(CM1,CIN1)+MPDE+HNDNM1
$R0093 C CHARACTERISTIC CONSTANTS FOR NON-DIMENSIONALIZATION
$R0094 CALL CHAR(CIN1,CIN1)+MPDE
$R0095
$R0096 C THOM W=0.0
$R0097 DT =DT1
$R0098 IFLAG=0
$R0099 DO 2 J=1,NPDC
$R0100 JL=J-1;JSL=JSL+1
$R0101 J C(IN1(J),CM1)=JL-1
$R0102 DO 3 O=0
$R0103 J TMIN=O.0
$R0104 AMIN=O.0
$R0105 MININ=O.0
$R0106 M=MININ
$R0107 K =-2

```



```

SIZE005 C      COMMON/GRID /DX1,EP5,EP51,EP52
SIZE006 C      COMMON/ITER /ITMAX,CRIT
SIZE007 C      COMMON/STEP /DT1
SIZE009 C
SIZE010 C      READ (5,100) DX1,DX1,EP5
SIZE011 C      WRITE (4,200) DT1,DT1,EP5
SIZE012 C      READ (5,101) ITMAX,CRIT
SIZE013 C      WRITE (4,201) ITMAX,CRIT
SIZE014 C
SIZE015 C      100 FORMAT(2E10.0,10.0)
SIZE016 C      101 FORMAT(11E10.0)
SIZE017 C      200 FORMAT(1H0,23HTHE NUMERICAL CONSTANTS/
SIZE018 C      1      SK,21HTHE FIRST STEP SIZE =,EP,3/
SIZE019 C      2      SK,21HTHE FIRST GRID SIZE =,EP,3/
SIZE020 C      3      SK,21HTHE INCREMENT FACTOR =,EP,3/
SIZE021 C      201 FORMAT(3X,21HMAX NO OF ITERATION =,10/
SIZE022 C      1      3X,21HMAX TOLERANCE =,EP,3)
SIZE023 C
SIZE024 C      RETURN
SIZE025 C      END

```

```

LOAD001 C      SUBROUTINE LOAD(A,NOMAX)
LOAD002 C
LOAD003 C      LOAD CONSTANT VECTORS FOR DIFFERENCE OPERATORS
LOAD004 C      (NOT PROBLEM NOR WATER DEPENDENT)
LOAD005 C
LOAD006 C      COMMON/GRID /DX1,EP5,EP51,EP52
LOAD007 C      DIMENSION A(NMAX*7)
LOAD008 C
LOAD009 C      EP51=1,EP52
LOAD010 C      EP52=1,EP51,EP52
LOAD011 C      I=2
LOAD012 C      A(I,1)=2,EP52,2,EP51/(EP51*EP52)
LOAD013 C      A(I,2)=-2,EP52,2,EP51,EP52/(EP51*EP52)
LOAD014 C      A(I,3)=2,EP52,2,EP51,EP52/(EP51*EP52)
LOAD015 C      A(I,4)=2,EP51,EP52,2,EP51,EP52/(EP51*EP52)
LOAD016 C      A(I,5)=EP51,EP52
LOAD017 C      A(I,6)=EP51,EP52
LOAD018 C      A(I,7)=1,EP51,EP52
LOAD019 C      DO 1 J=1,4
LOAD020 C      DO 1 I=3,NMAX
LOAD021 C      1 A(I,J)=A(I-1,J)/EP52
LOAD022 C      DO 2 J=5,7
LOAD023 C      DO 2 I=3,NMAX
LOAD024 C      2 A(I,J)=A(I-1,J)/EP5
LOAD025 C      I=1
LOAD026 C      A(I,1)=0.0
LOAD027 C      A(I,2)=-EP52,EP51,EP52
LOAD028 C      A(I,3)=EP51,EP52
LOAD029 C      A(I,4)=1,EP51,EP52
LOAD030 C      RETURN
LOAD031 C      END

```

```

SAP101 C      SUBROUTINE SHAPE1
SAP102 C
SAP103 C      READ SDT AND LSPUL
SAP104 C
SAP105 C      COMMON/SDT /N(2),LSPUL
SAP106 C      COMMON/PULS /SDT
SAP107 C
SAP108 C      READ (5,100) SDT,LSPUL
SAP109 C      WRITE (4,200) SDT,LSPUL
SAP110 C      100 FORMAT(1E10,1,110)
SAP111 C      200 FORMAT(1H0,23HSTEP PRINT OUT PARAMETERS/
SAP112 C      1      5X,4HSDT=,EP,2,4H 5CC,5X,4HLSPUL=,23)
SAP113 C
SAP114 C      RETURN
SAP115 C      END

```

```

SAP201 C      SUBROUTINE SHAPE2(EE,TT,LSPUL)
SAP202 C
SAP203 C      LASER NORMALIZED PULSE SHAPE
SAP204 C      (NOT PROBLEM NOR WATER DEPENDENT)
SAP205 C
SAP206 C      COMMON/PULS /SDT
SAP207 C      DIMENSION SH(LSPUL),TT(LSPUL)
SAP208 C
SAP209 C      DO 1 I=1,LSPUL
SAP210 C      1 TT(I)=SDT*FLOAT(I-1)
SAP211 C      READ (5,101) (SH(I),I=1,LSPUL)
SAP212 C      WRITE (4,201)
SAP213 C      WRITE (4,202)
SAP214 C      WRITE (4,203)
SAP215 C      WRITE (4,204) (I,TT(I),I,SH(I),I=1,LSPUL)
SAP216 C
SAP217 C      101 FORMAT(7F10.3)
SAP218 C      201 FORMAT(1H1,55(1H0))
SAP219 C      202 FORMAT(1H 1X,53HISTOGRAM OF TEMPORAL SHAPE OF PULSE FROM POWER Y
SAP220 C      1X,4E1)
SAP221 C      203 FORMAT(1H 55(1H0))
SAP222 C      204 FORMAT(1H 10X,3HTT(,12,3H) =,EB,2X,3HSH(,12,3H) =,F,4,3)
SAP223 C
SAP224 C      RETURN
SAP225 C      END

```

```

SKC001 C      FUNCTION SOURCE1TIME,FF,LSPUL)
SKC002 C
SKC003 C      TEMPORAL SHAPE FROM LASER POWER TRACE
SKC004 C
SKC005 C      COMMON/PULS /SDT
SKC006 C      DIMENSION PULS(LSPUL)
SKC007 C
SKC008 C      X=11M,TIME
SKC009 C      1 X1
SKC010 C      11,1,4 1,10M 12 4000 1

```

```

SRC011 SOURCE=FF(I+1)+(X1-1)*DIFF(I+2)-FF(I+1)
SRC012 RETURN
SRC013 1 SOURCE=0.0
SRC014 RETURN
SRC015 END

CAG001 SUBROUTINE CONVERG(M,MLAST,MIT,M,NREGBT,BT,MPDE,MGNAX)
CAG002 C
CAG003 C TEST THE CONVERGENCY OF ITERATIONS
CAG004 C (NOT PROBLEM NOR WATER DEPENDENT)
CAG005 C
CAG006 C
CAG007 C DIMENSION M(MGNAX,3,MPDE),MLAST(MPDE)
CAG008 IF (MIT.EQ.0) GOTO 2
CAG009 DO 10 I=1,MPDE
CAG010 RATIO=ABS((M(I,3,2)-M(I,3,1)))/(M(I,2,1)-MLAST(I))
CAG011 IF (RATIO.GE.1.) GOTO 1
CAG012 10 CONTINUE
CAG013 GOTO 2
CAG014 C IF NOT CONVERGE, DT WAS TOO LARGE
CAG015 C DT REDUCED TO HALF
CAG016 1 DT=DT/2.
CAG017 M=NRESET
CAG018 MIT=0
CAG019 RETURN
CAG020 C IF CONVERGE, (I,3) TRANSFER TO (I,2), ITERATION GOES ON
CAG021 2 CONTINUE
CAG022 DO 20 I=1,MPDE
CAG023 20 MLAST(I)=M(I,3,2)
CAG024 L1=M2
CAG025 DO 3 I=1,MPDE
CAG026 DO 3 J=1,L1
CAG027 M(J,2)=M(J,3,2)
CAG028 3 CONTINUE
CAG029 MIT=MIT+1
CAG030 RETURN
CAG031 END

TSTP01 SUBROUTINE TSTEP(DT,DVS,M)
TSTP02 C
TSTP03 C DETERMINATION OF TIME INCREMENT BY COMPARING SUCCESSIVE TIME
TSTP04 C DERIVATIVE OF SURFACE TEMPERATURE (NOTE THAT ONLY TEMP IS USED)
TSTP05 C
TSTP06 C (NOT PROBLEM NOR WATER DEPENDENT)
TSTP07 C
TSTP08 DTOLD=DT
TSTP09 DMDT=MVS/DTOLD
TSTP10 IF (N.EQ.1) GOTO 1
TSTP11 DMDT=ABS(MDT-MLD)/ABS(MDT)
TSTP12 C 10% OF D(MDT) TOLERANCE
TSTP13 IF (D1.GT.1.0*DTOLD) M1=1.0*DTOLD
TSTP14 IF (DT.GT.1.0*DT1) M1=1.0*DT1

```

```

TSTP15 IF (DT.LT.DT1) DT=DT1
TSTP16 A1=0
TSTP17 10 K1=A1+1
TSTP18 POWER=10.*K1
TSTP19 IF (AINT(DT*POWER).EQ.0) GOTO 10
TSTP20 DT=AINT(DT*POWER+.5)/POWER
TSTP21 1 DMDT=DMDT
TSTP22 IF (N.GT.1) RETURN
TSTP23 DT=DT
TSTP24 MLTUM
TSTP25 ENL

EOK001 SUBROUTINE ERROR(N)
EOK002 C
EOK003 WRITE (6,1) N
EOK004 1 FORMAT('3INERRROR#0 STORAGE EXCEEDED BY ',I4)
EOK005 STOP
EOK006 END

MCL001 SUBROUTINE MICAL(M,THOM,DT,MCH,M,A,FF,LSPL,M,N,IFLAG)
MCL002 1 DVS=M,MLAST,MPDE,MGNAX)
MCL003 C
MCL004 C NUMERICAL CALCULATION OF PARTIAL DIFFERENTIAL EQUATION AT SOME
MCL005 C TIME STEP "THOM" WITH CRANK-NICOLSON FINITE DIFFERENCE METHOD.
MCL006 C PREDICTOR-CORRECTOR SCHEME FOR THE FIRST GUESS: NEWTON-RAPHSON
MCL007 C METHOD FOR ITERATION AND GAUSSIAN ELIMINATION METHOD FOR SOLVING
MCL008 C LINEARIZED EQUATIONS.
MCL009 C (NOT PROBLEM NOR WATER DEPENDENT)
MCL010 C
MCL011 COMMON/ITER /ITMAX,CRIT
MCL012 DIMENSION M(MGNAX,3,MPDE)
MCL013 DIMENSION A(MGNAX,7)
MCL014 DIMENSION DS(MPDE),MCH(MPDE)
MCL015 DIMENSION MLAST(MPDE)
MCL016 DIMENSION FF(1:LSPL)
MCL017 C
MCL018 TOLD=THOM
MCL019 PRESET=M
MCL020 1 MIT=0
MCL021 T=TOLD+DT/2.
MCL022 THOM=T+DT/2.
MCL023 C
MCL024 C PREDICTOR
MCL025 M1=0.0
MCL026 CALL CRANK(M,1,DT,2.,MCH,M,A,FF,LSPL,MFLD,MPDE,MGNAX)
MCL027 C
MCL028 C CORRECTOR
MCL029 M1=0.1
MCL030 CALL CRANK(M,THOM,DT,MCH,M,A,FF,LSPL,MFLD,MPDE,MGNAX)
MCL031 C
MCL032 1 MIT=MIT+1
MCL033 IF (MIT .GT. ITMAX) CRIT

```


[illegible]

NO	ACLO34	ACLO35	ACLO36	ACLO37	ACLO38	ACLO39	ACLO40	ACLO41	ACLO42	ACLO43	ACLO44	ACLO45	ACLO46	ACLO47	ACLO48	ACLO49	ACLO50	ACLO51	ACLO52	ACLO53	ACLO54	ACLO55	ACLO56	ACLO57	ACLO58	ACLO59	ACLO60	ACLO61	ACLO62	ACLO63	ACLO64	ACLO65	ACLO66	ACLO67	ACLO68	ACLO69	ACLO70	ACLO71	ACLO72	ACLO73	ACLO74	ACLO75	ACLO76	ACLO77	ACLO78	ACLO79	ACLO80	ACLO81	ACLO82	ACLO83	ACLO84	ACLO85	ACLO86	ACLO87	ACLO88	ACLO89	ACLO90	ACLO91	ACLO92	ACLO93	ACLO94	ACLO95	ACLO96	ACLO97	ACLO98	ACLO99	ACLO100	ACLO101	ACLO102	ACLO103	ACLO104	ACLO105	ACLO106	ACLO107	ACLO108	ACLO109	ACLO110	ACLO111	ACLO112	ACLO113	ACLO114	ACLO115	ACLO116	ACLO117	ACLO118	ACLO119	ACLO120	ACLO121	ACLO122	ACLO123	ACLO124	ACLO125	ACLO126	ACLO127	ACLO128	ACLO129	ACLO130	ACLO131	ACLO132	ACLO133	ACLO134	ACLO135	ACLO136	ACLO137	ACLO138	ACLO139	ACLO140	ACLO141	ACLO142	ACLO143	ACLO144	ACLO145	ACLO146	ACLO147	ACLO148	ACLO149	ACLO150	ACLO151	ACLO152	ACLO153	ACLO154	ACLO155	ACLO156	ACLO157	ACLO158	ACLO159	ACLO160	ACLO161	ACLO162	ACLO163	ACLO164	ACLO165	ACLO166	ACLO167	ACLO168	ACLO169	ACLO170	ACLO171	ACLO172	ACLO173	ACLO174	ACLO175	ACLO176	ACLO177	ACLO178	ACLO179	ACLO180	ACLO181	ACLO182	ACLO183	ACLO184	ACLO185	ACLO186	ACLO187	ACLO188	ACLO189	ACLO190	ACLO191	ACLO192	ACLO193	ACLO194	ACLO195	ACLO196	ACLO197	ACLO198	ACLO199	ACLO200	ACLO201	ACLO202	ACLO203	ACLO204	ACLO205	ACLO206	ACLO207	ACLO208	ACLO209	ACLO210	ACLO211	ACLO212	ACLO213	ACLO214	ACLO215	ACLO216	ACLO217	ACLO218	ACLO219	ACLO220	ACLO221	ACLO222	ACLO223	ACLO224	ACLO225	ACLO226	ACLO227	ACLO228	ACLO229	ACLO230	ACLO231	ACLO232	ACLO233	ACLO234	ACLO235	ACLO236	ACLO237	ACLO238	ACLO239	ACLO240	ACLO241	ACLO242	ACLO243	ACLO244	ACLO245	ACLO246	ACLO247	ACLO248	ACLO249	ACLO250	ACLO251	ACLO252	ACLO253	ACLO254	ACLO255	ACLO256	ACLO257	ACLO258	ACLO259	ACLO260	ACLO261	ACLO262	ACLO263	ACLO264	ACLO265	ACLO266	ACLO267	ACLO268	ACLO269	ACLO270	ACLO271	ACLO272	ACLO273	ACLO274	ACLO275	ACLO276	ACLO277	ACLO278	ACLO279	ACLO280	ACLO281	ACLO282	ACLO283	ACLO284	ACLO285	ACLO286	ACLO287	ACLO288	ACLO289	ACLO290	ACLO291	ACLO292	ACLO293	ACLO294	ACLO295	ACLO296	ACLO297	ACLO298	ACLO299	ACLO300	ACLO301	ACLO302	ACLO303	ACLO304	ACLO305	ACLO306	ACLO307	ACLO308	ACLO309	ACLO310	ACLO311	ACLO312	ACLO313	ACLO314	ACLO315	ACLO316	ACLO317	ACLO318	ACLO319	ACLO320	ACLO321	ACLO322	ACLO323	ACLO324	ACLO325	ACLO326	ACLO327	ACLO328	ACLO329	ACLO330	ACLO331	ACLO332	ACLO333	ACLO334	ACLO335	ACLO336	ACLO337	ACLO338	ACLO339	ACLO340	ACLO341	ACLO342	ACLO343	ACLO344	ACLO345	ACLO346	ACLO347	ACLO348	ACLO349	ACLO350	ACLO351	ACLO352	ACLO353	ACLO354	ACLO355	ACLO356	ACLO357	ACLO358	ACLO359	ACLO360	ACLO361	ACLO362	ACLO363	ACLO364	ACLO365	ACLO366	ACLO367	ACLO368	ACLO369	ACLO370	ACLO371	ACLO372	ACLO373	ACLO374	ACLO375	ACLO376	ACLO377	ACLO378	ACLO379	ACLO380	ACLO381	ACLO382	ACLO383	ACLO384	ACLO385	ACLO386	ACLO387	ACLO388	ACLO389	ACLO390	ACLO391	ACLO392	ACLO393	ACLO394	ACLO395	ACLO396	ACLO397	ACLO398	ACLO399	ACLO400	ACLO401	ACLO402	ACLO403	ACLO404	ACLO405	ACLO406	ACLO407	ACLO408	ACLO409	ACLO410
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PT0046 SPEC(23)=SPEC(23)-0.3
PT0045 CALL TITLE(UNIT,INCHOR EI = JOULES,SPECB)
PT0044 C PLOT DATA POINTS WITH SOLID LINE
PT0047 CALL PLOT(11,X,Y,HIGH,LOW,SPECB)
PT0048 C PLOT LINES FOR TOP AND RIGHT EDGES OF PLOT AREA
PT0049 SIDE(1)=SIDE(4)
PT0050 SIDE(1)=SIDE(5)
PT0051 SIDE(2)=SIDE(5)
PT0052 SIDE(2)=SIDE(5)
PT0053 SIDE(3)=SIDE(3)
PT0054 SIDE(3)=SIDE(4)
PT0055 SPEC(13)=3.0
PT0056 CALL SPLIT(1,SIDE,SIDE,SPECB)
PT0057 C
PT0058 C FINISH PLOT IF IT IS THE LAST ONE
PT0059 C OTHERWISE, CALL NEXT FRAME
PT0060 IF (IFRAME.EQ.1) GOTO 1000
PT0061 CALL NEXTFRAME(SPEC)
PT0062 RETURN
PT0063 1000 CALL OSEND(SPEC)
PT0064 RETURN
PT0065 END

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PIX001 SUBROUTINE PLOT(X,Y,M,IFRAME)
PIX002 C
PIX003 C PLOT TEMPERATURE PROFILE VS SPACE
PIX004 C
PIX005 DIMENSION SPEL(10),X(1),Y(1)
PIX006 DIMENSION SIDE(1),SIDE(3)
PIX007 DIMENSION BU(1000),BU(1000)
PIX008 C
PIX009 C SET ADJUST YDIST (IN INCHES)
PIX010 SPEC(1)=1.5
PIX011 SPEC(2)=1.5
PIX012 C SET HEIGHT, LEFT, TOP, YBOT
PIX013 SPEC(3)=50.0
PIX014 SPEC(4)=0.0
PIX015 SPEC(5)=5000.0
PIX016 SPEC(6)=0.0
PIX017 C SET PLOTting LENGTH (IN INCHES)
PIX018 SPEC(7)=5.0
PIX019 SPEC(8)=5.0
PIX020 C SET NO. OF DIVISIONS
PIX021 SPEC(9)=10.0
PIX022 SPEC(10)=10.0
PIX023 C SPECIFIC TOL
PIX024 SPEC(11)=1.0
PIX025 C SPECIFY INTERMEDIATE FILE NO.
PIX026 SPEC(12)=00.
PIX027 CALL MAX(1,SIDE)
PIX028 C SPECIFY NO. OF DATA POINTS
PIX029 SPEC(13)=10000
PIX030 C SET X&YPS, Y&XPS

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PIX031 SPEC(14)=1.
PIX032 SPEC(15)=1.
PIX033 C SET FONT,FONTH,SPACER,ROTATE,FONTH
PIX034 SPEC(17)=0.0
PIX035 SPEC(18)=0.0
PIX036 SPEC(19)=0.12
PIX037 SPEC(19)=0.04
PIX038 SPEC(20)=0.0
PIX039 SPEC(21)=2.0
PIX040 C SET ZUMER
PIX041 SPEC(24)=0.1
PIX042 SPEC(24)=0.05
PIX043 C ANNOTATE HORIZONTAL AXIS
PIX044 SPEC(28)=0.4
PIX045 SPEC(29)=2.0
PIX046 SPEC(29)=5.0
PIX047 CALL MODL(SPECB)
PIX048 C ANNOTATE VERTICAL AXIS
PIX049 SPEC(28)=1.0
PIX050 SPEC(29)=4.0
PIX051 SPEC(30)=10.0
PIX052 CALL MODL(SPECB)
PIX053 C CONSTRUCT TITLE FOR X-AXIS
PIX054 CALL TITLE(1,INCHOR(SPECB),SPECB)
PIX055 C CONSTRUCT TITLE FOR Y-AXIS
PIX056 CALL TITLE(1,TEMPERATURE (K),SPECB)
PIX057 C CONSTRUCT ANNOTATION OF GRAPH TITLE
PIX058 C ALSO USE FONT 2
PIX059 C SPECIFY XSTART, YSTART (INCHES)
PIX060 SPEC(22)=3.0
PIX061 SPEC(23)=5.0
PIX062 HOLT=1.0
PIX063 CALL TITLE(1,INCHOR(TEMPERATURE PROFILE,SPECB)
PIX064 SPEC(23)=SPEC(23)-0.3
PIX065 CALL TITLE(1,INCHOR(TEMP MAX SURFACE TEMP,SPECB)
PIX066 SPEC(23)=SPEC(23)-0.3
PIX067 CALL TITLE(1,INCHOR(EI = JOULES,SPECB)
PIX068 C PLOT DATA POINTS WITH SOLID LINE
PIX069 CALL PLOT(11,X,Y,HIGH,LOW,SPECB)
PIX070 C PLOT LINES FOR TOP AND RIGHT EDGES OF PLOT AREA
PIX071 SIDE(1)=SIDE(4)
PIX072 SIDE(1)=SIDE(5)
PIX073 SIDE(2)=SIDE(5)
PIX074 SIDE(2)=SIDE(5)
PIX075 SIDE(3)=SIDE(3)
PIX076 SIDE(3)=SIDE(4)
PIX077 SPEC(13)=3.0
PIX078 CALL SPLIT(1,SIDE,SIDE,SPECB)
PIX079 C
PIX080 C FINISH PLOT IF IT IS THE LAST ONE
PIX081 C OTHERWISE, CALL NEXT FRAME
PIX082 IF (IFRAME.EQ.1) GOTO 1000
PIX083 CALL NEXTFRAME(SPEC)
PIX084 RETURN

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PTX005 1000 CALL DENSEND(SPEC5)
PTX006 RETURN
PTX007 END

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PR5003 SUBROUTINE PLOTX(x,y,n,IFRM)
PR5002 C
PR5003 C PLOT SURFACE COMPOSITION VS TIME
PR5004 C
PR5005 DIMENSION SPEC(10),X(1),Y(1)
PR5006 DIMENSION SIDER(3),SIDEY(3)
PR5007 DIMENSION BUF X(500),BUF Y(500)
PR5008 C
PR5009 C SET X(1), Y(1) (IN INCHES)
PR5010 SPEC(1)=1.5
PR5011 SPEC(2)=1.5
PR5012 C BLT X(10), X(LEFT), Y(10), Y(TOP)
PR5013 SPEC(13)=0.5
PR5014 SPEC(14)=0.0
PR5015 SPEC(15)=2.0
PR5016 SPEC(16)=1.5
PR5017 C SET XLNGTH, YLNGTH (IN INCHES)
PR5018 SPEC(17)=5.0
PR5019 SPEC(18)=2.0
PR5020 C SET NO. OF DIVISIONS
PR5021 SPEC(19)=10.
PR5022 SPEC(19)=10.
PR5023 C SPECIFY TOOL
PR5024 SPEC(11)=1.0
PR5025 C SPECIFY INTERMEDIATE FILE NO.
PR5026 SPEC(12)=99.
PR5027 CALL AXILL(SPEC3)
PR5028 C SPECIFY NO. OF DATA POINTS
PR5029 SPEC(13)=FLOATHIN
PR5030 C BLT X(10), Y(10)
PR5031 SPEC(14)=1.
PR5032 SPEC(15)=1.
PR5033 C SET FUNIT, FOMTH, SPACER, ROTATE, FONTNO
PR5034 FUNIT=2
PR5035 SPEC(17)=0.00
PR5036 SPEC(18)=0.12
PR5037 SPEC(19)=0.04
PR5038 SPEC(20)=0.0
PR5039 SPEC(21)=2.0
PR5040 C SET J(1)
PR5041 SPEC(24)=0.1
PR5042 SPEC(24)=0.1
PR5043 C ANNOTATE HORIZONTAL AXIS
PR5044 SPEC(26)=2.0
PR5045 SPEC(27)=2.0
PR5046 SPEC(28)=5.0
PR5047 CALL MODL(SPEC5)
PR5048 C ANNOTATE VERTICAL AXIS
PR5049 SPEC(29)=4.0

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PR5050 SPEC(12)=5.0
PR5051 SPEC(10)=5.0
PR5052 CALL MODL(SPEC3)
PR5053 C CONSTRUCT TITLE FOR X-AXIS
PR5054 CALL TITLE(1)TIME (INSEC),SPEC3)
PR5055 C CONSTRUCT TITLE FOR Y-AXIS
PR5056 CALL TITLE(1)SURFACE COMPOSITION,SPEC3)
PR5057 C CONSTRUCT ANNOTATION OF X-AXIS TITLE
PR5058 C FUNIT 2 ALSO
PR5059 C SPECIFY X(10), Y(10) (IN INCHES)
PR5060 SPEC(12)=3.0
PR5061 SPEC(13)=3.5
PR5062 NUL=1.0
PR5063 CALL TITLE(RULE,2)HRSURFACE COMPOSITION VS TIME,SPEC3)
PR5064 SPEC(23)=SPEC(23)-0.3
PR5065 CALL TITLE(RULE,1)HOUR E1 = J(1),SPEC3)
PR5066 C PLOT DATA POINTS WITH SOLID LINE
PR5067 CALL PLT(1,X,Y,BUF X,BUF Y,SPEC3)
PR5068 C PLOT LINES FOR TOP AND RIGHT EDGES OF PLOT AREA
PR5069 SIDER(1)=SPEC(4)
PR5070 SIDER(2)=SPEC(5)
PR5071 SIDER(3)=SPEC(6)
PR5072 SIDER(4)=SPEC(7)
PR5073 SIDER(5)=SPEC(8)
PR5074 SIDER(6)=SPEC(9)
PR5075 SPEC(13)=3.0
PR5076 CALL SLL(1,SIDEX,SIDEY,SPEC3)
PR5077 C FINISH PLOT IF IT IS THE LAST.
PR5078 C OTHERWISE CALL NEXT FRAME
PR5079 IF (IFRM.EQ.1) GO TO 1000
PR5080 CALL NEXT(SPEC3)
PR5081 RETURN
PR5082 1000 CALL DENSEND(SPEC5)
PR5083 RETURN
PR5084 END

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PRX001 SUBROUTINE PLOTX(x,y,n,IFRM)
PRX002 C
PRX003 C PLOT COMPOSITION PROFILE VS SPACE
PRX004 C
PRX005 DIMENSION SPEC(30),X(1),Y(1)
PRX006 DIMENSION SIDER(3),SIDEY(3)
PRX007 DIMENSION BUF X(500),BUF Y(500)
PRX008 C
PRX009 C SET X(1), Y(1) (IN INCHES)
PRX010 SPEC(1)=1.5
PRX011 SPEC(2)=1.5
PRX012 C SET XLNGTH, YLNGTH, Y(TOP), Y(TOP)
PRX013 SPEC(13)=0.0
PRX014 SPEC(14)=0.0
PRX015 SPEC(15)=2.0
PRX016 SPEC(16)=1.5
PRX017 C SET XLNGTH, YLNGTH (IN INCHES)

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PRF038 101 FORMAT(4F10.0,E10.0)
PRF039 102 FORMAT(2F10.0,E10.0)
PRF040 110 FORMAT(F10.0,2E10.0)
PRF041 200 FORMAT(1H0,23HNE MATERIAL PROPERTIES/
PRF042 1 3/2)CHEMICAL WEIGHT OF COMP 1 =F9.2/
PRF043 2 3/2)CHEMICAL WEIGHT OF COMP 2 =F9.2/
PRF044 201 FORMAT(1H0,4X,3HIN=F5.0,5X,5HMSUR=F5.0,5X,5HNTUS=F5.0,
PRF045 1 5X,4HMISS=F5.3,5X,14HAI(ROOM TEMP)=E10.3)
PRF046 202 FORMAT(1H0,4X,3HIN=F7.0,5X,5HNEHEX=E9.2,5X,3HSL=F7.0,
PRF047 1 5X,3HLEK=E9.2)
PRF048 210 FORMAT(1H0,21HSLASH CHARACTERISTICS/
PRF050 1 5X,2HR=F5.3,5X,5HTPL=F10.4,4H SEC,
PRF051 2 5X,5HAEFF=E10.4,4H CM2)
PRF052 C RETURN
PRF053 END

TBL001 SUBROUTINE TABLE
TBL002 C
TBL003 C TABULATED AS A3 AND COND BELOW MELTING TEMP
TBL004 C (PROBLEM DEPEND BUT NOT WATER DEPEND)
TBL005 C
TBL006 COMMON/MATP /TM,HISUB,HFUS,ENHIS,A3RM
TBL007 COMMON/TAB /SCON(500),BA1(500),BA3(500)
TBL008 C
TBL009 IS=AIMT(1TN-300.)/10.)*2
TBL010 DO 1 I=1,IS
TBL011 TEMP=300.+(10.*FLOT(I)-1)
TBL012 SCON(I)=COND(TEMP)
TBL013 C BPT CORRECTED WITH MELTING BY TREATING HFUS AS DELTA FUNCTION
TBL014 ARG=((TEMP-TM)/50.)*882
TBL015 IF(ARG.LT.20.) GOTO 11
TBL016 CFF=0.0
TBL017 GOTO 12
TBL018 11 CFF=HFUS/(50*(1.1413)*850.)*EXP(-ARG)
TBL019 12 SALT=HFUS*(TEMP-415)*PT(TEMP)/CFF/SCON(I)
TBL020 IF(I.EQ.1) GOTO 13
TBL021 SA3(I)=(SCON(I)-SCON(I-1))/(10.*SCON(I))
TBL022 GOTO 1
TBL023 C THE VALUE OF A3 AT ROOM TEMP (300 K)
TBL024 13 SA3(I)=A3RM
TBL025 1 CONTINUE
TBL026 RETURN
TBL027 END

ITL001 SUBROUTINE INITIAL(4,WO,NPDE,HQMAX)
ITL002 C
ITL003 C INITIAL CONDITIONS
ITL004 C (PROBLEM DEPEND BUT NOT WATER DEPEND)
ITL005 C
ITL006 DIMENSION W(NQMAX,3,NPDE)
ITL007 DIMENSION WQ(MPDE)

ITL008 C
ITL009 READ (3,100) (WQ(I),I=1,NPDE)
ITL010 WRITE (4,200)
ITL011 WRITE (4,201) (I,WQ(I),I=1,NPDE)
ITL012 DO 1 K=1,NPDE
ITL013 DO 1 I=1,NQMAX
ITL014 W(I,1,K)=1.0
ITL015 W(I,2,K)=1.0
ITL016 W(I,3,K)=1.0
ITL017 1 CONTINUE
ITL018 WRITE (4,300)
ITL019 WRITE (4,201) (I,W(I,1,I),I=1,NPDE)
ITL020 C
ITL021 100 FORMAT(F10.0)
ITL022 200 FORMAT(1H0,40HNE INITIAL QUANTITY (WITH DIMENSION) OF/)
ITL023 201 FORMAT(1H0,27HPARTIAL DIFFERENTIAL EQN NO,12,2H =F10.4)
ITL024 300 FORMAT(1H0,40HNE INITIAL QUANTITY (DIMENSIONLESS) OF/)
ITL025 C
ITL026 RETURN
ITL027 END

CHRO01 SUBROUTINE CHAR(WO,MCH,NPDE)
CHRO02 C
CHRO03 C DEFINE THE CHARACTERISTIC TERMS
CHRO04 C (PROBLEM DEPEND AND MAYBE WATER DEPEND)
CHRO05 C
CHRO06 COMMON/CHAR/XCH,TCH
CHRO07 COMMON/LASER/EI,OP,R,TPL,AEFF
CHRO08 COMMON/MATP /TM,HISUB
CHRO09 DIMENSION MCH(NPDE),WO(NPDE)
CHRO10 C
CHRO11 C THERMAL DIFFUSIVITY AT MELTING TEMPERATURE
CHRO12 TDFM=COND(TM)/(RHO(TM)*SPHT(TM))
CHRO13 C
CHRO14 TCH=TPL
CHRO15 XCH=SORT(TDFM/TCH)
CHRO16 DO 1 I=1,NPDE
CHRO17 MCH(I)=WQ(I)
CHRO18 WRITE (4,100)
CHRO19 WRITE (4,200) TCH,XCH,(I,MCH(I),I=1,NPDE)
CHRO20 C
CHRO21 100 FORMAT(1H2,34HNE CHARACTERISTIC QUANTITIES ARE(1/
CHRO22 200 FORMAT(1H0,29HCCHARACTERISTIC TIME =E10.4,4H SEC/
CHRO23 1 11X,2YHCHARACTERISTIC DISTANCE =E10.4,4H CM/
CHRO24 2 11X,25HCHARACT QUANTITY OF PDE NO,12,2H =F10.4)
CHRO25 RETURN
CHRO26 END

HFXX01 SUBROUTINE MFLUX(1TME,T,R,F,F1,F2,IFLAG)
HFXX02 C
HFXX03 C CALCULATION OF THE EVAPORATED MASS FLUXES (G/CM2-SEC)
HFXX04 C (PROBLEM AND MATERIAL DEPENDENT)

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MF005 C COMMON/MDLN /MH(5:5)
MF006 DIMENSION P(5:5)
MF007
MF008 C DO 9 N=1:5
MF009 DO 9 M=1:5
MF010
MF011 P(M,N)=0.
MF012 CALL ROOT(TMR,P(3,1),P(2,1),P(1,2),P(2,2),P(3,2),P(4,2))
MF013 P(1,1)=0.
MF014 S0=0.
MF015 S1=0.
MF016 S2=0.
MF017 DO 10 N=1:5
MF018 DO 10 M=1:5
MF019 IF (N.EQ.1.AND.M.EQ.1) GO TO 10
MF020 P(1,1)=P(1,1)+P(M,N)
MF021 S0=S0+SORT(WR(N,M))/P(N,M)
MF022 S1=S1+FLOAT(N-1)*P(N,M)/SORT(WR(N,M))
MF023 S2=S2+FLOAT(N-1)*P(N,M)/SORT(WR(N,M))
MF024 10 CONTINUE
MF025 C=44.2/SORT(I)*80.82
MF026 C=44.2-3.5E22*740/4.023E23-FLUX IN O/CN2-BEC
MF027 F=C*50
MF028 FL=C*WR(2,1)*80.82
MF029 F2=C*WR(1,1)*80.82
MF030 IF (FLAQ.LT.1.OR.IFLAQ.GT.3) RETURN
MF031 DO 10 (1,2,3),IFLAG
MF032
MF033 C 1 CONTINUE
MF034 WRITE (4,100)
MF035 WRITE (4,101) P(4,2),P(3,2),P(2,2),P(1,2),P(3,1),P(2,1),P(1,1)
MF036 RETURN
MF037 C 2 CONTINUE
MF038 WRITE (4,200)
MF039
MF040 C 3 CONTINUE
MF041 WRITE (4,300) TIME,TMR,P(4,2),P(3,2),P(2,2),P(1,2),P(3,1),P(2,1),
MF042 P(1,1)
MF043 1
MF044 100 FORMAT(1H2,4X,46HTHE EQUILIBRIUM VAPOR PRESSURES AT MAXIMUM SURFACE
MF045 1E TEMPERATURE ARE//
MF046 2 10X,5HP-U03,5X,5HP-U02,6X,4HP-U07X,3HP-U,6X,4HP-U02,
MF047 3 7X,3HP-U,3X,7HP-TOTAL)
MF048 101 FORMAT(1H 4X,7E10,3I)
MF049 200 FORMAT(1H 4X,4E11,5E11,3X,5HTS(1X),6X,3HP-U,5X,5HP-U03,5X,5HP-U02,
MF050 1 6X,4HP-U,7X,3HP-U,6X,4HP-U02,7X,3HP-U,
MF051 2 3X,7HP-TOTAL,7X,4HT(1X))
MF052 300 FORMAT(1H 4E11,3F8,0,F9,4,F7E10,3I,E11,3)
MF053
MF054 C RETURN
MF055 END
MF056

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U02001 SUSPENSION ROOT(TEMP=DU,P02/PD,PU,P02,P03)
U02002 C
U02003 C CALCULATION OF PARTIAL VAPOR PRESSURES OF URANIUM-OXYGEN
U02004 C SYSTEM BASED ON BLACKBURN'S MODEL.
U02005 C MODIFIED FROM SUMOGRAM DEVELOPED BY BLACKBURN FOR MIXED OXIDE
U02006 C
U02007 COMMON/CC/A1,A2,A3,A4,A5,A6,A7
U02008 C
U02009 Y=DU-2.0
U02010 IF (Y.LT.0.) Y=0.0
U02011 OOH=ROU(100)
U02012 O02=O0982
U02013 TN=3133.0
U02014 C
U02015 C THERMODYNAMIC DATA
U02016 IF (TEMP.DI.TN) GO TO 40
U02017 DD=EXP(178240./TEMP-12.64)
U02018 EE=EXP(16350./TEMP-4.94-54.08Y)
U02019 FF=EXP(47000./TEMP-11.25)
U02020 GG=EXP(25100./TEMP-4.92)
U02021 CU0C=EXP(-187200./TEMP+133.43)
U02022 CU0C=EXP(-128700./TEMP+27.3)
U02023 CU02C=EXP(-72400./TEMP+20.3)
U02024 CU03C=EXP(-28500./TEMP+10.5)
U02025 O010=70
U02026 DD=EXP(178240./TEMP-12.64)
U02027 EE=EXP(16350./TEMP-4.94-54.08Y)
U02028 FF=EXP(47000./TEMP-11.25)
U02029 GG=EXP(25100./TEMP-4.92)
U02030 CU0C=EXP(-167900./TEMP+27.3)
U02031 CU0C=EXP(-108400./TEMP+20.8)
U02032 CU02C=EXP(-51500./TEMP+13.4)
U02033 CU03C=EXP(-3100./TEMP+2.8)
U02034 C
U02035 70 DEF=DD*EE*FF
U02036 F0=FF*DD
U02037 MH=DEF*DD/OU2+13.0/OU-1.0
U02038 PP=DEF/OU+11.54+13.0/OU-1.0
U02039 DD=DD/OU+13.08E/OU+2.08F0/OU+10B/OU+1E-4B)
U02040 KR=OU*F0/DOH+12.0/OU-1.0
U02041 SS=4.0/OU+DB+2.0/OU+1E-DB-F0)
U02042 TT=(1.0/DOH-DOH)*FF
U02043 A1=HR
U02044 A2=PP
U02045 A3=GO
U02046 A4=RR
U02047 A5=SS
U02048 A6=TT
U02049 A7=1.0-OU
U02050 C
U02051 G=0.0
U02052 G151=0
U02053 G1=0.15F
U02054 DU=1.0-1.4E

```

```

002086 FUNCTION F(X)
002087 COMMON/CC/A1,A2,A3,A4,A5,A6,A7
002088 F=1/(((A1*(X+A2)*(X+A3)*(X+A4)*(X+A5)*(X+A6)*(X+A7
002089 RETURN
002090 END

```

```

UU2091      FUNCTION FFI(X)
UU2092      COMMON/CC/A1,A2,A3,A4,A5,A6,A7
UU2093      FFI=((16.0*A1*K15.0+A21*X14.0+A33)*X13.0+A44)*X12.0+A55)*X11.0
UU2094      RETURN
UU2095      END

```

```

U02094      FUNCTION RHO(TEMP)
U02097      C
U02098      C DENSITY OF UO2 IN G/CM3
U02099      C
U02099      C GIVEN BY CHAGANDOU(1975) AND LEIBOWITZ(1974)
U02100      C
U02101      DATA A-R-C-D/9.0E-06,0.01-9.3,0.0E-12,10.78
U02102      DATA F1-A-B1/13.1,1.6,4.8,-0.0009E+04

```

```

002125      FUNCTION COND(TEMP)
002126      C
002127      C THERMAL CONDUCTIVITY OF UO2 IN W/CM-K
002128      C GIVEN BY SCHMIDT(1971)
002129      C

```

```

U02130      DATA A,B,10,80,0.0218,
U02131      DATA X,CV0,T0,2.00E-9,.299,2050.,
U02132      DATA IF,1721200.,72860.,
U02133      DATA IF0,273.,
U02134      IF1=GC,T01  GOTO 1
U02135      COND=1,/(A+B)*T01
U02136      RETURN
U02137      1  IF,1,GT,T01  GOTO 2
U02138      P=(SPH1*(T01)-X)/(CV0)/CVD
U02139      Y=1+(2-1/(1+P))/(T01+T01)/2.
U02140      COND=(1,/(A+B)*Y)/(1,1)
U02141      RETURN
U02142      2  COND=0.037
U02143      RETURN
U02144      IF0=

```

```

002145      FUNCTION DIFF(TEMP)
002146      C
002147      C OXYGEN DIFFUSIVITY IN UO2 IN CM2/SEC
002148      C COEFFICIENTS READ FROM SUBROUTINE PROP
002149      C
002150      COMMON/DIFFL/EN,EMCY,EL,ELEY

```



```

002151      IF (TEMP.GE.21.33.) GO2152
002152      RETURN
002153      RETURN
002154      DEF=MAX(XRXP,-CH/TEMP)
002155      RETURN
002156      END

```

3

B.3 User's Manual

B.3.1 Program Input Data Cards

- (1) PROGRAM INITIATION AND TITLE (2(A5,5x),6A5,5x,2A5) - One card, read in subroutine TLCDE.

Column	Variable	Description
1-5	MODE	Problem initiation flag; "START" for initiation and "STOP" for termination
6-10	--	Blank
11-15	MTYPE	Problem type; "SURFT" for heat conduction only and "STAR" for conduction and diffusion
16-20	--	Blank
21-50	HED	Problem title for labeling output (an array of dimension 6)
51-55	--	Blank
56-65	DATE	Date of the run

- (2) DIMENSION SPECIFICATION (2I5) - One card, read in subroutine TLCDE.

Column	Variable	Description
1-5	NPDE	Number of partial differential equations to solve; 1 for "SURFT" and 2 for "STAR"
6-10	NGMAX	Estimated maximum number of grid points needed for space variable (normally 300)

- (3) MATERIAL PROPERTIES - read in subroutine PROP.

- (i) ATOMIC WEIGHTS (2F10.0)

Column	Variable	Description
1-10	WA(1)	Atomic weight of component 1 (g/g-atom)
11-20	WA(2)	Atomic weight of component 2 (g/g-atom); zero or blank for single component materials or congruently vaporizing materials

- (ii) OTHER PROPERTIES (4F10.0,E10.0)

Column	Variable	Description
1-10	TM	Melting temperature (K)
11-20	HSUB	Heat of sublimation of solid phase (J/g)
21-30	HFUS	Heat of fusion (J/g)
31-40	EMISS	Total normal optical emissivity
41-50	A3RM	Coefficient A3 ($1/k dk/dT$) of heat conduction equation at room temperature

(iii) DIFFUSION COEFFICIENT (2(F10.0,E10.0)) - Blank for "SURFT" when no diffusion process is engaged.

Column	Variable	Description
1-10	EH	Diffusion activation energy / Gas constant of the mobile component above some temperature (normally melting point)
11-20	EHEX	Pre-exponential factor of diffusion coefficient of the mobile component above some temperature
21-30	EL	Diffusion activation energy / Gas constant of the mobile component below some temperature
31-40	ELEX	Pre-exponential factor below the temperature

(4) LASER PARAMETERS (F10.0,E10.0,F10.0) - One card, read in subroutine PROP.

Column	Variable	Description
1-10	R	Reflectivity of target material to the laser light
11-20	TPUL	Effective laser pulse width (sec)
21-30	AEFF	Effective surface area (cm^2) of laser exposure spot

(5) CONTROL CARDS FOR NUMERICAL STABILITY - Two cards, read in subroutine SIZE.

(i) STEP AND GRID SIZES (2E10.0,F10.0)

Column	Variable	Description
1-10	DT1	First time step size (dimensionless); also the lower limit of the following step sizes
11-20	DX1	First space grid size (dimensionless); grid sizes are geometrically increasing
21-30	EPS	Geometric factor for increasing grid sizes; a constant greater than 1 (normally 1.035)

(ii) ITERATION CONTROL (I10,E10.0)

Column	Variable	Description
1-10	ITMAX	Maximum number of iterations allowed
11-20	CRIT	Error tolerance for iteration termination

(6) TABULATED LASER PULSE

(i) SIZE AND DIVISION OF TABULATION - One card, read in subroutine
SHAPE1.

Column	Variable	Description
1-10	SDT	Step size of tabulation
11-20	LSPUL	Number of divisions of tabulation

(ii) NORMALIZED LASER PULSE SHAPE (7F10.0) - As many cards as needed to
specify the tabulated normalized pulse shape (LSPUL/7 or
LSPUL/7+1), read in subroutine SHAPE2.

Column	Variable	Description
1-10	SS(1)	Normalized digital pulse shape
...	...	...

(7) TERMINATION OF STEP DO LOOP (2E10.0) - One card, read in the main
program.

Column	Variable	Description
1-10	TSTOP	Time to stop the time step do loop (sec)
11-20	TCYCL	Time of a cycle for repetitive pulse (sec); Default (if zero or blank) for single pulse source is set to 1.e10 which is supposed to be approximately infinite

(8) INCIDENT TOTAL ENERGY (F10.0) - One card, read in the main program.

Column	Variable	Description
1-10	EI	Incident total laser energy (J)

(9) INITIAL CONDITIONS (F10.0) - NPDE cards, read in subroutine INITAL.

Column	Variable	Description
1-10	WO(I)	Initial condition for I-th partial differential equation; e.g. WO(1) - initial temperature and WO(2) - initial composition

(10) RESTART OR TERMINATION CARD (I1) -One card, read in the main program.

Column	Variable	Description
1	ICTL	Control character; 0 or blank: STOP 1: same material, another run for different laser energy 2: different material, start from the right beginning

B.3.2 Subroutines to be Supplied by the User

(1) BNDRY: The boundary condition(s) of the problem. Input the surface value(s) of the function(s) and output the surface gradient(s) of the function(s).

(2) ACALC: The coefficients of the partial differential equation(s) and the form of function(s) PHAI (ψ_T in eq. (1) or/and ψ_r in eq. (2)).

- (3) TABLE: To tabulate the coefficients of the partial differential equation(s).
- (4) INITIAL: The initial condition(s).
- (5) CHAR: The characteristic quantities for the non-dimensionalization of the boundary value problem.
- (6) MFLUX: The calculation of the convective term (due to moving boundary) and the surface heat loss due to surface recession. Also, some printout formats are to be provided.
- (7) A function library providing the physical properties, such as RHO (density), SPHT (specific heat), COND (thermal conductivity), DIF (diffusion coefficient), etc.

B.4 Sample Analyses for UO_2 Vaporization

Fig. B.1 shows an example of the input deck of the program "STAR" with total laser energy of 30 Joules and the digital pulse shape measured by the photodiode. The physical properties used in the program are provided from Appendix C.

B.4.1 Program Test with Analytic Solutions

The program was tested by comparing the results with the analytic

Fig. B.1 An Example of The Input Deck for "STAR"

start	star	uo2 surf temp and comp calc					12/03/80
2	300						
	16.	238.					
	3133.	2234.	275.	0.83	-.190e-2		
	5033.	2.22e-02	28550.	.115e01			
	0.05	1.9208e-4	4.964e-1				
	1.0e-03	.2e-1	1.035				
	5	1.0e-4					
	1.0e-5	51					
	0.0	0.275	0.691	0.851	0.926	0.980	0.996
	1.000	0.990	0.970	0.925	0.890	0.845	0.820
	0.760	0.710	0.670	0.618	0.575	0.525	0.485
	0.440	0.400	0.364	0.330	0.298	0.270	0.238
	0.214	0.190	0.170	0.145	0.126	0.105	0.092
	0.078	0.063	0.050	0.041	0.030	0.025	0.017
	0.010	0.004	0.0	0.0	0.0	0.0	0.0
	0.0	0.0					
	0.25e-3						
	3.e1						
	1400.						
	2.00						

solutions assuming no ablation of the surface ($v=0$, $j=0$), no radiation heat loss ($\epsilon=0$) and constant properties ρ , C_p and k (at 4000K) when analytic solutions are possible.

The heat source was assumed either rectangular shape with pulse width equal to t_{pul} or triangular shape with total pulse width equal to $2t_{pul}$ and maximum at t_{peak} .

The test runs assumed total energy of $E = 30$ Joules, $t_{pul} = 1.9208 \times 10^{-4}$ sec, and $t_{peak} = 0.7 \times 10^{-4}$ sec.

Without ablation and radiation heat loss and with constant properties, the heat conduction can be solved analytically for rectangular and triangular surface heat sources: ⁽¹⁴⁾

(i) Rectangular:

$$T_s(t) - T_o = \frac{2q(Kt)^{1/2}}{k} \quad 0 < t < t_{pul}$$

$$= \frac{2q(K/\pi)^{1/2}}{k} [t^{1/2} - (t - t_{pul})^{1/2}] \quad t > t_{pul} \quad (B-1)$$

(ii) Triangular:

$$T_s(t) - T_0 = \frac{4q(K/\pi)^{1/2}}{3k} \left[\frac{t^{2/3} t_2 (t - t_1)^{2/3}}{t_1 t_1 t_2 - t_1} H(t - t_1) + \frac{(t - t_2)^{2/3}}{t_2 - t_1} H(t - t_2) \right] \quad (B-2)$$

$$\begin{aligned} \text{where } H(t - t_i) &= 1 & t > t_i \\ &= 0 & t < t_i \end{aligned}$$

where q is the surface heat flux (W/cm^2), k is the thermal conductivity, and K is the thermal diffusivity ($k/\rho C_p$).

Figs. B.2 and B.3 show the results of the calculation. In both of the figures, the solid lines mean the analytic solutions and the dashed lines, which fall almost exactly on the solid lines, supposedly are the numerical solutions. This means that the numerical solutions have very good agreement with the analytic solutions at least when vaporization process and radiation are not significant and properties are not temperature dependent.

B.4.2 Numerical Solutions

Figs. B.4, B.5, B.6 and B.7 show the results of the "STAR" computer run in which the incident laser total energy is 30 Joules with the pulse shape shown in Fig. B.8, and the effective laser spot area is 0.4964 cm^2 .

In Figs. B.4 and B.6, there also show the results of the "SURFT" computer run which assumed no composition change so that only heat conduction equation is solved and the laser has the same characteristics as

the run in "STAR".

B.4.3 Discussions

The difference of the maximum surface temperature from "STAR" and "SURFT" is about 200K, which is mainly due to the vaporization effect in the ablation terms and the fact that the rate of vaporization is not only a function of temperature but also a function of the composition.

Fig. B.9 shows the maximum surface temperature as a function of the total incident laser energy calculated from both "STAR" and "SURFT". At low energy (or low temperature) range where vaporization effect is relatively unimportant, the two calculations give the consistent results, while at high energy, the discrepancy as much as 200 - 300 K attributes to the ablation effect.

B.5 Other Applications:

The computer program has been designed to deal with (i) thermal conduction problem only ("SURFT") or diffusion process at constant temperature only ("SURFR") or both ("STAR"), (ii) same kind of problems with different materials and (iii) different kind of problems (e.g. different differential equations, boundary conditions, or/and different type of heat sources - laser or ion beam, surface or volumetric source) with different materials.

For instance, the possible applications might be the thermal analysis of the first wall in CTR, the thermal analysis of laser driven micro-explosion, the thermal analysis of electron beam heating, the thermal and mass transport analysis of laser annealing or electron beam

annealing in semiconductor ion implantation and other applications of pulse heating techniques in material processing, etc.

In this report, the thermal analysis of the first wall design is presented and compared with the current design based on approximate analytic solutions.

B.5.1 Problem Formulation

Consider the first wall design of an Inertial Confinement Fusion Reactor with proposed nuclear grade H-451 graphite.⁽³⁸⁻⁴⁰⁾ The heat loading on the first wall has been assumed consisting of 4000 MJ total with 2% in x-rays, 30% in debris and the remainder in neutrons. The debris pulse has been taken as 500 μ s duration which may be treated as a surface heat flux. The x-ray, which occurs long before the debris pulse, is assumed to be uniformly deposited on the first wall over a time of a few microseconds and produce initial temperature of 840K before the debris pulse. The neutron flux which penetrates mostly into the bulk will deposit only a small amount of energy in the first wall and is neglected here⁽³⁹⁾.

B.5.2 Analytic Solutions

Assuming no ablation, no radiation heat loss and constant properties, the analytic solution for a rectangular shape of 500 μ sec duration pulse is given in eq. (B-1) with t_{pul} equal to 500 μ s. The thermal conductivity is taken as constant at temperature of 1300K (0.58 W/cm-K for unirradiated graphite and 0.34 W/cm-K for irradiated one), the density is 1.74 g/cm³, the heat capacity is 1.89 J/g-K for unirradiated and irradiated graphite, and t is the time from the beginning of the debris

pulse.

Figs. B.10 and B.11 show the surface temperatures as a function of time for three sizes of radius (12m, 13m and 14m) for unirradiated and irradiated graphite.

The thermal evaporation rate is approximated by

$$\dot{m}_R = 2.14 \times 10^{12} T_s^{-1/2} \exp(-92,000/T) \quad \text{mm/sec} \quad (\text{B-3})$$

where T_s is the surface temperature given by eq. (B-1).

The total erosion rate per second is obtained by numerically integrating eq. (B-3) assuming one second pulse repetition rate. Fig. B.12 shows the erosion rate per year with first wall radius for unirradiated and irradiated graphite.

B.5.3 Numerical Solutions

The computer code "SURFT" has been applied to this problem with ablation and radiation heat loss and variable properties. The properties are given from Ref. 39. Figs. B.13 and B.14 show the surface temperatures as a function of time for three size of radius (12m, 13m and 14m) for unirradiated and irradiated graphite.

The erosion rate is calculated along with the temperature evolution in "SURFT", and the result is shown in Fig. B.15 for unirradiated and irradiated graphite.

B.5.4 Discussions

To compare the numerical solutions with the analytic solutions, Table B.1, B.2, Figs. B.16, B.17, B.18 and B.19 show the maximum surface

temperatures and erosion rate for different reactor radius design.

Given the erosion rate of 5 mm/yr as the design criterion of first wall, the analytic calculation by Hopkins⁽³⁹⁾ gives radius of 12.54 meters for unirradiated H-451 graphite and 14.35 meters for irradiated H-451 graphite, while the numerical calculation of this work gives radius of 13.38 meters for unirradiated one and 13.35 meters for irradiated one. Figs. B.18 and B.19 show so high sensitivity of the erosion rate to the radius that a difference of 1 meter in radius may result in almost 2 orders of magnitude difference in erosion rate, which leads to 2 orders of magnitude difference in the first wall thickness with same amount of life time. Therefore, it is concluded that the accurate calculation of thermal analysis is necessary.

It is assumed here that the vaporization is the major factor of material erosion and the erosion rate is the design criterion for first wall; however, it has been pointed out that the thermal stress due to the high thermal gradient may impose another design limitation.⁽⁴¹⁾ After all, the purpose of this report is to show a useful application of the computer code developed here and to compare the results with the analytic solutions.

TABLE B.1: Erosion Rate of Unirradiated H-451 Graphite

RADIUS (meters)	$T^{\max}$ (K)		$\dot{r}$ (mm/yr)	
	Analytic ^s	Numerical	Analytic	Numerical
10	4289	4314	3.98×10^4	9.15×10^4
12	3223	3766	3.43×10^1	1.50×10^3
13	2865	3227	0.988	24.7
14	2580	2814	2.94×10^{-2}	0.384
15	2351	2499	9.30×10^{-4}	6.43×10^{-3}
16	2163	2264	3.19×10^{-5}	1.45×10^{-4}
18	1877	1911	5.15×10^{-8}	8.61×10^{-8}
20	1672	1661	1.35×10^{-10}	6.48×10^{-11}

TABLE B.2: Erosion Rate of Irradiated H-451 Graphite

RADIUS (meters)	$T^{\max}$ (K)		$\dot{r}$ (mm/yr)	
	Analytic ^s	Numerical	Analytic	Numerical
10	5357	4016	2.83×10^6	1.32×10^4
12	3965	3387	6.93×10^3	1.37×10^2
13	3497	3110	3.16×10^2	11.8
14	3125	2870	14.1	0.985
15	2825	--	0.635	--
16	2580	2466	2.94×10^{-2}	5.01×10^{-3}
18	2207	2168	7.35×10^{-5}	2.94×10^{-5}
20	1939	--	2.46×10^{-7}	--

SURFACE TEMPERATURE OF LASER PULSING ON UO_2

Fig. B.2 The Surface Temperature UO_2 with Time for Rectangular Surface Heat Source with No Ablatron, No Radiation Heat Loss and Constant Properties

SURFACE TEMPERATURE OF LASER PULSING ON UO₂

NO ABL NO RAD CONST PRCP
TRIANGULAR E = 30 Joules

— ANALYTIC SOLUTION
- - - NUMERICAL SOLUTION

Fig. B.3 The Surface Temperature of UO₂ with Time for
Triangular Surface Heat Source with No Ablation,
No Radiation Heat Loss and Constant Properties

SURFACE TEMPERATURE OF LASER PULSING ON UO_2

NUMERICAL SOLUTIONS
E = 30 Joules

— FROM STAR
- - - FROM SURFT

Fig. B.4 The Surface Temperature of UO_2 with Time for Real Laser Pulse with Ablatron, Radiation Heat Loss and Variable Properties

SURFACE COMPOSITION OF LASER PULSING ON UO_2

CALCULATED FROM S¹X
E = 30 Joules

Fig. B.5 The Surface Composition of UO_2 with Time for Real Laser Pulse with Incongruent Vaporization

Fig. B.6 The Temperature Profile of UO_2 Subject to Total Energy 30 Joule Laser Surface Heating

COMPOSITION PROFILE OF LASER PULSING ON UO₂

CALCULATED FROM STAR
E = 30 Joules

Fig. B.7 The Oxygen-to-Uranium Ratio Profile of UO₂ Subject to Total Energy 30 Joule Laser Surface Heating with Incongruent Vaporization

LASER PULSE SHAPE FROM POWER TRACE

Fig. B.8 The Normalized Laser Pulse Shape With Time

MAX SURFACE TEMPERATURE OF LASER PULSING ON UO₂

Fig. B.9 The Maximum Surface Temperature of UO₂ vs Incident Total Energy

SURFACE TEMPERATURE OF UN-IRRADIATED H-451 GRAPHITE

ANALYTIC SOLUTION
for $E = 1200 \text{ MJ/m}^2$ & $\tau_{PUL} = 500 \text{ } \mu\text{s}$

from top to bottom
RADIUS = 12.0, 13.0, 14.0 m

Fig. B.10 The Surface Temperature of Un-irradiated H-451 Graphite With Time for Rectangular Surface Heat Load with No Ablatron, No Radiation Heat Loss and Constant Properties

SURFACE TEMPERATURE OF IRRADIATED H-451 GRAPHITE

ANALYTIC SOLUTION
for $E = 1200 \text{ MJ}$ & $\text{TPUL} = 500 \text{ us}$

from top to bottom
RADIUS-12.0, 13.0, 14.0 m

Fig. B.11 The Surface Temperature of Irradiated H-451 Graphite with Time for Rectangular Surface Heat Load with No Ablation, No Radiation Heat Loss and Constant Properties

CTR FIRST WALL H-451 GRAPHITE

Fig. B.12 The First Wall Erosion Rate with Different Fusion Core Size for Rectangular Heat Load with no Ablatron, No Radiation Heat Loss and Constant Properties

SURFACE TEMPERATURE OF UN-IRRADIATED H-451 GRAPHITE

CALCULATED FROM SIIRFT
for $E = 1200 \text{ MJ}$ & $\text{TPUL} = 500 \text{ } \mu\text{s}$

from top to bottom
RADIUS = 12.0, 13.0, 14.0 m.

Fig. B.13 The Surface Temperature of Irradiated H-451 Graphite with Time for Angular Surface Heat Load with Ablatron, Radiation Heat Loss and Variable Properties

SURFACE TEMPERATURE OF IRRADIATED H-451 GRAPHITE

CALCULATED FROM SURFY
for $E = 1200 \text{ MJ}$ & $TPUL = 500 \text{ ps}$

from top to bottom
RADIUS=12.0,13.0,14.0 m.

Fig. B.14 The Surface Temperature of Irradiated H-451 Graphite with Time for Rectangular Surface Heat Load with Ablatron, Radiation Heat Loss and Variable Properties

CTR FIRST WALL H-451 GRAPHITE

NUMERICAL SOLUTION
 $E = 1200 \text{ MJ} \text{ \& } TPUL = 500 \mu s$

— UN-IRRADIATED
 - - - IRRADIATED

Fig. B.15 The First Wall Erosion Rate with Different Fusion Core Size for Rectangular Heat Load (with Ablatron, Radiation Heat Loss and Variable Properties)

CTR FIRST WALL H-451 GRAPHITE

Fig. B.16 Comparison of the Analytic and Numerical Calculation of the Maximum Surface Temperature with Different Fusion Core Size for Un-Irradiated H-451 Graphite

CTR FIRST WALL H-451 GRAPHITE

Fig. B.17 The Maximum Surface Temperature with Different Fusion Core Size for Irradiated H-451 Graphite

CTR FIRST WALL H-451 GRAPHITE

Fig. B.18 Comparison of the Analytic and Numerical Calculation of the First Wall Erosion Rate with Different Fusion Core Size for Un-Irradiated H-451 Graphite

CTR FIRST WALL H-451 GRAPHITE

IRRADIATED GRAPHITE
 $E = 1200 \text{ MJ/g}$ & $TPUL = 500 \text{ } \mu\text{s}$

— ANALYTIC SOLUTION
 - - - NUMERICAL SOLUTION

Fig. B.19 Comparison of the Analytic and Numerical Code of the First Wall Erosion Rate with Different Fusion Core Size for Irradiated H-451 Graphite

APPENDIX C: THERMODYNAMIC DATA AND MATERIAL PROPERTIES OF UO_2

C.1 Density

The density of solid UO_2 is given by M. G. Chasanov et al⁽²⁰⁾ as:

$$\rho = (10.98) / [1 + 9 \times 10^{-6}(T-273) + 6 \times 10^{-9}(T-273)^2 + 3 \times 10^{-12}(T-273)^3]$$

for $T < 3140$ K (C-1)

The density of liquid UO_2 is given by L. Leibowitz et al⁽²¹⁾ as:

$$\rho = 10.658 - 6.3609 \times 10^{-4} T \quad \text{for } T > 3140 \text{ K} \quad (C-2)$$

where ρ is in g/cm^3 and T is in K.

C.2 Specific Heat

The specific heat of solid UO_2 is given by Kerrisk and Clifton⁽²²⁾ as:

$$C_p = \frac{K_1 \theta^2 e^{\theta/T}}{T^2 (e^{\theta/T} - 1)^2} + 2K_2 T + \frac{K_3 E_D}{RT^2} e^{-E_D/RT} \quad \text{for } T < 3140 \text{ K} \quad (C-3)$$

where $\theta = 535.285$ K

$$E_D = 1.578 \times 10^5 \text{ J/mole}$$

$$K_1 = 0.2968 \text{ J/g-K}$$

$$K_2 = 1.217 \times 10^{-5} \text{ J/g-K}^2$$

$$K_3 = 8.750 \times 10^{-8} \text{ J/g}$$

$$R = 8.314 \text{ J/mole-K}$$

and C_p is in J/g-K and T is in K.

The specific heat of liquid UO_2 is given by Leibowitz⁽²³⁾ as:

$$C_p = 0.503 \text{ J/g-K} \quad \text{for } T > 3140 \text{ K} \quad (C-4)$$

At the melting point, the enthalpy increment exhibits a discontinuity due to the phase change. This heat of phase transition is

handled by an effective heat capacity term in the vicinity of the melting temperature.⁽²⁴⁾

The molar enthalpy of a material at temperature T can be expressed in terms of the heat of fusion and the molar heat capacity at constant pressure C_p as follows:

$$H^0(T) = \begin{cases} H_{298}^0 + \int_{298}^T C_p dT' & \text{if } T < T_{mp} \\ H_{298}^0 + \Delta H_f + \int_{298}^T C_p dT' & \text{if } T > T_{mp} \end{cases} \quad (C-5)$$

Define an apparent molar heat capacity $\tilde{c}_p$:

$$\tilde{c}_p(T) = C_p(T) + \Delta H_f \delta(T - T_{mp}) \quad (C-6)$$

so that

$$H^0(T) = H_{298}^0 + \int_{298}^T \tilde{c}_p(T') dT' \quad \text{for every } T \quad (C-7)$$

Since discontinuities in material properties are not desirable in the numerical solution of the conduction equation, the heat capacity is approximated by a continuous function of temperature; that is, in stead of using a delta function to account for the heat of fusion, a Gaussian function of finite width centered at the melting point is used.

Let $C_{pp} = \Delta H_f \delta(T - T_{mp})$ be approximated by a function of Gaussian form:

$$C_{pp} = \frac{\Delta H_f}{\sqrt{\pi} \sigma_1} \exp \left[-\left(\frac{T - T_{mp}}{\sigma_1} \right)^2 \right] \quad (C-8)$$

where ΔH_f = heat of fusion of UO_2 = 274.4 J/g⁽²³⁾

T_{mp} = 3140 K⁽²⁵⁾

σ_1 = half width of the heat of fusion peak = 50 K
and the apparent specific heat

$$C_p = C_p + C_{pp} \quad (C-9)$$

C.3 Thermal Conductivity

The thermal conductivity of solid UO_2 is given by Schmidt⁽²⁶⁾ as:

$$k = 1/[10.80 + 0.0218(T-273)] \quad \text{for } T < 1473 \text{ K} \quad (C-10)$$

$$k = (1+F)/(10.80 + 0.0218 \theta) \quad \text{for } 1473 \text{ K} < T < 3140 \text{ K} \quad (C-11)$$

$$\theta = [T + 1777 - (T - 273) \tanh(T - 2323)]/2$$

$$F = [C_p - 2 \times 10^{-5}(T - 273) - 0.299]/0.299$$

The thermal conductivity of liquid UO_2 is obtained from measured thermal diffusivity of liquid UO_2 (constant value of $2.6 \times 10^{-2} \text{ cm}^2/\text{sec}$) by Leibowitz⁽²⁷⁾:

$$k = 0.139 - 8.32 \times 10^{-6} T \quad \text{for } T > 3140 \text{ K} \quad (C-12)$$

where k is in W/cm-K , T is in K , and C_p in eq. (C-11) is in J/g-K .

A.4 Vapor Pressure

A computer subroutine following Blackburn's model⁽²⁸⁾, "ROOT", is used to calculate the partial pressure of each vapor species as a function of both composition and temperature.

A.5 Heat of Vaporization

The heat of vaporization is assumed constant below the melting temperature:

$$\Delta H_{\text{vap}} = 2234 \text{ J/g} \quad \text{for } T < 3140 \text{ K} \quad (\text{C-13})$$

given by Bogensberger et al⁽²⁹⁾ in which the averaged vapor pressure curve of Ohse⁽³⁰⁾ and Tetenbaum and Hunt⁽³¹⁾ was fitted to the Claysius-Clapeyton equation.

Given the heat of fusion, 274.4 J/g, from L. Leibowitz, et al.,⁽³²⁾, the heat of vaporization above the melting point is:

$$\Delta H_{\text{vap}} = 2234 - 274.4 = 1959.6 \text{ J/g} \quad \text{for } T > 3140 \text{ K} \quad (\text{C-14})$$

C.6 Diffusion Coefficient of Oxygen

The diffusion coefficient of oxygen in solid UO_2 is given by J. Belle⁽³³⁾ as:

$$D_o = 1.15 \exp(-28550/T) \quad \text{for } T < 3140 \text{ K} \quad (\text{C-15})$$

Since the diffusion coefficient of oxygen in the oxide melt is not available, estimates of upper and lower limits are made in this calculation. An upper estimate can be arrived at with the aid of kinetic gas theory.⁽³⁴⁾ In a ideal gas the diffusion coefficient is $D = \frac{1}{3} \bar{l} \bar{v}$, where $\bar{l}$ is the mean free path and $\bar{v}$ is the mean velocity of the gas molecules. If the interionic distance in the oxide melt 10 Å is introduced for $\bar{l}$, then $D_o = 7.37 \times 10^{-3} \text{ cm}^2/\text{sec}$ at 3140 K and $1.14 \times 10^{-2} \text{ cm}^2/\text{sec}$ at 7560 K (the estimated critical temperature by Ohse⁽³⁵⁾). In this case, D_o is fitted to be approximately

$$(D_o)_{\text{max}} = 1.59 \times 10^{-2} \exp(-2516/T) \quad \text{for } T > 3140 \text{ K} \quad (\text{C-16})$$

where $(D_o)_{\text{max}}$ is in cm^2/sec and T is in K. This sets up an upper limit of diffusion coefficient in the temperature of interest, since the ions in the oxide melt cannot move unimpeded between collisions in the same way as gas molecules. Instead, molecules in

the liquid migrate from one potential site to an adjacent one, remaining in each site for a certain time. A lower estimate is based upon the fact that the diffusion coefficient in the liquid oxide is higher than that in the solid.

$$(D_o)_{\min} = 3.03 \times 10^{-1} \exp(-24800/T) \text{ for } T > 3140 \text{ K} \quad (C-17)$$

This equation is fitted by taking $D_o = 1.1 \times 10^{-4} \text{ cm}^2/\text{sec}$ from Belle⁽³³⁾ at 3140 K and $1.14 \times 10^{-2} \text{ cm}^2/\text{sec}$ at 7560 K.

C.7 Optical Emissivity

The spectral emissivity at $\lambda = 6500 \text{ \AA}$ is given by Held and Wilder⁽³⁶⁾ as:

$$\epsilon_{\lambda=6500\text{\AA}} = 0.83 \quad (C-18)$$

which is found insensitive to the temperature over a wide temperature range (450 - 2400 K). It is proposed⁽²¹⁾ that $\epsilon_{\lambda=6500\text{\AA}}$ be constant up to the melting point.

Above the melting point $\epsilon_{\lambda=6500\text{\AA}}$ is measured by Bober⁽³⁷⁾ as:

$$\epsilon_{\lambda=6500\text{\AA}} = 0.81 \text{ to } 0.87 \text{ for } 3140 \text{ K} < T < 3700 \text{ K} \quad (C-19)$$

No data is available for the total hemispherical emissivity of UO_2 with temperature, so an average value of 0.83 is adopted, i.e.

$$\epsilon_t \approx 0.83 \text{ at all temperatures.}$$

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